

Supporting Information

Pd-Catalyzed Cyclocarbonylation of Allylic Alcohol under Benign Conditions with Ionic Liquid as Stabilizer

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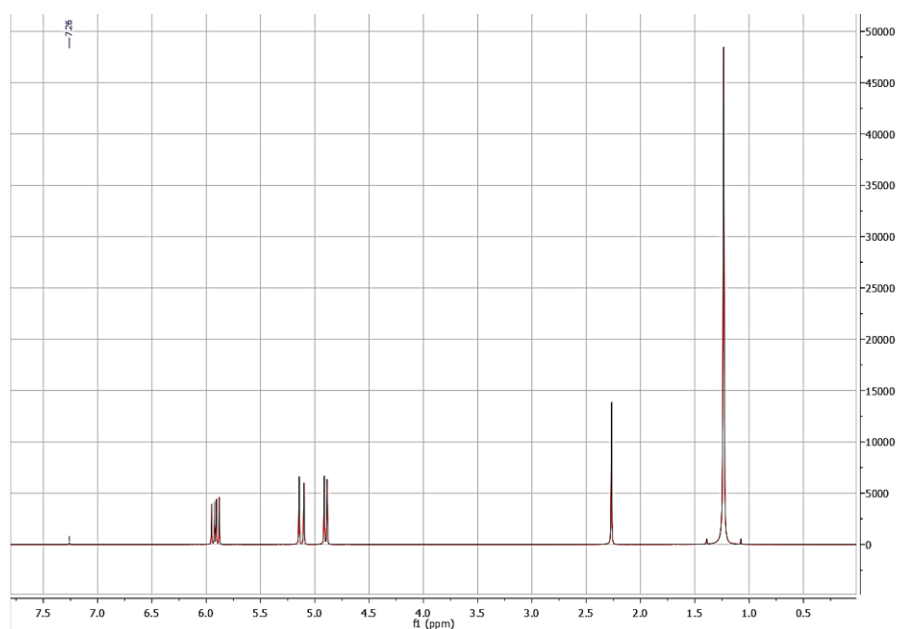
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(a)



(b)

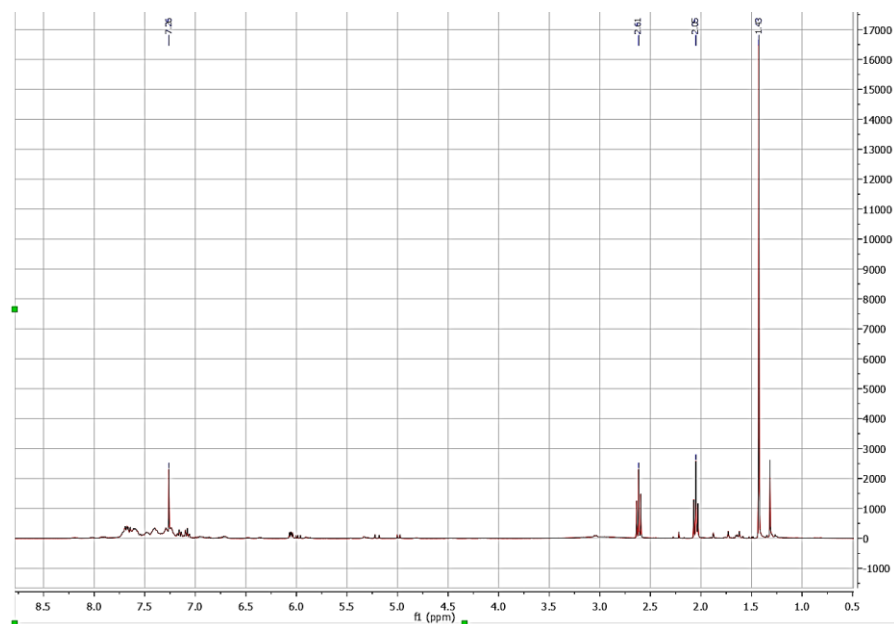


Figure S1. ^1H NMR spectra (400 MHz, CDCl_3) of (a) 2-methyl-3-buten-2-ol (δ/ppm : 1.2 (s, $2\times\text{CH}_3$, 6H), 2.3 (s, OH, 1H), 4.8 (d, HCH, 1H), 5.2 (d, HCH, 1H), 5.9 (dd, CH, 1H) and (b) reaction mixture after cyclocarbonylation with the product 4,4-dimethyl- γ -butyrolactone (δ/ppm : 1.43 (s, CH_3 , 6H), 2.05 (t, CH_2 , 2H), 2.61 (t, COCH_2 , 2H) and by-product(s) (δ/ppm : 1.63 (d), 3.06 (m), 5.28 (q)). Reaction conditions: 0.1 M 2-methyl-3-buten-2-ol, 4 mol% $\text{Pd}(\text{OAc})_2$, 16 mol% DPEPhos, 5 ml DCM, 95 $^\circ\text{C}$, 28 bar (CO/H_2 : 23/5), 18 h.

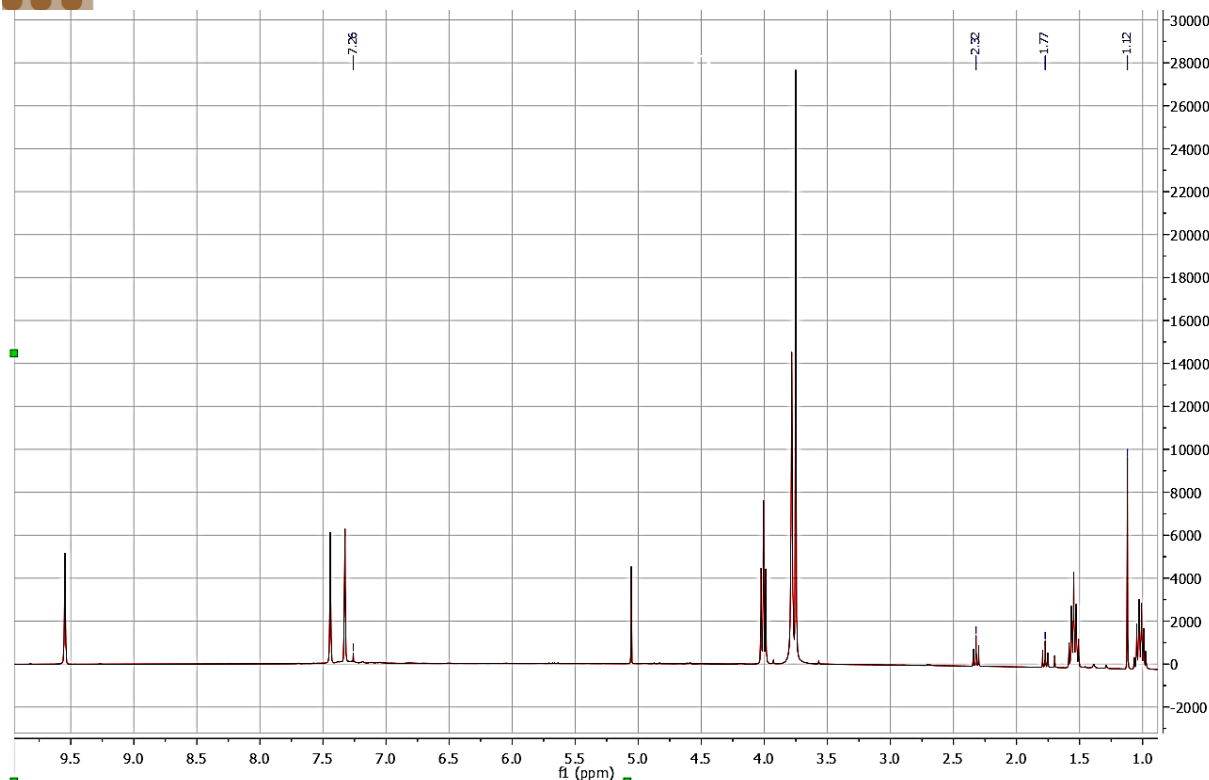


Figure S2. ^1H NMR spectrum (400 MHz, CDCl_3) of the reaction mixture after cyclocarbonylation of 2-methyl-3-buten-2-ol with presence of the IL [BMIM]Cl (δ/ppm : 1.05 (m, $\text{N-CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$, 5H), 1.55 (q, $\text{N-CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$, 2H), 4.0 (t, $\text{N-CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$, 2H), 5.1 (s, N-CH_3 , 3H), 7.3 (s, $\text{N-CH}_2\text{CH}_2\text{-N}$, 2H), 7.45 (s, $\text{N-CH}_2\text{CH}_2\text{-N}$, 2H), 9.6 (s, N-CH-N , 1H) and the product 4,4-dimethyl- γ -butyro-lactone (δ/ppm : 1.15 (s, CH_3 , 6H), 1.75 (t, CH_2 , 2H), 2.3 (t, COCH_2 , 2H). Reaction conditions: 0.1 M 2-methyl-3-buten-2-ol, 1.0 g [BMIM]Cl, 4 mol% $\text{Pd}(\text{OAc})_2$, 16 mol% DPEPhos, 5 ml DCM, 95 $^\circ\text{C}$, 28 bar (CO/H_2 : 23/5), 18 h.

(a)



(b)



(c)



(d)



Figure S3. Reaction mixture after (a) first reaction run (no Pd-black), (b) second reaction run (no Pd-black), (c) third reaction run (no Pd-black) and (d) fourth reaction run (Pd-black) during recycling of the IL Pd-DPEPhos catalyst system in the cyclocarbonylation of 2-methyl-3-buten-2-ol. Reaction conditions: 0.1 M 2-methyl-3-buten-2-ol, 4 mol% Pd(OAc)₂, 16 mol% DPEPhos, 1.0 g [BMIM]Cl, 5 ml DCM, 100 °C, 28 bar (CO/H₂/N₂: 20/5/3), 18 h.



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