

# Photocatalytic Properties of ZnO: Al/MAPbI<sub>3</sub>/Fe<sub>2</sub>O<sub>3</sub> Heterostructure: First-Principles Calculations

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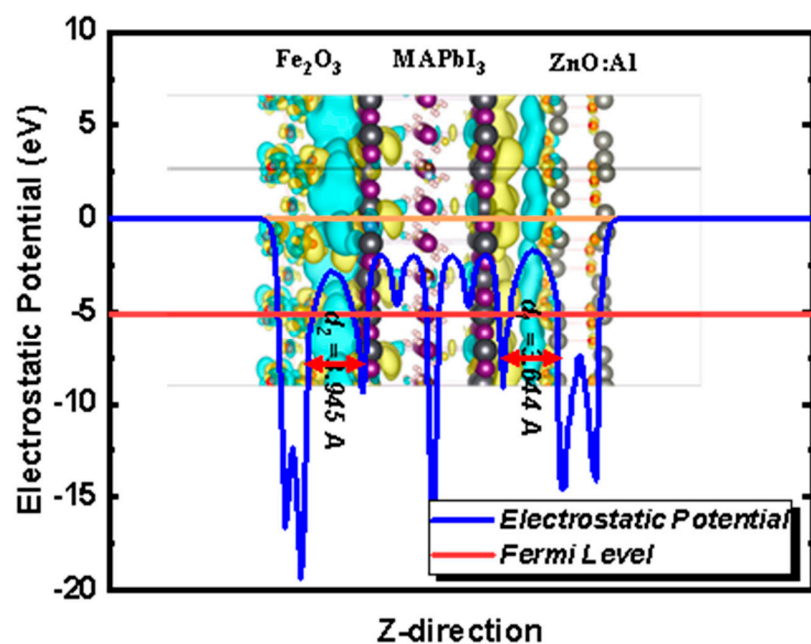


Figure S1: The work function of  $\text{Fe}_2\text{O}_3/\text{MAPbI}_3/\text{ZnO:Al}$  heterostructure. The red and yellow lines denote the Fermi level  $E_F$ , and the vacuum energy level  $E_{\text{vac}}$ , respectively and planar-averaged electron density difference in blue.

Equations used to compute different physical properties given in the manuscript

$$p_{s_2} = p_{v_2} \times e^{\frac{qV_{bi2}}{k_B T}}$$

$$p_{v_2} = N_{c2} \times e^{\frac{E_{F02} - E_{C02}}{k_B T}}$$

$$V_{bi_{12}} = E_{F02} - E_{F01}$$

$$n_{s_2} = n_{v_2} \times e^{\frac{-qV_{bi2}}{k_B T}}$$

$$n_{v_2} = N_{c2} \times e^{\frac{E_{F02} - E_{C02}}{k_B T}}$$

$$V_{bi_{12}} = E_{F02} - E_{F01}$$

**Equations used to compute for H<sub>2</sub> production yield by rate of generation of excess carriers (unit volume/time):**

$$\Delta i_{ph} = V \Delta G$$

$$\Delta G = \frac{A}{w} \Delta \sigma$$

$$\Delta \sigma = q(\mu_n + \mu_p) \Delta n$$

$$\Delta n = \tau_n \Delta G_L$$

Generation rate due to light:

$$\Delta G_L = \frac{\eta \Delta P}{h\nu A w}$$

$$\Delta i_{ph} = V \frac{A \tau_n}{w} q(\mu_n + \mu_p) \frac{\eta \Delta P}{h\nu A w}$$

$$\Delta i_{ph} = \frac{\Delta Q}{\Delta t} = \frac{\eta \Delta P}{h\nu} q \left( \frac{\tau_n}{t} \right)$$

$$n_e = \frac{\Delta Q}{q} = \frac{\eta \Delta P}{h\nu} \left( \frac{\tau_n}{t} \right) \Delta t$$

$$\Delta n = \Delta p$$

$$n_0 = N_C e^{\frac{-(E_C - E_F)}{k_B T}}$$

$$\Delta n = n_0 e^{\frac{-qV_{bi}}{k_B T}}$$

$$N_C = 2 \left( \frac{2\pi m_e^* k_B T}{h^2} \right)^{\frac{3}{2}}$$

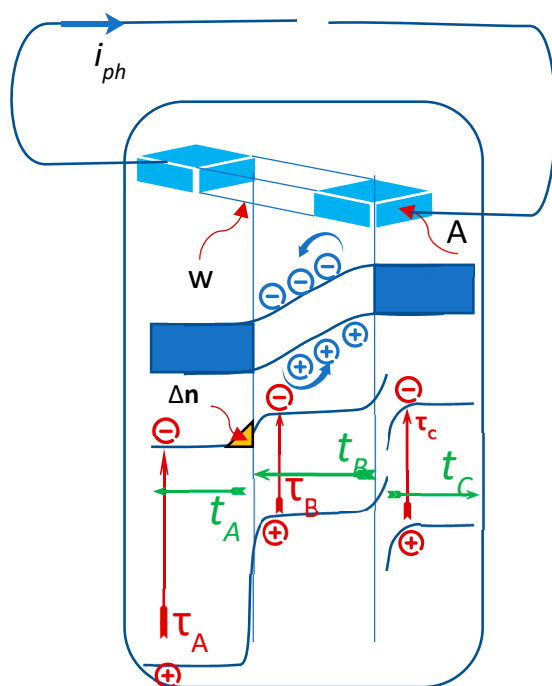


Figure S2:  $\text{Fe}_2\text{O}_3/\text{MAPbI}_3/\text{ZnO:Al}$  heterostructure schematic and photocatalytic activity measurements set up