



Article New Monoterpenoids and Polyketides from the Deep-Sea Sediment-Derived Fungus Aspergillus sydowii MCCC 3A00324

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Abstract: Chemical study of the secondary metabolites of a deep-sea-derived fungus *Aspergillus sydowii* MCCC 3A00324 led to the isolation of eleven compounds (1–11), including one novel (1) and one new (2) osmane-related monoterpenoids and two undescribed polyketides (3 and 4). The structures of the metabolites were determined by comprehensive analyses of the NMR and HRESIMS spectra, in association with quantum chemical calculations of the ¹³C NMR, ECD, and specific rotation data for the configurational assignment. Compound **1** possessed a novel monoterpenoid skeleton, biogenetically probably derived from the osmane-type monoperpenoid after the cyclopentane ring cleavage and oxidation reactions. Additionally, compound **3** was the first example of the *α*-pyrone derivatives bearing two phenyl units at C-3 and C-5, respectively. The anti-inflammatory activities of **1–11** were tested. As a result, compound **6** showed potent inhibitory nitric oxide production in lipopolysaccharide (LPS)-activated BV-2 microglia cells with an inhibition rate of 94.4% at the concentration of 10 μ M. In addition, a plausible biosynthetic pathway for **1** and **2** was also proposed.

Keywords: polyketides; monoterpenoids; deep-sea fungus; Aspergillus sydowii; ECD calculation

1. Introduction

Deep-sea-derived fungi inhabiting extreme sea environments have proven to be a new source for a wide array of structurally intriguing and biologically active secondary metabolites [1–3]. As of December 2019, about 700 new secondary metabolites were reported from deep-sea-derived fungi, such as terpenoids, polyketides, alkaloids, and steroids. Some of the metabolites featured a broad range of biological activities, for example, cytotoxicity [4,5] as well as antiviral [6,7], antibacterial [8,9], anti-inflammatory [10], and antiallergic effects [11]. In our continuing efforts to discover new or bioactive secondary metabolites from deep-sea-derived fungi [12–16], the fungus *Aspergillus sydowii* MCCC 3A00324, isolated from the South Atlantic Ocean deep-sea sediment (2246 m), attracted our attention due to the abundantly metabolic profile obtained by HPLC and TLC analysis (Figure S1, Supplementary Materials). A literature retrieval discovered that 52 out of 59 new compounds were isolated from the marine-derived *Aspergillus sydowii* fungus. The bisabolane-type sesquiterpenoids [17–24], xanthones [18,23,25], diphenyl ethers [20,26–28], and diketopiperazine alkaloids [29–31] were common secondary metabolites of the *A. sydowii*,

some of which exhibited a wide range of pharmacological activities. To date, only three undescribed polyketides, namely asperentin B [32], 2,3,5-trimethyl-6-(3-oxobutan-2-yl)-4H-pyran-4-one, and (2R)-2,3-dihydro-7-hydroxy-6,8-dimethyl-2-[(E)-propl-enyl] chromen-4-one [33], have been reported from the deep-sea-derived A. sydowii. On the basis of the seldomly carried out chemical investigation and the abundantly metabolic profile of the A. sydowii MCCC 3A00324, a chemical study of the MCCC 3A00324 fungus was carried out. In our previous research, 17 undescribed and 10 known sesquiterpenoids were obtained from the fermented cultures of the fungus [34]. In our continuous research on this fungus, eleven compounds (1–11) (Figure 1), including one novel (1) and one new (2) monoterpenoids as well as two new polyketides (3 and 4) were discovered. Compounds 1 and 2 were the first representatives of the osmane-related monoterpenoids discovered from the fungi, while **3** was the first example of the α -pyrone derivatives with two phenyl units at C-3 and C-5, respectively. As terpenoids usually possess anti-inflammatory activities [35–37], all compounds were subsequently evaluated for their anti-inflammatory effects. As a result, compound 6 exhibited strong inhibitory nitric oxide production in lipopolysaccharide (LPS)-activated BV-2 microglia cells with a 94.4% inhibition rate at 10 µM. Herein, the isolation, structure elucidation, and anti-inflammatory activities of these metabolites as well as the biosynthetic pathway of 1 and 2 are reported.



Figure 1. Chemical structures of 1-11 from Aspergillus sydowii MCCC 3A00324.

2. Results and Discussion

Aspermonoterpenoid A (1), colorless oil, has the molecular formula of $C_{10}H_{14}O_4$ as determined by the ¹³C NMR data and the HRESIMS spectrum (*m*/*z* 221.0784, [M + Na]⁺), indicating four degrees of unsaturation. The ¹H NMR spectrum showed two olefinic protons (δ_H 5.77, 6.66), a methine (δ_H 3.36), and three methyls (δ_H 1.24, 1.88, 2.14) (Table 1), while the ¹³C NMR and HSQC spectra exhibited four sp² carbons (δ_C 117.0, 130.2, 144.5, 162.0) for two double bonds, two ester carbonyl carbons (δ_C 170.4, 171.5), one methine (δ_C 43.8), and three methyls (δ_C 12.8, 17.0, 19.0) (Table 1). As all degrees of unsaturation were accounted for by two carbonyl carbons and two double bonds, an acyclic structure was required in 1. The COSY data of H-4 (δ_H 3.36)/H₃-9 (δ_H 1.24) and the HMBC cross-peaks from H₃-8 (δ_H 1.88) to C-5 (δ_C 144.5)/C-6 (δ_C 130.2)/C-7 (δ_C 171.5), H₃-9 to C-3 (δ_C 162.0)/C-4 (δ_C 43.8)/C-5, H₃-10 (δ_H 2.14) to C-2 (δ_C 117.0)/C-3/C-4, and from H-2 (δ_H 5.77) to C-1 (δ_C 170.4)/C-4/C-10 (δ_C 17.0) established the planar structure of 1 as 2,4,5-trimethylhepta-2,5-dienedioic acid (Figure 2). The *E* geometries at Δ^2 and Δ^5 were resolved on the basis of the NOESY cross-peaks from H-2 to H-4 and from H₃-8 to H-4 (Figure 2), which were further corroborated by the chemical shifts of Me-8 ($\delta_{\rm C}$ 12.8) and Me-10 ($\delta_{\rm C}$ 17.0). The sole chiral center of C-4 in **1** was resolved by the calculation of the specific rotation data using Gaussian 09 [38]. The specific rotation data of (4*S**)-**1** was calculated at the B3LYP/6-311+G(2d,p) level with the CPCM model in methanol, which has the same sign ([α]_D +138) as that of the experimental data ([α]_D²⁵ +54), revealing the 4*S* configuration. Therefore, the structure of **1** was determined to be (2*E*,5*E*,4*S*)-2,4,5-trimethylhepta-2,5-dienedioic acid, which was named aspermonoterpenoid A. Noteworthily, **1** possessed a novel chained monoperpenoid skeleton, biogenetically probably derived from the osmane-type monoperpenoid after the cyclopentane ring cleavage and oxidation reactions [39,40].

No.	1		2		
	δ _C	δ_{H}	$\delta_{\rm C}$	$\delta_{ m H}$	
1	170.4 <i>,</i> C		211.6, C		
2	117.0, CH	5.77, s	130.1, CH	5.89, brs	
3	162.0, C		184.4, C		
4	43.8, CH	3.36, m	44.8, CH	2.76, m	
5	144.5 <i>,</i> CH	6.66, d (9.0)	58.4, CH	2.21, dd (4.9, 2.5)	
6	130.2, C		41.2, CH	2.94, m	
7	171.5 <i>,</i> C		178.5, C		
8	12.8, CH ₃	1.88, d (0.6)	15.3, CH ₃	1.29, d (7.2)	
9	19.0, CH ₃	1.24, d (6.8)	18.9, CH ₃	1.28, d (7.1)	
10	17.0, CH ₃	2.14, s	17.1, CH ₃	2.13, s	

Table 1. ¹H (400 MHz) and ¹³C (100 MHz) NMR spectroscopic data of 1 and 2 in CD₃OD.



Figure 2. Key COSY, HMBC, and NOESY correlations of 1 and 2.

The molecular formula of aspermonoterpenoid B (2) was established to be $C_{10}H_{14}O_3$ on the basis of the HRESIMS spectrum at the sodium adduct ion peak at m/z 205.0845 (calcd for $C_{10}H_{14}O_3Na$, 205.0841), requiring four degrees of hydrogen deficiency. The ¹H NMR spectrum showed one olefinic proton (δ_H 5.89), three methines (δ_H 2.21, 2.76, 2.94), and three methyls (δ_H 1.28, 1.29, 2.13). Analyses of the ¹³C NMR spectrum, with the aid of the HSQC spectrum, revealed ten carbon signals attributable to two olefinic carbons (δ_C 130.1, 184.4) for a double bond, two carbonyl carbons for a keto (δ_C 211.6) and a carboxylic (δ_C 178.5) functionalities, three methines (δ_C 41.2, 44.8, 58.4), and three methyls (δ_C 15.3, 17.1, 18.9), which accounted for three degrees of unsaturation. The remaining one double bond equivalent indicated that a monocyclic skeleton was required in **2**. The COSY cross-peaks of H-4 (δ_H 2.76)/H-5 ($\delta_{\rm H}$ 2.21)/H-6 ($\delta_{\rm H}$ 2.94) and the HMBC correlations from H₃-8 ($\delta_{\rm H}$ 1.29) to C-5 ($\delta_{\rm C}$ 58.4)/C-6 $(\delta_{C} 41.2)/C-7$ ($\delta_{C} 178.5$), H₃-9 ($\delta_{H} 1.28$) to C-3 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 44.8$)/C-5, H₃-10 ($\delta_{H} 2.13$) to C-2 ($\delta_{C} 184.4$)/C-4 ($\delta_{C} 18$ 130.1)/C-3/C-4, and from H-2 ($\delta_{\rm H}$ 5.89) to C-1 ($\delta_{\rm C}$ 211.6)/C-4/C-5 deduced the structure of 2 to be 3,4-dimethyl-5-isopropanic acidated cyclopentenone (Figure 2). The relative configuration of C-4 and C-5 in cyclopentenone ring moiety was determined as $4S^*$ and $5R^*$ on the basis of the NOESY correlation from H₃-9 to H-5, in addition to the very weak interaction between H-4 and H-5 (Figure 2). Additionally, the relative configuration of C-6 was unreliable to be deduced by the NOESY data because of its location at the soft side chain. In order to assign the relative configuration, the two possible epimers $(4S^*, 5R^*, 6S^*)$ -2 (2a) and $(4S^*, 5R^*, 6R^*)$ -2 (2b) were subjected to the ¹³C NMR chemical shift theoretical calculation at the mPW1PW91/6-311+G(2d,p) level in methanol using Gaussian 09. As shown in Figure 3, the calculated ¹³C NMR data of **2a** were nearly identical to those of the experimental NMR data, as corroborated by the correlation coefficient (R^2) and root mean square error (RMSE) of **2a** $(R^2 = 0.9992, RMSE = 2.4436)$ and **2b** $(R^2 = 0.9988, RMSE = 2.7912)$, suggesting the 4S*, 5R*, and 6S* configurations. The absolute configuration of 2 was resolved on the basis of the ECD calculation, which was carried out at the B3LYP/6-311+G(2d,p) level in MeOH using the optimized geometries at the B3LYP/6-31+G(d,p) level in gas phase after conformational searches via the OPLS3 force field using Maestro 10.2. The ECD spectrum of 2a was calculated by the TDDFT method, which was in good agreement with the experimental ECD curve, indicating the 4S, 5R, and 6S configurations (Figure 4). Interestingly, 2 was the first representative of the osmane-type monoterpenoids discovered from the fungi.



Figure 3. Linear regression analysis of the experimental ¹³C NMR data of **2** and calculated ¹³C NMR chemical shift data of **2a**.

Compound **3** was isolated as a colorless oil, and its molecular formula was assigned to be $C_{18}H_{14}O_5$ on the basis of the HRESIMS spectrum at the sodium adduct ion peak at *m*/*z* 333.0741, implying 12 indices of hydrogen deficiency. The ¹H NMR spectrum exhibited four aromatic protons with dual intensity at δ_H 6.85 (d, *J* = 8.5 Hz, 2H), 7.01 (d, *J* = 8.6 Hz, 2H), 7.29 (d, *J* = 8.5 Hz, 2H), and 7.33 (d, *J* = 8.6 Hz, 2H), indicating the presence of two para-substituted benzene rings. Moreover, a methoxy (δ_H 3.85) and one olefinic (δ_H 7.59) protons were also observed (Table 2). The ¹³C NMR spectrum showed a total of 18 carbon resonance signals, including twelve aromatic carbons (δ_C 115.1 × 2, 116.3 × 2, 123.5, 124.6, 131.9 × 2, 133.2 × 2, 158.9, and 160.9) for two benzene units, four olefinic carbons (δ_C 107.1, 120.4, 149.5, and 167.1) for two double bonds, a carbonyl carbon (δ_C 166.2), and one methoxy group (δ_C 55.7) (Table 2). The COSY data of H-1" (δ_H 7.33) and H-2" (δ_H 7.01) and the HMBC cross-peaks from H-1"/H-5" to C-3" (δ_C 160.9) and C-3 (δ_C 107.1), H-2"/H-4" to C-3" and C-6" ($\delta_{\rm C}$ 124.6), and from the methoxy protons ($\delta_{\rm H}$ 3.85) to C-3" deduced the *p*-methoxyphenyl moiety (ring C) to be tethered at C-3 (Figure 5). Additional COSY correlation between H-1' ($\delta_{\rm H}$ 7.29)/H-2' ($\delta_{\rm H}$ 6.85) and the HMBC correlations from H-1'/H-5' to C-3' ($\delta_{\rm C}$ 158.9) and C-5 ($\delta_{\rm C}$ 120.4) and from H-2'/H-4' to C-3' and C-6' ($\delta_{\rm C}$ 123.5) in association with the chemical shift of C-3' established the *p*-hydroxyphenyl unit (ring A) to be resided at C-5. The remaining five olefinic carbons ($\delta_{\rm C}$ 107.1, 120.4, 149.5, 166.2, and 167.1) constructed an α -pyrone moiety with a hydroxy at C-4 position, according to the HMBC cross-peaks from H-6 ($\delta_{\rm H}$ 7.59) to C-2 ($\delta_{\rm C}$ 166.2), C-4 ($\delta_{\rm C}$ 167.2), C-5 ($\delta_{\rm C}$ 120.4), and C-6', as well as their chemical shifts and the molecular formula. Therefore, the structure of **3** was elucidated to be 4-hydroxy-5-(4-hydroxyphenyl)-3-(4-methoxyphenyl)-2*H*-pyran-2-one, which was named asperphenylpyrone. It is of note that **3** was the first example of the α -pyrone derivatives bearing two phenyl units at C-3 and C-5, respectively.



Figure 4. Experimental (Exp.) and calculated (Cal.) ECD spectra of **2** in methanol at the B3LYP/6-311+ G(2d,p) level.



Figure 5. Selected COSY and HMBC correlations of compounds 3 and 4.

No	3 ^a			4 ^b		
	δ _C	$\delta_{ m H}$	No	δ _C	$\delta_{ m H}$	
2	166.2 ^{<i>c</i>} , C		2	156.6, C		
3	107.1, CH		3	144.8, C		
4	167.1 ^c , C		4	113.4, CH	7.49, s	
5	120.4, C		4a	120.4, C		
6	149.5 <i>,</i> CH	7.59, s	5	128.9, CH	8.23, d (1.8)	
1′	131.9 <i>,</i> CH	7.29, d (8.5)	6	127.8, C		
2'	116.3 <i>,</i> CH	6.85, d (8.5)	7	129.5, CH	7.95, dd (8.6, 1.8)	
3'	158.9, C		8	116.5, CH	7.45, d (8.6)	
4'	116.3 <i>,</i> CH	6.85, d (8.5)	8a	152.2, C		
5'	131.9 <i>,</i> CH	7.29, d (8.5)	9	166.9, C		
6'	123.5, C		OMe	56.8, CH ₃	3.86, s	
1″	133.2, CH	7.33, d (8.6)				
2″	115.1 <i>,</i> CH	7.01, d (8.6)				
3″	160.9, C					
$4^{\prime\prime}$	115.1 <i>,</i> CH	7.01, d (8.6)				
5″	133.2, CH	7.33, d (8.6)				
6″	124.6, C					
OMe	55.7, CH ₃	3.85, s				

Table 2. ¹H (400 MHz) and ¹³C (100 MHz) NMR spectroscopic data of 3 and 4.

^a Measured in CD₃OD. ^b Measured in DMSO-d₆. ^c Assignments in column could be interchanged.

Compound 4 has the molecular formula of $C_{11}H_8O_5$ as determined by the positive HRESIMS spectrum (*m*/*z* 243.0268, [M + Na]⁺) and the ¹³C NMR data, indicating eight degrees of unsaturation. The ¹H NMR spectrum showed three aromatic protons (δ_H 7.45 (d, *J* = 8.6 Hz, H-8), 7.95 (dd, *J* = 8.6, 1.8 Hz, H-7), and 8.23 (d, *J* = 1.8 Hz, H-5)) for a 1,2,4-trisubstituted benzene ring, a singlet olefinic proton (δ_H 7.49), and one methoxy (δ_H 3.86), whereas the ¹³C NMR spectrum exhibited 11 carbon signals, including six aromatic carbons (δ_C 116.5, 120.4, 127.8, 128.9, 129.5, and 152.2) for a benzene ring, two olefinic carbons (δ_C 113.4, 144.8) for a double bond, two ester carbonyl carbons (156.6, 166.9), and a methoxy group (δ_C 56.8). The aforementioned NMR data were very similar to those of the known coumarin derivative [41], with the exception of the presence of an additional aromatic methine (δ_H 7.95, δ_C 129.5) in 4 instead of an oxygenated nonprotonated sp² carbon. The additional olefinic proton was appointed at C-7 (δ_C 129.5) on the basis of the COSY cross-peaks from H-7 (δ_H 7.95) to H-8 (δ_H 7.45) and H-5 (δ_H 8.23) as well as the HMBC correlations from H-5 to C-4 (δ_C 113.4)/C-7/C-9 (δ_C 166.9)/C-8a (δ_C 152.2) and from H-7 to C-5 (δ_C 128.9), C-9, and C-8a (Figure 5). Thus, the structure of 4 was established as a 7-deoxylated derivative of the above known compound, which was given the trivial name of aspercoumarine acid.

In addition, seven known compounds were established to be pestalotiolactone A (5) [42,43], 3,7-dihydroxy-1,9-dimethyldibenzofuran (6) [44], diorcinol (7) [45], cordyol C (8) [46], 4-(3',4'-dihydroxyphenyl)-2-butanone (9) [47], 3,4,5-trimethoxybenzoic acid (10) [48], and 4-hydroxybenzoic acid (11) [49] on the basis of comparison of their NMR and specific rotation data with those reported in the literature.

Compound **1** was a novel chained monoterpenoid. The plausible biosynthetic pathway for **1** and **2** was proposed (Scheme 1). Starting from the GPP, hydrosis, oxygenation, and cyclization reaction occurrence constructed the monocyclic ring monoterpenoid osmane. The osmane might undergo carbon–carbon bond cleavage to form a key intermediate **A**, which was further oxygenated to yield **1**. Additionally, the osmane might also undergo oxygenation to form **2**.



Scheme 1. Hypothetical biogenetic pathways for compounds 1 and 2.

As terpenoids usually possess the anti-inflammatory activities, all isolated metabolites were evaluated for their inhibitory effects against NO secretion in LPS-activated BV-2 microglia cells. As a result, none of them showed obvious cytotoxic activities against BV-2 microglia cells at the concentration of 20 μ M by the CCK-8 Kit, and all compounds exhibited dose-dependent inhibitory effects against NO production induced by the LPS at the concentrations of 20 and 10 μ M, respectively (Table 3). Interestingly, compound **6** (10 μ M) showed potent anti-inflammatory activities with the inhibition rate of 94.4%.

Common la	Anti-N	IO (%)	Cell Viability Inhibition (%)	
Compounds	20 µM	10 µM	20 µM	10 µM
1	33.5 ± 1.5	10.2 ± 2.0	0.7 ± 0.1	0.3 ± 1.2
2	34.0 ± 1.4	22.7 ± 1.4	4.6 ± 2.6	-1.1 ± 0.4
3	22.7 ± 1.5	13.2 ± 1.3	4.1 ± 7.6	3.3 ± 3.1
4	28.3 ± 0.7	18.0 ± 2.0	3.5 ± 3.7	2.01 ± 1.4
5	39.1 ± 1.6	25.1 ± 0.8	10.0 ± 0.2	3.2 ± 1.7
6	101.4 ± 2.4	94.4 ± 0.0	-1.6 ± 5.1	-0.8 ± 3.6
7	55.0 ± 1.4	35.4 ± 2.4	4.1 ± 3.8	-1.5 ± 3.5
8	30.7 ± 0.8	18.0 ± 0.8	1.8 ± 4.7	-0.1 ± 8.2
9	39.3 ± 0.7	30.4 ± 1.9	2.3 ± 0.1	-1.1 ± 3.7
10	44.8 ± 0.7	33.0 ± 0.7	0.4 ± 1.3	0.2 ± 1.8
11	42.7 ± 1.3	30.8 ± 2.6	1.9 ± 2.4	-0.5 ± 3.1

Table 3. Inhibitory effects of **1–11** against NO production in LPS-activated BV-2 microglia cells and their cytotoxicities against BV-2 microglia cells.

3. Materials and Methods

3.1. General Experimental Procedures

UV spectra were recorded using a UV8000 UV/Vis spectrophotometer (Shanghai Metash Instruments Inc., Shanghai, China). Optical rotation data were measured on the basis of the Rudolph Autopol IV automatic polarimeter (Rudolph Reaearch Analytical, Newburgh, NY, USA). ECD spectra were measured using a Chirascan spectrometer (Applied Photophysics inc., Leatherhead, Surrey, UK). HRESIMS data were measured using a Xevo G2 Q-TOF mass spectrometer (Waters, Milford, MA, USA). The ¹H, ¹³C, HSQC, COSY, HMBC, and NOESY spectra were measured on the basis of the Bruker Avance 400 FT NMR spectrometer (Bruker Company, Fällanden, Switzerland) with tetramethylsilicane (TMS) as an internal standard. Semi-preparative HPLC was performed using an Alltech LS class pump with a model 201 variable wavelength UV/Vis detector, and a YMC packed ODS-A (250 × 10 mm, 5 μ m) column (YMC Co., Ltd. Kyoto, Japan) was used for the purification. Column chromatography (CC)

was carried out using a Sephadex LH-20 (Amersham Biosciences, San Francisco, CA, USA), ODS-A-HG (YMC Co., Ltd. Kyoto, Japan), and silica gel (Qingdao Marine Chemistry Co., Ltd., Qingdao, China). The TLC analyses were carried out with the precoated silica gel plates by heating after spraying with vanillin sulfuric acid chromogenic reagent (Xilong Scientific Co., Ltd., Shantou, China).

3.2. Fungal Material and Identifiation

The fungus was isolated from the deep-sea sediment at the depth of 2246 m sampling from the South Atlantic Ocean (W13.6639°, S14.2592°), and it was identified to be *Aspergillus sydowii* on the basis of the morphology and the internal transcribed spacer (ITS) region of the rDNA sequence. The ITS gene sequence was deposited in GenBank and assigned the accession no. MN918102. The fungus was preserved at the Marine Culture Collection of China (MCCC), and assigned the accession no. MCCC 3A00324. Therefore, the producing fungus was named *Aspergillus sydowii* MCCC 3A00324.

3.3. Fermentation, Extraction, and Isolation

The fungus was cultivated in a potato dextrose agar (PDA) plate under 25 °C for four days, and then the fresh mycelia and spores were inoculated into 500 mL Erlenmeyer flasks (×2), each containing 100 mL potato dextrose broth (PDB) medium and followed by cultivation in a rotary shaker under 25 °C at 200 rpm for four days. The seed cultures were subsequently inoculated to 30 Erlenmeyer flasks (1 L) (each containing 80 g rice and 120 mL sea water) after autoclaving at 121 °C for 22 min. The fermentation was performed under static conditions at 25 °C for 26 days.

The fermented material was fragmented using a stick and extracted successively with EtOAc three times, and then evaporated under reduced pressure to get an EtOAc extract (16.2 g). The extract was subjected to the vacuum liquid chromatography column on silica gel eluting with a gradient of CH₂Cl₂ and MeOH (1:0 to 0:1) to furnish fractions A and B. The fraction B (8.5 g) was subsequently separated by CC on ODS with MeOH/H₂O elution (30–100%) to obtain fourteen fractions (Fr.1-Fr.14). Fraction Fr.3 (147 mg) was separated by CC on silica gel eluting with petroleum ether (PE)/EtOAc (10:1) to obtain two subfractions. The former was further purified by semi-preparative HPLC with the mobile phase of MeOH/H₂O (3:7) to yield 2 (1.8 mg) and 5 (21.0 mg), whereas the latter was separated by semi-preparative HPLC (CH₃CN/H₂O, 1:9) to get **11** (0.5 mg). Fraction Fr.5 (333 mg) was chromatographed by CC on silica gel eluting with CH₂Cl₂/MeOH (30:1) to furnish two subfractions, whereas the former was further separately purified by semi-preparative HPLC using CH₃CN/H₂O (19:81) elution to obtain 10 (4.0 mg), and the latter using MeOH/H₂O (7:18) elution to yield 4 (3.0 mg). Fraction Fr.7 (489 mg) was subjected to silica gel CC with CH_2Cl_2 and MeOH gradient elution (1:0 \rightarrow 0:1) to yield two subfractions (Fr.7-1 and Fr.7-2). Subfraction Fr.7-1 was separated by semi-preparative HPLC purification (CH₃CN/H₂O, 23:77) to get 1 (1.4 mg) and 3 (3.8 mg). Subfraction Fr.7-2 was purified by semi-preparative HPLC (CH₃CN/H₂O, 37:63) to yield 9 (1.2 mg). Fraction Fr.11 (146 mg) was separated by silica gel CC eluting with CH₂Cl₂/acetone gradient from 1:0 to 0:1 to furnish two subfractions, whereas the former was further separated using CC on a Sephadex LH-20 (MeOH) and semi-preparative HPLC (CH₃CN/H₂O, 3:7) to obtain 7 (18.5 mg) and 8 (3.0 mg), respectively. Fraction Fr.13 (545 mg) was subjected to CC on silica gel eluting with PE/EtOAc (1:0 \rightarrow 0:1) to get four subfractions (Fr.13-1-Fr.13-4). Subfraction Fr.13-1 was subjected to semi-preparative HPLC with the mobile phase of CH_3CN/H_2O (39:61) to yield 6 (1.0 mg).

Aspermonoterpenoid A (1): colorless oil; $[\alpha]_D^{25}$ +54 (*c* 0.06, MeOH); UV (MeOH) λ_{max} (log ε) 223 (1.59) nm; ¹H and ¹³C NMR data, see Table 1; HRESIMS *m*/*z* 221.0784 [M + Na]⁺ (calcd. for C₁₀H₁₄O₄Na, 221.0790).

Aspermonoterpenoid B (2): colorless oil; $[\alpha]_D^{25}$ –38 (*c* 0.07, MeOH); UV (MeOH) λ_{max} (log ε) 229 (0.96) nm; ECD (MeOH) λ_{max} ($\Delta \varepsilon$) 199 (-21.88), 222 (-5.77), 231 (-13.46), 260 (-0.15) nm; ¹H and ¹³C NMR data, see Table 1; HRESIMS *m*/*z* 205.0845 [M + Na]⁺ (calcd. for C₁₀H₁₄O₃Na, 205.0841).

Asperphenylpyrone (3): colorless oil; UV (MeOH) λ_{max} (log ε) 204 (1.67), 248 (0.85), 315 (0.28) nm; ¹H and ¹³C NMR data, see Table 2; HRESIMS *m*/*z* 333.0741 [M + Na]⁺ (calcd. for C₁₈H₁₄O₅Na, 333.0739).

Aspercoumarine acid (4): white powder; UV (MeOH) λ_{max} (log ε) 204 (0.38), 296 (0.15) nm; ¹H and ¹³C NMR data, see Table 2; HRESIMS *m*/*z* 243.0268 [M + Na]⁺ (calcd. for C₁₁H₈O₅Na, 243.0269).

3.4. BV-2 Cell Culture and Treatment

The BV-2 microglia cells were cultured in DMEM medium containing 10% fetal bovine serum and antibiotics (100 units/mL of penicillin and 100 g/mL of streptomycin) and maintained in a humidified 5% CO₂ incubator (Beijing Luxi Technology Co., Ltd., Beijing, China) at 37 °C. For the experiment, cells were seeded into 24-well plates (2×10^4 cells/well) overnight. Next day, cells were incubated with fresh culture medium containing indicated concentration of the tested compounds for half an hour and following LPS treatment (1 µg/mL). Cells were treated vehicle (DMSO, 0.1%) as control.

3.5. Nitrite Quantification

The concentration of nitrite in culture medium was determined using a Griess Reagent Kit (Thermo Fisher, Shanghai, China). Briefly, 75 μ L of cell culture supernatants were reacted with an equal volume of Griess Reagent Kit for 30 min at room temperature, and absorbance of diazonium was obtained at a wavelength of 560 nm. Nitrite production by vehicle stimulation was designated as 100% inhibition compared to LPS stimulation for the experiment.

3.6. Computational Details

3.6.1. ¹³C NMR Calculation of 2

Conformational searches were carried out using the Maestro 10.2 program (Schrödinger Inc., NY, USA) at the OPLS3 molecular mechanics force field within an energy window of 3.0 kcal/mol. The results exhibited 8 conformers for $(4S^*,5R^*,6S^*)$ -2 and $(4S^*,5R^*,6R^*)$ -2, respectively. The conformers were further optimized at the B3LYP/6-31+G(d,p) level in gas phase using Gaussian 09 (Gaussian, Inc., Wallingford, CT, USA) [38]. The conformers with a Boltzmann population over 1% were selected for the NMR calculations. The NMR data were calculated by the GIAO method at the mPW1PW91/6-311+G(2d,p) level with the IEFPCM model in methanol. Finally, the calculated NMR data were averaged according to the Boltzmann distribution for each conformer and then fitted to the experimental values by linear regression.

3.6.2. ECD Calculation of 2

The eight conformers of $(4S^*, 5R^*, 6S^*)$ -2 were optimized by density functional theory (DFT) calculations at the B3LYP/6-31+G(d,p) level in gas phase using Gaussian 09. The energies, oscillator strengths, and rotational strengths of the first 60 electronic excitations were calculated by the TDDFT method at the B3LYP/6-311+G(2d,p) level in methanol. The ECD spectrum was simulated using SpecDis (version1.70, Berlin, Germany) by applying the Gaussian band shapes with sigma = 0.3 eV. Finally, the calculated ECD data were weighted and then summed up each stable conformer on the basis of the Boltzmann population.

4. Conclusions

In summary, four new compounds, including one novel (1) and one new (2) monoterpenoids and two new polyketides (3 and 4), were obtained from the EtOAc extract of the deep-sea-derived fungus *Aspergillus sydowii* MCCC 3A00324, together with seven known compounds (5–11). The structures of metabolites were determined by comprehensive analyses of the NMR and HRESIMS spectra, in association with quantum chemical calculations of the ECD, ¹³C NMR, and specific rotation data for their configurational assignment. Compound 1 possessed a novel monoterpenoid skeleton,

bearing two phenyl units at C-3 and C-5, respectively, indicating that the deep-sea-derived fungi are a unique source of the structurally novel compounds. Compound **6** exhibited significant inhibitory effects against NO secretion in LPS-activated BV-2 microglia cells (94.4% inhibition rate, 10 μ M), suggesting the potential application for the anti-inflammatory agents.

Supplementary Materials: The following are available online at http://www.mdpi.com/1660-3397/18/11/561/s1, Figure S1: The HPLC and TLC metabolic profile of the study fungus, Figures S1-1–S4-7: HRESIMS, ¹H, ¹³C, HSQC, COSY, and HMBC spectra of new compounds 1–4, and NOESY spectra of 1 and 2.

Author Contributions: S.N., Z.S., and S.P. isolated and identified the fungus. S.N. isolated and elucidated the structures. L.Y. and T.C. performed the anti-inflammatory assay. B.H. and G.Z. designed and coordinated the study. S.N., B.H., and G.Z. wrote and revised the manuscript. All authors have read and agreed to the published version of the manuscript.

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