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Acetylugenol from *Acacia nilotica* (L.) Exhibits a Strong Antibacterial Activity and Its Phenyl and Indole Analogues Show a Promising Anti-TB Potential Targeting PknE/B Protein Kinases

Abubakar Abdulhamid ¹, Talal Ahmed Awad ² , Abdalla E. Ahmed ³, Faisal Hammad Mekky Koua ^{4,*} 
and Amar Mohamed Ismail ^{5,*} 

¹ Department of Biochemistry, Faculty of Science, Kebbi State University of Science and Technology, PMB 1144 Aliero, Kebbi State, Nigeria; abuaugie132@yahoo.com

² Department of Pharmaceutical Chemistry, Faculty of Pharmacy, University of Ibn Sina, Aljerif West Bank Block 5, PO Box 10995 Khartoum, Sudan; talaladlan@hotmail.com

³ National University Biomedical Research Institute, National University-Sudan, Air St. PO Box 3783 Khartoum, Sudan; abdallah_elseer@yahoo.com

⁴ Biotechnology Park, African City of Technology, Khartoum 11111, Sudan

⁵ Department of Biochemistry and Molecular Biology, Faculty of Science and Technology, Al Neelain University, El Baladiya Ave, PO Box 12702 Khartoum, Sudan

* Correspondence: faisalkoua@gmail.com (F.H.M.K.); amarqqqu@yahoo.com (A.M.I.)



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Abstract: Acetylugenol is a phytochemical compound with broad effects against infectious diseases and tumors. Here, we extracted, characterized, and elucidated the structure of acetylugenol, for the first time, from the leaves of *Acacia nilotica* (L.)—a well-known medicinal plant. The broad antibacterial potential of acetylugenol was first confirmed against seven bacterial clinical isolates, which reveal a strong activity against *Proteus* sp., *Salmonella typhi*, *Staphylococcus aureus*, and *Streptococcus pneumoniae* with similar or better zone of inhibition comparing to that of the control amoxicillin. To further investigate its effect against *Mycobacterium tuberculosis*, acetylugenol and its indole and phenyl analogues were subjected to molecular docking experiments against two potential tuberculosis drug targets—*MtPknE* and *MtPknB* Ser/Thr protein kinases. The results reveal that all of the analogs have improved docking scores compared to the acetylugenol. The indole analogues EUG-1 and EUG-3 were more effective with better docking scores for *MtPknE* with −11.08 and −10.05 kcal/mol, respectively. Similar results were obtained for the *MtPknB*. In contrast, only the EUG-2 phenyl analogue has given rise to similar docking scores for both targets. This opens the door for further comprehensive studies on these acetylugenol analogues with in vitro and in vivo experiments to validate and get more insights into their mechanisms of action.

Keywords: acetylugenol; *Acacia nilotica*; anti-tuberculosis; Serine/Threonine kinases; multi-drug resistant TB

1. Introduction

Tuberculosis (TB), caused by *Mycobacterium tuberculosis*, is among the top 10 causes of death worldwide [1], and represents a serious challenge that faces global disease eradication policy owing to the growing antimicrobial resistance [2,3]. TB kills annually around 1.5 million people, with over 80% in developing countries including India, China, Indonesia, Philippines, Pakistan, Nigeria, Bangladesh, and South Africa [4]. TB treatment regimen is a complex procedure that involves the use of several antibiotics for a long period of time and requires special awareness from patients. The lack of such awareness leads often to the emergence of multi-drug resistant tuberculosis (MDR-TB), which is a consequence of the lack of good education and poor hygienic conditions and other environmental factors, especially in developing countries [5]. This has also increased the toll of death due to TB among

HIV-positive patients [6,7]. The drug resistance developed by *M. tuberculosis* against first line anti-tuberculosis (anti-TB)—Isoniazid and Rifampicin—has become a major problem worldwide [8–10]. This has been complicated by the emergence of resistance against other anti-TB drugs including, for example, the quinolones and aminoglycosides [11,12]. This emergence of MDR-TB has driven the urgent need for the discovery of novel therapeutic agents to overcome the TB challenges worldwide [9,12].

Medicinal plants that have well-documented applications as natural remedies for the treatment of many infectious diseases, including those infecting the respiratory system, are considered a major source in drug discovery [13]. Plants produce many secondary metabolites in tremendous amounts, many of which have been used as templates or scaffolds for synthesis of novel pharmaceutical drugs [13–15]. In the developing countries, plants remain the major source of medicines for centuries [16,17]. Sub-Saharan Africa is rich in medicinal plants, which have been used extensively to treat infectious diseases, especially those of the respiratory system such as tuberculosis, asthma, influenza, and pneumonia [17]. For example, in South Africa alone, there are over 300 medicinal plants, with ~6% species belong to the family *Fabaceae*, that have been used in remedies of various respiratory infections [18]. The medicinal plant *Acacia nilotica* (L.)—native to Sudan—from the family *Fabaceae* is well-known for its medicinal applications in the treatment of various respiratory system diseases and other diseases including malaria and Hepatitis C virus, as well as wound healing and disinfection [19–21]. *A. nilotica* (L.) is rich in several groups of active phytochemical constituents including alkaloids, flavonoids, tannins and saponins, gallic acid, kaempferol, umbelliferone, and niloticane [22–24]. Its pharmacological activities are broad which are attributed mainly to the presence of these phytochemicals. It involves antibacterial, antifungal, anti-mutagenic, anti-inflammatory and antioxidant activities [19,22,25].

Eugenol, 4-allyl-2-methoxybenzene, is a natural product that is produced by many angiosperms and known for its aromatic flavor [26,27]. It is a phenolic compound that belongs to the group of phenylpropanes, which is biosynthesized via the shikimic acid pathway [26]. Its derivative acetyleugenol, with the formula $C_{12}H_{14}O_3$ and molecular weight 206.238 Da, is one of the major components of clove *Syzygium aromaticum* (L.) essential oil and is also produced by many other plants such as cinnamon, pimento, ylang ylang, and rose [28]. Eugenol has also been detected in the chloroform extracts of the *A. nilotica* (L.) leaves without further chemical description [29]. Eugenol is slightly acidic and very reactive with a stronger action than alcohols [28]. It has a wide range of applications in the food and cosmetic industry as a flavor and antioxidant, as well as in traditional medicine [27,28,30]. Eugenol and its derivatives have also been reported to have potential activities as anti-inflammatory, anti-tumor, antimicrobial, anti-pyretic, analgesic, and anesthetic [31]. Its benzoate and 2-nitrobenzoate derivatives, as well as the synergistic effects with approved anti-TB drugs have been shown to exhibit strong activity against *Mycobacterium* spp. and *M. tuberculosis* [32]. Its hydrazone derivatives were studied by molecular docking against the enoyl-ACP reductase as a molecular target, and validated by in vitro assay, which showed a docking score of -10.393 kcal/mol with minimum inhibitory activity at $25\mu\text{g}/\text{mL}$ [31,33].

The *M. tuberculosis* Serine/Threonine (S/T) protein kinases *MtPknA-L*, except *PknC*, are considered as potential anti-TB drug targets [34]. The *M. tuberculosis* S/T protein kinases, belong to the *Pkn2* family, are eukaryote-like S/T kinases and function in phosphorylation-dependent signal transduction manner by transducing extracellular stimuli into a cellular response to ensure *M. tuberculosis* cell division, transcription regulation, stress response, regulation of multiple metabolic processes and pathogenesis [34,35]. These *Mt* kinases can be categorized based on their similarity into five clades; Clade-1 (*PknA*, *PknB*, *PknL*), Clade-2 (*PknH*, *PknE*, *PknD*), Clade-3 (*PknF*, *PknI*, *PknJ*), Clade-4 (*PknK*) and Clade-5 (*PknG*) [36]. *MtPknB* and *MtPknE* have been implicated in important biological roles in *M. tuberculosis*, which include the determination of cell shape, morphology, cell division, and biofilm formation for *PknB*, and inhibition of apoptosis in infected macrophages for

PknE [37–40]. Thus, both PknB and PknE are considered as important molecular targets for effective anti-TB drug search [34,41,42]. The *MtPknB* and *MtPknE* crystal structures were reported previously, and their structures have been used as molecular targets for inhibitors screening and drug discovery [43–45]. Herein, we applied bioactivity-guided screening of acetylugenol isolated from *A. nilotica* (L.) leaves against various pathogenic bacterial isolates and performed molecular docking studies using *MtPknB* and *MtPknE* as targets. Based on the preliminary results of isolated acetylugenol, we designed and in silico synthesized phenyl and indole derivatives using acetylugenol as a chemical scaffold and studied them with docking experiments against the same targets.

2. Materials and Methods

2.1. Plant Collection and Authentication

Fresh disease-free leaves of *A. nilotica* (L.) were separately collected from Bodinga, Sokoto State, Nigeria. The specimens were identified and authenticated by Botanists at the Department of Biological Sciences, Usmanu Danfodiyo University, Sokoto, Nigeria. The specimens identified as *A. nilotica* (L.) has given the voucher number UDUH/ANS/0247. Specimens were then shed-dried, ground finely, and kept in air-tight containers until further use.

2.2. Preparation of Ethyl Acetate Extracts

Crude methanolic extraction was initially performed, followed by solvent partitioning on the residue (marc) to obtain the ethyl acetate extract. In brief, the methanol extraction was carried out by soaking 1 kg of the specimen in 6 L of 90% methanol for 72 h and was filtered using a clean cloth and Whatman No. 1 filter paper. The filtrate was then concentrated in vacuo at 30 °C and stored in sterile containers at 4 °C until further use. For the solvent partitioning, about 100 g of the crude extract of *A. nilotica* was dissolved in 500 mL sterile distilled water in a separatory funnel and extracted with *n*-hexane. The resulting *n*-hexane phase was concentrated to dryness, and the powder was kept in a freezer in an air-tight container. The aqueous phase was further extracted with ethyl acetate. The fraction obtained from the ethyl acetate extraction was concentrated to dryness and the powder was kept at 4 °C until used. The same procedure was applied to obtain the *n*-butanol fraction.

2.3. Isolation and Purification of the Bioactive Compounds

Thin Layer Chromatography (TLC), Column Chromatography (CC) and preparative TLC techniques were used for the isolation and purification of the bioactive compounds as described previously [46,47].

2.4. Thin Layer Chromatography

TLC was carried out on aluminum TLC sheets pre-coated with silica gel (60 PF254) with a layer thickness of ~0.2 mm. The one-way ascending spotting and development technique was followed. Spots were applied manually on silica gel aluminum plates using a capillary tube and dried using an air blower. The TLC was developed at a room temperature using a Shandon chromatographic tank covered with a lid during the run. The solvent system contained a mixture of *n*-hexane and ethyl acetate solvent with a ratio of 70:30 (see Supplementary Materials Figure S1).

2.5. Column Chromatography

The ethyl acetate extract was loaded on a stationary phase containing silica gel (60–120 mesh size) using the wet loading method in a glass column (75 × 3.5 cm dimensions) packed with a wet slurry silica gel as described previously [46]. The extract was first dissolved in a small amount of *n*-hexane mixed with a small quantity of silica gel and then loaded on top of the pre-packed column. Different solvent systems were used to elute the silica gel column by gradient elution method—(i) 100% *n*-hexane, (ii) *n*-hexane/ethyl

acetate mixtures (5:95, 10:90, 15:85, 20:80, 25:75, 30:70, 35:65, 40:60, 45:55, 50:50, 55:45; 60:40, 65:35, 70:30, 75:25, 80:20, 85:15, 90:10, 95:5), (iii) 100% ethyl acetate, or (iv) ethyl acetate/methanol mixtures with similar ratios to that of *n*-hexane/ethyl acetate. The best solvent system from which the fraction was eluted is *n*-hexane/ethyl acetate (75:25), which was confirmed by TLC using *n*-hexane/ethyl acetate solvent with 70:30 ratio.

The ethyl acetate extract—the most effective in the antibacterial sensitivity assays—was further subjected to CC fractionation as described above. In brief, 3.0 g of the extract was chromatographed on a silica gel column eluted with a mobile phase containing; (i) 100% *n*-hexane, (ii) *n*-hexane/ethyl acetate mixtures (90:10, 80:20; 70:30, 60:40, 50:50, 40:60, 30:70, 20:80, 10:90), (iii) 100% ethyl acetate, or (iv) ethyl acetate/methanol mixtures (98:20, 95:5, (0:10, 80:20), which were gave rise to a total of 110 fractions. These 110 fractions were pooled together based on similarity in their TLC profile and their retention factors (R_f), calculated according to equation (1), to give rise into 5 sub-fractions. The sub-fractions were then subjected to TLC and a single spot was observed for each fraction, confirming the similarity and assuring the pooling of sub-fractions. Sub-fractions were dried, weighed, and labelled before used for bioactivity assay.

$$R_f = \frac{\text{Distance traveled by spot}}{\text{Distance traveled by solvent}} \quad (1)$$

2.6. Gas Chromatography/Mass Spectrometry Analysis

For GC/MS analysis, 0.5 mg of each pure compound was placed in a labeled vial and diluted with an appropriate solvent of HPLC grade up to a final concentration of 1.0 μM . The solution in each case was filtered to remove any insoluble material before injecting into the column. An aliquot of 2.0 mL of each sample solution was then injected into a GC/mass spectrometer equipped with a GC-MSD detector (Agilent GC 7890B, MSD 5977A, Agilent Technologies, USA) and ran for 27 min. The column oven and injection temperature were set at 80 $^\circ\text{C}$ and 250 $^\circ\text{C}$, respectively, and the column pressure was maintained at 108 kPa during the run. The spectra were analyzed using the program GCMS-QP2010 PLUS.

2.7. Nuclear Magnetic Resonance Spectroscopy

Proton (^1H) and carbon-13 isotope (^{13}C) NMR experiments alongside two-dimensional (2D) NMR spectroscopy including Correlation Spectroscopy (COSY), Nuclear Overhauser Effect Spectroscopy (NOESY), Heteronuclear Single-Quantum Correlation Spectroscopy (HSQC), and Heteronuclear Multiple-Bond Correlation Spectroscopy (HMBC) were employed to get structural information on the isolated compounds. For each sample, 10 mg of the pure compound was dissolved in 1.0 mL of 100% dimethyl sulfoxide (DMSO, Sigma-Aldrich) as a solvent and filtered through a Pasteur pipette equipped with a glass wool plug that was discharged into a labelled 5.0 mm NMR tube. The filtration serves to remove undissolved residues and dust from the solution that affects the resolution and line shape of NMR spectra. NMR spectra were recorded on 400 MHz Agilent-VNMRS-NMR spectrometer (Agilent Technologies, Santa Clara, USA). Chemical shift values were reported in parts per million (δ -ppm). All spectra were analyzed, and the results obtained were compared with published data in order to elucidate the structures of the isolated compounds.

2.8. Bioactivity Antibacterial Sensitivity Assays

2.8.1. Microorganisms, Culture Media and Inoculum Preparations

The microorganisms used in the present study were locally isolated and identified by the Department of Microbiology, Faculty of Science, Kebbi State University of Science and technology, Aliero, Nigeria. All isolates were purified and maintained in slant cultures of Mueller Hinton Agar (MHA). These isolates are *Escherichia coli*, *Klebsiella pneumoniae*, *Proteus* sp., *Pseudomonas aeruginosa*, *Staphylococcus aureus*, *Salmonella typhi* and *Streptococcus pneumoniae*. The sensitivity tests were performed on either Mueller Hinton Agar or Mueller Hinton Broth (MHB), which were all prepared freshly under aseptic conditions.

To prepare the inoculum of each microorganism a 0.5 McFarland turbidity standard was used to standardize the microorganism. The scale was prepared by adding 50 μL of 1% BaCl_2 to 9.95 mL of 1% H_2SO_4 . The suspensions of the microorganisms were prepared in normal saline and compared with 0.5 McFarland turbidity standard against a white background with contrasting black lines in the presence of good lighting [48]. The suspensions were diluted with normal saline if the density is higher than that of the standard, and additional microorganism suspension was added in case of lower density. This continues until the density of the suspension matches with that of 0.5 McFarland turbidity standard, which corresponds to $\sim 1.5 \times 10^{18}$ CFU/mL.

2.8.2. Antibacterial Sensitivity Assay of the Extracts

Agar well diffusion method was used to test the antibacterial activity of column fractions as reported previously [49,50]. The assay was performed using stock concentrations of 10 mg/mL. The standardized inoculums of the isolates were uniformly inoculated on freshly prepared MHA plates by streaking the swab extensively over the entire surface of the plates. A sterile Cork borer of 6.0 mm diameter was used to make two wells per plates and labelled appropriately. An aliquot of 20 μL of the extract was added in the well and allowed to diffuse into the agar and incubated at 37 °C for 24 h. The zone of inhibition diameter was measured in millimeter (mm) as an indication of antibacterial activity of the extract against the bacterial isolate. The antibiotic amoxicillin was used as a positive control with a final concentration of 1.0 mg/mL.

2.8.3. Minimum Inhibitory Concentration and Minimum Bactericidal Concentration

The Minimum Inhibitory Concentration (MIC) determination was performed on the microorganisms that were sensitive to the fractions using the broth dilution method [51]. Different concentrations of the extracts were prepared and added to test tubes containing MHB. Bacterial isolates were inoculated into each tube and incubated at 37 °C for 24 h. The minimum concentrations of the extract that showed no turbidity in the MHB was considered as the MIC. The Minimum Bactericidal Concentration (MBC) was performed to test the viability of microorganisms upon treatment with the compound. The MIC cultures of different concentrations were inoculated on MHA plates and incubated at 37 °C for 24 h. After incubation, the plates were observed for the growth of colonies and data were recorded. The MBC represents the MHA plate with a minimum concentration of the extract that displays no bacterial growth [52].

2.9. Molecular Docking Experiment

2.9.1. Ligands and Protein Preparations

The 3D structure of acetyleugenol was obtained from the PubChem database (www.pubchem.ncbi.nlm.nih.gov). Other eugenol derivatives—a total of four molecules containing eugenol scaffold—were sketched and converted into 3D structures using ChemDraw (ChemDraw 2017 PerkinElmer Informatics Inc.). All molecules were prepared using the Molecular Operating Environment (MOE) program (www.chemcomp.com, Chemical computing group, Montreal, Canada). Partial charges were added to the compounds using MMFF94 force field before subjecting them to energy minimization to obtain optimized geometries.

For the protein targets preparation, X-ray crystal structures of the *M. tuberculosis* Serine/Threonine kinases enzymes *MtPknE* and *MtPknB* with PDB codes 2H34 and 2FUM, respectively, were retrieved from the protein databank (www.rcsb.org). The structural models were prepared by removing water molecules, adding hydrogen atoms, and applying energy minimization. The binding sites of the *MtPknE* and *MtPknB* were predicted using homology modelling. Target protein sequences were submitted to the SWISS-Model server and homolog templates were obtained by BLAST. The template selected for the next experiments was the plant Cipk family of protein kinases (PDB ID: 4CZU). Target proteins and templates were superimposed in the MOE software and the binding sites

were searched for using the Site-finder function of the MOE. We purposely selected the binding sites that coincide with the co-crystallized ligand of the template proteins.

2.9.2. Docking Procedure

The docking experiments were performed using the docking suite of the MOE software. The compound structures were transferred and saved in MOE database. A Gaussian contact surface was drawn around the binding sites of the target protein and the receptor was verified as a receptor and the site as ligand atoms. The placement method used was triangle matcher. The first scoring function was set to London dG/MOE and the refinement to force field. The docking process was then started by retaining 100 poses. The final refined poses were ranked by the MM/GBV1 binding free energy estimation. The results depiction was generated by PyMOL APBS tools and MOE.

3. Results and Discussion

3.1. Isolation and Structural Elucidation of Eugenol and Its Antibacterial Activity

A. nilotica (L.) is a medicinal plant that is well-known for its applications in traditional medicines and has been used extensively to treat various respiratory diseases and other infectious diseases [19–21]. Here, we applied bioactivity guided approach to justify the wide-spread application of *A. nilotica* (L.) by conducting systematic phytochemical analysis on the leaves and tested its antimicrobial activity against several pathogenic bacteria including *E. coli*, *K. pneumoniae*, *Proteus* sp., *P. aeruginosa*, *S. aureus*, *S. typhi* and *S. pneumonia* (Supplementary Materials Figure S1). First, we extracted the chemical compounds from the leaves using methanol and used the methanol extract and marc for further fractionation with *n*-hexane, ethyl acetate, and *n*-butanol consecutively. The ethyl acetate extract was found to be the most effective extract against these bacterial isolates; hence it was selected for further phytochemical analysis in order to isolate the active compound(s). TLC and column chromatography were subsequently applied to fractionate *A. nilotica* (L.) leaves extracts by using solvent systems containing various ratios of *n*-hexane, methanol and ethyl acetate. Fractions were pooled and categorized based on their similarity in colors and retention factors into 5 sub-fractions named Fractions I–V (Supplementary Materials Table S1). Fractions IV and V show either null or neglectable antibacterial activities therefore were excluded without further discussion.

Table 1 shows the antibacterial activity of ethyl acetate fractions that were prepared from the ethyl acetate extract. The most prominent fraction with strong antibacterial activity was fraction I, which exhibits potential activity against all bacterial isolates with noticeable effects against *Proteus* sp., *S. typhi*, *P. aeruginosa* and *S. pneumonia*. Low activities were observed on fractions II and V and no activities were observed for fractions III and IV. The antibacterial activity for fraction I was similar or even higher in some cases from that of the positive control amoxicillin. Therefore, fraction I was selected for the determination of the MIC and MBC, which further confirmed the antibacterial activity of the fraction (Table 2). These strong antibacterial activities agree well with previous reports on *A. nilotica* (L.), which can be attributed to the presence of several active phytochemicals in the different parts of *A. nilotica* (L.) such as flavonoids, alkaloids, terpenes, and others active ingredients [20,22,24]. To elucidate the chemical structure and properties of the active ingredient in fraction I; GC-MS, NMR spectroscopy, FTIR and UV-visible spectroscopy were applied. The GC-MS spectrum indicated that the most prominent compound in fraction I is acetyeugenol with the formula $C_{12}H_{14}O_3$ —a eugenol derivative (Supplementary Materials Figure S2). This the first report in the purification of acetyeugenol, to our knowledge, from the leaves of *A. nilotica* (L.). 1H NMR, ^{13}C NMR and 2D NMR analyses confirmed the structure of fraction I—acetyeugenol (Supplementary Materials Figures S3–S6). The ^{13}C NMR spectral data of fraction I was compared with a reference ^{13}C NMR spectrum of pure acetyeugenol compound, as well as theoretically predicted ^{13}C NMR chemical shifts for the acetyeugenol (δ -ppm) [53,54]. These comparisons revealed similar chemical shifts for the 12 carbon positions in the backbone of acetyeugenol structure (Supplementary Materials Table S2). In

light of these results along with the GC-MS, we concluded that the major compound of fraction I is an acetyleugenol (Table 3). Fraction I was homogenous and showed maximum absorption peak (λ_{\max}) at 365 nm (Supplementary Materials Figure S7) and its FTIR spectrum indicated the presence of characteristic peaks of well-documented eugenol compounds (Supplementary Materials Figure S8) [55]. Previous results on eugenol isolated from other plants indicated significant molecular interactions with Extended Spectrum Beta-Lactamase enzymes (ESBL) and showed significant antibacterial activity against *E. coli* and *K. pneumoniae* [56]. This might explain the strong susceptibility of these bacterial isolates to the acetyleugenol (Tables 1 and 2). Eugenol oil, which contains about 10% acetyleugenol, has also been shown to exhibit a strong antimicrobial activity against *E. coli*, *S. aureus*, and *Candida albicans* and less activity against *Enterococcus faecalis* [57]. These studies clearly indicate that acetyleugenol may have broad-spectrum antimicrobial activity. Thus, it is important to explore its potential with different drug targets to understand its mechanism of actions. Such knowledge can help to develop novel compounds based on the acetyleugenol or eugenol scaffolds with stronger antimicrobial potential.

Table 1. Antibacterial activity of pure fractions (I-III) from the ethyl acetate extraction of the *A. nilotica* (L.) leaves.

Bacterial Isolates	Zone of Inhibition (mm)				
	Fraction-I	Fraction-II	Fraction-III	Methanol 5%	Amox.
<i>E. coli</i>	22.0 ± 3.00	5.33 ± 1.52	4.33 ± 2.52	n.d	28.7 ± 1.15
<i>K. pneumoniae</i>	15.0 ± 4.58	5.67 ± 2.31	5.33 ± 1.53	n.d	26.7 ± 0.58
<i>Proteus sp.</i>	29.0 ± 2.00	4.67 ± 2.08	3.33 ± 1.53	n.d	30.7 ± 1.15
<i>P. aeruginosa</i>	24.7 ± 3.51	7.67 ± 3.06	4.33 ± 1.15	n.d	25.4 ± 1.53
<i>S. aureus</i>	24.3 ± 4.16	6.00 ± 1.73	3.61 ± 2.08	n.d	23.0 ± 1.00
<i>S. typhi</i>	28.0 ± 5.57	2.00 ± 1.00	3.00 ± 2.00	n.d	28.0 ± 2.00
<i>S. pneumonia</i>	26.7 ± 1.53	5.33 ± 1.53	3.33 ± 1.52	n.d	21.7 ± 1.53

Amox. = Amoxicillin antibiotic (positive control). 5% Methanol = negative control. n.d = no detected activity. Data presented as mean ± std. (n = 3).

Table 2. The Minimum inhibitory concentration (MIC) and the minimal bactericidal concentration (MBC) of fraction I acetyleugenol.

Bacterial Isolates	MIC (mg/mL)	MBC (mg/mL)
<i>E. coli</i>	0.31	5.00
<i>K. pneumoniae</i>	1.25	10.0
<i>Proteus sp.</i>	0.63	5.00
<i>P. aeruginosa</i>	1.25	5.00
<i>S. aureus</i>	0.63	10.0
<i>S. typhi</i>	1.25	n.d
<i>S. pneumonia</i>	2.50	n.d

Table 3. GC/MS analysis profile of the pure ethylacetate extract from *A. nilotica* (L.) leaves.

Peak#	Name	Formula	Molecular Weight (g/mol)	Retention Time (min)	Retention Index	Area (%)
1	Phenol, 2-methoxy-4-(2-propenyl)-, acetate (acety Eugenol)	C ₁₂ H ₁₄ O ₃	206	8.242	1552	25.66
2	Dodecanoic acid	C ₁₂ H ₂₄ O ₂	200	11.425	1570	7.11
3	Tetradecanoic acid	C ₁₄ H ₂₈ O ₂	228	13.642	1769	3.01
4	Hexadecanoic acid, methyl ester	C ₁₇ H ₃₄ O ₂	270	15.625	1878	3.35
5	<i>n</i> -Hexadecanoic acid	C ₁₆ H ₃₂ O ₂	256	16.817	1968	22.99
6	3,4,5-Trihydroxybenzaldehyde	C ₇ H ₆ O ₄	154	18.325	1644	1.47
7	11-Octadecanoic acid, methyl ester	C ₁₉ H ₃₆ O ₂	296	18.900	2085	3.19
8	Heneicosanoic acid, methyl ester	C ₂₂ H ₄₄ O ₂	340	19.267	2375	1.09
9	Erucic acid	C ₂₂ H ₄₂ O ₂	338	19.958	2572	19.87
10	Octadecanoic acid	C ₁₈ H ₃₆ O ₂	284	20.150	2167	9.54
11	1,E-11,Z-13-Octadecatriene	C ₁₈ H ₃₂	248	21.617	1817	1.52
12	13-Oxabicyclo[9.3.1]pentadecane	C ₁₄ H ₂₆ O	210	23.492	1690	0.38
13	Methyl hexacosanoate	C ₂₇ H ₅₄ O ₂	410	23.858	2872	0.45
14	1-Gala-1-ido-octose	C ₈ H ₁₆ O ₈	240	25.567	2221	0.39

3.2. Molecular Docking of Acety Eugenol and Its Analogues Against MtPknB and MtPknE

To study the effect of acety Eugenol at a molecular level and explore its antimicrobial potential, two important *M. tuberculosis* drug targets were chosen; (i) MtPknB and (ii) MtPknE S/T protein kinases, which play pivotal roles in *M. tuberculosis* development and pathogenesis [34,35]. First, we used the acety Eugenol and conducted molecular docking experiments against the MtPknE protein kinase (PDB ID: 2H34). The results indicated relatively moderate docking score with a binding energy at -8.53 kcal/mol (Figure 1). The docking revealed that the acety Eugenol interacts with Ile96 residue of MtPknE via single hydrogen bond with 2.8 Å, which is formed between the carbonyl oxygen of the ester group of acety Eugenol and the hydrogen of the backbone NH group of Ile96 residue (Figure 1a). The two alkyl groups of the acety Eugenol were found to be embedded in a hydrophobic pocket that is formed by several hydrophobic residues, including Leu95, Ile96, Ala43, Val22, Val30, Val77, Val156, Phe158, Leu146, and Ile160 (Figure 1b). The vinyl group ($-\text{CH}=\text{CH}_2$) of the acety Eugenol is located in a solvent accessible region of the binding cavity and show no interaction with any of the residues in this cavity. This result indicates that the active functional group of the acety Eugenol in the interaction with MtPknE is the ester carbonyl group. A previous docking study on eugenol showed that both phenolic hydroxyl and methoxy oxygen of eugenol can interact with residues on the active site, as demonstrated in the case of *C. albicans* squalene epoxidase [58]. Thus, the acety Eugenol has smaller size relative to the binding cavity of the MtPknE kinases, which might potentially affect its orientation and binding. This might explain the relatively low binding energy and docking score.

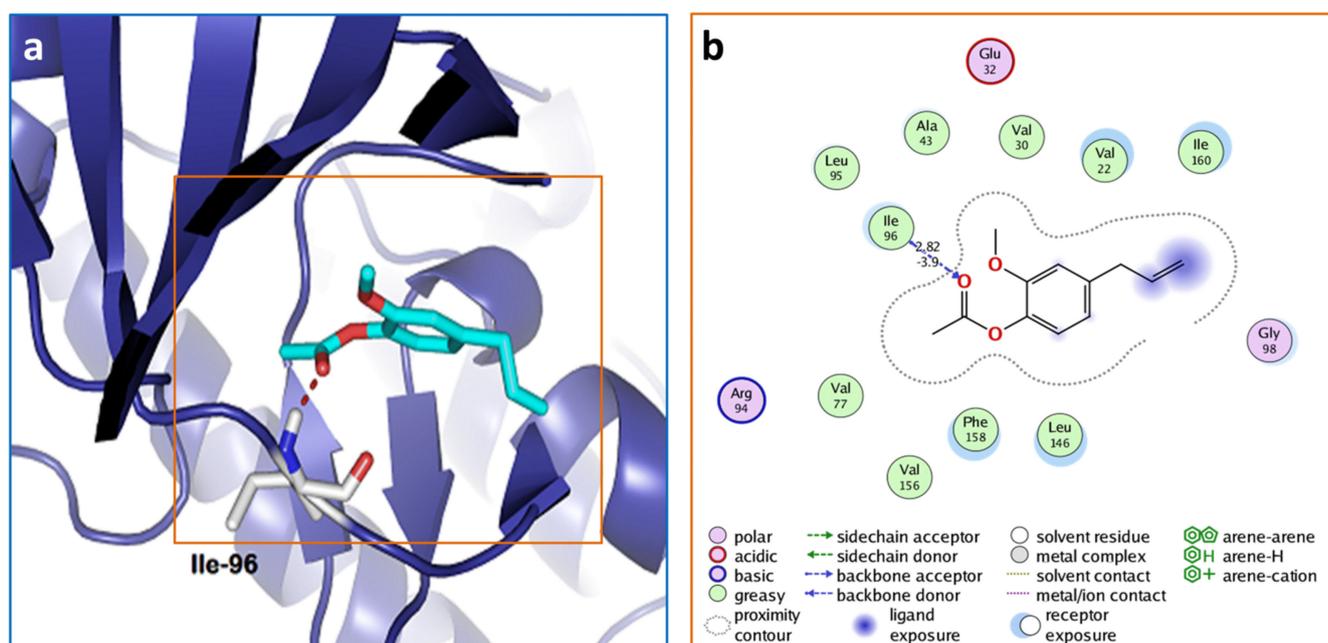


Figure 1. Molecular docking of the eugenol acetate showing (a) the binding interactions in the binding site of *MtPknE* kinase. The eugenol acetate shown as stick (cyan), and (b) the 2D representation of the docking generated by MOE software; the amino acid residues are colored according to their chemical nature shown in the key. The 2D views shows the hydrogen bonding network and the Van der Waal contacts with the hydrophobic residues (green).

In order to optimize the molecular interactions of the acetyleugenol for best orientation and binding in the hydrophobic binding pocket of *MtPknE*, we performed in silico synthesis for new compounds based on acetyleugenol skeleton modifications by adding either phenyl ring or indole to the ester moiety or the vinyl sites. This gave rise to five acetyleugenol analogues including the acetyleugenol and named compounds 1–5 as depicted in Figure 2. The compounds 1 and 3, which were named EUG-1 and EUG-3, respectively, are indole analogues with indole group at either the vinyl double bond site (EUG-1) or at the ester site (EUG-3) and similarly, the phenyl group was added to make compound 2 (EUG-2) and compound 4 (EUG-4). This is the first report, to our knowledge, to design and study the docking profile of these newly proposed acetyleugenol analogues.

To get insights into the mechanism of interactions and docking-based structure–activity relationships of these newly designed compounds *M. tuberculosis* protein Ser/Thr kinases *MtPknB* and *MtPknE* were used as molecular targets [34]. The binding energy values of the docking experiments are listed in Table 4. The docking results of these compounds appear to be similar between *MtPknB* and *MtPknE* with slightly higher selectivity to *MtPknE* than *MtPknB*. The most selective ligand with high ligand–protein affinity is the EUG-1, which has the lowest binding energy, i.e., the best interaction and docking score. The results indicate that the addition of an indole group to the double bond of the vinyl site has significantly improved the docking score in comparison with the acetyleugenol (Table 4). This incorporation of an indole ring to the acetyleugenol skeleton in the case of EUG-1 provides an additional hydrogen bond donor via the indole amino group to an existing hydrogen bond acceptor of the ester carbonyl group. The EUG-1 also provides a molecular bulk necessary for Van der Waals and hydrophobic interactions, which further favored the EUG-1 binding energetics and improved the docking scores with both *MtPknE* and *MtPknB* (Figure 3). The EUG-1 also reveals similar binding energies to *MtPknE* and *MtPknB* with -11.08 kcal/mol and -10.46 kcal/mol, respectively (Table 4). A recent study by Ali et al. [42] has examined several propolis compounds as potential anti-TB drugs including flavonoids pachypodol and pinobanksin-3-(E)-caffeate, which showed a lower affinity with docking scores at -9.1 kcal/mol binding energy against the *MtPknB* comparison to the control inhibitor mitoxantrone, which exhibited a better docking score with

−10.8 kcal/mol binding energy. When the same propolis were examined against other *M. tuberculosis* enzymes involving *MtPank*, *MtDprE1*, and *MtKasA*, the results indicate similar or better docking scores than that of mitoxantrone [42]. This is a clear indication that the indole analogue EUG-1 of the acetyeugenol is a potential anti-TB that can act against both *MtPknE* and *MtPknB* kinases. Another compound named IMB-YH-8 has also provided promising clinical trials and was shown to act specifically against the *MtPkn* enzymes with the most effect against the *MtPknB* [59].

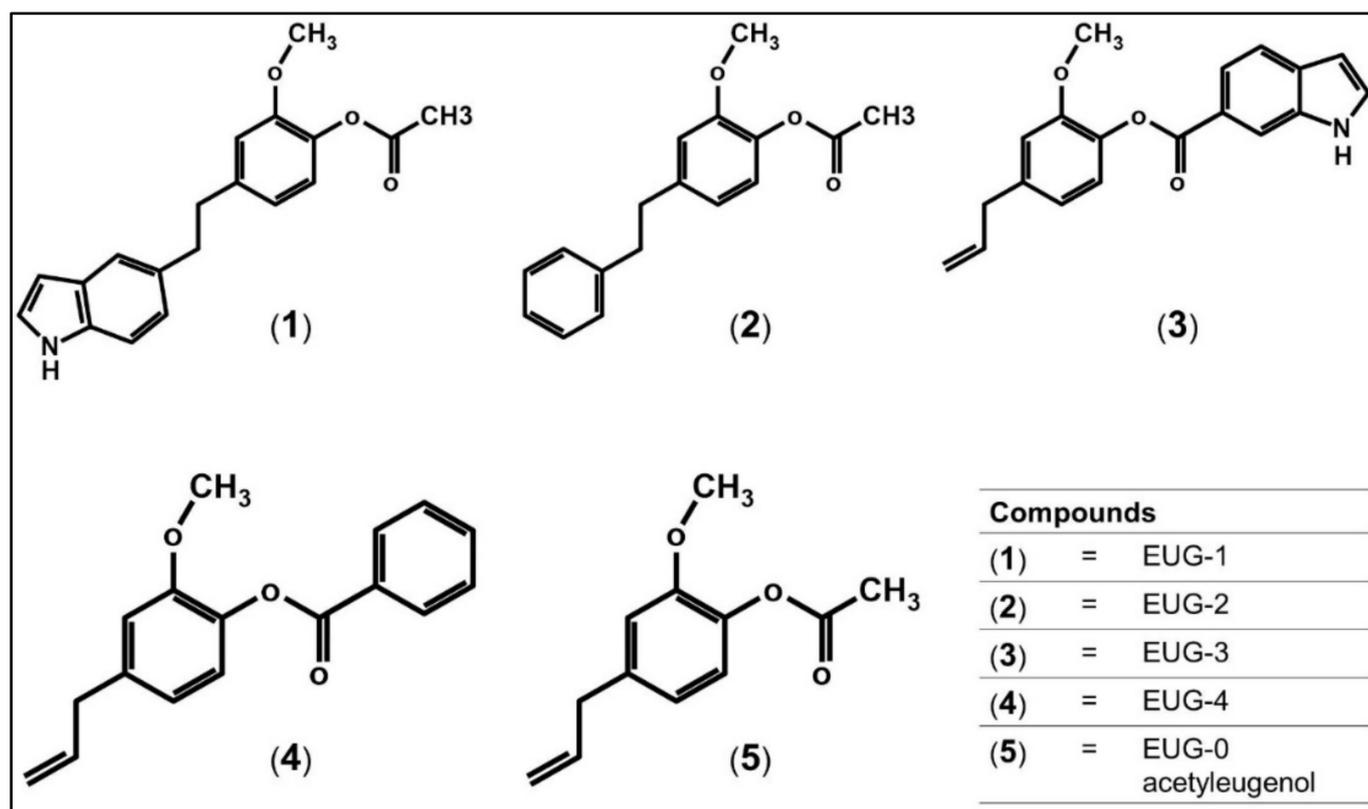


Figure 2. The acetyeugenol (5) from *A. nilotica* (L.) and the in silico synthesized phenyl and indole analogues.

Table 4. Molecular docking binding free energy of the acetyeugenol compound and its phenyl and indole analogues against *MtPknB* and *MtPknE* molecular targets.

Compound	Binding Energy (kcal/mol)		Reference
	<i>MtPknE</i>	<i>MtPknB</i>	
EUG-1	−11.08	−10.46	This study
EUG-2	−10.29	−10.19	This study
EUG-3	−10.05	−9.64	This study
EUG-4	−9.25	−9.35	This study
Acetyeugenol	−8.53	−7.60	This study
Mitoxantrone	n.a *	−10.8	[42]

* n.a = not available.

The docking results indicate that the EUG-1 interacts via its indole amino group and the carbonyl oxygen to form energetically favorable binding affinity with the drug target. It forms three hydrogen bonds with *MtPknE*; the first with NH group and the side chain of Asp100 with 3.07 Å bond length, while the remaining hydrogen bonds are formed between the carbonyl oxygen and the backbone of Leu95 and Ile96 with 3.31 Å and 2.87 Å bond length, respectively (Figure 3a). The EUG-1 interacts with *MtPknB* via a hydrogen bond that is formed between the indole ring of EUG-1 and Val95 with 2.67 Å bond length and another hydrogen bond between the EUG-1 carbonyl group with Phe19 with 3.41 Å bond

length. Furthermore, the EUG-1 also interacts with *MtPknB* via a relatively weak H- π interaction between its indole pyrrole moiety and a hydrogen of Leu17 (Figure 3b).

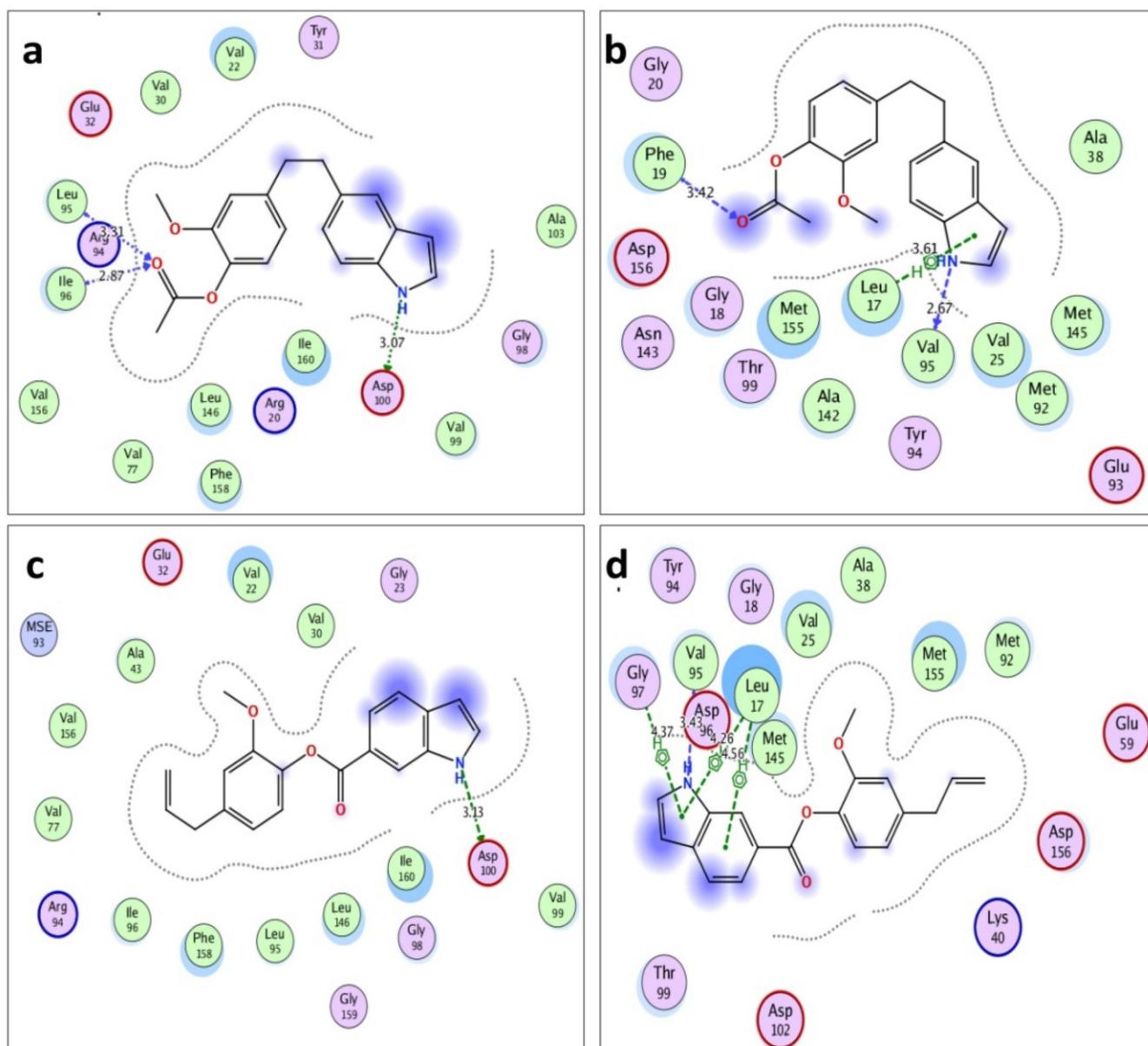


Figure 3. Binding interactions of eugenol-indole analogues with *MtPknE* and *MtPknB* S/T kinases; (a) EUG-1 with *MtPknE*, (b) EUG-1 with *MtPknB*, (c) EUG-3 with *MtPknE*, and (d) EUG-3 with *MtPknB*.

EUG-3, which has an indole group attached to the ester end of acetyeugenol, reveals improved docking scores relative to the acetyeugenol compound for both enzymes *MtPknE* and *MtPknB* with -10.29 kcal/mol and -10.19 kcal/mol binding energy, respectively (Table 1). These values are only slightly higher than that of mitoxantrone standard inhibitor for *MtPknB* that scores -10.8 kcal/mol with 0.8 μ M IC_{50} [42]. Unlike EUG-1, the carbonyl group of EUG-3 doesn't interact with any of the amino acid residues in the binding pocket owing most likely to the shielding provided by the bulky indole moiety that prevents the carbonyl from approaching any nearby residues as illustrated in the case of *MtPknE* (Supplementary Materials Figure S5c). EUG-3 interacts with *MtPknB* via one hydrogen bond that is formed between the amino group of the indole pyrrole and Val95. Additionally, we observed three weak H- π interactions among the EUG-3 aromatic benzyl and pyrrole rings and the enzyme Gly97, Leu17 and Met145 (Figure 3d). In contrast, the

phenyl analogues EUG-2 and EUG-4, which have a phenyl ring attached either to the ester group site or to the double bond of the vinyl group of the acetylenol skeleton, show weaker binding affinities for both enzymes comparing with that of the EUG-1 and EUG-3 (Table 1). EUG-2 has slightly better binding affinity comparing to EUG-4 with -10.05 kcal/mol binding energy for *MtPknE* and -9.64 kcal/mol for *MtPknB*. Altogether, all of the modifications applied in the present study have improved the binding affinity and docking score when compared with our starting compound acetylenol. Generally, replacing the methyl group of the ester moiety by indole rather than phenyl group has robustly influenced the binding energetics and interactions with the hydrophobic pocket of the enzymes. On the other hand, the choice of the site of substitution has also played a paramount role in the improvement of the binding affinity, hence the docking scores, especially with the *MtPknE* target. Therefore, these compounds are more likely to be good leads for the discovery of potential anti-tuberculosis drugs.

4. Conclusions

Here, we isolated and elucidated the molecular structure of acetylenol from the leaves of *A. nilotica* (L.) and confirmed its antibacterial potential against seven clinical isolates. We further demonstrated, for the first time, the potential anti-TB properties of the acetylenol analogues, which were designed, in silico synthesized, and used for molecular docking experiments against *M. tuberculosis* Ser/Thr protein kinases—the *MtPknE* and *MtPknB*. Our docking results indicate that the indole analogues EUG-1 and EUG-3 are potential inhibitors targeting these kinases with a promising outcome. The effective interactions of EUG-1 and EUG-3 with both targets suggest a dual mechanism of action for these compounds, and propose a possible interference with the pathogenesis and cell division of *M. tuberculosis* at different stages based on the functions of PknE and PknB in *M. tuberculosis* development. This urgently requires comprehensive in vivo and in vitro experiments to understand their mechanism of actions against *M. tuberculosis* or other mycobacterial models such as *M. smegmatis* to enable designing an effective anti-TB drug. The present study also encourages further investigation of other acetylenol derivatives against these kinases or other potential anti-TB targets.

Supplementary Materials: The following are available online at <https://www.mdpi.com/2036-7481/12/1/1/s1> Table S1: thin layer chromatography profile, Table S2 Comparison of the ^{13}C NMR spectra of fraction 1 with literature and theoretical values, Figure S1: phytochemical and antibacterial screening, Figure S2: GC-MS analysis of acetylenol, Figure S3: GC-MS spectral comparison of the major compound in subgroup 1 with the first hit of the NIST search library, Figure S4: ^{13}C NMR spectrum of acetylenol, Figure S5: ^1H NMR and the DEPT experiments of acetylenol, Figure S6: 2D COSY correlation of acetylenol, Figure S7: UV-visible spectrum of acetylenol, Figure S8: FT-IR analysis of the acetylenol.

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