

Article

Impact of Coated Zinc Oxide Nanoparticles on Photosystem II of Tomato Plants

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Table S1. Definitions of the chlorophyll fluorescence parameters used in the experiments.

Parameter	Definition	Calculation
Fv/Fm	Maximum efficiency of PSII photochemistry	$(Fm - Fo)/Fm$
Φ_{PSII}	Effective quantum yield of PSII photochemistry	$(Fm' - Fs)/Fm'$
Φ_{NPQ}	Quantum yield of regulated non-photochemical energy loss in PSII	$Fs/Fm' - Fs/Fm$
Φ_{NO}	Quantum yield of nonregulated energy loss in PSII	Fs/Fm
Fv'/Fm'	Efficiency of open PSII centers	$(Fm' - Fo')/Fm'$
Fv/Fo	Efficiency of the oxygen evolving complex (OEC) on the donor side of PSII	$(Fm - Fo)/Fo$
ETR	Electron transport rate	$\Phi_{PSII} \times PAR \times c \times abs$, where PAR is the photosynthetically active radiation, c is 0.5, and abs is the total light absorption of the leaf taken as 0.84
qp	Photochemical quenching, representing the redox state of quinone A (Q_A), or in other words the fraction of PSII reaction centers in open state	$(Fm' - Fs)/(Fm' - Fo')$
NPQ	Non-photochemical quenching reflecting the dissipation of excitation energy as heat	$(Fm - Fm')/Fm'$
EXC	Excess excitation energy	$(1 - qp) \times Fv'/Fm'$

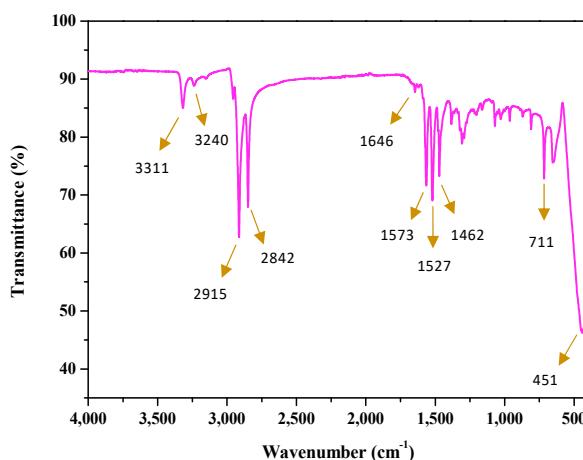


Figure S1. Fourier transform infrared spectroscopy (FT-IR) spectra of the ZnO@OAm NPs.

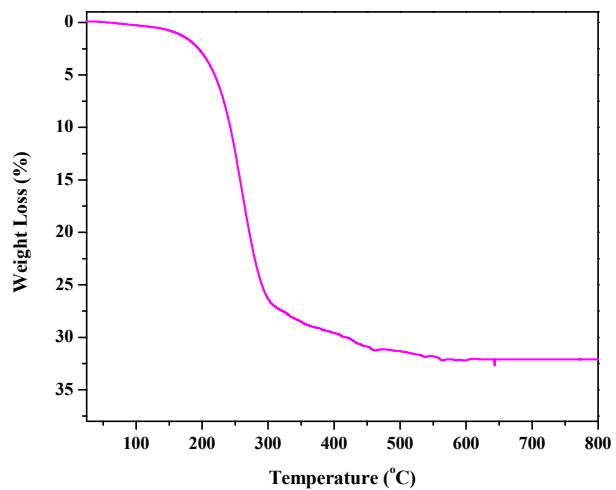


Figure S2. Thermogravimetric analysis curve of ZnO@OAm NPs with the main steps of organic coating weight loss.

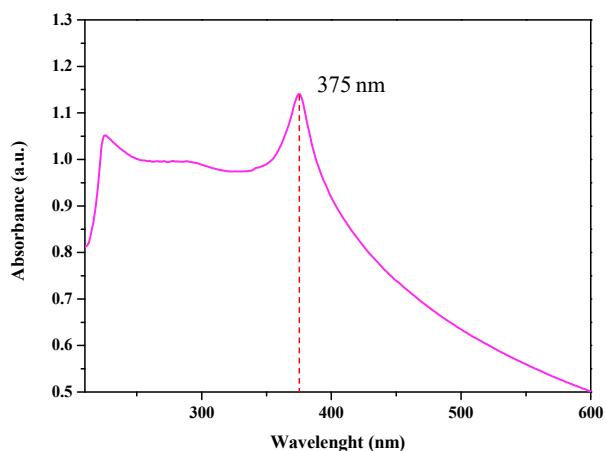


Figure S3. UV-Vis absorption spectrum of the ZnO@OAm NPs after preparation of NPs in a water-ethanol media.

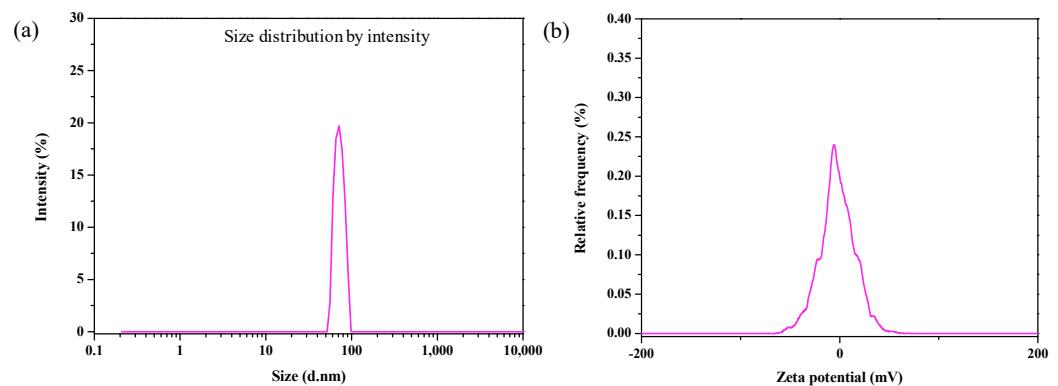


Figure S4. Size distribution (a) and ζ -potential (b) of the solvothermally prepared ZnO@OAm NPs.