

Supplementary Information

Article

Cobalt Impregnation on Titania Photocatalysts Enhances Vis Phenol Photodegradation

**Soukayna Belekbir¹, Mohammed El Azzouzi¹, Laura Rodríguez-Lorenzo², Adnane El Hamidi¹,
Juan Arturo Santaballa^{3,*} and Moisés Canle^{3,*}**

¹ Laboratory of Nanomaterials, Nanotechnologies and Environment, Center of Materials Science, Faculty of Sciences, Mohammed V University in Rabat, Rabat BP 1014, Morocco

² INL—International Iberian Nanotechnology Laboratory, Water Quality Group, Av. Mestre José Veiga, 4715-330 Braga, Portugal

³ React! Group, Department of Chemistry, Faculty of Sciences & CICA, University of A Coruña, E-15071 A Coruña, Spain

* Correspondence: arturo.santaballa@udc.es (J.A.S.); moises.canle@udc.es (M.C.)

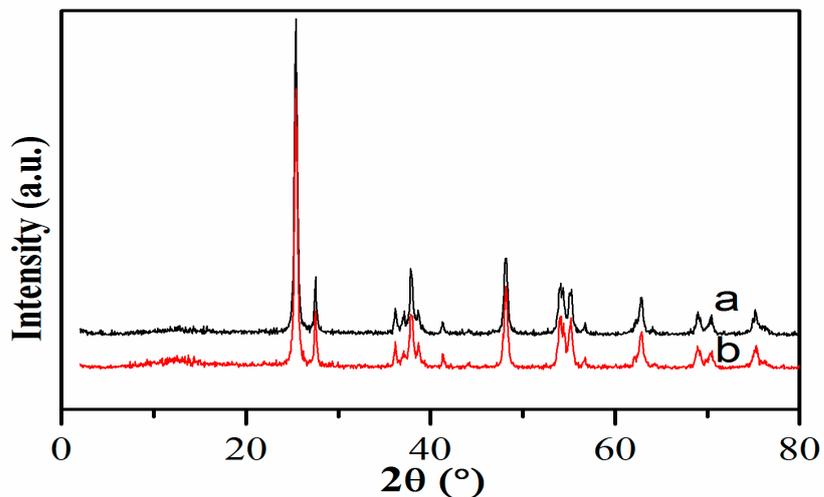


Figure S1. Diffractograms of Co(0.1%)/TiO₂, a: before leaching test, b: after leaching test.

Table S1: Raman peak centred at 143 cm⁻¹ as a function of laser line and laser's power.

<i>Laser power</i>	<i>Laser line Photocatalyst</i>	633 nm		785 nm	
		P25	Co(0.1%)/TiO ₂	P25	Co(0.1%)/TiO ₂
21-28 mW	<i>Peak position (cm⁻¹)</i>	141	141	144	144
	<i>Height (CCD cts)</i>	10160	12954	9620	18381
	<i>FWHM (cm⁻¹)</i>	4.34	19.7	14.1	14.4
10 mW	<i>Peak position (cm⁻¹)</i>	141	141	n.a.	n.a.
	<i>Height (CCD cts)</i>	1304	2250	n.a.	n.a.
	<i>FWHM (cm⁻¹)</i>	15.07	15.63	n.a.	n.a.
35 mW	<i>Peak position (cm⁻¹)</i>	n.a.	n.a.	144	144
	<i>Height (CCD cts)</i>	n.a.	n.a.	15831	45907
	<i>FWHM (cm⁻¹)</i>	n.a.	n.a.	13.9	14.0

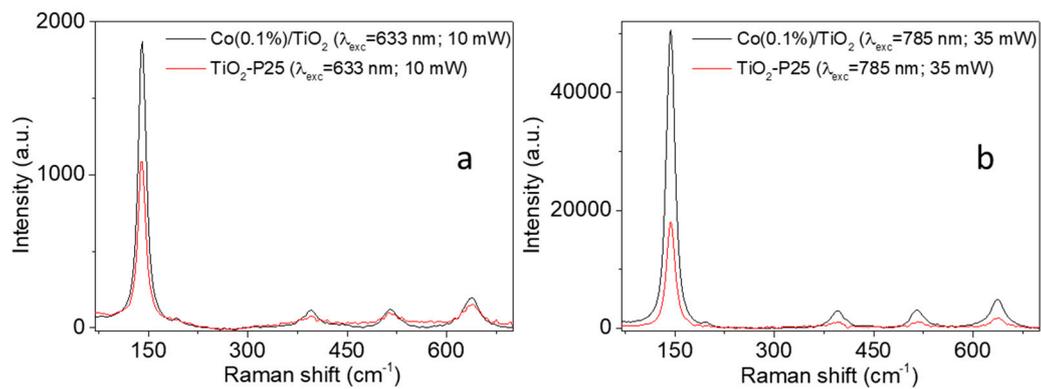


Figure S2: Raman spectra of TiO₂-P25 and Co(0.1%)/TiO₂ acquired using (a) 633 nm and (b) 785 nm excitation laser line.

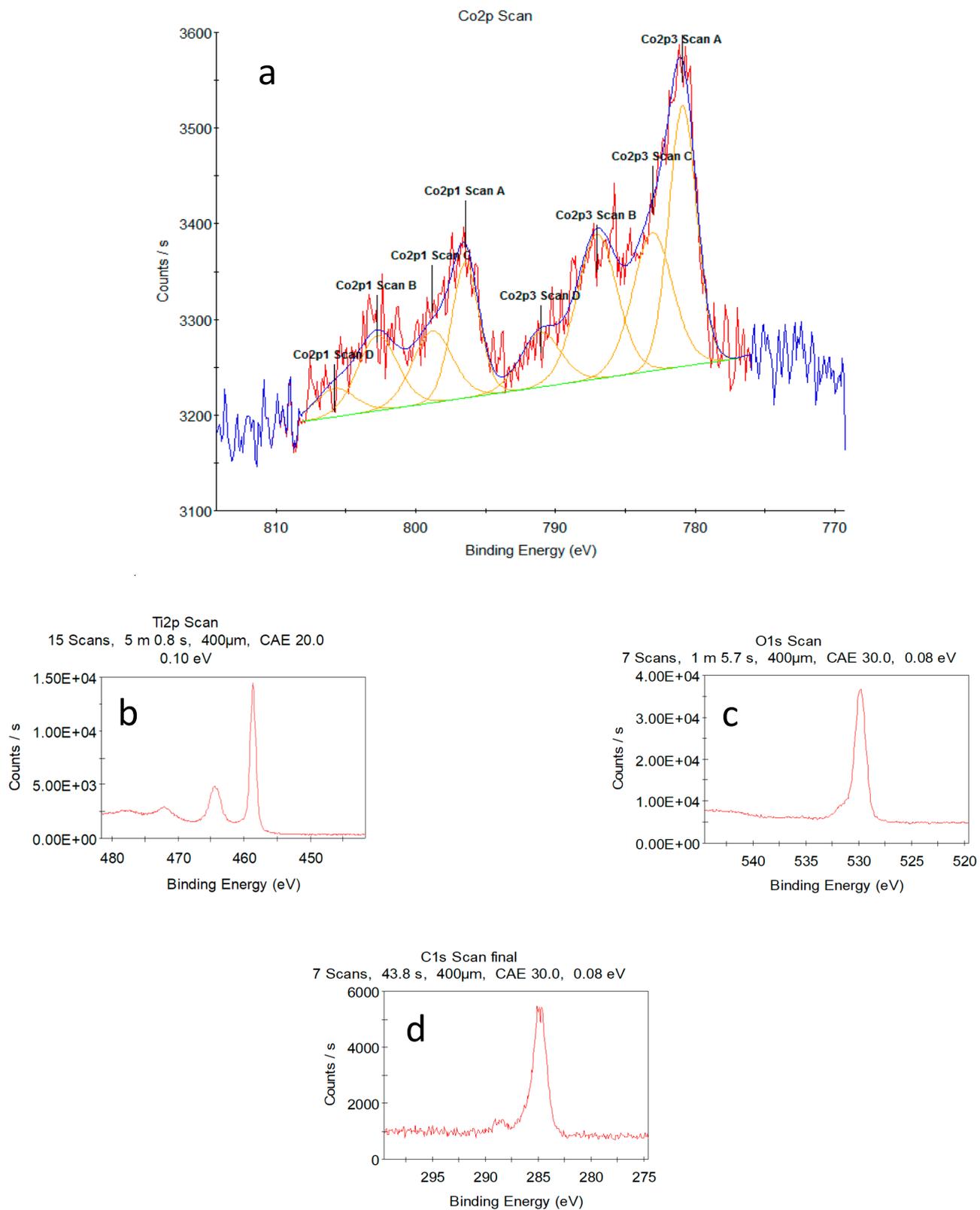


Figure S3. XPS spectra (a) Co 2p scan and fitting; (b) Ti 2p scan; (c) O 1s scan; (d) C 1s scan.

Table S2: Band gap for TiO₂-P25 and Co doped TiO₂ photocatalysts synthesized by different methods.

Photocatalyst	Co (%wt)	Synthetic method	Band gap (eV)	Reference (DOI)
TiO ₂ -P25	--	--	3.30	This work
Co/TiO ₂	0.1	Impregnation	2.4	This work
Co/TiO ₂	0.1	Sol-gel	2.66	10.1007/s10971-015-3715-3
Co/TiO ₂	0.1	Sol-gel	2.77	10.3390/ma11101946
Co/TiO ₂	0.1	Coprecipitation	2.78	10.1016/j.matchemphys.2012.04.062
Co/TiO ₂	0.3	Impregnation	2.3	This work
Co/TiO ₂	0.3	Sol-gel	3.00	10.5004/dwt.2017.20205
Co/TiO ₂	0.5	Impregnation	2.3	This work
Co/TiO ₂	0.5	Sol-gel	2.48	10.1016/j.sajce.2017.10.001
Co/TiO ₂	0.5	Hydrothermal	3.19	10.1016/j.apcatb.2016.03.003 1-s2.0-S0926337316301692-main
Co/TiO ₂	0.5	Sol-gel & Precipitation	3.24	10.1016/j.chemosphere.2020.128931
Co/TiO ₂	0.67	Sol-gel	3.02	10.1016/j.jwpe.2017.02.015 1-s2.0-S2214714416305256-main
Co/TiO ₂	1.0	Impregnation	2.3	This work
Co/TiO ₂	1.0	Sol-gel	1.97	10.1007/s10971-015-3715-3
Co/TiO ₂	1.0	Hydrothermal	3.13	10.1016/j.apcatb.2016.03.003 1-s2.0-S0926337316301692-main
Co/TiO ₂	1.0	Sol-gel & Precipitation	3.05	10.1016/j.chemosphere.2020.128931

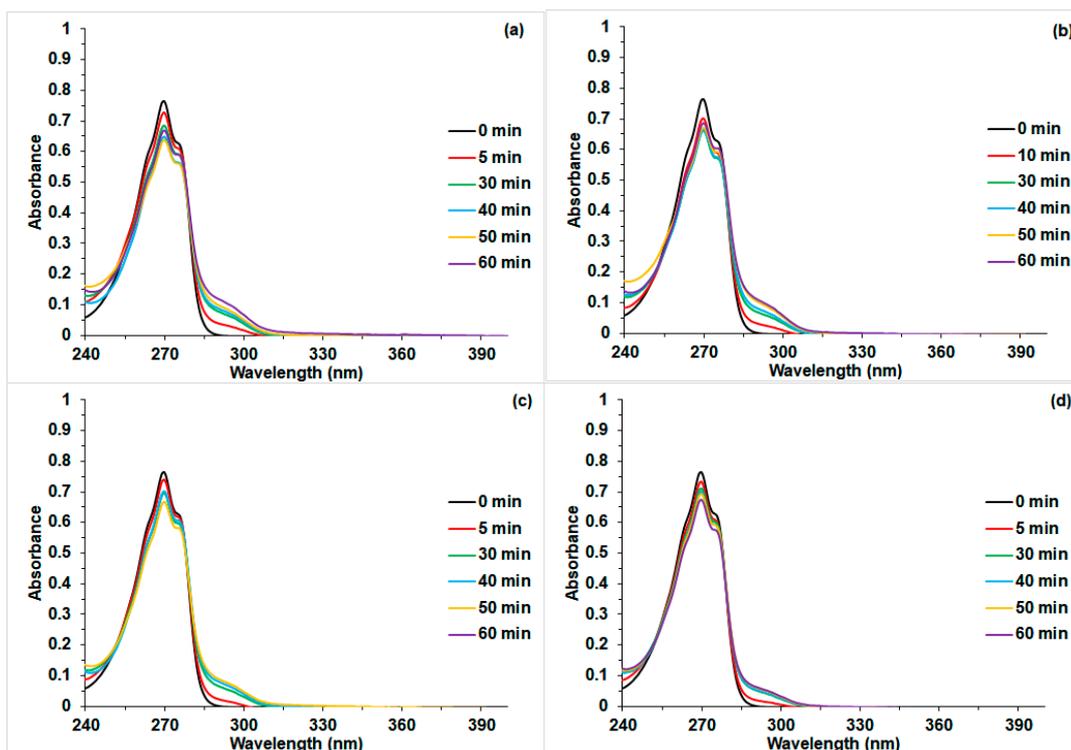


Figure S4: Time-resolved UV-Vis spectra of phenol during its UV irradiation in the presence of (a) Co(0.1%)/TiO₂, (b) Co(0.3%)/TiO₂, (c) Co(0.5%)/TiO₂, (d) Co(1%)/TiO₂. [Phenol] = 50 ppm; Photocatalysts = 1 g·L⁻¹, T *ca.* 25⁰C, natural pH.

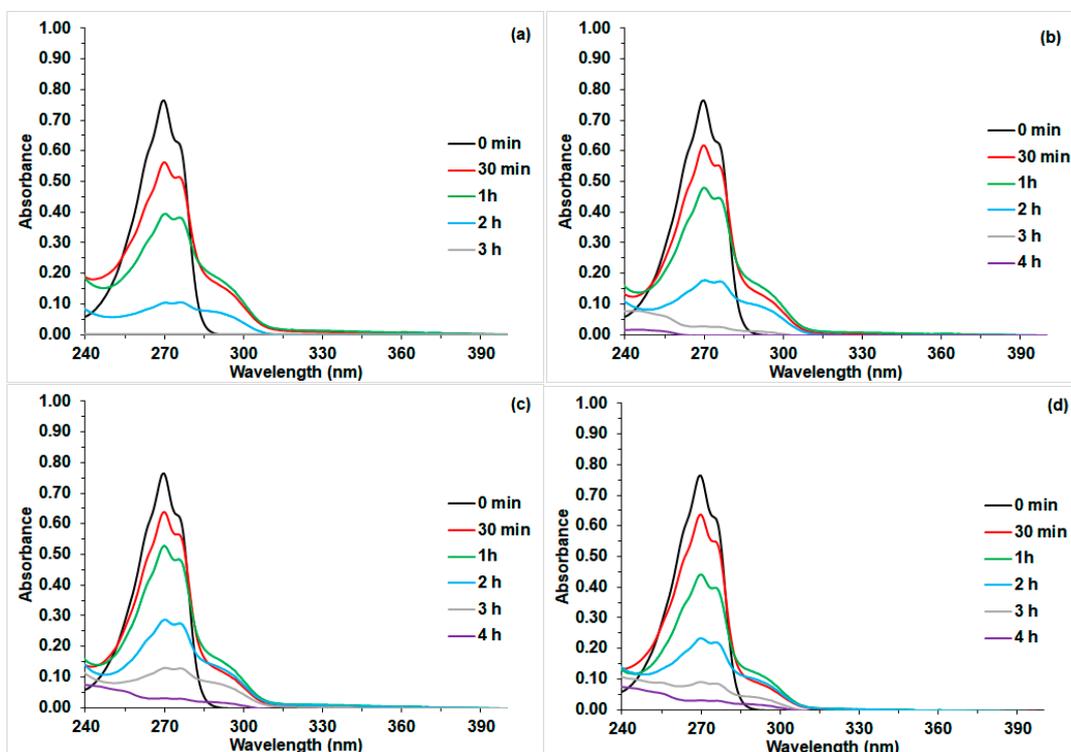


Figure S5: Time-resolved UV-Vis spectra of phenol during Vis irradiation in the presence of (a) Co(0.1%)/TiO₂, (b) Co(0.3%)/TiO₂, (c) Co(0.5%)/TiO₂, (d) Co(1%)/TiO₂. [Phenol] = 50 ppm; Photocatalyst = 1 g·L⁻¹, T *ca.* 25⁰C, natural pH.