



Article Formation and Characterization of Stable TiO₂/Cu_xO-Based Solar Cells

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Abstract: According to increasing demand for energy, PV cells seem to be one of the best answers for human needs. Considering features such as availability, low production costs, high stability, etc., metal oxide semiconductors (MOS) are a focus of attention for many scientists. Amongst MOS, TiO2 and Cu_xO seem to be promising materials for obtaining an effective photoconversion effect. In this paper, specific investigation, aimed at the manufacturing of the complete photovoltaic structure based on this concept is described in detail. A set of samples manufactured by DC magnetron sputtering, with various process parameters, is characterized by morphology comparison, layer structure and material composition investigation, and finally by the obtained photovoltaic parameters. Based on SEM studies, it was established that the films are deposited uniformly and complete their formation; without clearly defined faces, the conglomerates of the film grow individually. These are areas with a uniform structure and orientation of atoms. The sizes of conglomerates are in a normal direction range from 20 to 530 nm and increase with film thickness. The film thickness was in the range from 318 to 1654 nm, respectively. The I-V study confirms the photovoltaic behavior of thin film solar cells. The open-circuit voltage (V_{oc}) and short-circuit current density (J_{sc}) values of the photovoltaic devices ranged from 1.5 to 300 mV and from 0.45 to 7.26 μ A/cm³, respectively, which corresponds to the maximum efficiency at the level of 0.01%. Specific analysis of the junction operation on the basis of characteristics flow, R_s, and R_{sh} values is delivered.

Keywords: thin films; TiO_2/Cu_xO ; solar cells; renewable energy; DC magnetron sputtering

1. Introduction

People always needed effective and easily accessible sources of energy to function. Over the centuries, and especially recently, such a demand evidently grew, and this trend will continue. Renewable sources, including solar energy converted into electricity using photovoltaic cells, have an increasing share in the energy market. Photovoltaic cells often have connotations with the silicon, while scientists are still looking for more cost-effective solutions, characterized by non-toxicity, stability, and availability.

One of the ways of development of photovoltaic cells concerns Cu_2ZnSnS_4 (CZTS) thin films [1–7]. CZTS has the band gap of ~1.50 eV, being a very close value to the best band gap required by a semiconductor solar cell (according to Shockley–Queisser theory) -1.4-1.45 eV [8,9]. Thanks to features such as the direct band gap and a high absorption



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). coefficient, CZTS material can be applied as the absorption layer of thin film solar cells. Guo et al. [2] implemented Ti as an intermediate layer in the CZTS structure at the interface between Mo and Cu₂ZnSnS₄. The authors showed that such an implementation improved the Hall mobility, surface evenness, crystallinity, and absorptivity of CZTS, as well as inhibited the formation of voids at the interface between the Mo and Cu₂ZnSnS₄ layer. The authors showed also that the application of the Ti layer resulted in an increase of 32% of the open-circuit voltage of the CZTS as well as the increase of 57% in conversion efficiency. Lee et al. [4] investigated the FTO/TiO₂/In₂S₃/Cu₂ZnSnS₄ cell with the spray deposition method and received the efficiency of 3.34%. The authors concluded, however, that the aqueous-based solution, which they used, could result in surface oxidation and limit grain growth, which leads to a low device performance. Satale et al. [5] considered a FTO/TiO₂/CdS/CZTS/Au structure and showed that by optimizing the composition of Cu and Zn, it was possible to obtain the power conversion efficiency up to 1.04%. The authors made the CZTS film at the temp. 250 °C, without sulfurization, with CdS as an interface, and with TiO₂ nanorod arrays processed hydrothermally. Khalil et al. [6] reported about CZTS fabricated on flexible Mo foil substrates with the electrodeposition-annealing method. The authors obtained 0.55% power conversion efficiency.

Promising materials for PV implementation are metal oxide (MO) semiconductors (MOS), which, thanks to features such as high sensitivity, low power consumption, and ease of synthesis, are widely investigated [10–16].

Very good representatives of MO are titanium dioxide (TiO₂) and cupric oxide/cuprous oxide (CuO/Cu₂O), as they are cheap materials, chemically neutral, with environmentally friendly resources and highly resistant to photocorrosion [17–24]. On the other hand, as the titanium dioxide has a bandgap of 2.96–3.2 eV [25–27] (depending on the TiO₂ form), it may absorb only the ultraviolet, which is ~7% of the sunlight spectrum. Anyway, by appropriate application of CuO/Cu₂O with direct band gaps, it enables absorption of most of the solar spectrum. As bangaps of CuO/Cu₂O depend strongly on the deposition method, it is worth noting that it may take values from the range 1.0–2.1 eV for CuO [28–32] and 1.82–2.2 eV [33–36] for Cu₂O, respectively. In our research, for CuO structure, we received values from the range 1.02–1.28 [37]. As in the case of multi-layer solar cells, there appears a lattice mismatch between TiO₂ and Cu₂O of ~11%, while in the case of CuO and Cu₂O, it was ~8.9% [38] (for TiO₂/Cu₂O, Hussain et al. [26] reported the value of ~25%). In [40] we showed that a lattice mismatch problem revealed a break in the continuity of the crystal structure of samples prepared in our lab.

Pavan et al. [18] investigated a TiO_2/Cu_2O all-oxide heterojunction solar cell with the use of the spray pyrolysis method. The authors applied a combinatorial approach and high throughput electrical and optical characterization, and derived a direct band gap of 2.5 eV, which revealed a strong absorption coefficient. The authors showed that almost every photon with energy above 2.5 eV can be absorbed within a Cu₂O layer (300 nm thick).

The highest conversion efficiency for TiO_2/CuO solar cells, 1.62%, was reported by Rokhmat et al. [19]. The authors reached that by depositing copper particles using fix-current electroplating and with a copper particle content of 7.93% mass.

Patel et al. [22] produced Ag/CuO_x/TiO₂/FTO solar cells, with a p–n junction of 150 nm. The authors investigated several devices, with different oxygen vacancies in CuO_x, and studied how it influences the performance of cells. They showed that thanks to average transmittance of 28–52%, such devices become transparent for the visible spectrum of the solar light, and semitransparent for the visual notion. The solar cell characteristics were carried out for the UV light of 365 nm, and revealed a linear increase in V_{oc} (open circuit voltage) from 0.266 to 1.01 V, and the change in CuO_x phases from Cu₂O to CuO was observed. The best solar cell with the CuO as an active layer was obtained for the following parameters: open circuit voltage—1.01 V, current density—3.48 mA/cm², and fill factor—31%. Such a combination resulted in the overall efficiency of 16.22% (for an input power of 7 mW/cm²). The authors also investigated the influence of TiO₂ as a

buffer layer and showed that the CuO/TiO_2 device ensures the highest efficiency due to large conduction band offset. Cu₂O and TiO₂ were applied in dye-sensitized solar cells (DSSCs), which also are promising solar devices with high effiency [41-47]. Ursu et al. [43] have investigated p-type DSSC based on full Cu_2O electrodes with the use of traditional dye and a redox mediator. The authors prepared the counter electrode and the semiconductor material for the cathode based on Cu_2O in single low-temperature hydrothermal technology. Such a structure provided twice as much JSC and VOC as a DSSC with the use of platinum as a counter electrode. By replacing conventional Pt with a Cu₂O/Cu counter electrode, the power conversion efficiency was improved by app. 60%. Baptayev et al. [47] considered In-doped TiO₂ photoanode for DSSC, prepared with the use of a simple surface-doping technique by soaking the TiO_2 film in acidic In^{3+} solution at 70 °C in a different time period, followed by sintering at 450 °C. Structural characterization of such a structure revealed the successful attachment of indium to the surface of TiO_2 films and showed that the amount of In dopant was proportional to the soaking time. The authors reported that photoelectric conversation efficiency of such doped devices, for a soaking time of 30 min, resulted in an increase of 18.0% compared to the cells without doping. Indium doping resulted in the downward shift of the TiO₂ conduction band as well as a decrease in electron recombination, which caused the significant increase in the Voc and fill factor values.

It is also worth mentioning that Cu₂O was widely studied, e.g., in the context of photoelectrochemical (PEC) cells, which can be applied for solar water splitting [48,49]. Bai et al. [48] showed that Cu₂O/Ni(OH)₂, used as a photocathode, results in a larger photocurrent and smaller onset potential compared with Cu₂O. The authors obtained the photoconversion efficiency at the level of 0.20% under one sun illumination. Chen et al. [49] considered a dual Cu₂O cell and showed that the deposition of the TiO_x overlayer on the Cu₂O supplies an additional photovoltage, which makes the hydrogen evolution reaction proceed faster. For the tandem device of n-TiO_x/p-Cu₂O//n-Cu₂O, the authors obtained the solar-to-hydrogen efficiency of 0.32%.

Yet another application of TiO₂ and Cu_xO can be found in the context of perovskite solar cells [50–56]. TiO₂ is used as an electron transport layer (ETL), while Cu_xO is used as a hole transport layer (HTL). Zhu et al. [51] applied magnetron sputtering technology on FTO glass to create different thicknesses and shapes of TiO2 ETLs, and based on this technology, prepared planar heterojunction perovskite solar cells. The authors observed a decrease in the surface roughness of the TiO2 film together with an increase in the sputtering time, while the electrical homogeneity and square resistance of the film increased. They obtained the maximum film transmittance value of 82.29% and the highest photoelectric conversion efficiency (PCE) of perovskite solar cells of 12.42%. Chen et al. [55] examined perovskite solar cells based on CuO-Cu₂O thin films, fabricated on ITO-coated glass. The authors applied a CH₃NH₃PbI₃ perovskite absorber on top of a CuO-Cu₂O structure with the use of a one-step spin-coating process with a toluene washing treatment. By applying different temperatures and thicknesses of Cu-Cu₂O film, they reached the maximum PCE value of 8.1% and concluded that such a structure is a good candidate for alternative HTL in large-scale perovskite solar cell production.

Among many heterojunction solar cells being considered in literature (see e.g., [57–63]), our group deals with TiO_2/Cu_xO and we already published several papers with investigations of TiO_2 and Cu_xO cells fabricated with the direct current magnetron sputtering (DC-MS) with different process conditions [38,40,64–66]. In [64] TiO_2 , Cu_2O , as well as TiO_2/Cu_2O , structures were investigated, with the wide variety of measurements carried out: morphology, cross-section, topography, roughness, and transmission spectra of the films. Despite the selection of different parameters, no photovoltaic effect was detected. The first success of obtaining the TiO_2/CuO solar cell was reported in [65]. The authors improved the performance of TiO_2 and CuO structures by increasing the oxygen flow rates and substrate temperature during the deposition of the cells mentioned above, compared with [64]. It allowed us to obtain the efficiency of 0.24%. In 2021, again by applying different

parameters, such as substrate temperature, oxygen and argon flow rates, as well as time deposition for the $TiO_2/Cu_2O/CuO/Cu_2O$ cell, the authors obtained the efficiency of 0.9%, which is the best result so far [38]. In [66], the authors investigated $TiO_2/CuO/Cu_2O$ solar cells with the following measurements: X-ray diffraction, scanning electron microscopy, and current–voltage characteristics. The main goal was to check how the deposition times of the top contact point of Cu and the TiO_2 layer influence the structural and electrical properties of solar cells being investigated. Such an approach did not result in detecting expected efficiency. The paper [40] contains an analysis of the $TiO_2:ZnO/CuO$ structure performance depending on the Cu diffusion in the layer formation process. It was shown that the presence of a Cu buffer layer, as well as a smooth start to the deposition procedure of the CuO layer, play a crucial role in the cell production procedure. Investigations revealed that the high resistance and inconsistency must be eliminated to improve the parameters of the structure.

In the present paper, with the use of the direct current magnetron sputtering (DC-MS) method, the authors investigate the influence of different parameters, such as layer thickness and sputtering time, on the properties and efficiency of TiO_2 and Cu_xO thin film layers.

2. Materials and Methods

 TiO_2/Cu_xO thin film solar cells were deposited using the direct current (DC) reactive magnetron sputtering method.

Titanium dioxide and copper oxide thin films were sputtered by two metallic targets: Ti (99.995% purity) and Cu (99.995% purity), respectively, with a diameter of each 25.4 mm (Kurt J. Lesker Company, Leonards-On-Sea, UK) on commercial silicon plates (N-type Si, 100) or glass-coated indium-tin oxide (ITO $100 \times 100 \times 1.1$ mm, Kavio).

Initially, the n-type TiO_2 emitter layer was deposited on substrates for 25 min or 24 min (for the sample #38) in a mixture of argon: oxygen (Ar:O₂) with a ratio of 1:2.5. The magnetron was powered by 100 W.

Next, a thin film with Cu buffer was deposited for 5 s (#28) or 7 s (samples #29, #32, #35, #37, and #38) for each sample with argon flow rates of only 4 cm³/s using a Cu target. Then, the magnetron shutter was closed, flows were set for deposition of the Cu_xO layer, the plasma beam was stabilised for 20 s (samples #28, #32), 30 s (samples #29, #38), or 60 s (samples #35, #37) with the shutter closed, and the p-type Cu_xO absorber layers were grown. A mixed atmosphere Ar:O₂w had a ratio of 1:3.5 and the magnetron was powered by 70 W.

For the thin films of TiO₂ and Cu_xO, the pressure during deposition was kept at $\sim 1 \times 10^{-2}$ mbar, the distance between the substrate, the target was equal to 58 mm, and the substrate temperature during thin film deposition was kept at 300 °C. Finally, a thin Cu film was deposited on top of the Cu_xO as a contact. Details of the sputtering conditions and values of the thickness of the TiO₂ and Cu_xO layers are listed in Tables 1 and 2. A scheme of the TiO₂/Cu_xO thin film heterostructure is shown in Figure 1.

	#28	#29	#32	#35	#37	#38
Time [min]			25			24
Power [W]	100					
Pressure [mbar]	$1.15 imes 10^{-2}$	$1.05 imes 10^{-2}$	$1.11 imes 10^{-2}$	$1.08 imes 10^{-2}$	$1.06 imes 10^{-2}$	$1.11 imes 10^{-2}$
Distance between the source and substrate [mm]	58					
Oxygen flow rate [cm ³ /s]	2.5					
Argon flow rate [cm ³ /s]	1					
Substrate temperature [°C]	300					
Thickness [nm]	49	41	54	43	49	45

Table 1. Growth parameters of the TiO₂ window layer.

	#28	#29	#32	#35	#37	#38
Time [min]	30	22		25		22
Power [W]	70					
Pressure [mbar]	$1.11 imes 10^{-2}$	$1.11 imes 10^{-2}$	$1.14 imes 10^{-2}$	$1.11 imes 10^{-2}$	$1.11 imes 10^{-2}$	$1.10 imes 10^{-2}$
Distance between the source and substrate [mm]	58					
Oxygen flow rate [cm ³ /s]	3.5					
Argon flow rate [cm ³ /s] Substrate temperature [°C]			30	1 00		
Thickness [nm]	1654	1201	1260	1305	1419	318

Table 2. Growth parameters of the CuO absorber layer.



Figure 1. Scheme of the sputtered layers $\text{TiO}_2,$ Cu buffer, Cu_xO, and Cu contact.

To create PV devices, two Cu electrodes were attached to the upper Cu contacts and the ITO by Ag (silver) conductive glue.

The authors invite readers to view a YouTube video of the PREVAC apparatus and PVD laboratory, which are linked in the Supplementary Materials section of this paper.

3. Results

3.1. Cross Section Analysis of TiO_2/Cu_xO Thin Films

Figures 2–7 show the cross-section of TiO_2/Cu_xO layers deposited on (Si) silicon substrates according to the growth parameters shown in Tables 1 and 3. It is obvious that the films have a similar structure. In the case of CuO, small TiO₂ clusters were formed on the CuO surface.



Figure 2. The SEM cross-section image (a,b) of TiO₂/Cu_xO layers and EDX spectra of the sample #28 (c).



Figure 3. The SEM cross-section image (a,b) of TiO₂/Cu_xO layers and EDX spectra (c) of the sample #29.



Figure 4. The SEM cross-section image (a,b) of TiO₂/Cu_xO layers and EDX spectra (c) of the sample #32.







Figure 6. The SEM cross-section image (a,b) of TiO₂/Cu_xO layers and EDX spectra (c) of the sample #37.



Figure 7. The SEM cross-section image (a,b) of TiO₂/Cu_xO layers and EDX spectra (c) of the sample #38.

	#28	#29	#32	#35	#37	#38
V _{oc} [mV]	176	300	75	14	23	71
Isc [µA]	21	38	4.3	7	1.5	5.3
μ[%]	0.01	0.01				
Contact area [mm ²]	30	35	45		36	40
V _{oc} /CuO thickness [mV/nm]	0.10	0.25	0.06	0.01	0.02	0.22
V _{oc} /TiO2 thickness [mV/nm]	3.59	7.32	1.39	0.33	0.47	1.58

Table 3. Basic I-V parameters of the manufactured samples.

 Cu_xO forms a fine-grained film (Figure 2a,b) with a non-monotonic distribution of these crystallites in thickness; the structure resembles a fibrous one, as evidenced by the alternation of dark and light areas in the images. Higher density is characteristic of the initial layers, and the following conglomerates are grouped in the vertical direction of film growth.

For sample #28 (Figure 2a,b), it should be noted that the thickness of 1654 nm of the CuO layer is the largest. The structure of the film is more uniform and completed its forming; without clearly defined faces, the columns of the film grow in different directions. The clearly observed grains are already present in Figure 3a,b, and the growth of individual conglomerates is observed. These are areas with a uniform structure and orientation of atoms. The sizes of such grains vary from 20 to 960 nm. The formation of grains occurs through the processes of nucleation and growth. A small number of nuclei or germs of grains is formed randomly in the process of nucleation [67,68]. These nuclei can be formed

as a result of thermal fluctuations or other external influences. After the formation of nuclei, grains begin to grow through the diffusion of material inside the nucleus from the core to the margins. This process leads to an increase in grain size and the development of its crystal structure. Grain growth can occur by various mechanisms, such as diffusion growth, grain boundary migration, etc. Factors that affect the formation of grains in a thin film include the chemical composition of the material, temperature, growth time, deposition, or growth rate, atmospheric conditions, and other production parameters [69].

Elemental or chemical analysis was performed with EDX using the SEM approach. The EDX spectrum for samples #28, #29, #32, #35, #37, and #38 of different thicknesses is shown in Figures 2–7. The EDX analysis was carried out in a low vacuum mode. In particular, the percentages of atomic weight are Cu (72.84%) and O (27.16%) for sample 28 (Figure 1c).

For all other samples (Figures 2–6), a high percentage contribution of silicon atoms, Si, as the main element of the substrate was observed in the spectrum. Al and Fe atoms are present in the spectrum as the material of the installation table on which the sample was placed, but their contribution is not significant and should be neglected.

The content of 1.75% Ti is sufficient for the film structure to resemble a lamellar structure (Figure 3), and for the surface to be rougher. When such films are split in the transverse direction, the individual plates are separated parallel to the fault plane.

With an increase in the Cu content to 43.25% and the absence of Ti (Figure 4c), the size of the plates decreases significantly; they are randomly arranged (Figure 4a,b). Only a significant surface roughness remains.

At a Cu content of 30.25%, and with the absence of Ti (Figure 5c), the lamellar structure is preserved with the orientation of the planes in the direction perpendicular to the sample surface (Figure 5a,b). The surface roughness of such a film is close to the morphology of the sample #29.

In Figures 6 and 7, thin film island growth is a process in which a thin film is grown on a substrate or layer while maintaining structural correspondence with the substrate. During this process, atoms or molecules of the film material diffuse or deposit on the surface of the substrate, forming a new film layer. Diffusion processes regulate the homogeneity of the film in the lateral direction. Transport from the vapor phase (layer growth) controls the homogeneity of the film in the normal direction [66]. The formation of a uniform film structure can occur only when, in the process of growth, atoms can freely pass between the layers of the film. This means that it is energetically more profitable for an atom that can settle on the top of the island to move to the lower layer, where the potential energy is lower [70].

The presence of Ti (1.6%) leads (Figure 6c) to the formation of hollow areas and introduces randomness in the orientation of lamellar crystallites (Figure 6a,b). The surface of such a film is more irregular and has a more complex relief. The conglomerate structure is preserved, but it becomes more disordered.

An increase in the amount of Ti to 2.15% leads to the separation of Cu_xO crystallites (Figure 7c), enlargement of their size, and compaction of the material to form more massive crystallites (Figure 7a,b). In general, there is an increase in the number and size of cavities between the substrate and the film.

3.2. Solar Cell Performance

As the final verification of the proposed concept, complete photovoltaic structures were tested according to their opto-electrical parameters. I-V characteristics were measured under AM 1.5 spectrum in STC conditions by a Quantum Design solar simulator. Table 3 presents specific values achieved for various samples, which demonstrate a photovoltaic effect. For one of the most promising samples (#28 and #29), I-V and P-V characteristics are presented in Figures 8 and 9, respectively. In Table 3, the comparison of electrical parameters for the whole set of samples is demonstrated.



Figure 8. Voltage-current and power-current characteristics of the #28 solar cell.



Figure 9. Voltage-current and power-current characteristics of the #29 solar cell.

For better understanding of the sample's parameters, the shunt resistance (Rsh) and series resistance (R_s) of two selected cells with the best parameters were calculated and presented separately in Table 4.

Table 4. Series (R_s), shunt resistance (R_{sh}), and fill factor value of the two selected samples.

	#28	#29	
$R_s[\Omega]$	27 k	55 k	
R _{sh} [Ω]	6.2 k	270 M	
FF [%]	18.5	16.6	

4. Discussion

The Stranski–Krastanov (layer plus island) growth mode was observed in samples #28 and #29. The general feature of layer plus island growth mode is that after the completion of the formation of the two-dimensional layer, the growth of three-dimensional islands takes place. The thickness and orientation of islands are very different in lateral and normal directions. For the sample #28, after 30 min of the deposition process, the film thickness of CuO reached 1654 nm (Table 2), and for the TiO₂ layer, 49 nm was deposited during

25 min. The next sample, #29, was obtained with a shorter deposition time of 22 min and accordingly reached the thickness of 1201 nm, and for the TiO_2 layer, 41 nm was deposited during 25 min. Thus, this mode is most often observed in heteroepitaxial thin films, which leads to a change in physical properties and has important applications in semiconductor systems [71,72].

The basis responsible for the structure of thin film formation is the processes of nucleation and growth. It is generally accepted that in the simplest cases, when mutual diffusion does not occur, three growth models are possible. In some cases, the bond (with energy E_b) of condensing adatoms is much stronger among themselves than with the substrate (E_a) [73–78]. As a rule, when the energy $E_b < E_a$ adatoms begin to diffuse with low diffusion energy, which leads to fusion with other adatoms, at the same time, small stable clusters begin to form. The formed small cluster of adatoms, which are connected to each other by the energy E_i , can often consist of only one atom, which corresponds to the critical stage of nucleation. In this case, the formed pairs of adatoms have energy $E_i = 0$ and are stable on the thin film surface. In another case, when the energy E_a is greater than several E_b , layer-by-layer growth begins. During such growth, the reevaporation processes does not occur, which leads to a linear increase in the concentration of one atom (n_i). Subsequent nucleation processes are rather more complicated, due to the fact that several modes compete for these adatoms.

Observing the presented characteristics, one may clearly indicate the presence of the photovoltaic effect in the manufactured structures. The general flow of the I-V curves confirm the p-n junction operation, but the I_{sc} and V_{oc} values are limited far below the available theoretical values, and additionally, the fill factor of these devices is even below 25%, which greatly reduces maximum power and efficiency. A falling of the FF value below 25% points out the improper tilt of the characteristics, and is caused by reverse-polarized junction. This fact is especially visible even in the case of the sample #29 with a very low FF value of 16.5%. This phenomenon occurs often in the junction regions, where instead of ohmic contacts, a reverse-polarized Schottky barrier appears. For the record samples (#28 and #29), achievement of FF at the level of 60% would increase the efficiency to the level of 0.032% for #28 and 0.036% for #29, respectfully. This would also result in the power at the level of 7 μ W. This fact prompts the contact structure optimization.

For better understanding of the technological process parameters' influence on the sample performance as well as cell construction importance, the Voc/thickness ratios were calculated for all samples (Table 3). These values were referred to both CuO and TiO₂ layers. Excluding sample #38 with the extremely low absorber thickness, one may observe that the peak open circuit voltage was present when the low thickness of the window layer (41 nm in the case of sample #29) meets a sufficient, but not too thick, absorber layer (1200 nm in case of sample #29). The second in the efficiency gradation #28 sample presents the thickest absorber and a relatively thick window layer, but the proper crystalline structure is also an important factor in this case.

Considering the influence of the resistance on the device parameters, the two best performance solar cells (samples #28 and #29) were compared. In the cases of both these devices, one may observe relatively high series resistance; which, in the case of the sample #29, is two times higher. This resulted in lower energy production of the sample #29; but in both cases, such high R_s values dramatically reduced the FF value. The shunt resistance of the sample #29 is much superior, being several orders of magnitude higher. Summarizing, for the final performance, the value of low series resistance appeared more important for the FF value and the final power gain than limited shunt losses; however, for future development, all these values shall be improved.

5. Conclusions

 TiO_2/Cu_xO thin film solar cells were deposited using the direct current (DC) reactive magnetron sputtering method. From cross-section analysis, it is obvious that the films have a similar structure and growth of individual conglomerates, with grains from 20 to

960 nm. The best crystalline structure of the samples and optimal material composition results in the highest I-V parameters range. Such a growth mechanism corresponds to the Stranski–Krastanov (layer plus island) growth mode. The general feature of this growth mode is that virtually any impediment to continuous layer growth results in the formation of islands on top of (or in some cases within) an intermediate layer, whose thickness and orientation is very different in the different systems. A noticeable photovoltaic effect was achieved in all samples, with the top values of power at the level of a few microwatts. The future optimization towards higher I-V values by an increase in the fill factor value and elimination of a counter-junction effect is planned in the near-future research.

Supplementary Materials: The Supplementary information can be found at: Link I: https://www. youtube.com/watch?v=-0Sn4UbiKaE (accessed on 28 June 2023); Link II: https://www.youtube.com/ watch?v=Lavsm1CIqhY (accessed on 28 June 2023); Link III: https://www.youtube.com/watch?v= iei5bn2UAzg\046t=35s (accessed on 28 June 2023); Link IV: https://www.youtube.com/watch?v= 0TjWJwxLZYk\046t=4s (accessed on 28 June 2023).

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