

## SUPPLEMENTARY MATERIALS

# Fabricating Polymer/Surfactant/Cyclodextrin Hybrid Particles for Possible Nose-To-Brain Delivery of Ropinirole Hydrochloride: In Vitro and Ex Vivo Evaluation

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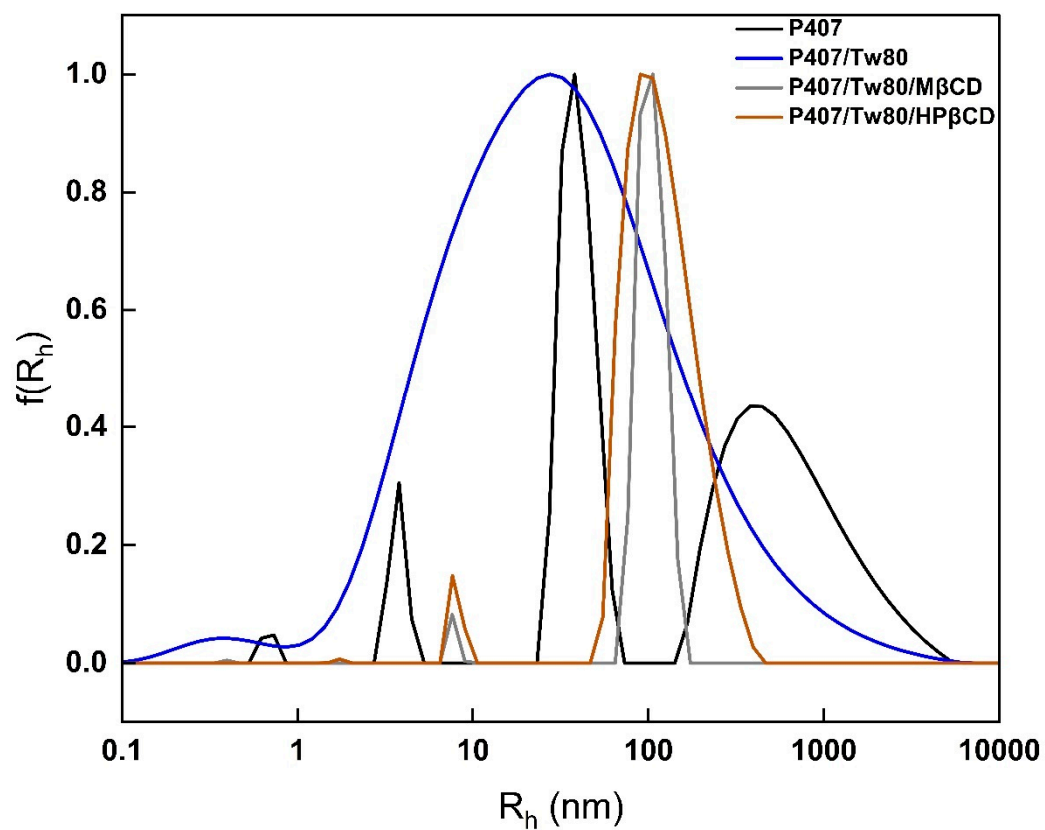
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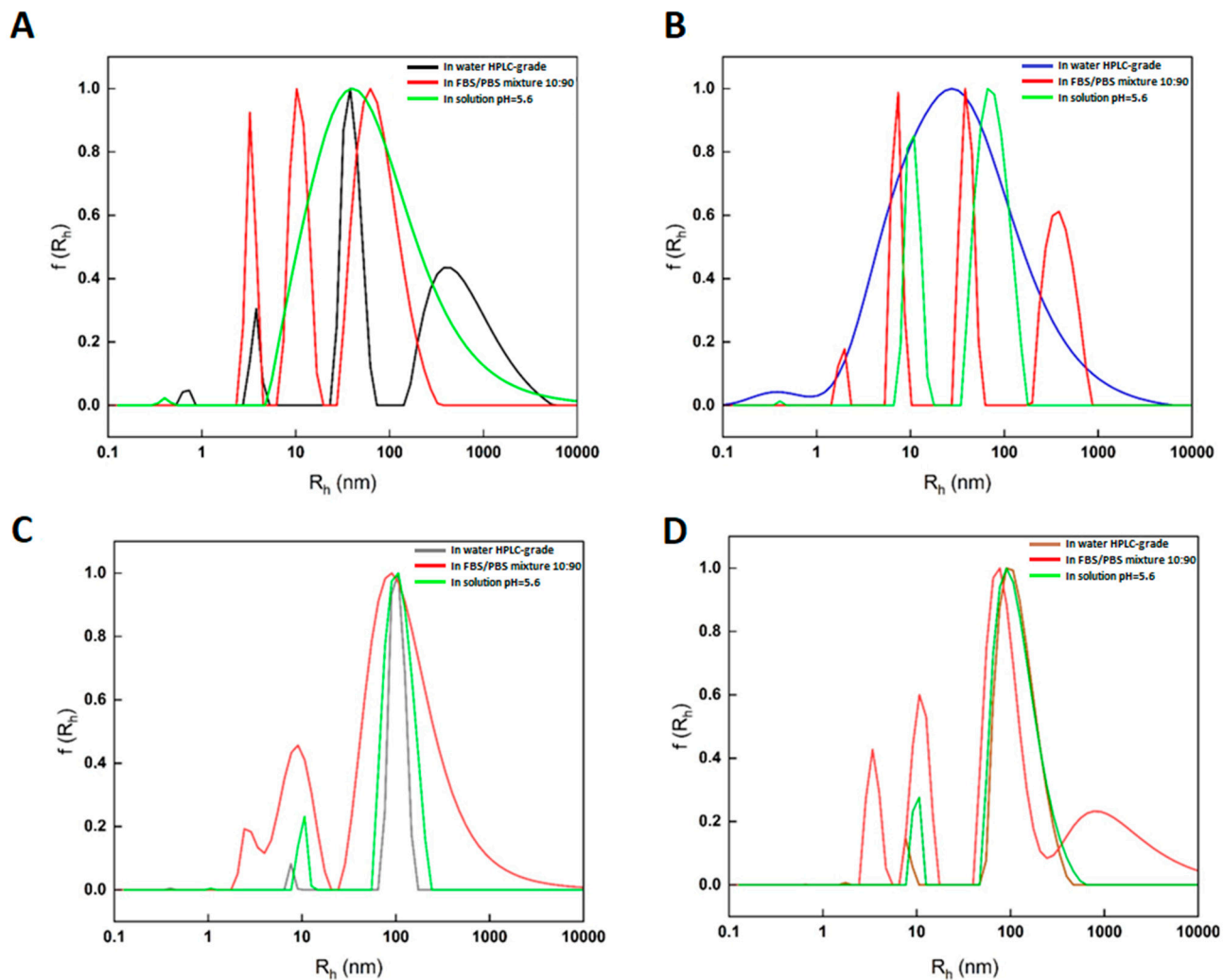
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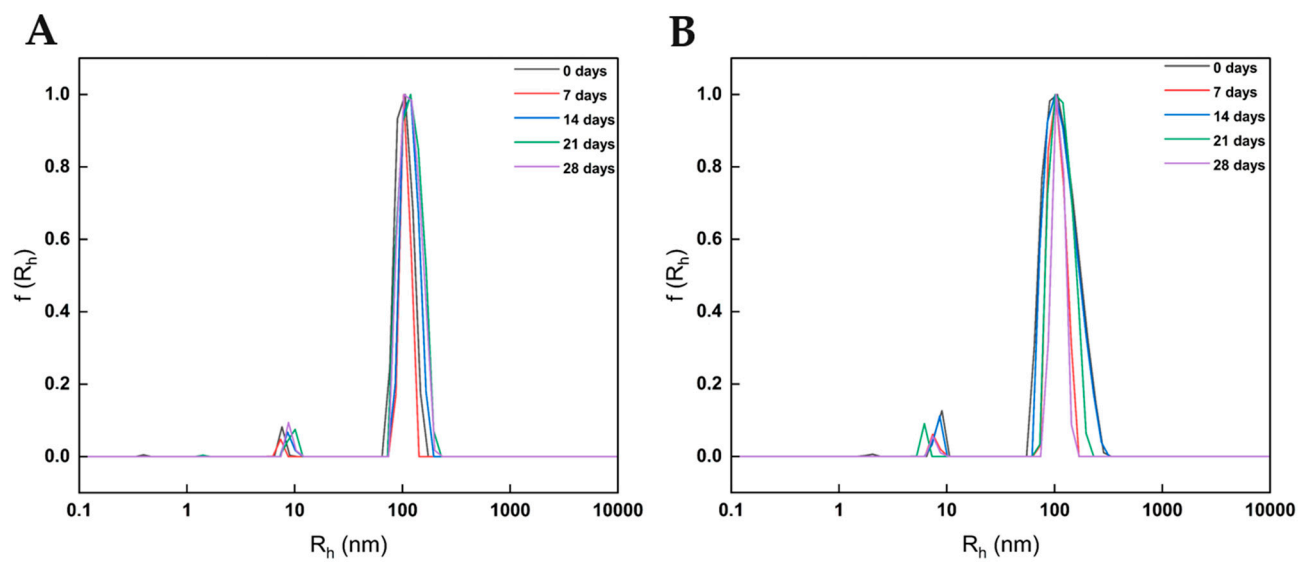
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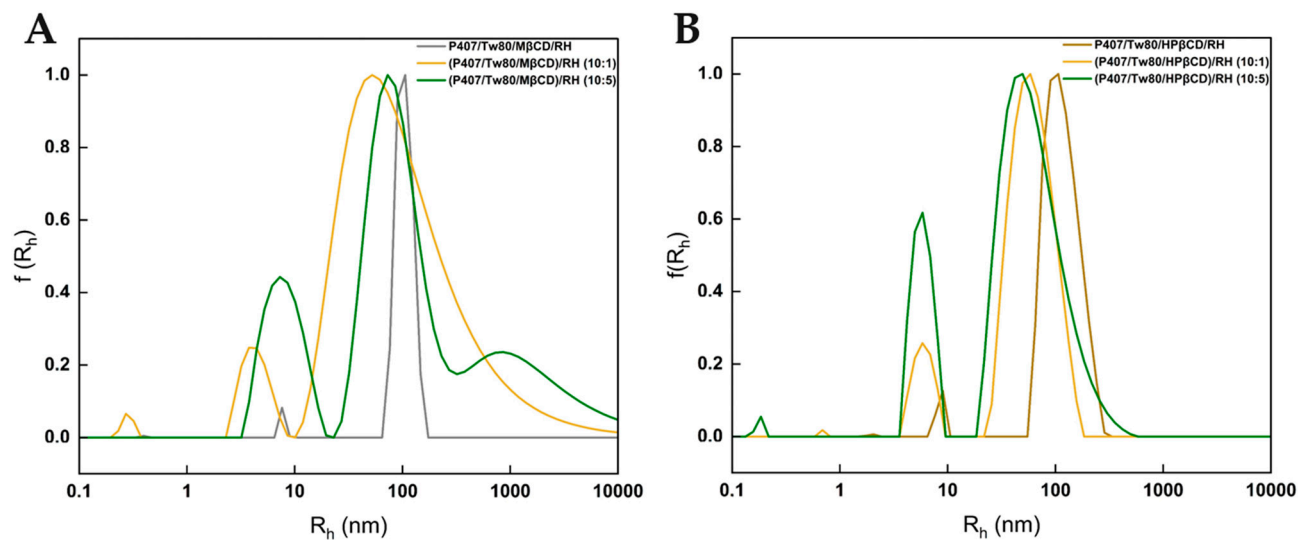
**Figure S1.** Size distributions for P407, P407/Tw80, P407/Tw80/M $\beta$ CD, and P407/Tw80/HP $\beta$ CD systems in aqueous dispersions prepared using the thin-film hydration method (t=0 days).



**Figure S2.** Size distributions for (A) P407, (B) P407/Tw80, (C) P407/Tw80/M $\beta$ CD and (D) P407/Tw80/HP $\beta$ CD systems in different dispersion media prepared using the thin-film hydration method ( $t=0$  days).



**Figure S3.** Stability assessment of (A) P407/Tw80/M $\beta$ CD and (B) P407/Tw80/HP $\beta$ CD hybrid systems.



**Figure S4.** Size distribution for the (A) P407/Tw80/M $\beta$ CD system, (P407/Tw80/M $\beta$ CD)/RH in weight ratios of 10:1 and 10:5 and (B) P407/Tw80/HP $\beta$ CD system, (P407/Tw80/HP $\beta$ CD)/RH in weight ratios of 10:1 and 10:5 in aqueous dispersions prepared using the thin-film hydration method ( $t=0$  days).

**Table S1.** Calorimetric heating profiles of pure compounds and their mixtures at the solid state. These calorimetric parameters correspond to the thermograms in Figure 2.

Sample	T <sub>onset,m</sub> (°C) <sup>a</sup>	T <sub>m</sub> (°C) <sup>b</sup>	ΔT <sub>1/2,m</sub> (°C) <sup>c</sup>	ΔH <sub>m</sub> (KJ mol <sup>-1</sup> ) <sup>d</sup>	T <sub>onset,s</sub> (°C)	T <sub>s</sub> (°C)	ΔT <sub>1/2,s</sub> (°C)	ΔH <sub>s</sub> (KJ mol <sup>-1</sup> )	T <sub>onset,t</sub> (°C)	T <sub>t</sub> (°C)	ΔT <sub>1/2,t</sub> (°C)	ΔH <sub>t</sub> (KJ mol <sup>-1</sup> )
P407	53.06	57.17	4.25	-46.38	151.56	157.83	6.53	7.83	–	–	–	–
Tw80	25.43	59.50	39.05	-2.07	–	–	–	–	–	–	–	–
MβCD	145.13	179.50	17.63	-1.29	68.20	75.33	6.91	-0.02	–	–	–	–
HPβCD	142.17	147.33	20.30	-5.70	–	–	–	–	–	–	–	–
Tw80/MβCD	123.78	147.50	27.10	0.50	184.83	187.00	6.65	-1.07	–	–	–	–
Tw80/HPβCD	121.29	132.33	16.19	-0.74	–	–	–	–	–	–	–	–
P407/Tw80	45.28	52.83	6.11	-31.08	–	–	–	–	–	–	–	–
P407/MβCD	46.19	52.83	5.01	-33.54	145.37	181.33	98.99	109.92	–	–	–	–
P407/HPβCD	47.63	52.83	4.22	-36.08	155.03	162.33	7.92	9.85	–	–	–	–
P407/Tw80/MβCD	46.14	53.00	7.24	-38.79	140.27	153.33	12.90	9.98	207.45	210.00	7.33	-3.42
P407/Tw80/HPβCD	50.20	55.17	5.92	-32.29	–	–	–	–	–	–	–	–

<sup>a</sup>T<sub>onset</sub>: the temperature at which the thermal event starts.

<sup>b</sup>T: the temperature at which heat capacity (ΔC<sub>p</sub>) at constant pressure is maximum.

<sup>c</sup>ΔT<sub>1/2</sub>: half width at the half peak height of the transition.

<sup>d</sup>ΔH: transition enthalpy normalized per mol of each system. m: main transition; s: secondary; t: ternary transition.

**Table S2.** The physicochemical characteristics of hybrid systems in A. FBS/PBS mixture (10:90) and B. buffer solution (pH=5.6 at the temperature of 34 °C).

<b>A. Dispersed in FBS/PBS mixture</b>						
<b>Colloidal dispersions</b>	<b>w/w</b>	<b>R<sub>h</sub> (Cumulant) (nm)<sup>1</sup></b>	<b>PDI<sup>2</sup></b>	<b>Number of peaks</b>	<b>R<sub>h</sub> (Contin) (nm)<sup>3</sup></b>	<b>Weight of Peak (%)</b>
P407	-	47	0.4 <sub>8</sub>	3	1) 3 2) 11 3) 75	1) 13% 2) 27% 3) 60%
P407/Tw80	70:30	38	0.5 <sub>0</sub>	1	1) 2 2) 7 3) 40 4) 389	1) 4% 2) 24% 3) 31% 4) 41%
(P407/Tw80)/MβCD	80:20	95	0.4 <sub>9</sub>	2	1) 8 2) 104	1) 3% 2) 97%
(P407/Tw80)/HPβCD	80:20	83	0.4 <sub>9</sub>	2	1) 9 2) 114	1) 3% 2) 97%
<b>B. Dispersed in buffer solution pH=5.6 at 34 °C</b>						
<b>Colloidal dispersions</b>	<b>w/w</b>	<b>R<sub>h</sub> (Cumulant) (nm)<sup>1</sup></b>	<b>PDI<sup>2</sup></b>	<b>Number of peaks</b>	<b>R<sub>h</sub> (Contin) (nm)<sup>3</sup></b>	<b>Weight of Peak (%)</b>
P407	-	42	0.5 <sub>0</sub>	3	62	100%
P407/Tw80	70:30	39	0.5 <sub>0</sub>	1	1) 11 2) 96	1) 27% 2) 73%
(P407/Tw80)/MβCD	80:20	90	0.4 <sub>1</sub>	2	1) 10 2) 108	1) 6% 2) 94%
(P407/Tw80)/HPβCD	80:20	86	0.4 <sub>2</sub>	2	1) 10 2) 118	1) 7% 2) 94%

<sup>1</sup> R<sub>h</sub> indicates the average hydrodynamic radius of three replicates of each sample obtained by the Cumulant method

<sup>2</sup> PDI indicates the average polydispersity index, and the first decimal number is the significant one

<sup>3</sup> R<sub>h</sub> indicates the average hydrodynamic radius of three replicates of each sample obtained by the Contin method

**Table S3.** Calorimetric parameters of pure RH, ternary systems and (P407/Tw80/CD)/RH at different weight ratios (10:0.1; 10:0.5; 10:1; 10:5; 10:10) at the solid state, using M $\beta$ CD or HP $\beta$ CD. These calorimetric parameters correspond to the thermograms of Figure 6.

Sample	Weight ratio	T <sub>onset,m</sub> (°C) <sup>a</sup>	T <sub>m</sub> (°C) <sup>b</sup>	$\Delta T_{1/2,m}$ (°C) <sup>c</sup>	$\Delta H_m$ (KJmol <sup>-1</sup> ) <sup>d</sup>	T <sub>onset,s</sub> (°C)	T <sub>s</sub> (°C)	$\Delta T_{1/2,s}$ (°C)	$\Delta H_s$ (KJmol <sup>-1</sup> )	T <sub>onset,t</sub> (°C)	T <sub>t</sub> (°C)	$\Delta T_{1/2,t}$ (°C)	$\Delta H_t$ (KJmol <sup>-1</sup> )
P407/Tw80/M $\beta$ CD	-	46.14	53.00	7.24	-18.62	140.27	153.33	12.90	4.79	207.45	210.00	7.33	-1.64
P407/Tw80/HP $\beta$ CD	-	50.20	55.17	5.92	-32.29	-	-	-	-	-	-	-	-
RH	-	244.62	248.50	3.59	-22.69	-	-	-	-	-	-	-	-
(P407/Tw80/M $\beta$ CD)/RH	10:0.1	42.51	49.50	4.94	-11.28	208.38	220.83	50.62	7.49	-	-	-	-
(P407/Tw80/M $\beta$ CD)/RH	10:0.5	41.13	48.50	6.67	-11.41	204.05	205.33	1.97	-1.42	-	-	-	-
(P407/Tw80/M $\beta$ CD)/RH	10:1	44.15	50.17	5.39	-11.38	237.33	234.42	2.28	-1.48	-	-	-	-
(P407/Tw80/M $\beta$ CD)/RH	10:5	42.49	49.83	7.69	-8.48	234.33	237.33	3.52	-7.49	-	-	-	-
(P407/Tw80/M $\beta$ CD)/RH	10:10	42.09	48.50	6.37	-6.98	224.21	230.83	5.74	-9.12	240.52	241.17	1.32	-3.17
(P407/Tw80/HP $\beta$ CD)/RH	10:0.1	46.76	50.00	4.02	-13.07	123.28	141.33	18.15	1.84	166.33	169.50	10.36	-4.24
(P407/Tw80/HP $\beta$ CD)/RH	10:0.5	43.56	51.83	8.75	-12.00	-	-	-	-	-	-	-	-
(P407/Tw80/HP $\beta$ CD)/RH	10:1	40.89	49.50	7.27	-11.40	117.94	143.83	29.49	4.33	234.72	237.50	5.27	-0.82
(P407/Tw80/HP $\beta$ CD)/RH	10:5	40.15	48.17	6.31	-8.37	120.98	144.67	17.18	2.22	206.48	217.33	9.73	-6.31
(P407/Tw80/HP $\beta$ CD)/RH	10:10	41.44	49.00	6.01	-6.12	214.00	225.67	14.20	-10.49	-	-	-	-

<sup>a</sup>T<sub>onset</sub>: the temperature at which the thermal event starts.

<sup>b</sup>T: the temperature at which heat capacity ( $\Delta C_p$ ) at constant pressure is maximum.

<sup>c</sup> $\Delta T_{1/2}$ : half width at the half peak height of the transition.

<sup>d</sup> $\Delta H$ : transition enthalpy normalized per mol of each system. m: main transition; s: secondary; t: trinary transition.



**Table S4.** The flux across the cellulose membrane ( $J_{CM}$ ) (mean  $\pm$  SD, n = 3), the flux ( $J_{NM}$ ) (mean  $\pm$  SEM, n = 4) and the apparent permeability ( $P_{app}$ ) across the nasal mucosa barrier of formulations F1-F4 and RH solution (0.5 mg/mL, PBS pH = 5.6). R-square of regression analysis of the amount of the drug permeated per unit area vs time, across the cellulose membrane and the nasal mucosa barrier are included in the table [ $R^2_{(CM)}$  and  $R^2_{(NM)}$ , respectively].

Formulation (F)	$J_{CM}$ ( $\mu\text{g}/\text{cm}^2/\text{min}$ ) $\pm$ SD	$R^2_{(CM)}$	$J_{NM}$ ( $\mu\text{g}/\text{cm}^2/\text{min}$ ) $\pm$ SEM	$R^2_{(NM)}$	$P_{app}$ (cm/min)
F1	$4.9 \times 10^{-4} \pm 5.1 \times 10^{-5}$	$0.9492 \pm 0.0053$	$2.0 \times 10^{-4} \pm 1.0 \times 10^{-5}$	$0.9645 \pm 0.0017$	0.40
F2	$5.9 \times 10^{-4} \pm 7.5 \times 10^{-5}$	$0.9256 \pm 0.0077$	$2.0 \times 10^{-4} \pm 1.0 \times 10^{-5}$	$0.9911 \pm 0.0008$	0.40
F3	$5.5 \times 10^{-4} \pm 5.2 \times 10^{-5}$	$0.9572 \pm 0.0054$	$1.7 \times 10^{-4} \pm 1.0 \times 10^{-5}$	$0.9825 \pm 0.0010$	0.35
F4	$5.9 \times 10^{-4} \pm 8.9 \times 10^{-5}$	$0.8978 \pm 0.0092$	$1.9 \times 10^{-4} \pm 1.0 \times 10^{-5}$	$0.9950 \pm 0.0006$	0.39
RH solution	$6.0 \times 10^{-4} \pm 8.2 \times 10^{-5}$	$0.9157 \pm 0.0085$	$1.4 \times 10^{-4} \pm 1.0 \times 10^{-5}$	$0.9769 \pm 0.0009$	0.28