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Synthesis of 3,5-di-(4-pyridyl)-1*H*-1,2,4-triazole

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Introduction

4-Amino-3,5-di-(4-pyridyl)-4*H*-1,2,4-triazole (1) and 3,5-di-(4-pyridyl)-1*H*-1,2,4-triazole (2) are potential multi-topic bridging ligands in preparation of coordination compounds of 1 – 3D structures.[1] In an early literature, a method to prepare compound 1 has been reported [2], which involves the reaction of isonicotinonitrile with excess amount of hydrazine in ethylene glycol at 130°C. Polymeric copper complexes of 2 have been prepared by Chen *et al.*; 2 was generated by an *in situ* cycloaddition of 4-pyridylnitrile and ammonia in presence of Cu²⁺ under hydrothermal condition.[3] However, no synthetic procedures about the compound 2 – the deaminated product of 1 has appeared in the literature to the best of our knowledge. In this paper, we describe a facile method to prepare compound 2 starting from isonicotinic acid and hydrazine.

Synthesis

Isonicotinic acid (2.46g,0.02mol) and 2ml of 80% hydrazine were mixed in a 15 ml Teflon-lined reactor and heated for 48 h at 186°C [4]. The resulting liquid was diluted by water, and then the pH was adjusted from 9 to 6 by addition of HCl solution. White precipitate was obtained, and recrystallised from ethanol. The product is actually a mixture of 4-amino-3,5-di-(4-pyridyl)-4*H*-1,2,4-triazole (1) and 3,5-di-(4-pyridyl)-1*H*-1,2,4-triazole (2) in *ca* 1:7 molar ratio, estimated from the HNMR spectrum.

Pure sample of **2** was prepared by a successive deamination process [5]. The mixture of **1** and **2** (2.38 g, *ca* 10 mmol) was dissolved in aqueous HCl (6M, 3 mL). To it was added slowly an aqueous solution of NaNO₂ (0.76 g, 11 mmol); a lot of bubbles were generated quickly. The mixture was stirred for 5~6 h, then the pH was adjusted from 4 to 6.5 by 10% NaOH solution. White precipitate appeared and it was filtered and dried in air. Recrystallisation of the crude product from ethanol afforded colorless needles of **2**. The overall yield of this method is 9%.

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OH + NH₂NH₂
$$\xrightarrow{186^{\circ}C}$$
 NH₂NH₂ $\xrightarrow{N-N}$ + NH₂NH₂ $\xrightarrow{N-N}$ (1)

Melting point: sublimated at 242°C

IR (KBr pellet, cm⁻¹): 1607 s, 1448 m, 1144 w, 983 m, 838 w, 724 m, 513 w.

¹H-NMR (400 MHz, DMSO-d₆): δ = 8.78 (4H, d, py-H), 8.02 (4H, d, py-H).

Elemental Analysis: Calculated for C₁₂H₉N₅·0.5H₂O: C 57.59, H 4.83, N 27.98%; Found: C 57.28, H 4.70, N 27.90%.

MS (m/z): 225.2 (15%), 224.2 (100%, M+H), 218.3 (5%).

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