

HPLC-UV-MS analysis of hydro-alcoholic propolis extract

The UV quantitation is made only if there is a match with MS spectra AND if the UV signal allows quantitation ($S/N > 5$). For gentisic acid, hyperoside and myricetin those criteria were not met, so only qualitative results were provided. The UV, together with corresponding MS chromatograms for the 3 compounds (aligned to time axis) are shown below: (chromatograms presented: UV, MS-gentisic acid, MS-myricetol, MS-hyperoside).

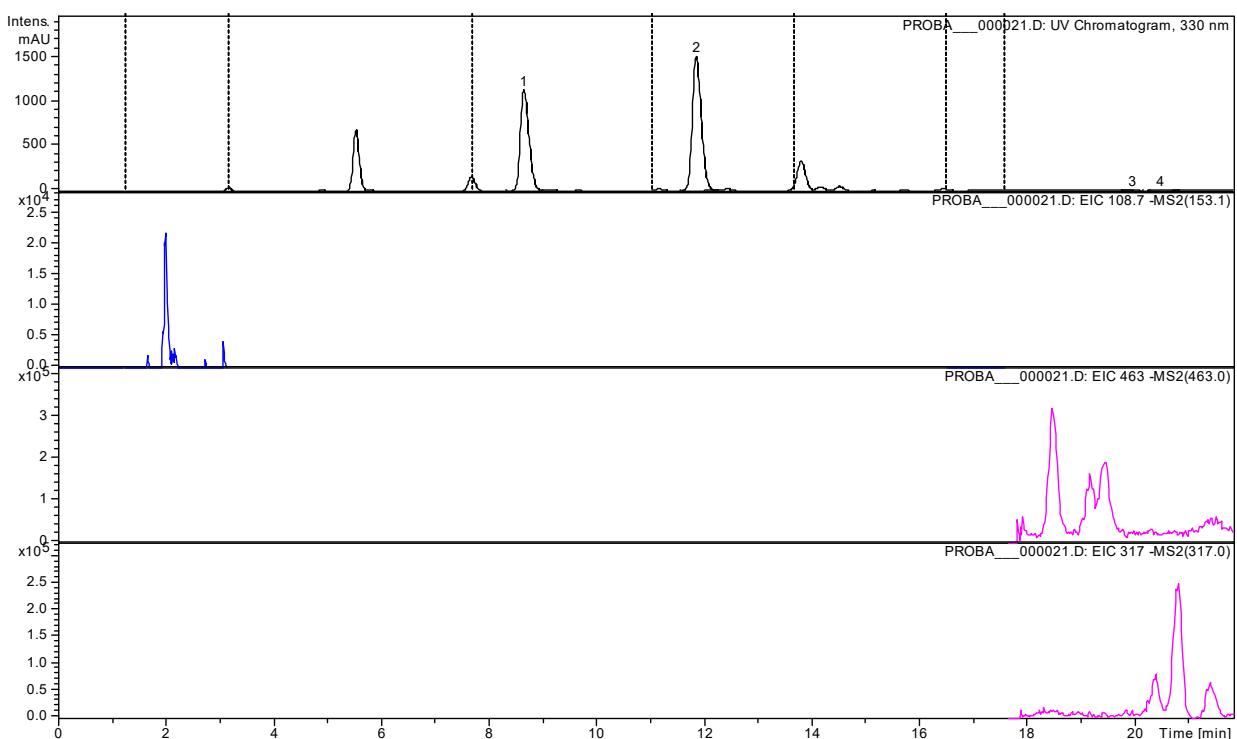


Figure S1. UV and MS chromatograms of gentisic acid, myricetol and hyperoside

HPLC-UV-MS analysis of oily propolis extract and honey sample

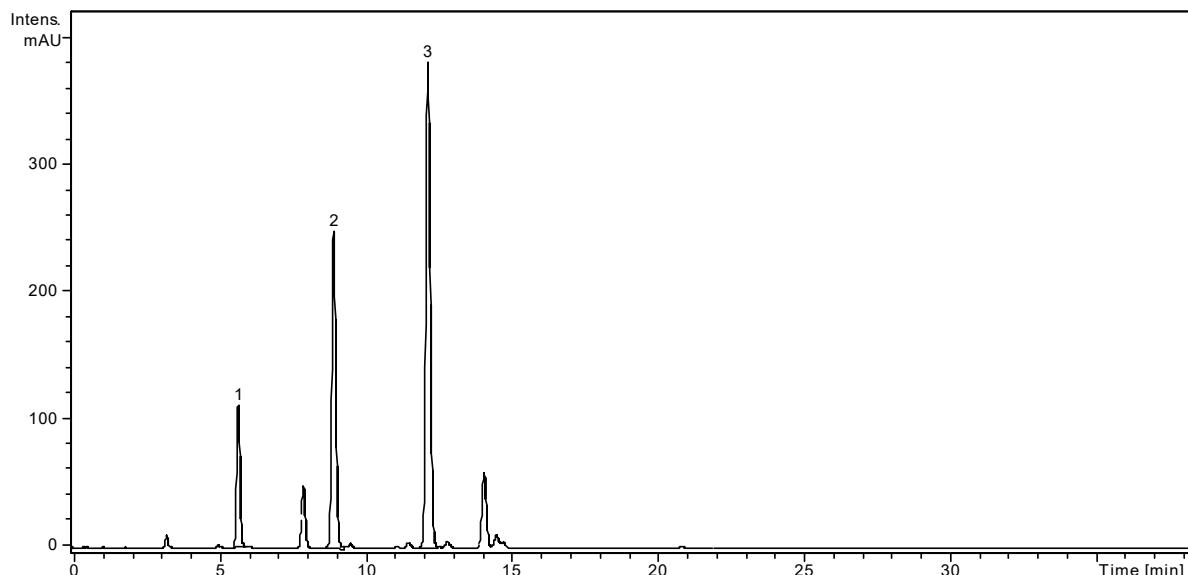


Figure S2. HPLC chromatogram of oily propolis extract (1-caffeic acid, 2-p-coumaric acid, 3-ferulic acid)

Table S1. Polyphenolic compounds identified in the oily propolis extract

No. on chromatogram	Compound	UV Identified	MS qualitatively identified	Concentration ($\mu\text{g}/\text{ml}$)
	Gentisic acid	NO	YES	qualitatively
1	Caffeic acid	YES	YES	19.129 ± 0.95
2	p-coumaric acid	YES	YES	71.661 ± 2.86
3	Ferulic acid	YES	YES	99.600 ± 2.98

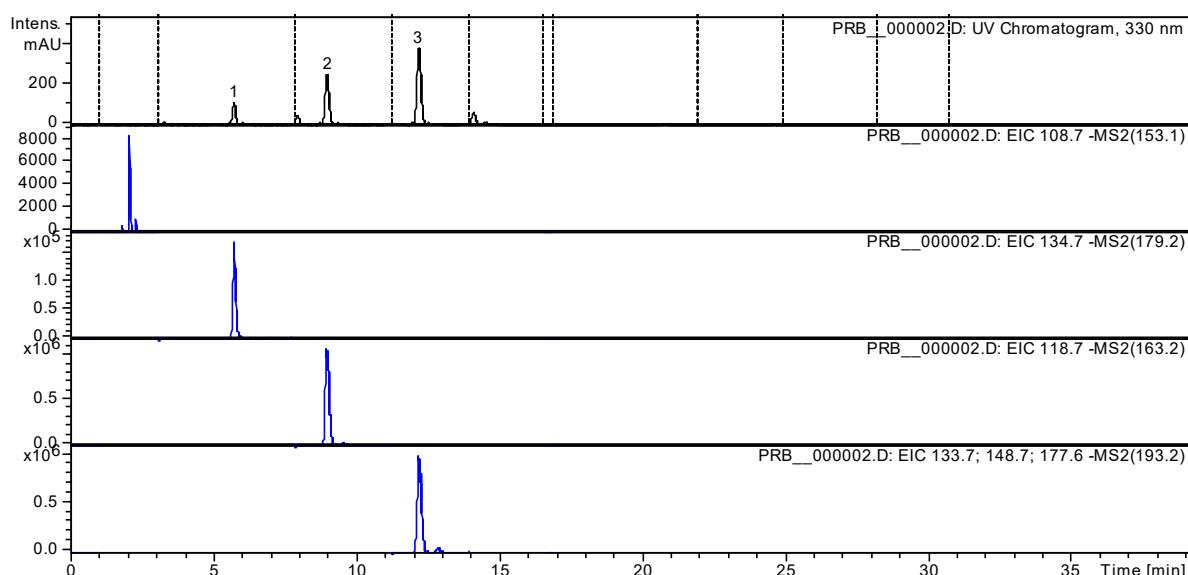


Figure S3. UV and MS chromatograms of identified compounds in the oily propolis extract

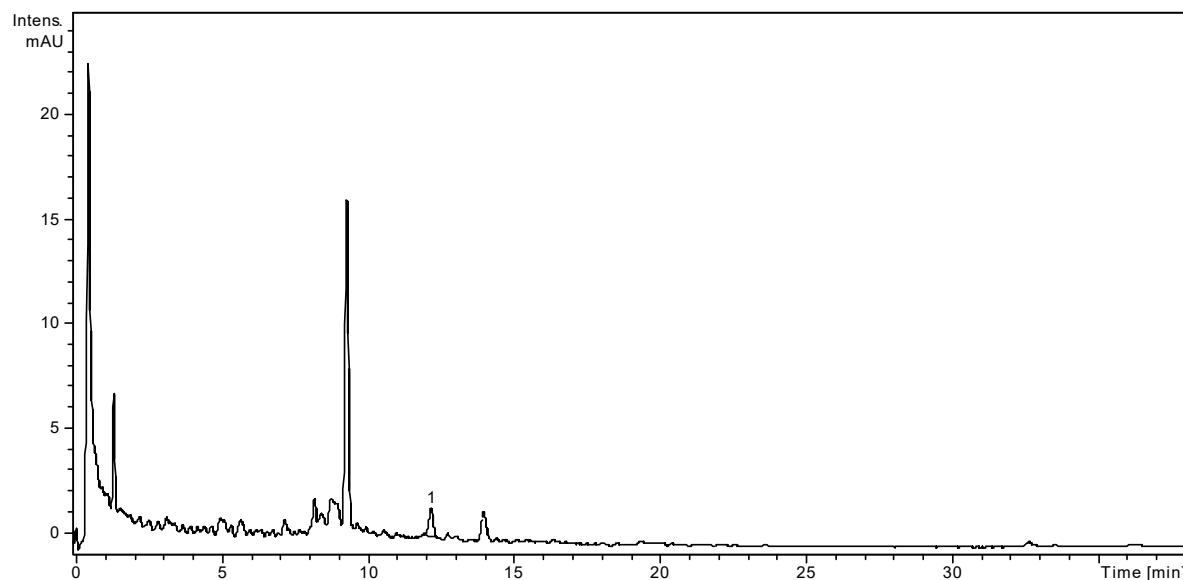


Figure S4. HPLC chromatogram of honey sample (1-gentisic acid, qualitatively identified)

Table S2. Organoleptic analysis and physico-chemical determination of honey – Bulletin no. 945/12.12.2016 (ICDA Bucharest - Research and Development Institute for Beekeeping)

Organoleptic properties of honey	Methods of analysis	Results
appearance	SR 784 / 3-2009	clean, homogeneous, without impurities, without foam
consistency	SR 784 / 3-2009	viscous mass, crystallized with coarse crystals
color	SR 784 / 3-2009	yellow
taste and odor	SR 784 / 3-2009	sweet taste, with a pleasant floral aroma

Physico-chemical determinations of honey	Methods of analysis	Results
free acidity	LCPA // DFC-PS-02 SR 784 / 3-2009 point 4.3	2.20 ± 0.351 mL NaOH 1N/100 g
reducing sugar expressed as invert sugar	LCPA // DFC-PS-06 // Elser SR 784 / 3-2009 pt.4.4	74.50 ± 0.7806 %, g/g

easily hydrolysable sugar expressed as sucrose	LCPA // DFC-PS-07 // Elser SR 784 / 3-2009 point 4.5	$1.90 \pm 0.1522 \text{ %, g/g}$
hydroxymethylfurfural	LCPA // DFC-PS-03 // Winkler SR 784 / 3-2009 point 4.9	$1.44 \pm 0.1713 \text{ mg/100 g}$
diastase index	LCPA // DFC-PS-05 // Gothe SR 784 / 3-2009 pt.4.7	$29.40 \pm 0.3596 \text{ Gothe units}$
ash	LCPA // DFC-PS-09	$0.1022 \pm 0.008 \text{ %, g/g}$
water	LCPA // DFC-PS-16 // refractometric method SR 784 / 3-2009 pt.4.1	$14.80 \pm 0.2828 \text{ %, g/g}$

Table S3. Physico-chemical determination of apilarnil – Bulletin no. 947/12.12.2016 (ICDA Bucharest - Research and Development Institute for Beekeeping)

Physico-chemical determinations of apilarnil	Methods of analysis	Results
total proteins	LCPA // DFC-PS-10	$7.5551 \pm 0.5130 \text{ %, g/g}$
total fats	LCPA // DFC-PS-07 //Soxhlet	$6.3916 \pm 0.2788 \text{ %, g/g}$
ash	LCPA // DFC-PS-09	$0.7508 \pm 0.008 \text{ %, g/g}$
water	LCPA//DFC-PS-12 //oven drying method	$71.4706 \pm 0.8388 \text{ %, g/g}$

Determination of vitamin C in honey sample was performed using a titrimetric method according to Association of Official Analytical Chemists. Vitamin C in Juices and Vitamin Preparations. Official Method 967.21. In AOAC Official Methods of Analysis, 18th; Association of Official Analytical Chemists: Gaithersburg, MD, USA, 2005; pp 45.1.14.

Result: $1.54 \pm 0.08 \text{ mg/100 g}$ honey