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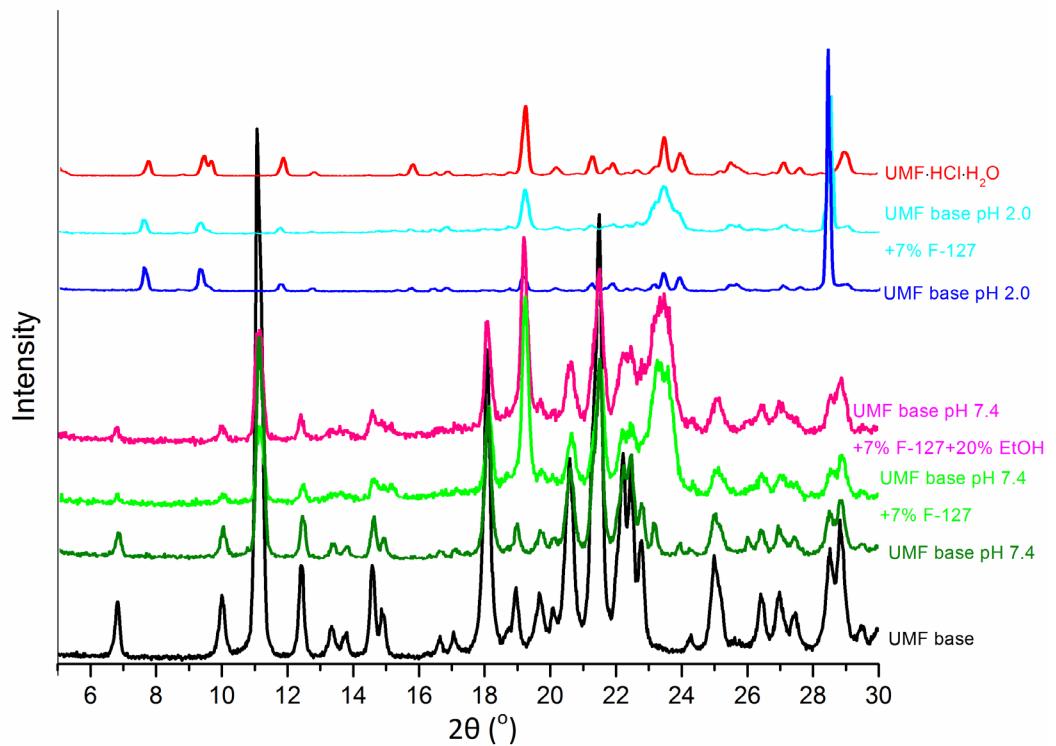


Figure S1. The PXRD patterns of: raw UMF base (black), raw UMF·HCl·H₂O (red), solid residuals after the solubility experiments of UMF base in buffer pH 2.0 (blue) and with 7% F-127 (cyan), in buffer pH 7.4 (olive), with 7% F-127 (green), and with 7% F-127 and 20% ethanol (pink).

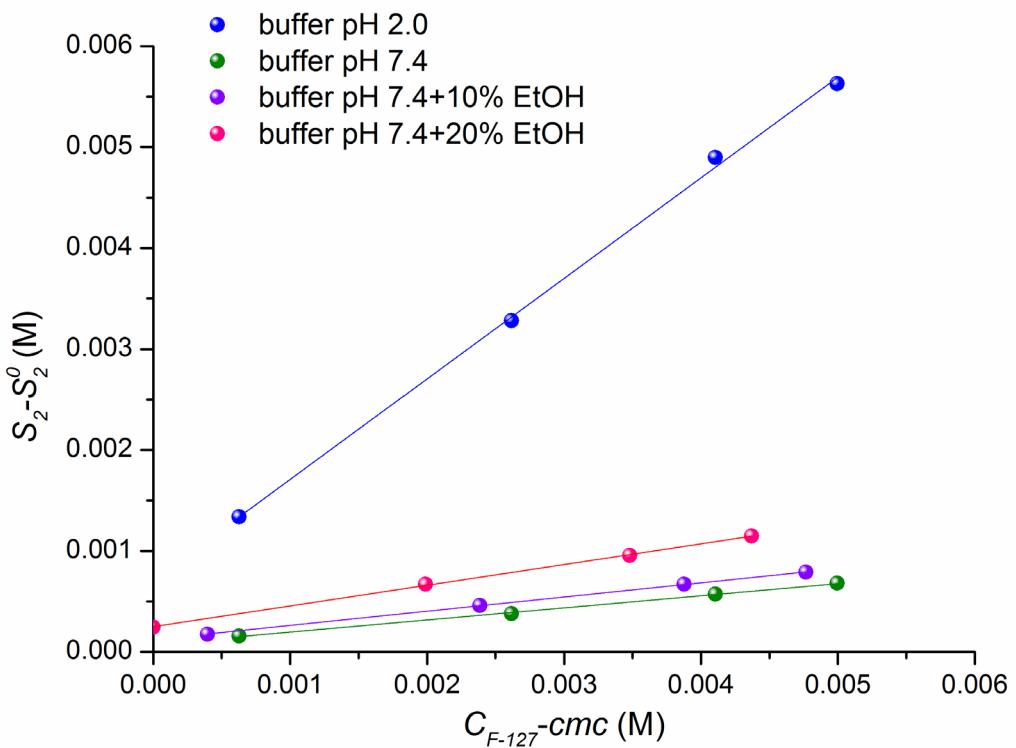


Figure S2. Plots of $(S_2 - S_2^0)$ on $(C_{F-127} - cmc)$ dependences at different concentrations of F-127 in buffers pH 2.0 (blue) and pH 7.4 (olive) and in the presence of 10 % (violet) and 20 % (pink) of ethanol.

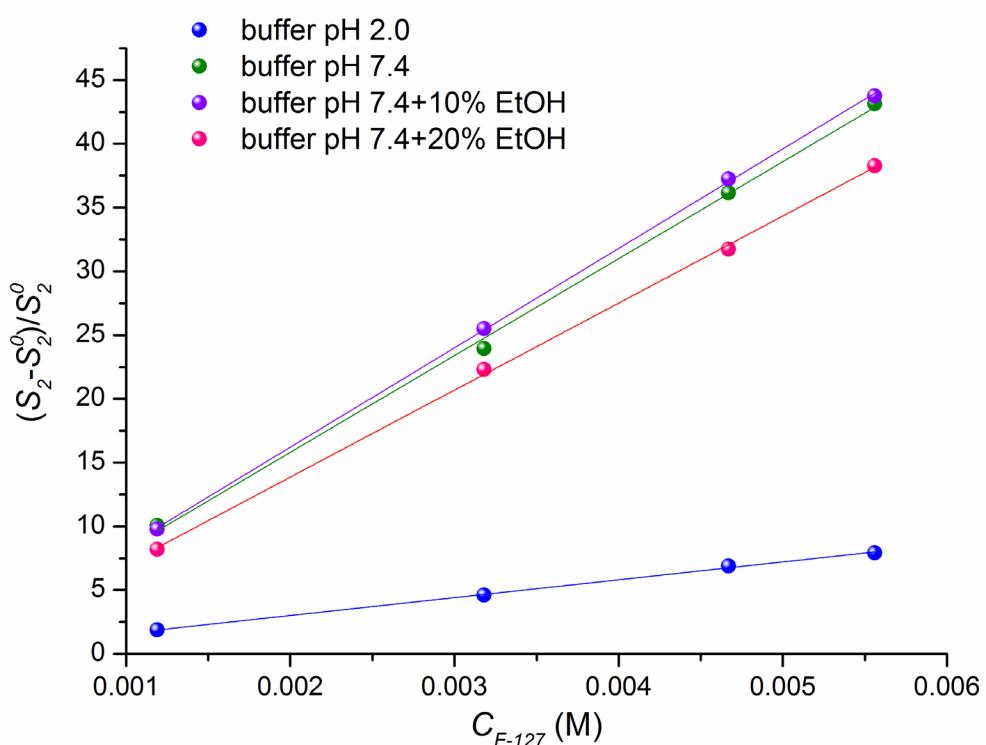


Figure S3. Plots correlating the UMF solubility in micellar F-127 solutions normalized by the aqueous solubility $((S_2 - S_2^0)/S_2^0)$ on F-127 concentration (C_{F-127}) in buffers pH 2.0 (blue) and pH 7.4 (olive) and in the presence of 10 % (violet) and 20 % (pink) of ethanol.

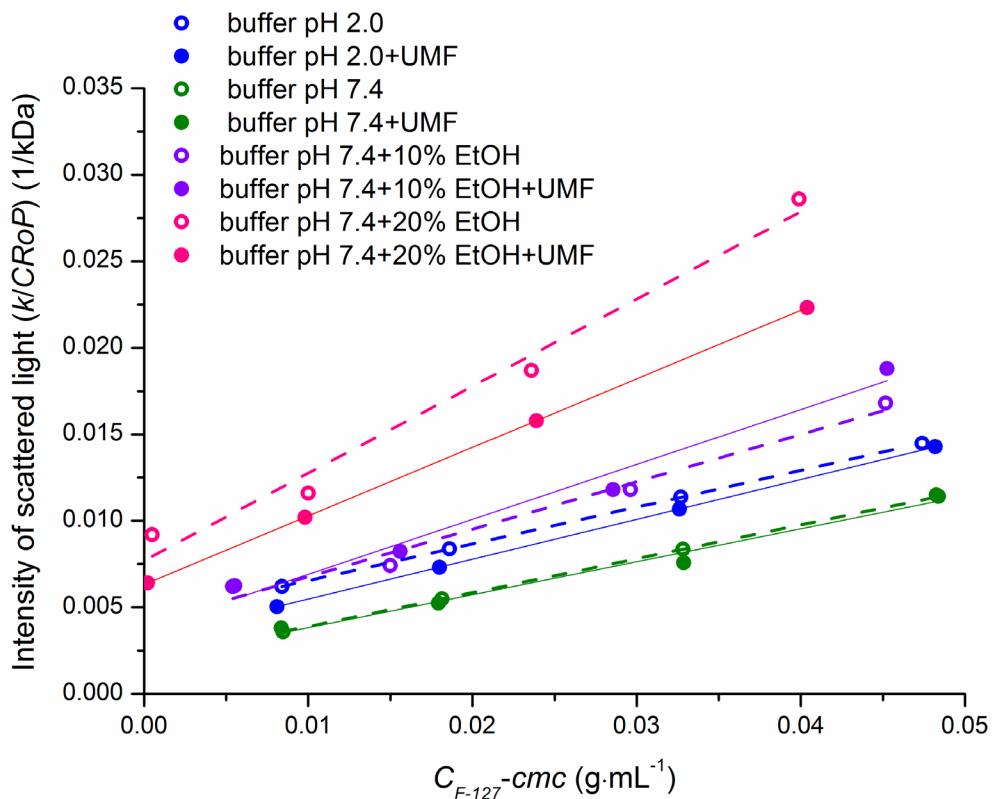


Figure S4. Debye-plots for systems of pluronic F-127 with and without UMF: buffer pH 2.0, buffer pH 7.4, buffer pH 7.4+10% EtOH and buffer pH 7.4+20% EtOH.

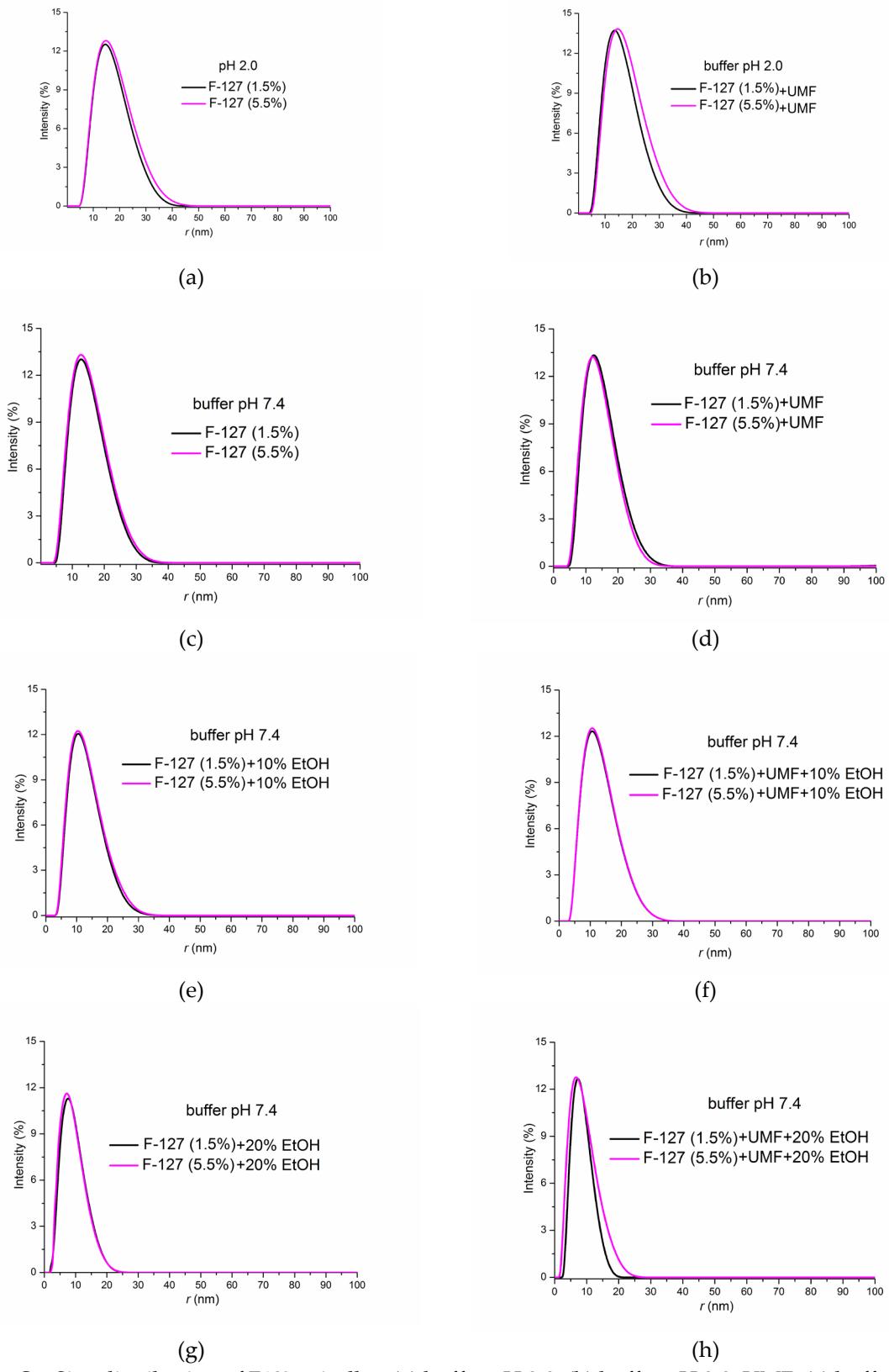


Figure S5. Size distribution of F127 micelles: (a) buffer pH 2.0; (b) buffer pH 2.0+UMF; (c) buffer pH 7.4; (d) buffer pH 7.4+UMF; (e) buffer pH 7.4+10% EtOH; (f) buffer pH 7.4+10% EtOH+UMF; (g) buffer pH 7.4+20% EtOH; (h) buffer pH 7.4+20% EtOH+UMF.

Table S1. UMF experimental solubility (S_{UMF}) in buffer pH 2.0, pH 7.4, and with the addition of F-127, and 10% or 20% ethanol at 310.15 K.

C_{F-127} (w/v%)	buffer	buffer	buffer pH 7.4	buffer pH 7.4
	pH 2.0	pH 7.4	+10% EtOH	+20% EtOH
	$S_{UMF} \cdot 10^3$ (M)	$S_{UMF} \cdot 10^4$ (M)	$S_{UMF} \cdot 10^4$ (M)	$S_{UMF} \cdot 10^4$ (M)
0	0.71±0.02	1.60±0.01	1.80±0.01	0.30±0.01
1.5	2.05±0.04	1.74±0.03	1.94±0.01	2.76±0.03
4.0	3.99±0.08	3.93±0.03	4.77±0.03	7.00±0.05
5.5	5.61±0.04	5.86±0.02	6.89±0.07	9.83±0.11
7.0	6.34±0.03	6.96±0.04	8.06±0.07	11.80±0.09

Each solubility value represents the mean ± SD ($n \geq 3$)

The standard uncertainties are $u(T) = 0.15$ K

Table S2. Donor solution concentrations (C_0), steady state flux (J), and permeability coefficients (P_{app}) of UMF, 310.15 K.

Investigated system	C_0 (M)	J ($\mu\text{mol}\cdot\text{cm}^{-2}\cdot\text{sec}^{-1}$)	P_{app} ($\text{cm}\cdot\text{s}^{-1}$)
pH 7.4 cellulose membrane MWCO 12-14 kDa			
UMF	1.04·10 ⁻⁵	4.36·10 ⁻⁷	(4.19±0.11)·10 ⁻⁵
UMF+10% EtOH	8.55 ·10 ⁻⁶	4.93·10 ⁻⁷	(5.76±0.14)·10 ⁻⁵
UMF+20% EtOH	1.25·10 ⁻⁵	6.25·10 ⁻⁷	(4.98±0.11)·10 ⁻⁵
UMF+1.5% F-127	5.24 ·10 ⁻⁵	1.22·10 ⁻⁶	(2.34±0.07)·10 ⁻⁵
UMF+4.0% F-127	5.75 ·10 ⁻⁵	1.02·10 ⁻⁶	(1.78±0.04)·10 ⁻⁵
UMF+10% EtOH+1.5% F-127	1.88 ·10 ⁻⁵	4.86·10 ⁻⁷	(2.59±0.06)·10 ⁻⁵
UMF+20% EtOH+1.5% F-127	7.77 ·10 ⁻⁵	3.58·10 ⁻⁷	(4.61±0.04)·10 ⁻⁶
pH 2.0 cellulose membrane MWCO 12-14 kDa			
UMF	1.41·10 ⁻⁴	2.76·10 ⁻⁶	(1.96±0.07)·10 ⁻⁵
UMF+1.5% F-127	3.57 ·10 ⁻⁴	4.19·10 ⁻⁶	(1.17±0.02)·10 ⁻⁵
UMF+4.0% F-127	1.26·10 ⁻³	7.09·10 ⁻⁶	(5.63±0.18)·10 ⁻⁶
pH 2.0 cellulose membrane MWCO 6-8 kDa			
UMF	1.80 ·10 ⁻⁴	3.27·10 ⁻⁶	(1.82±0.05)·10 ⁻⁵
UMF+1.5% F-127	3.07 ·10 ⁻⁴	3.36·10 ⁻⁶	(1.09±0.02)·10 ⁻⁵