

Copper(I)-chloroquine complexes: Interaction with DNA and Ferriprotoporphyrin, Inhibition of β -hematin Formation and Relation to Antimalarial Activity

Wilmer Villarreal ^{1,2}, William Castro ², Sorenlis González ², Marylin Madamet ^{3,4,5,6}, Rémy Amalvict ^{3,4,5,6}, Bruno Pradines ^{3,4,5,6}, Maribel Navarro ^{2,7*}

¹ Grupo de Química Inorgânica Medicinal e Reações Aplicadas. Instituto de Química, Universidade Federal do Rio Grande do Sul (UFRGS), Porto Alegre 91501-970, Brazil.

² Centro de Química, Instituto Venezolano de Investigaciones Científicas (IVIC). Caracas 1020-A, Venezuela.

³ Unité Parasitologie et Entomologie, Département Microbiologie et Maladies Infectieuses, Institut de Recherche Biomédicale des Armées, Marseille 13005, France.

⁴ Aix Marseille Unive, IRD, SSA, AP-HM, VITROME, Marseille 13005, France.

⁵ IHU Méditerranée Infection, Marseille 13005, France.

⁶ Centre National de Référence du Paludisme, Marseille 13005, France.

⁷ Laboratório de Química Bioinorgânica e Catálise. Departamento de Química, Instituto de Ciências Exatas, Universidade Federal de Juiz de Fora (UFJF), Juiz de Fora 36036-900, Brazil.

Supporting Information

Figure S1 – S5: ¹H, ¹³C{¹H}, ¹H-¹H COSY, ¹H-¹³C HMQC and ¹H-¹³C HMBC NMR spectrum of [Cu(CQ)(PPh₃)₂]⁺NO₃ (**1**) in CD₂Cl₂ at 298 K.

Figure S6: ³¹P{¹H} NMR spectrum of [Cu(CQ)(PPh₃)₂]⁺NO₃ (**1**) in CD₂Cl₂ at 298 K.

Table S1: Chemical shifts of the protons and carbons for the complex **1** and their variations when compared with the free chloroquine ligand.

Figure S7: Theoretical (a) and Experimental (b) isotopic distribution for [Cu(CQ)(PPh₃)₂]⁺.

Figure S8: EPR spectra for (a) [Cu(PPh₃)₂](NO₃), (b) [Cu(CQ)(PPh₃)₂]⁺NO₃ and (c) Cu(II) complex for comparison.

Figure S9: ¹H NMR spectra of the complex **2** in 90% DMSO-*d*₆ and 10% PBS-*D*₂O, obtained at 0 – 48 h after sample preparation. [Complex **2**] = 13.9 mM.

Figure S10: ¹H NMR spectra of the complex **1** in 90% DMSO-*d*₆ and 10% PBS-*D*₂O, obtained at 0 – 48 h after sample preparation. [Complex **1**] = 9.2 mM.

Figure S11: ³¹P{¹H} NMR spectra of the complex **1** in 90% DMSO-*d*₆ and 10% PBS-*D*₂O, obtained at 0 – 48 h after sample preparation. [Complex **1**] = 9.2 mM.

Figure S12: Absorption spectra of the complexes **1**(a) and **2**(b) in 10% DMSO and 90% PBS, obtained at 0 – 48 h after sample preparation. [Complex] = 10 μM.

M. Navarro / Wilmer Villarreal / MNWV-48. / CD₂Cl₂.
Protones.

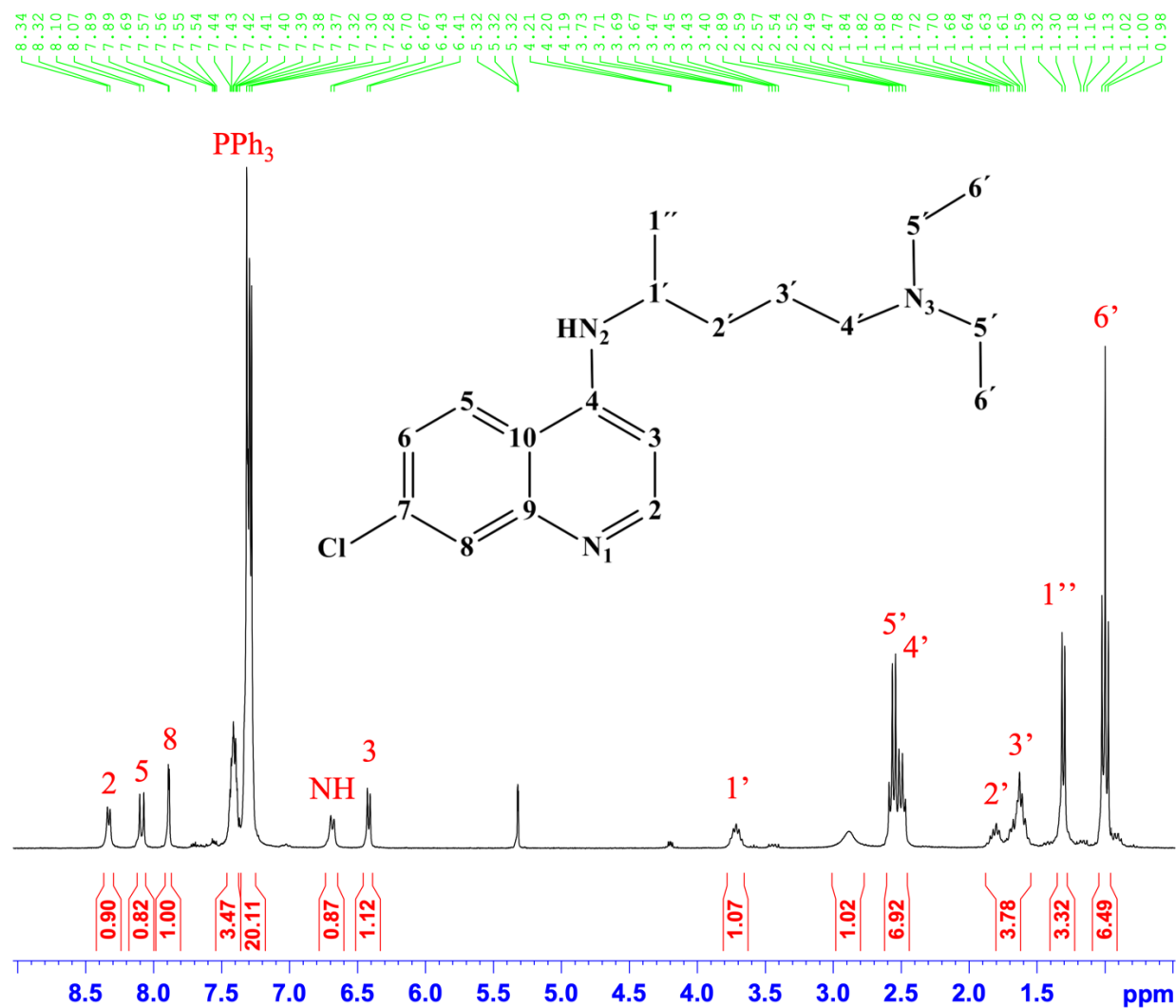


Figure S1 – The ¹H NMR spectrum of [Cu(CQ)(PPh₃)₂]⁺NO₃⁻ (1) in CD₂Cl₂ at 298 K

M. Navarro / W.Villarreal / MNWV-48 / CD₂Cl₂ / Carbone.

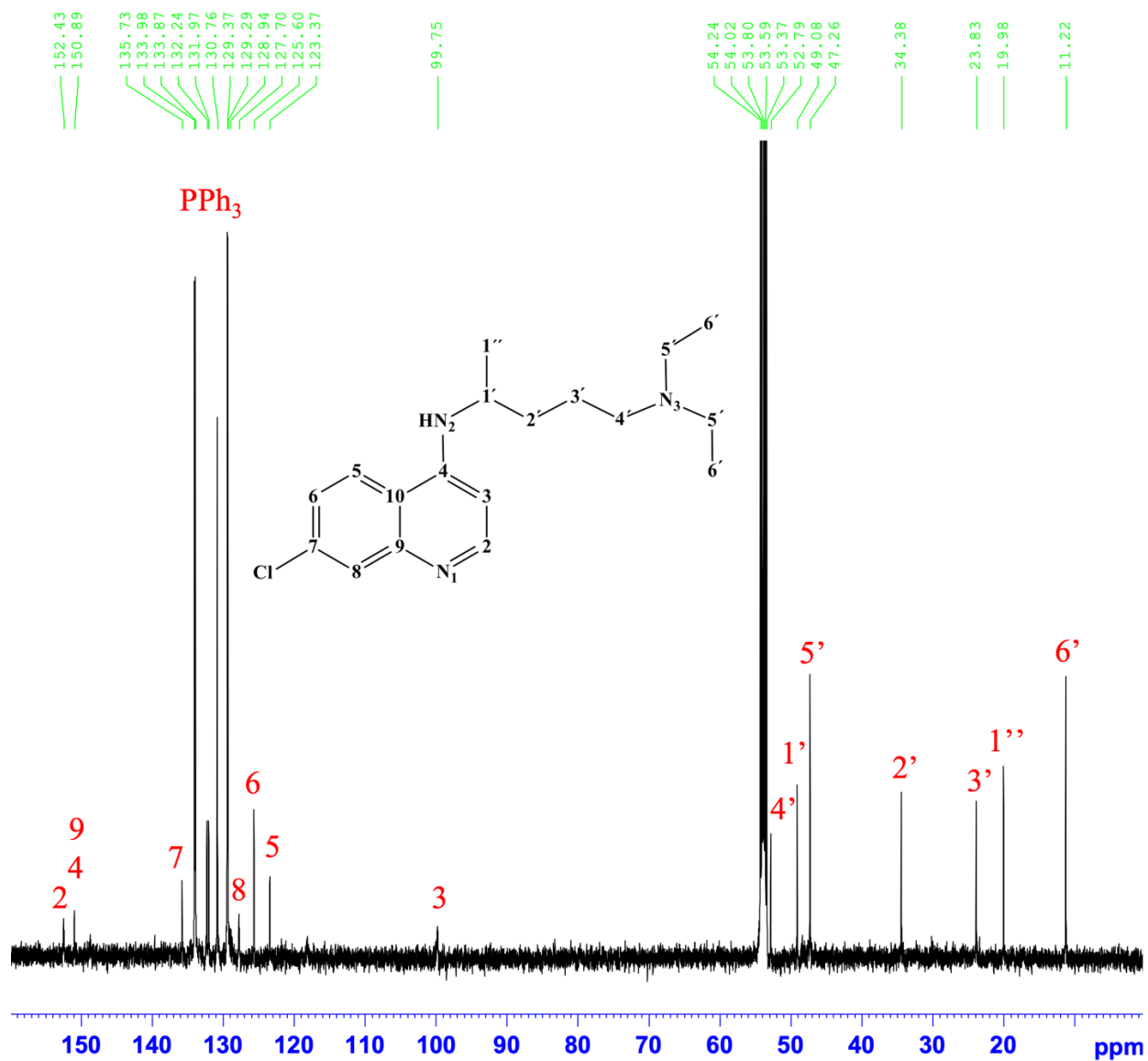


Figure S2 – The ¹³C{¹H} NMR spectrum of [Cu(CQ)(PPh₃)₂]⁺NO₃⁻ (1) in CD₂Cl₂ at 298 K

M. Navarro / Wilmer Villarreal / MNWV-48. / CD₂Cl₂.
Cosy.

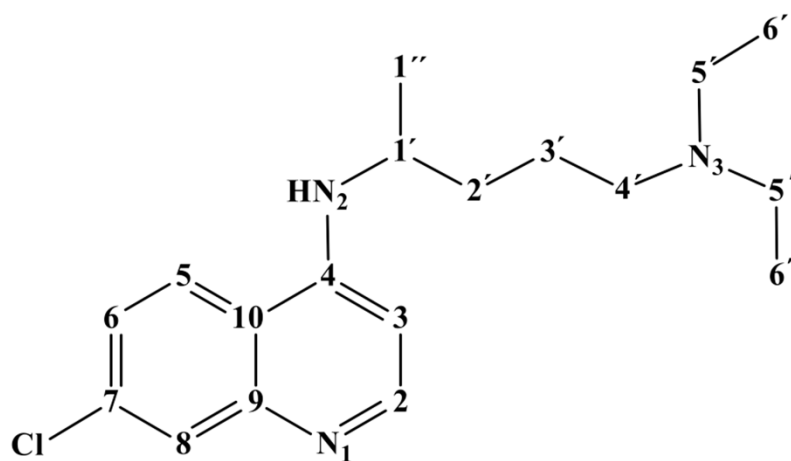
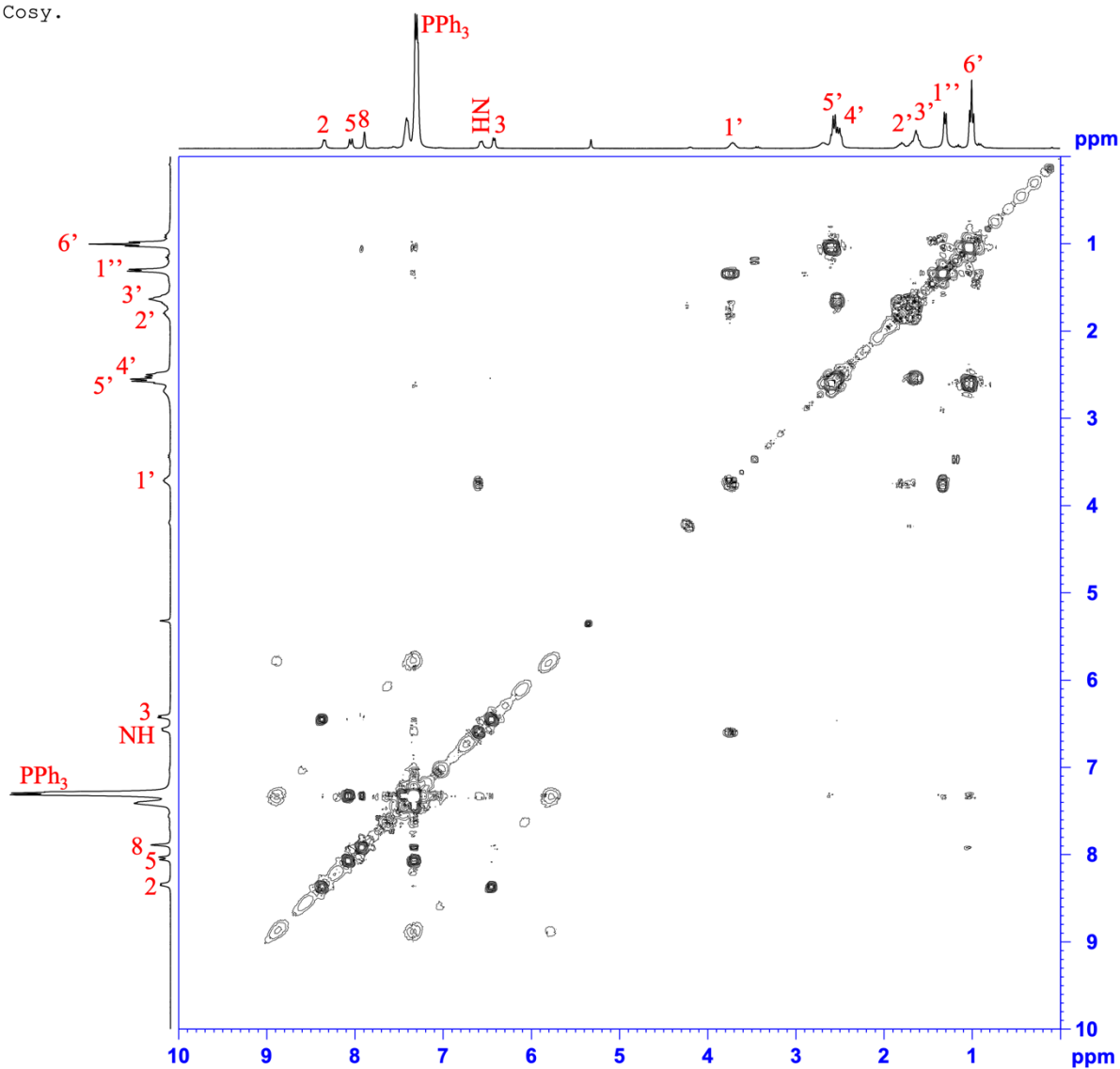


Figure S3 – The 2D homonuclear ¹H-¹H COSY NMR spectrum of [Cu(CQ)(PPh₃)₂]⁺NO₃⁻ (**1**) in CD₂Cl₂ at 298 K

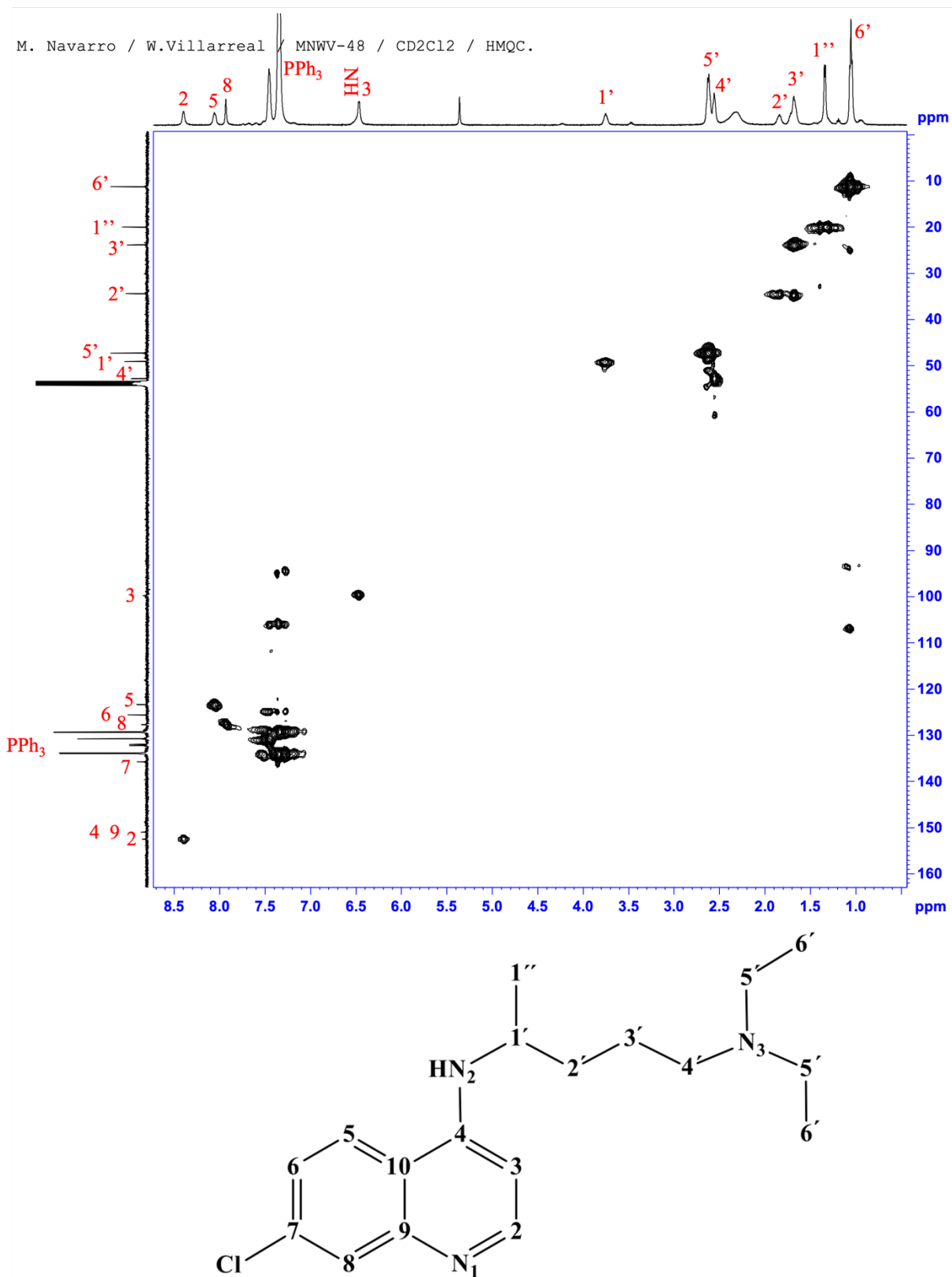


Figure S4 – The 2D heteronuclear ¹H-¹³C HMQC NMR spectrum of [Cu(CQ)(PPh₃)₂]NO₃ (**1**) in CD₂Cl₂ at 298 K

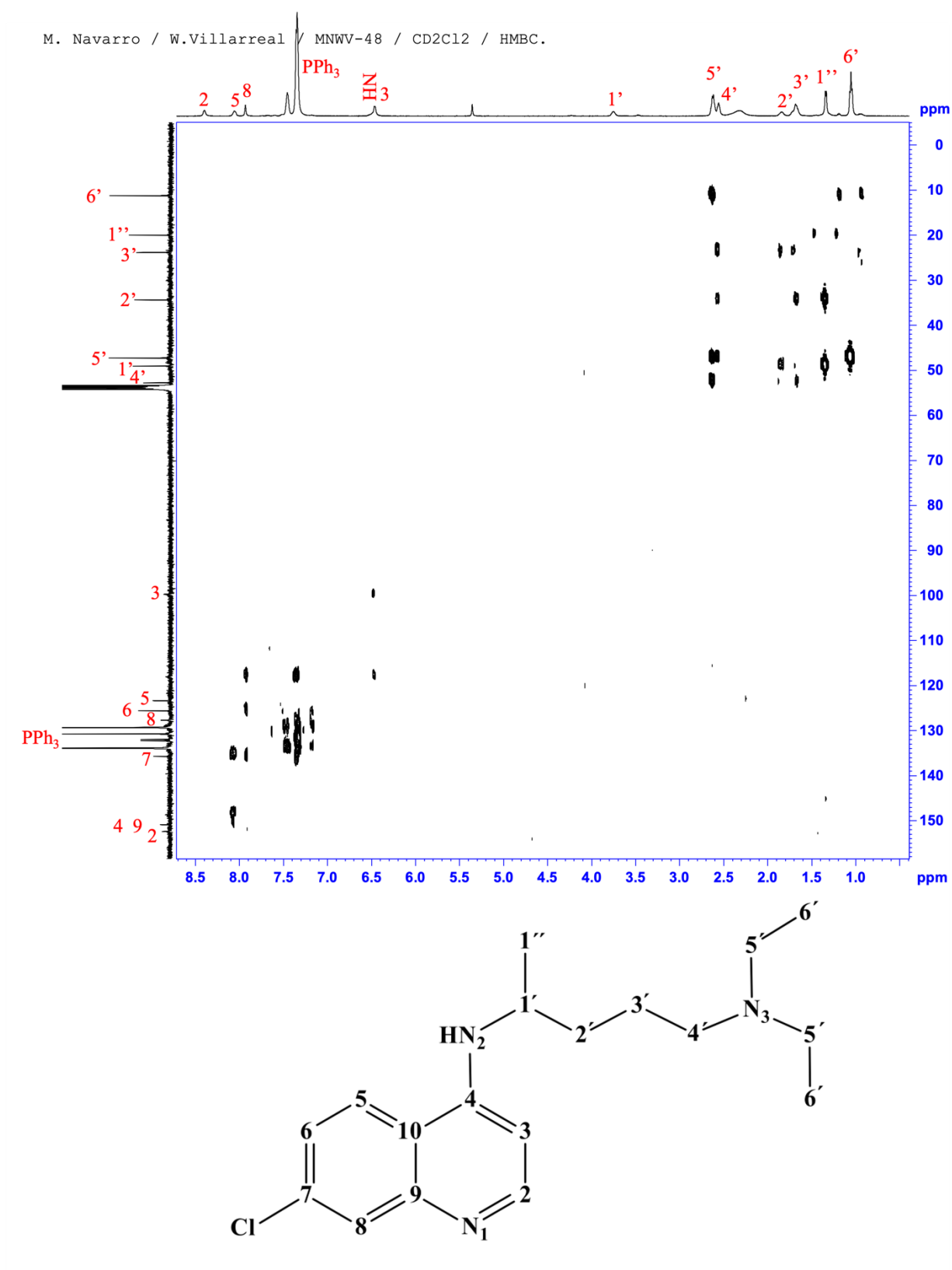


Figure S5 – The 2D heteronuclear ¹H-¹³C HMBC NMR spectrum of [Cu(CQ)(PPh₃)₂]NO₃ (**1**) in CD₂Cl₂ at 298 K

M. Navarro / Wilmer Villarreal / MNWV-48. / CD₂Cl₂.
Fosforo.

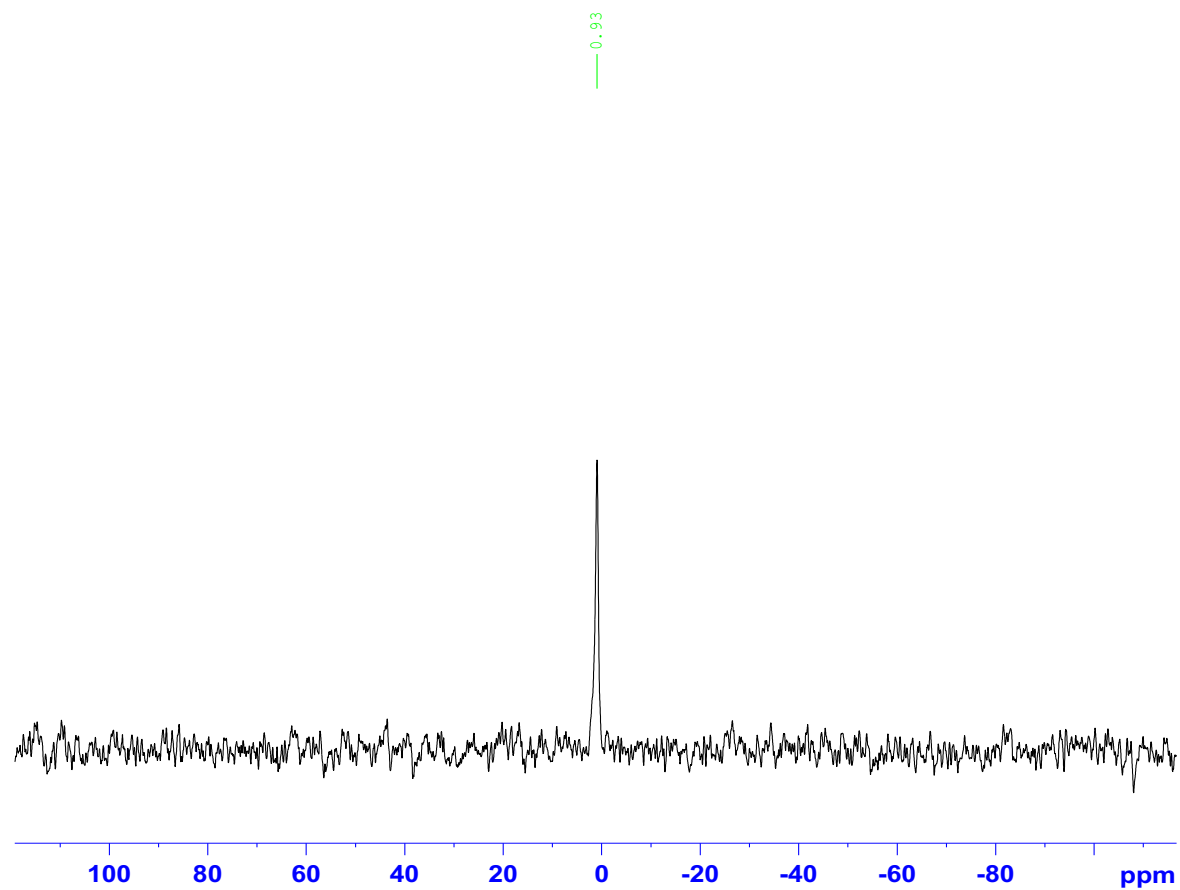


Figure S6 - $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $[\text{Cu}(\text{CQ})(\text{PPh}_3)_2]\text{NO}_3$ (**1**) in CD_2Cl_2 at 298 K

Table S1 - Chemical shifts of the protons and carbons for the complex **1** and their variations when compared with the free chloroquine ligand

Chemical shifts in ^1H NMR ($\text{CH}_2\text{Cl}_2-d_2$)			Chemical shifts in $^{13}\text{C}\{^1\text{H}\}$ NMR ($\text{CH}_2\text{Cl}_2-d_2$)		
Position	δ (ppm)	Δ (ppm)	Position	δ (ppm)	Δ (ppm)
6'	1.00	0.02	6'	11.22	0.40
1''	1.31	0.01	1''	19.98	0.22
2' and 3'	1.71	0.06	3'	23.83	0.36
4' and 5'	2.47	0.07	2'	34.38	0.39
1'	3.70	0.01	5'	47.26	0.08
NH	6.68	1.27	1'	49.08	0.33
3	6.42	0.04	4'	52.79	0.07
PPh ₃	7.38	-----	3	99.75	0.08
5	8.09	0.72	10	118.20	0.32
8	7.90	0.01	5	123.37	1.23
2	8.33	0.13	6	125.60	0.6
			8	127.70	1.11
			b	129.33	----
			a	130.76	----
			d	132.06	----
			c	133.55	----
			7	135.73	0.93
			9	150.89	1.24
			4	150.92	1.11
			2	152.43	0.15

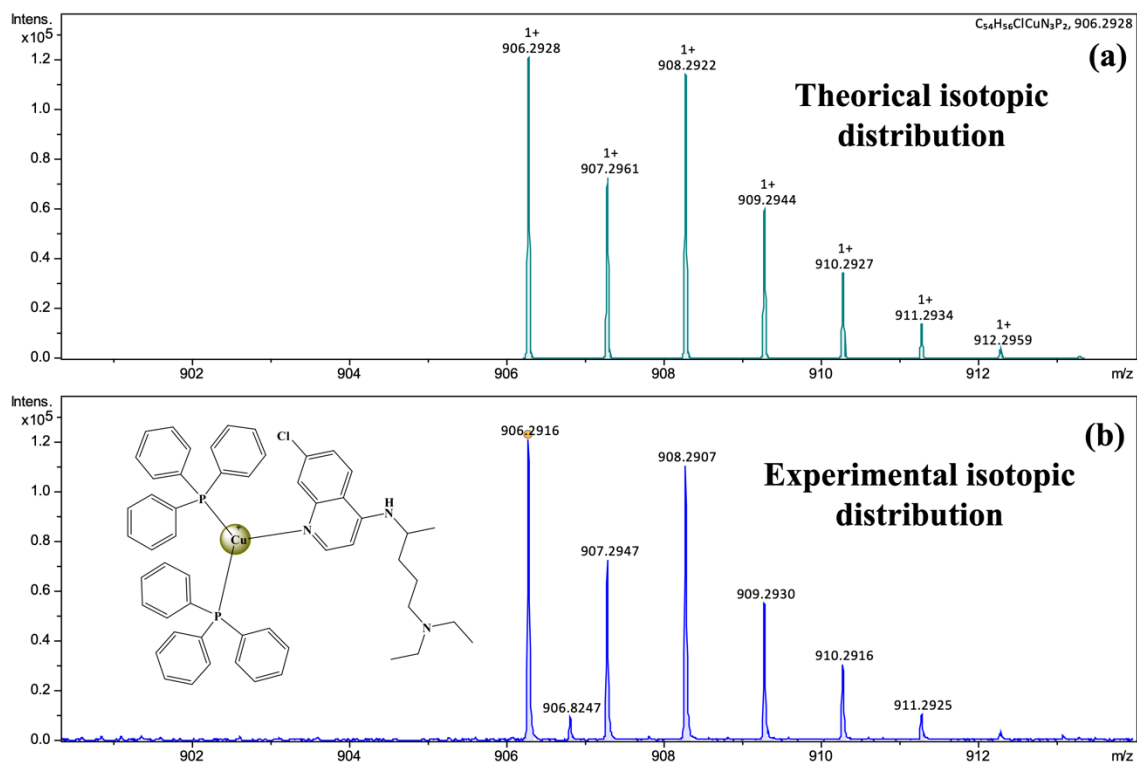


Figure S7 - Theoretical (a) and Experimental (b) isotopic distribution for $[\text{Cu}(\text{CQ})(\text{PPh}_3)_2]^+$.

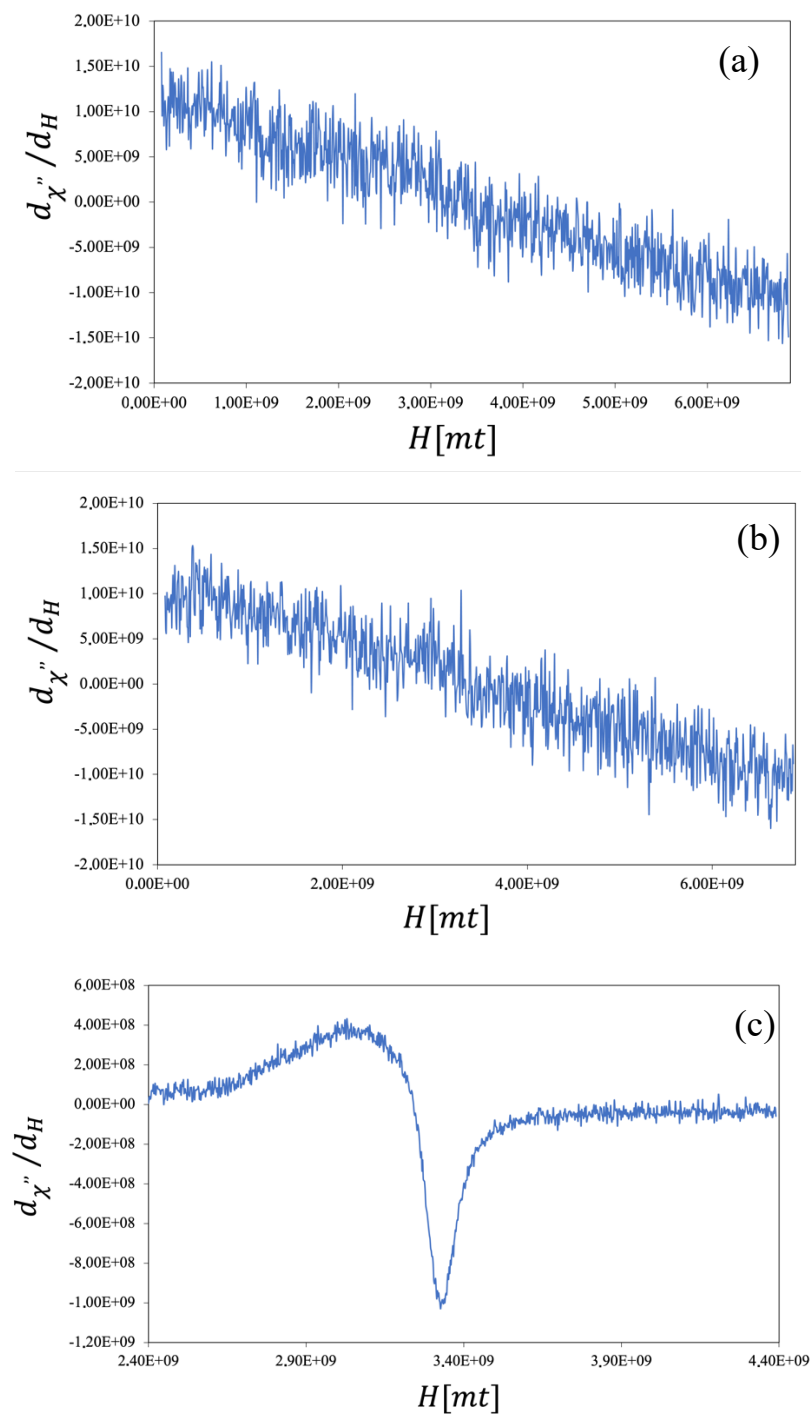


Figure S8 – EPR spectra for (a) $[\text{Cu}(\text{PPh}_3)_2(\text{NO}_3)]$, (b) $[\text{Cu}(\text{CQ})(\text{PPh}_3)_2]\text{NO}_3$ and (c) Cu(II) complex for comparison.

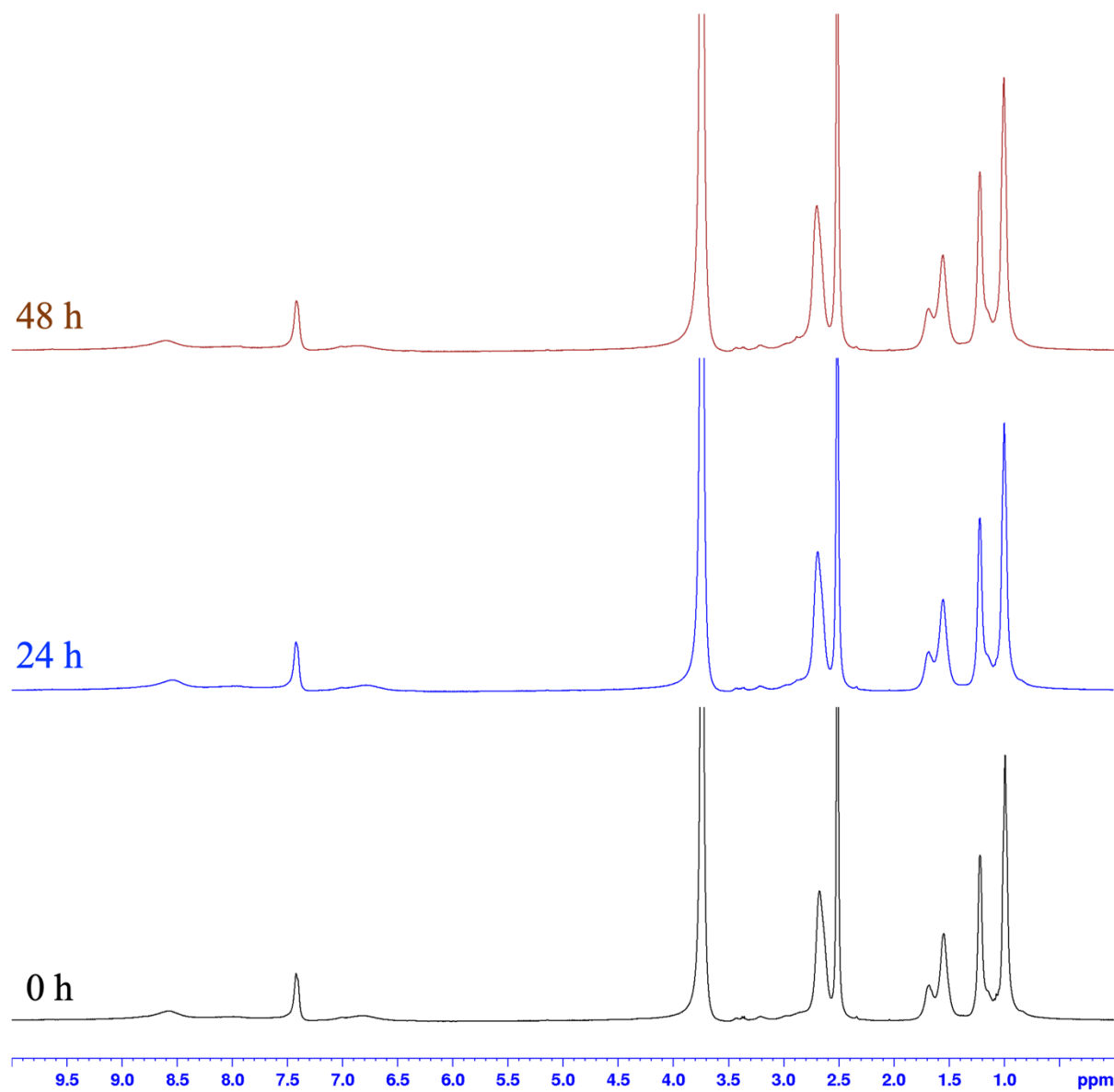


Figure S9 – ^1H NMR spectra of the complex **2** in 90% $\text{DMSO}-d_6$ and 10% $\text{PBS}-D_2O$, obtained at 0 – 48 h after sample preparation. $[\text{Complex } \mathbf{2}] = 13.9 \text{ mM}$.

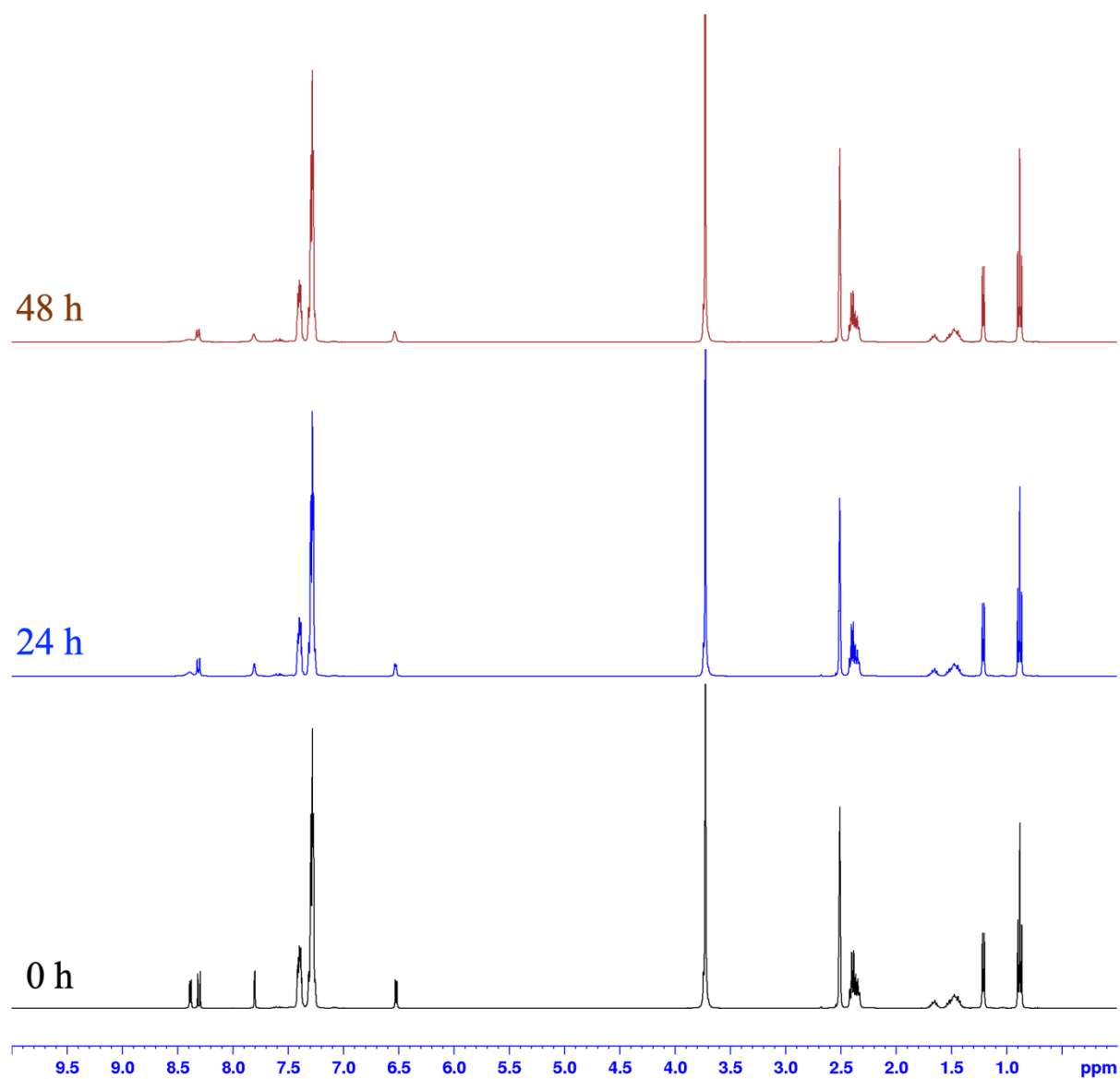


Figure S10 – ^1H NMR spectra of the complex **1** in 90% $\text{DMSO}-d_6$ and 10% $\text{PBS}-D_2O$, obtained at 0 – 48 h after sample preparation. $[\text{Complex } \mathbf{1}] = 9.2 \text{ mM}$.

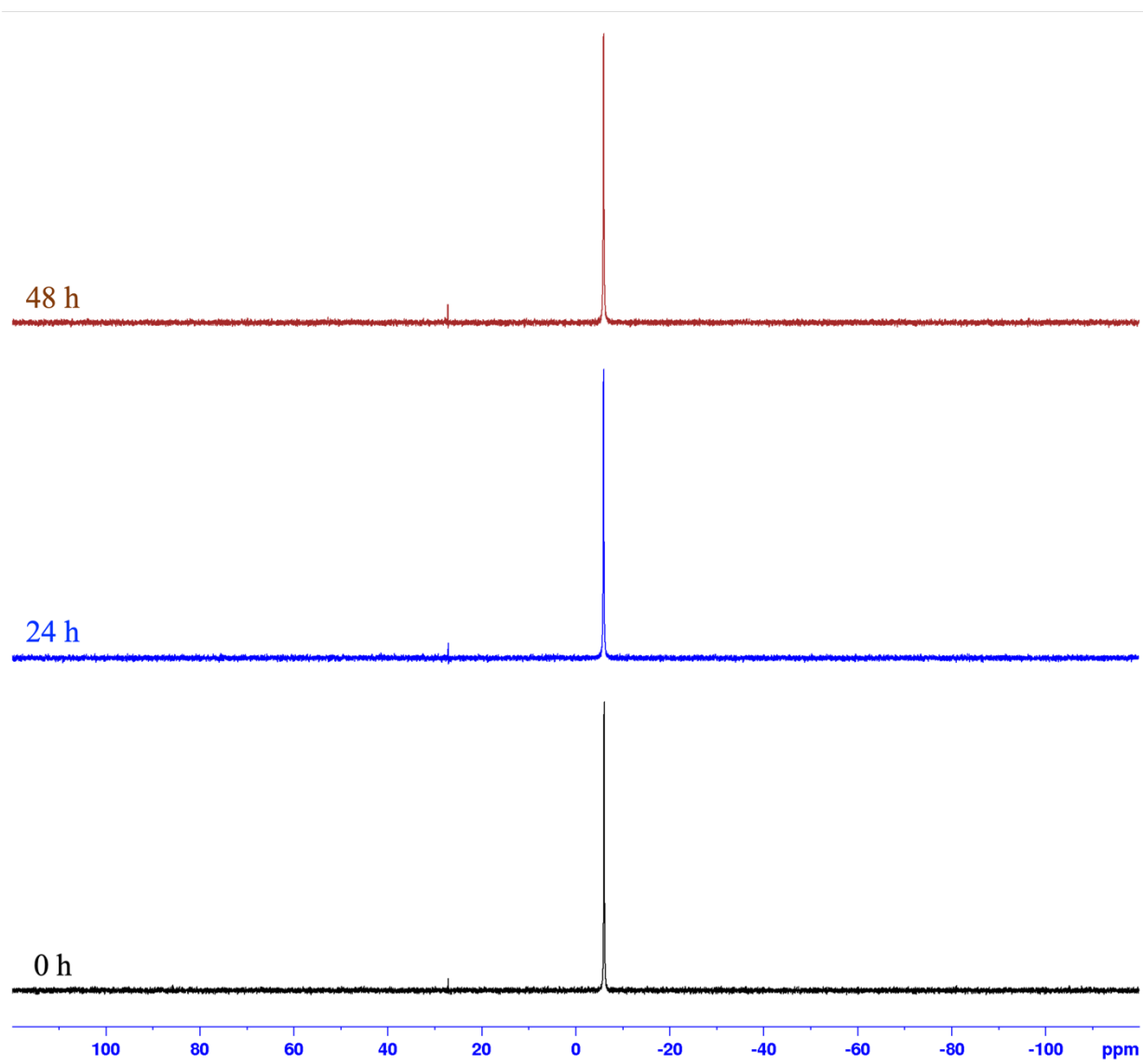


Figure S11 – $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of the complex **1** in 90% $\text{DMSO}-d_6$ and 10% $\text{PBS}-D_2\text{O}$, obtained at 0 – 48 h after sample preparation . $[\text{Complex } \mathbf{1}] = 9.2 \text{ mM}$.

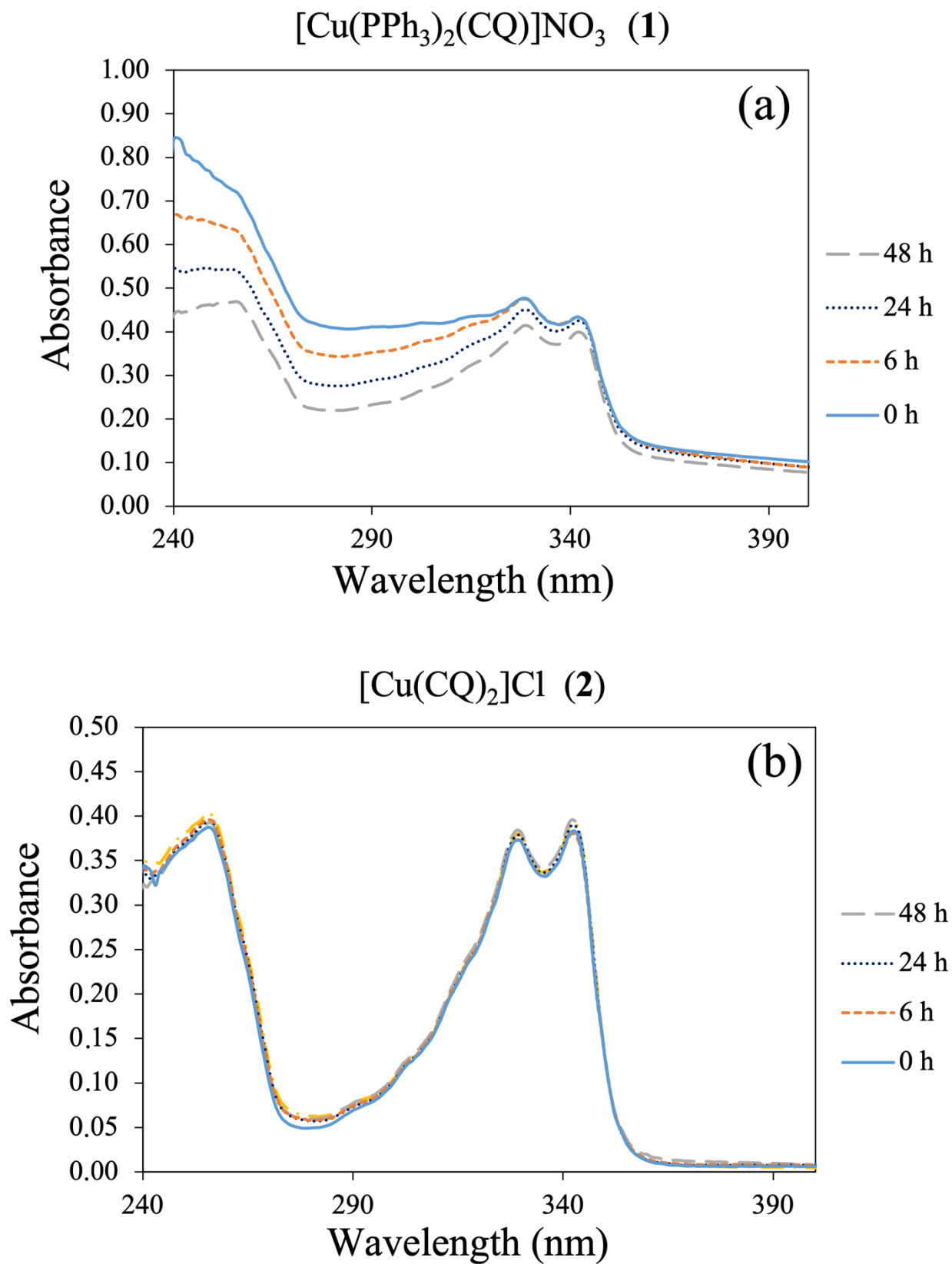


Figure S12 – Absorption spectra of the complexes **1**(a) and **2**(b) in 10% DMSO and 90% PBS, obtained at 0 – 48 h after sample preparation . $[\text{Complex}] = 1 \text{ } \mu\text{M}$.