

# ***Supplementary Information***

## **Highly Efficient and Exceptionally Durable Photooxidation Properties on Co<sub>3</sub>O<sub>4</sub>/g-C<sub>3</sub>N<sub>4</sub> Surfaces**

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### **Characterization of the photocatalysts**

X-ray diffraction (XRD, D8 X, Bruker) patterns were recorded with Cu K $\alpha$  radiation ( $\lambda$  = 1.54178 Å). X-ray photoelectron spectroscopy (ESCALAB MKII X, Thermo Fisher) measurements were performed. The field emission scanning electron microscopy (SEM) images were performed by using a JEOL-JSM-7800F. Transmission electron microscopy (TEM) and high-resolution TEM images were obtained by using a JEOL-2010. UV-vis diffuse reflectance spectroscopy (DRS) was recorded on a UV-3600 plus (Shimadzu, Japan). Fourier transform infrared (FT-IR) spectra were acquired on a Nicolet iS-50 spectrometer with KBr tablets, scanning from 4000 to 400 cm<sup>-1</sup> at room temperature (Thermo Fisher Scientific, America). Room temperature photoluminescence (PL) were recorded on a QM4m fluorescence spectrophotometer (QM4m, PTI). Electron spin resonance (ESR) patterns were recorded on a Bruker model ESR JES-FA200 spectrometer in methanol and water, respectively.

### **Photoelectrochemical test**

An electrochemical workstation (Chenhua Instruments Company, China, 660B) was used to perform electrochemical measurements using a three-electrode system. In the three-electrode system, the reference electrode was a silver and silver chloride, and the platinum wire was used as the counter electrode. The working electrode was prepared as follows: 5 mg of photocatalyst powder was dispersed in 0.5 mL ethylene glycol, then 20  $\mu$ L of the dispersed solution was coated in a 10  $\times$  5 mm indium tin oxide (ITO) glass and dried under 60°C to remove ethanol for 12 hours. The electrolyte solution is 0.2 M Na<sub>2</sub>SO<sub>4</sub>. The light source is a 500 W Xenon lamp (PLS-FX300HU, Beijing Perfectlight). Electrochemical impedance spectra (EIS) measurements were performed in 0.1 mol/L KCl solution containing 5 mM Fe(CN)<sub>6</sub><sup>3-</sup>/Fe(CN)<sub>6</sub><sup>4-</sup>.

### **Photocatalytic degradation activity test**

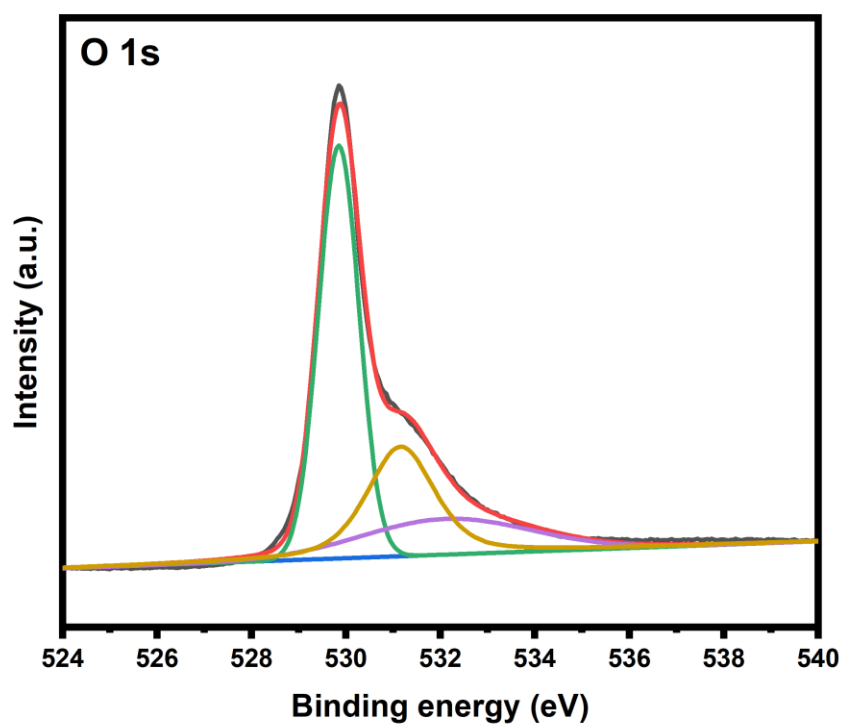
**Photocatalytic degradation of RhB:** 100 mL of 10<sup>-5</sup> M RhB solution containing 10 mg of catalyst was poured into a 100 mL reactor and kept in magnetic stirring throughout the experiment. The solution was first allowed to react darkly for 30 min to reach adsorption-desorption equilibrium, and then irradiated with a 300 W xenon lamp (PLS-SXE300, Porphyry) for degradation experiments. Every 10 min, 3 mL of supernatant was removed from the solid catalyst by high-speed centrifugation (8000 r/min), and the absorbance of the supernatant at 554 nm was measured by UV spectrophotometer.

**Photocatalytic cycle:** The photocatalytic cycle is similar to the photocatalytic degradation of RhB. After each cycle, the photocatalyst is washed three times with

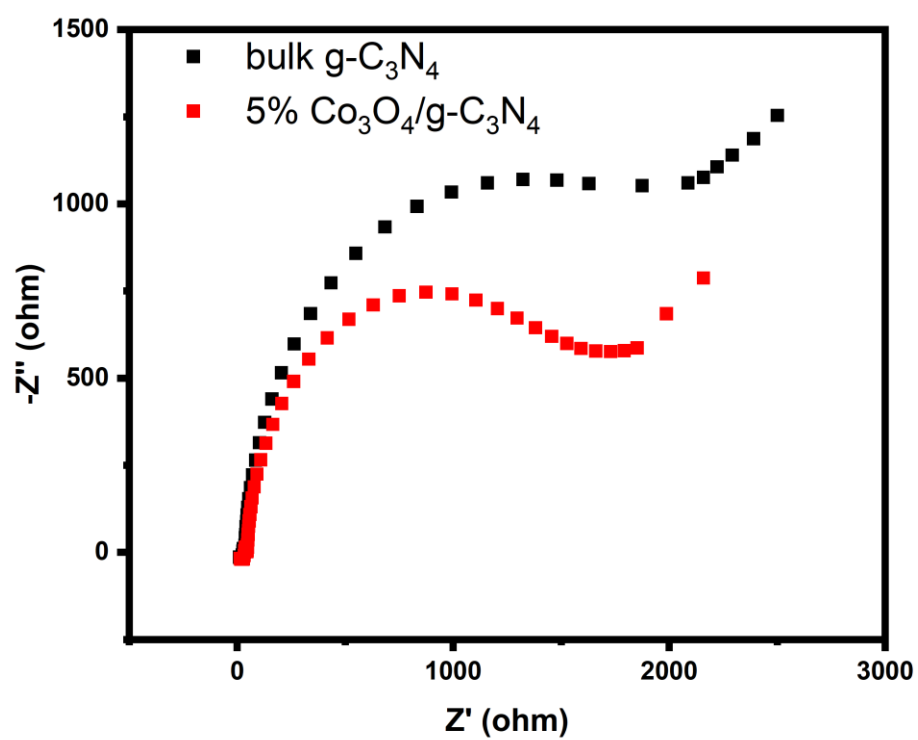
deionized water and anhydrous ethanol and dried under vacuum, and then a new cycle of reaction testing is started.

To investigate the active species produced in the photodegradation process of photocatalytic, hydroxyl radicals ( $\bullet\text{OH}$ ), holes ( $\text{h}^+$ ) and superoxide radical ( $\bullet\text{O}^{2-}$ ) were probed by adding 5 mM Tertiary butanol (TBA), triethanolamine (TEOA) and benzoquinone (BQ), respectively.

## Figure and Table



**Figure S1.** The high-resolution O 1s spectra of  $\text{Co}_3\text{O}_4$ .



**Figure S2.** EIS spectra under light conditions.

**Table S1.** Common photocatalyst and the effect of degrading organic pollutants in water.

Photocatalyst	Reaction medium	Degradation rate (min <sup>-1</sup> )	Ref.
Co <sub>3</sub> O <sub>4</sub> /g-C <sub>3</sub> N <sub>4</sub>	300 W Xe-lamp, 10 mg catalyst, 100 mL 10 <sup>-5</sup> M RhB	0.0703	This work
P-Doped g-C <sub>3</sub> N <sub>4</sub>	300 W Xe-lamp, 50 mg catalyst, 100 mL 10mg/L RhB	0.0659	S1
g-C <sub>3</sub> N <sub>4</sub>	300 W Xe-lamp, 20 mg catalyst, 100 mL 10mg/L RhB	0.051	S2
CQDs/Sb <sub>2</sub> WO <sub>6</sub>	300 W Xe-lamp, 100 mg catalyst, 100 mL 10mg/L RhB	0.01346	S3
MIL-125(Ti)/BiOI	250 W Xe-lamp, 15 mg catalyst, 100 mL 20mg/L RhB	0.034	S4
MoS <sub>2</sub>	300 W Xe-lamp, 50 mg catalyst, 50 mL 10mg/L RhB	0.212	S5
BiFeO <sub>3</sub> @B-rGO	300 W Xe-lamp, 20 mg catalyst, 40 mL 10mg/L RhB	0.0241	S6
CoFe <sub>2</sub> O <sub>4</sub> @Bi <sub>2</sub> WO <sub>6</sub> @BiOBr	400 W metal halide lamp, 65 mL 10mg/L RhB	0.01208	S7
BiOCl	86 W 420 nm LED, 50 mg catalyst, 100 mL 10mg/L RhB	0.1867	S8
SbVO <sub>4</sub> /g-C <sub>3</sub> N <sub>4</sub>	300 W Xe-lamp, 50 mg catalyst, 100 mL 10mg/L RhB	0.0587	S9
O-g-C <sub>3</sub> N <sub>4</sub> /SnO <sub>2</sub>	300 W Xe-lamp, 100 mg catalyst, 100 mL 10mg/L RhB	0.0458	S10

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