

Supplementary Materials for

Synthesis, physicochemical, labeling and in vivo characterization of ^{44}Sc -labeled DO3AM-NI as a hypoxia sensitive PET probe

Dániel Szücs^{1, 2, 3}, Tibor Csupász^{2,3}, Judit P. Szabó^{1,4}, Adrienn Kis¹, Barbara Gyuricza^{1, 3}, Viktória Arató^{1, 5}, Viktória Forgács^{1, 3}, Adrienn Vágner⁶, Gábor Nagy⁶, Ildikó Garai^{1, 6}, Dezső Szikra^{1, 6}, Imre Tóth², György Trencsényi^{1, 6}, Gyula Tircsó² and Anikó Fekete^{1, *}

¹ Division of Nuclear Medicine and Translational Imaging, Department of Medical Imaging, Faculty of Medicine, University of Debrecen, Nagyerdei krt. 98., H-4032 Debrecen, Hungary;
szucs.daniel@science.unideb.hu (D.S.); szabo.judit@med.unideb.hu (J.P.S); kis.adrienn@med.unideb.hu (A.K.); gyuricza.barbara@med.unideb.hu (B.G.); arato.viktoria@med.unideb.hu (V.A.); forgacs.viktoria@med.unideb.hu (V.F.); garai.ildiko@med.unideb.hu (I.G.); szikra.dezso@med.unideb.hu (D.S.); trencsenyi.gyorgy@med.unideb.hu (G.T.); fekete.aniko@science.unideb.hu (A.F.).

² Department of Physical Chemistry, Faculty of Science and Technology, University of Debrecen, Egyetem tér 1, H-4032 Debrecen, Hungary; csupasz.tibor@science.unideb.hu (T.C.); imre.toth@science.unideb.hu (I.T.); gyula.tircso@science.unideb.hu (G.T.).

³ Doctoral School of Chemistry, Faculty of Science and Technology, University of Debrecen, Egyetem tér 1., H-4032 Debrecen, Hungary;

⁴ Doctoral School of Clinical Medicine, Faculty of Medicine, University of Debrecen, Nagyerdei krt. 98., H-4032 Debrecen, Hungary;

⁵ Doctoral School of Pharmaceutical Sciences, Faculty of Pharmacy, University of Debrecen, Nagyerdei krt. 98., H-4032 Debrecen, Hungary;

⁶ Scanomed Ltd., Nagyerdei krt. 98., H-4032 Debrecen, Hungary; vagner.adrienn@scanomed.hu (A.V.); nagy.gabor@scanomed.hu (G.N.).

* Correspondence: fekete.aniko@science.unideb.hu (A.F.); Tel.: +36-52-255-510 (ext. 54470)

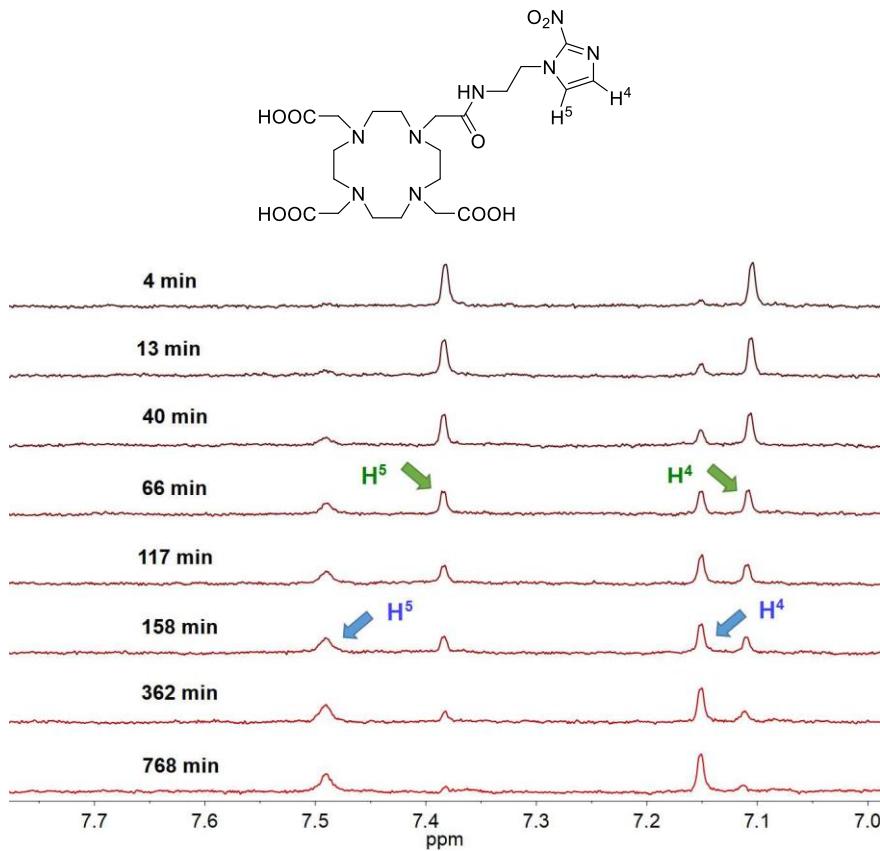


Figure S1. Aromatic region of the ^1H -NMR spectra of the samples used to probe the formation rate of Sc(DO3AM-NI) at pH = 1.52 ($c_L=3.8$ mM; $c_{\text{Sc}^{3+}}=4.9$ mM; $I=0.15$ M; 25 °C). Green arrows show the ^1H -NMR peaks of the ligand and blue arrows show the ^1H -NMR peaks of the scandium complex.

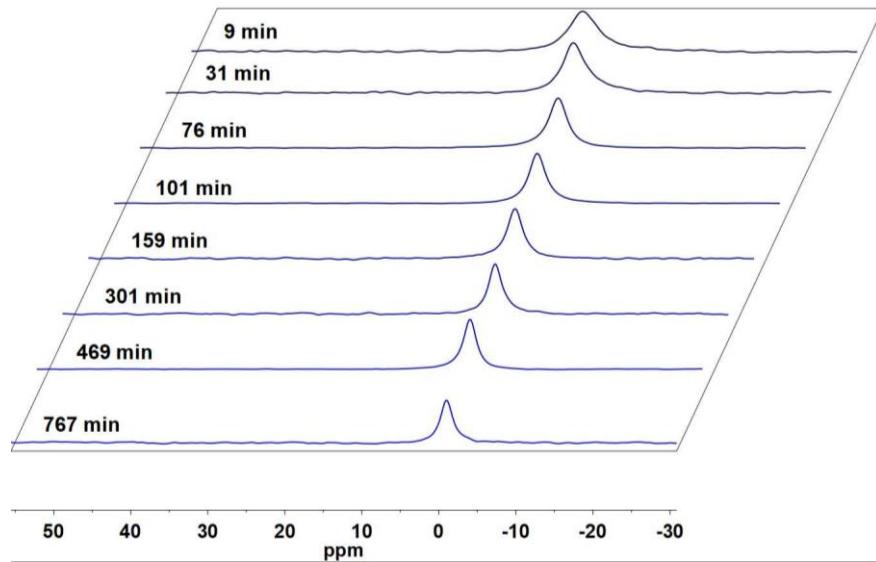


Figure S2. The ^{45}Sc -NMR spectra obtained as a function of time for the Sc(DO3AM-NI) complex formation at pH = 1.52 ($c_{\text{L}}=3.8 \text{ mM}$; $c_{\text{Sc}^{3+}}=4.9 \text{ mM}$; $I=0.15 \text{ M}$; 25 °C) ($\delta=+4.8 - -0.7 \text{ ppm}$). These spectra show just the free $\text{Sc}^{3+}_{(\text{aq})}$ peaks, the signal of the scandium complex is out of this chemical shift range.

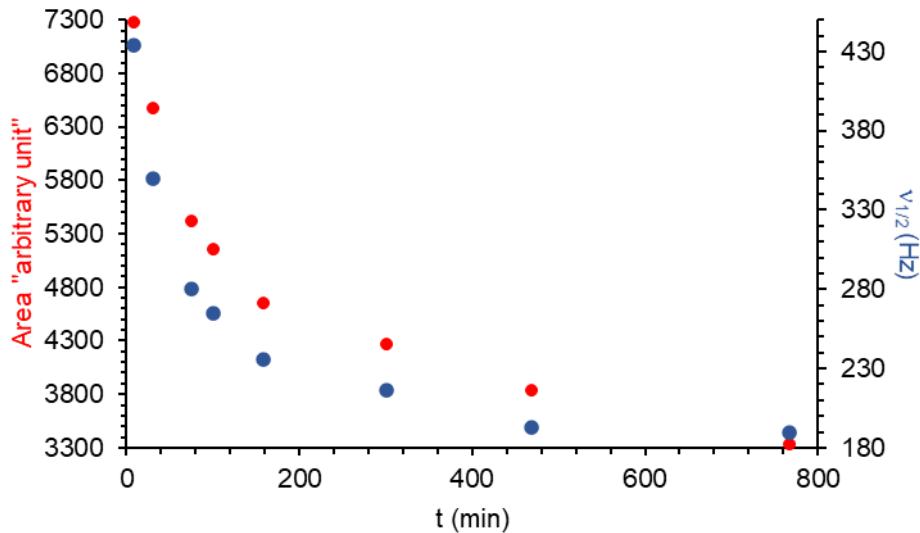


Figure S3. Peak area (●) and width (○) of the ^{45}Sc -NMR signal corresponding to the free Sc^{3+} as a function of time for the formation of $\text{Sc}(\text{DO3AM-NI})$ at $\text{pH} = 1.52$ ($c_{\text{L}}=3.8 \text{ mM}$; $c_{\text{Sc}^{3+}}=4.9 \text{ mM}$; $I=0.15 \text{ M}$; 25°C).

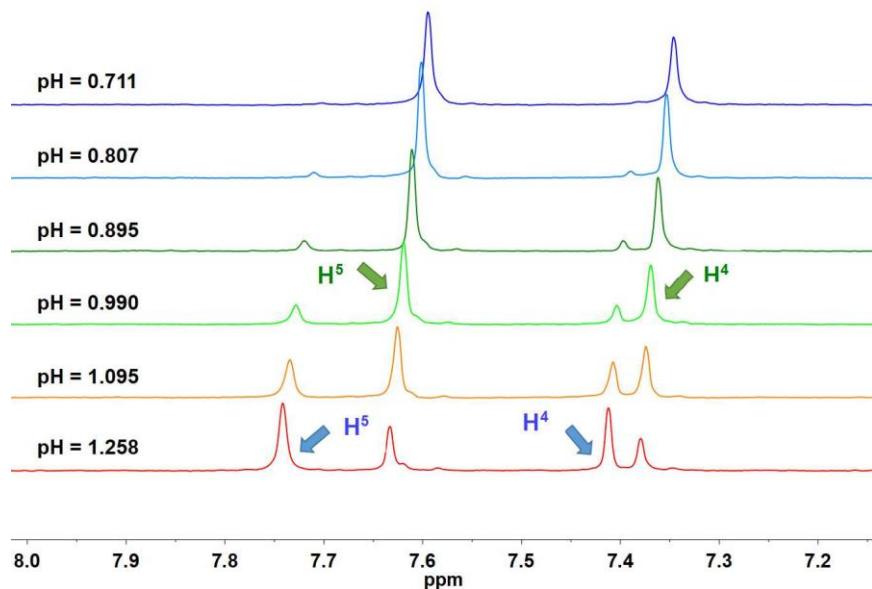


Figure S4. Aromatic region of the ^1H -NMR spectra of the samples used for the determination of the $\text{Sc}(\text{DO3AM-NI})$ complex ($c_{\text{L}}=c_{\text{Sc}^{3+}}=3.2 \text{ mM}$; $I=0.15 \text{ M}$; 37°C). Green arrows show the ^1H -NMR peaks of the ligand and blue arrows show the ^1H -NMR peaks of the scandium complex.

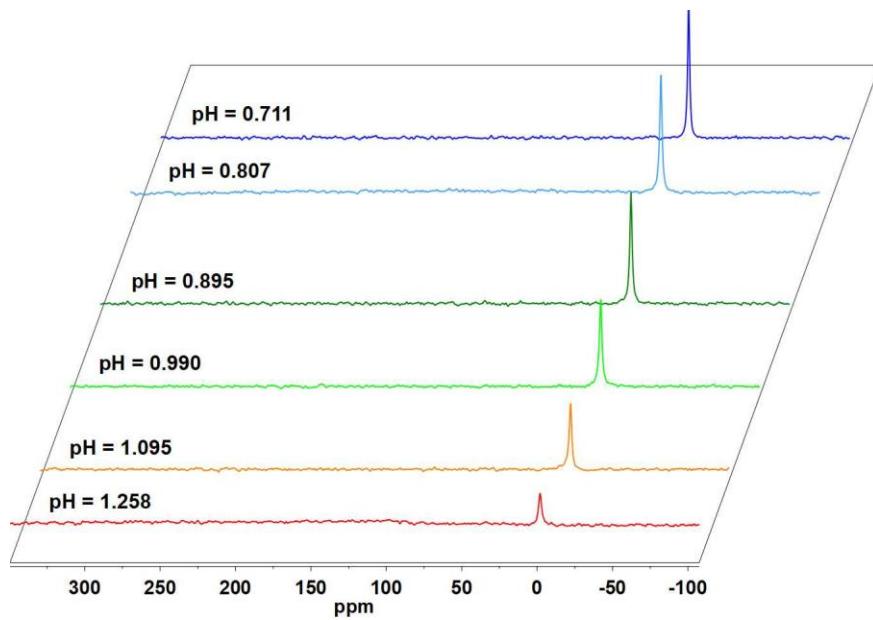


Figure S5. ^{45}Sc -NMR spectra of the samples used for the determination of the Sc(DO3AM-NI) complex ($c_{\text{L}}=c_{\text{Sc}^{3+}}=3.2 \text{ mM}$; $I=0.15 \text{ M}$; 37°C).

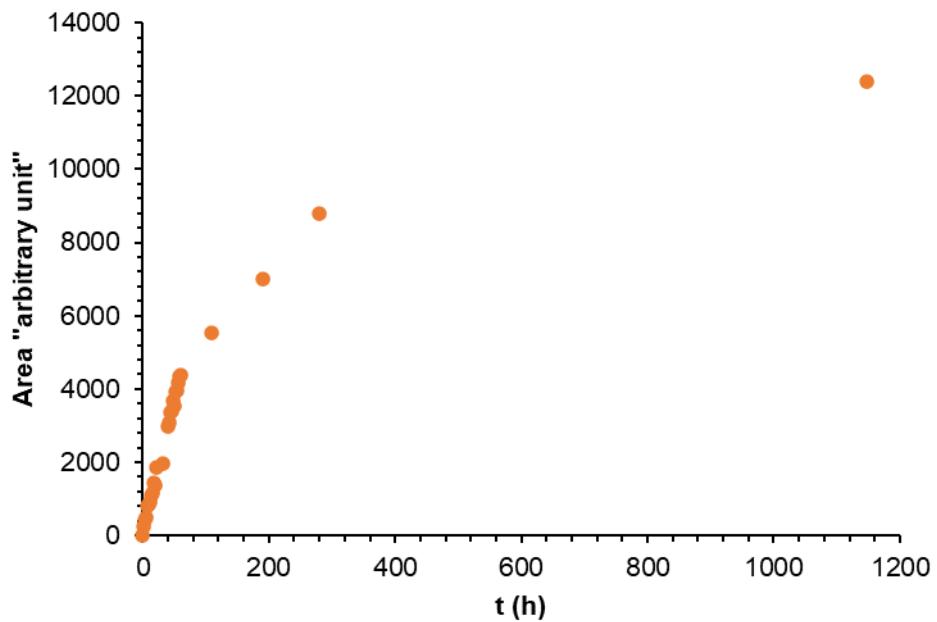


Figure S6. Peak area corresponding to the free Sc(III) as a function of time for the dissociation of Sc(DO3AM-NI) in 1.0 M HCl ($c_{\text{complex}}=7.30 \text{ mM}$; $I=1.03 \text{ M}$; 25°C).

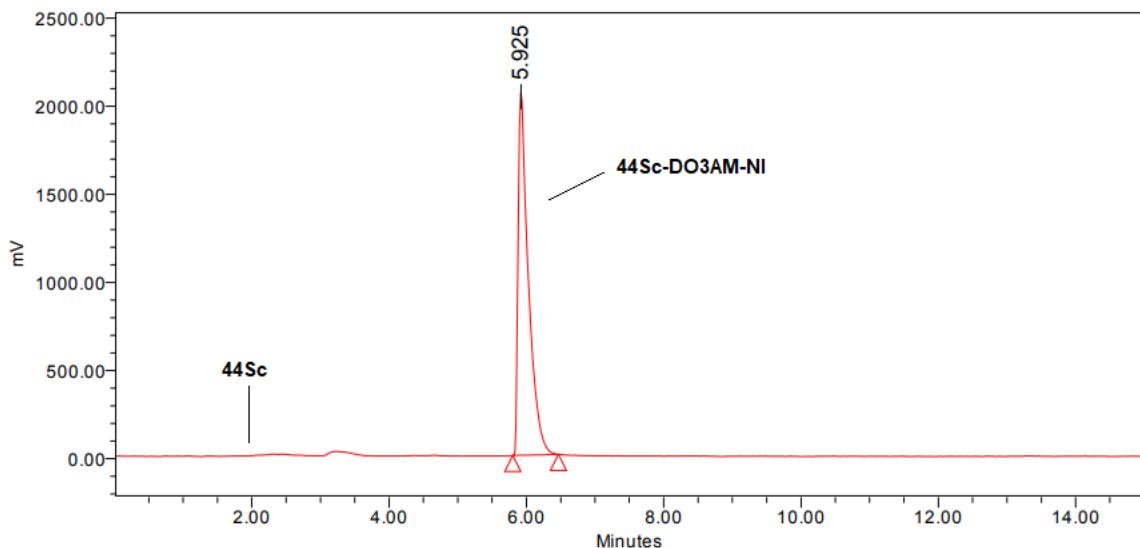


Figure S7. Radiochromatogram of purified $[^{44}\text{Sc}]\text{Sc}(\text{DO3AM-NI})$. Radio-HPLC was performed using a Luna C18 3 μm (150 x 4.6 mm) column, solvent A: oxalic acid (0.01 M pH=3); solvent B: acetonitrile.

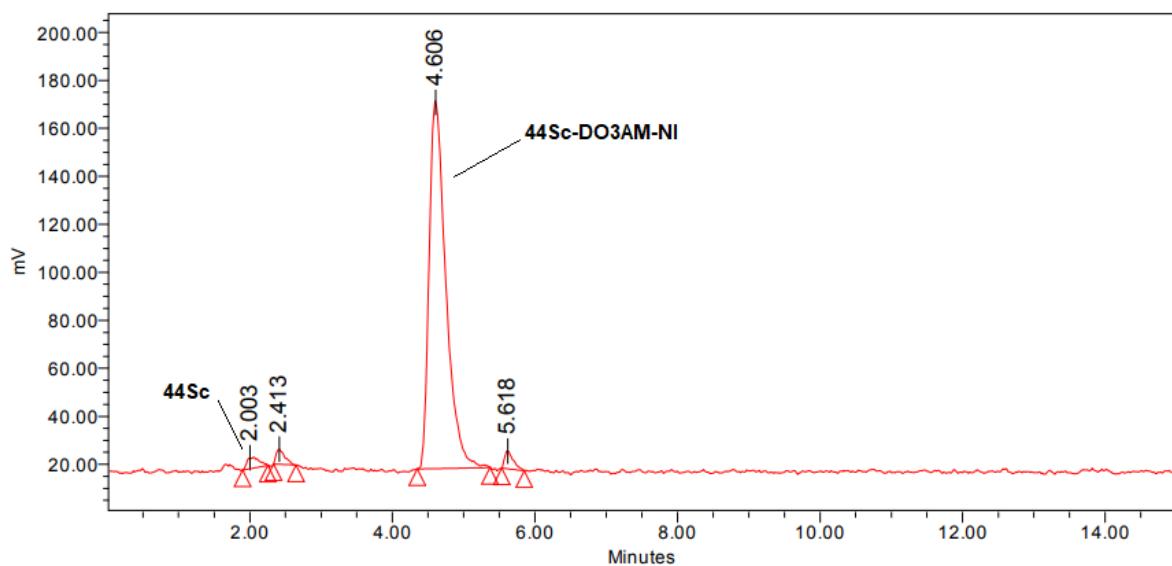


Figure S8. Radiochromatogram of the serum stability test of $[^{44}\text{Sc}]\text{Sc}(\text{DO3AM-NI})$ at 240 min. Radio-HPLC was performed using a Kinetex C18 2.6 μm (100 x 4.6 mm) column, solvent A: oxalic acid (0.01 M pH=3); solvent B: acetonitrile.