

Supplementary Material

Development of Novel Magnetoliposomes Containing Nickel Ferrite Nanoparticles Covered with Gold for Applications in Thermotherapy

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Evidence of stability of NPs dispersions

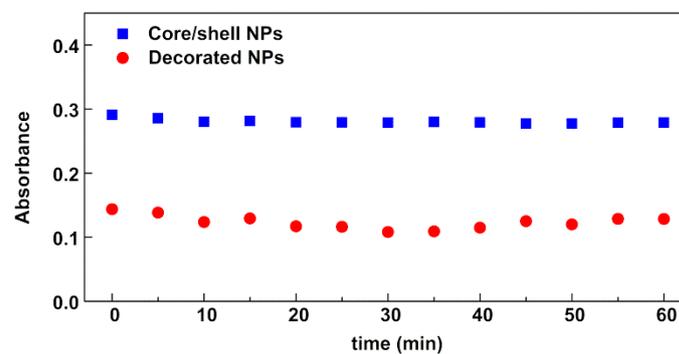


Figure S1. UV-Visible maximum absorbance of nanoparticles dispersions in PBS buffer (pH = 7.0) as function of time (5 minutes intervals).

Fluorescence intensity dependence on temperature in irradiation assays

Assuming an Arrhenius behaviour for the rate constant, k_{nr} , of non-radiative processes, it is possible to determine the temperature in each irradiation assay,

$$k_{nr} = k_0 e^{-\frac{E_a}{RT}} \quad (S1)$$

where k_0 is the preexponential factor; E_a is the activation energy, R the gas constant and T the absolute temperature.

The fluorescence quantum yield, Φ_F , is given by

$$\Phi_F = \frac{k_F}{k_F + k_{nr}} \Rightarrow \frac{1}{\Phi_F} - 1 = \frac{k_{nr}}{k_F} \quad (S2)$$

where k_F is the fluorescence rate constant.

Taking as reference the fluorescence quantum yield at room temperature, Φ_F^0 , it is obtained

$$\frac{\Phi_F^0}{\Phi_F} - \Phi_F^0 = \Phi_F^0 \frac{k_{nr}}{k_F} \quad (S3)$$

or, similarly,

$$\frac{I_F^0}{I_F} - \Phi_F^0 = \Phi_F^0 \frac{k_{nr}}{k_F} \quad (S4)$$

where I_F represents the fluorescence intensity at a given wavelength.

Therefore, it can be obtained

$$\frac{I_F^0}{I_F} - \Phi_F^0 = \Phi_F^0 \frac{k_0}{k_F} e^{-\frac{E_a}{RT}} \quad (S5)$$

Considering, as an approximation, that the fluorescence quantum yield of Rhodamine B near the nanoparticles is much lower than the quenching ratio, $\frac{I_F^0}{I_F}$, that is always above unity, it is obtained:

$$\frac{I_F^0}{I_F} \approx \Phi_F^0 \frac{k_0}{k_F} e^{-\frac{E_a}{RT}} \quad (S6)$$

Or

$$\frac{I_F^0}{I_F} \propto A e^{-\frac{E_a}{RT}} \quad (S7)$$



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