## In Situ Oxidation of $\mathrm{Cu}_{2} \mathrm{O}$ Crystal for Electrochemical Detection of Glucose

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Figure S1. Comparison of different glucose sensors performance.


Figure S2. The SEM images of 5M-0.1-300, 5M-0.1-1200, 3M-0.1-300, 5M-0.1-1200.


Figure S3. The Raman spectrum of the $\mathrm{Cu}_{2} \mathrm{O}$ electrodes fabricated under different conditions.
The Raman spectra of $3 \mathrm{M}-0.1,3 \mathrm{M}-0.3,5 \mathrm{M}-0.1,5 \mathrm{M}-0.3$ and $5 \mathrm{M}-0.5$ electrodes exhibit characteristic peaks at $108.8 \mathrm{~cm}^{-1}, 148.7 \mathrm{~cm}^{-1}, 218.3 \mathrm{~cm}^{-1}, 520 \mathrm{~cm}^{-1}$ and $630 \mathrm{~cm}^{-1}$, which are attributed to $\mathrm{Cu} \mathrm{C}_{2} \mathrm{O}$ [1,2]. The Raman spectra of $3 \mathrm{M}-0.5,3 \mathrm{M}-0.3$ and $1 \mathrm{M}-0.1$ exhibit the characteristic peaks of $\mathrm{Cu}(\mathrm{OH})_{2}$ at $290 \mathrm{~cm}^{-1}$ and $490 \mathrm{~cm}^{-1}$, according to previous researches [2-4]. It should be noticed that the Raman spectrum of 3M-0.3 exhibits the both characteristic peaks of $\mathrm{Cu}_{2} \mathrm{O}$ and $\mathrm{Cu}(\mathrm{OH})_{2}$. This agrees with the SEM in Figure S1 that shows the existence of nanoparticles and nanotubes.

## References

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