

Supplementary Information

Article

In Situ Construction of Nitrogen-Doped and Zinc-Confined Microporous Carbon Enabling Efficient Na⁺-Storage Abilities

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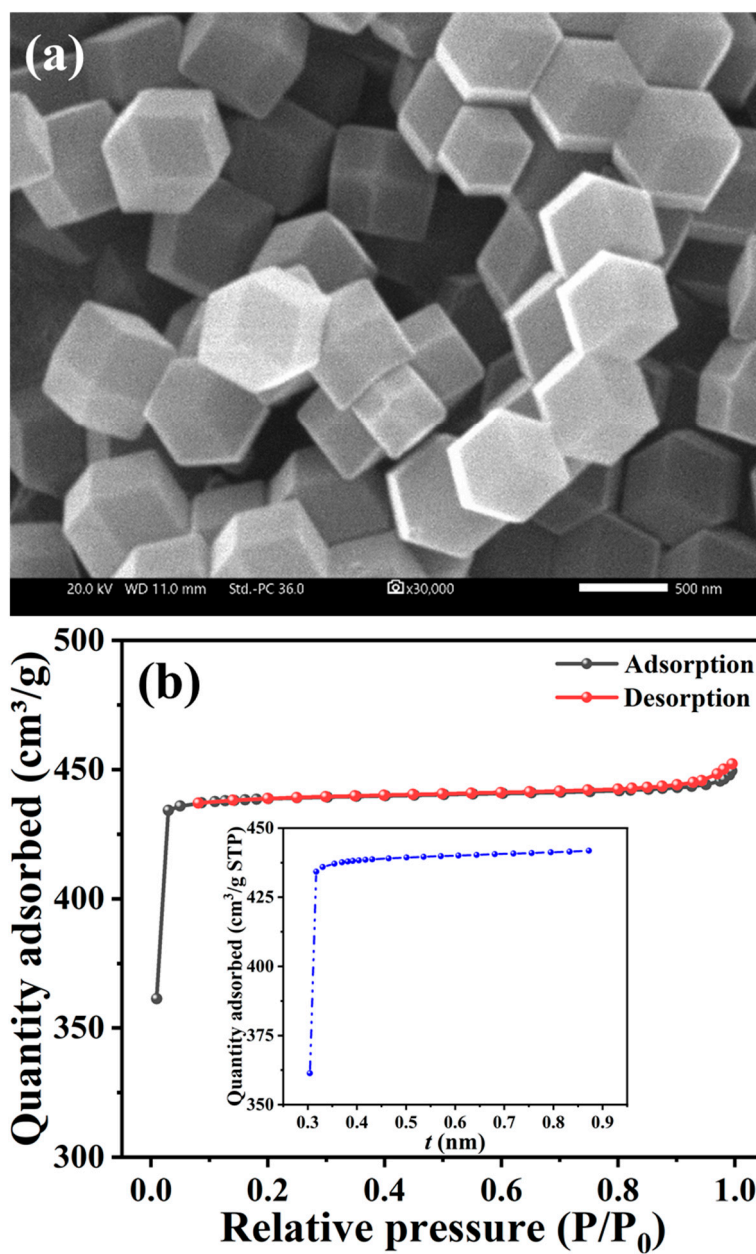


Figure S1. (a) SEM micrograph and (b) nitrogen adsorption-desorption isotherm of ZIF-8. Scale bar of (a) is 500 nm, whereas the inset of (b) shows the pore-size distribution analyzed by a *t*-plot method.

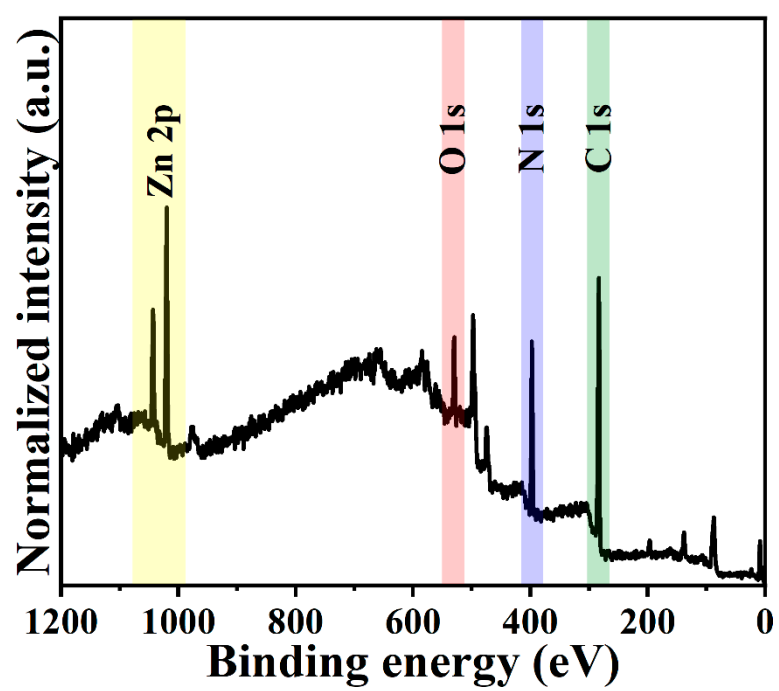


Figure S2. Full-range XPS spectrum of ZIF-8.

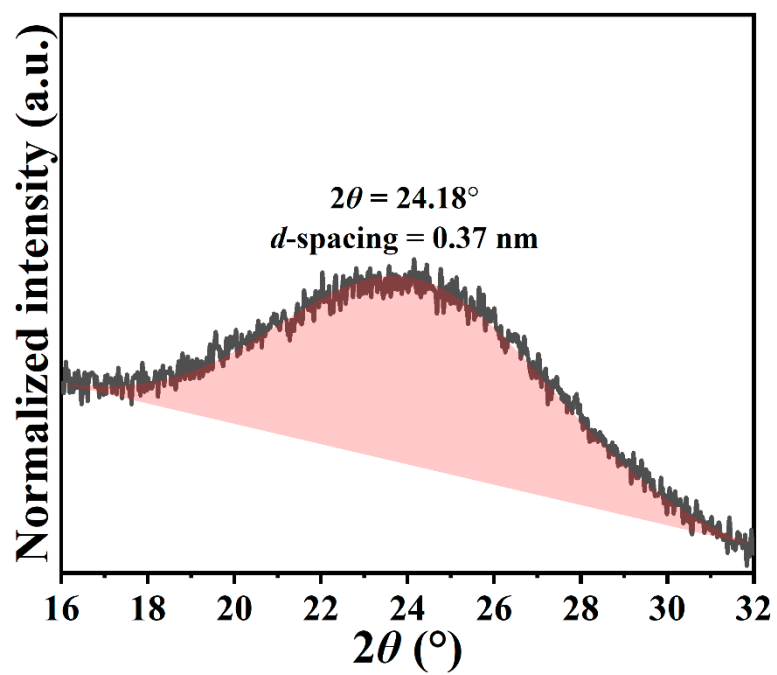


Figure S3. The magnified XRD pattern of (002) peak for *N,Z*-MPC analyzing by Gaussian-Lorentzian method.

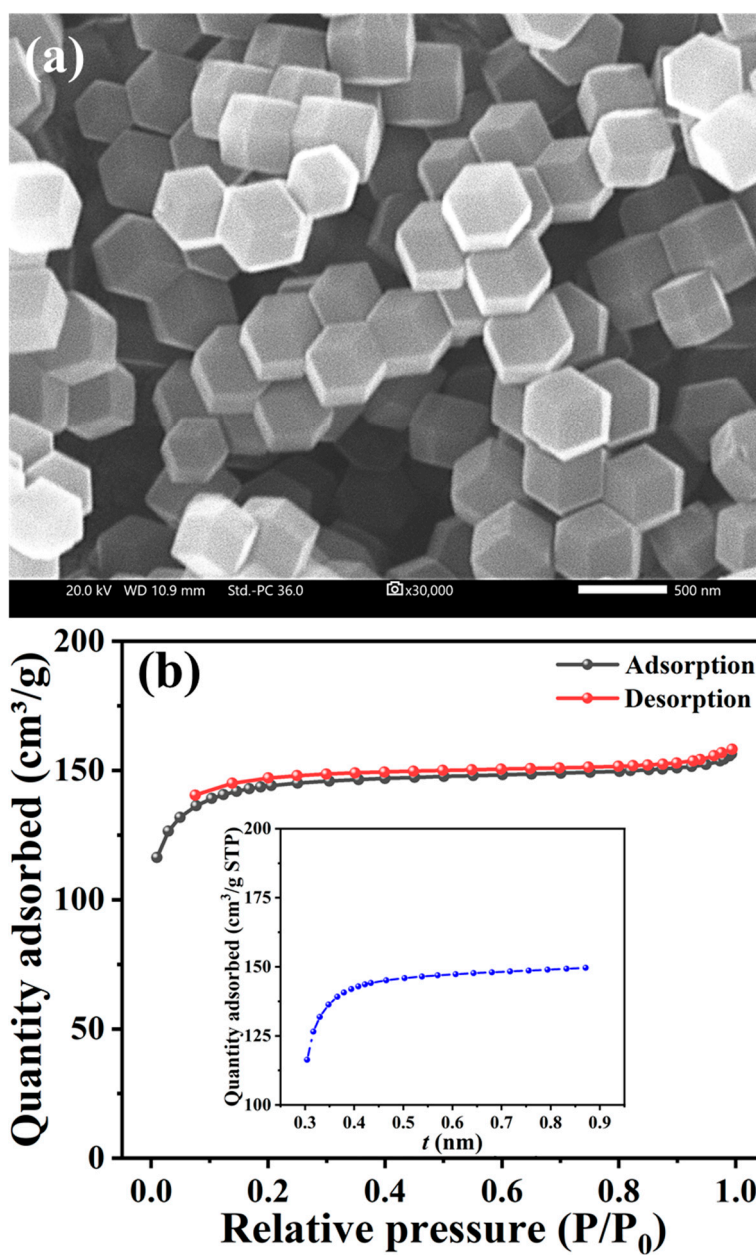


Figure S4. (a) SEM micrograph and (b) nitrogen adsorption-desorption isotherm of *N,Z*-MPC. Scale bar of (a) is 500 nm, whereas the inset of (b) shows the pore-size distribution analyzed by a *t*-plot method.

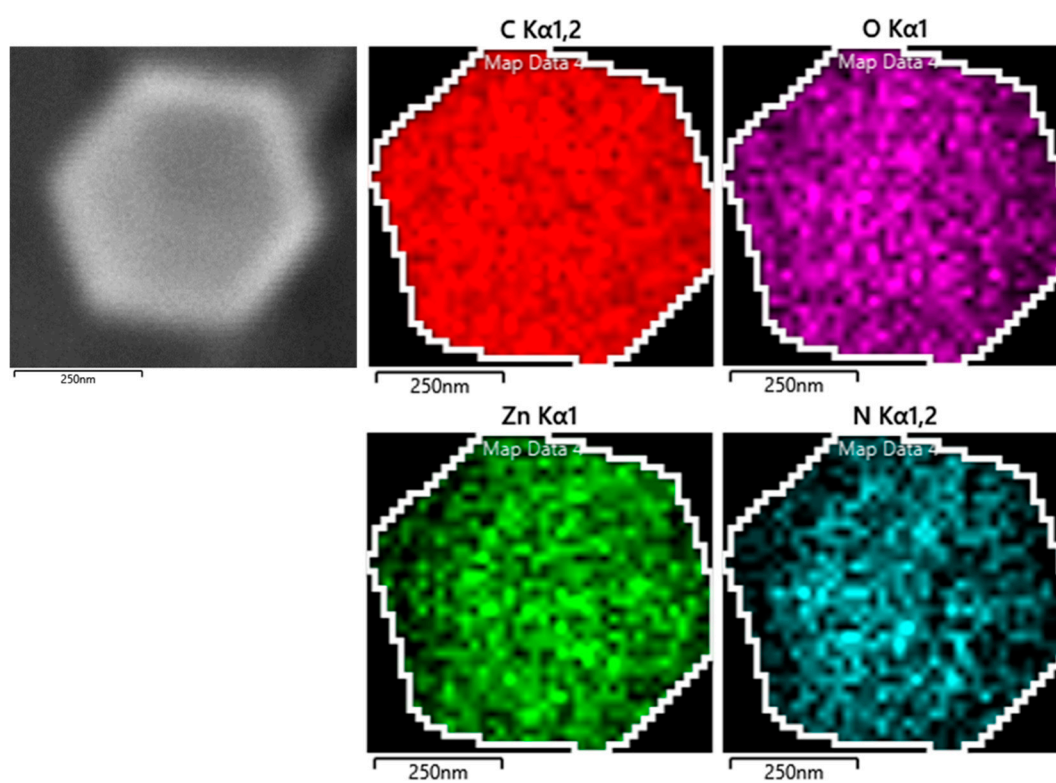


Figure S5. SEM-EDX analysis of *N,Z*-MPC, scale bar: 250 nm.

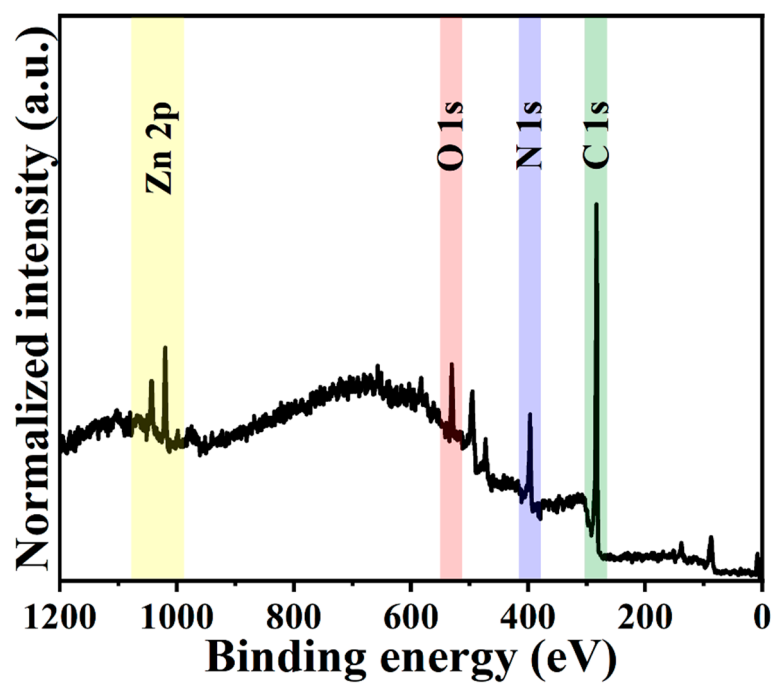


Figure S6. Full-range XPS spectrum of *N,Z*-MPC.

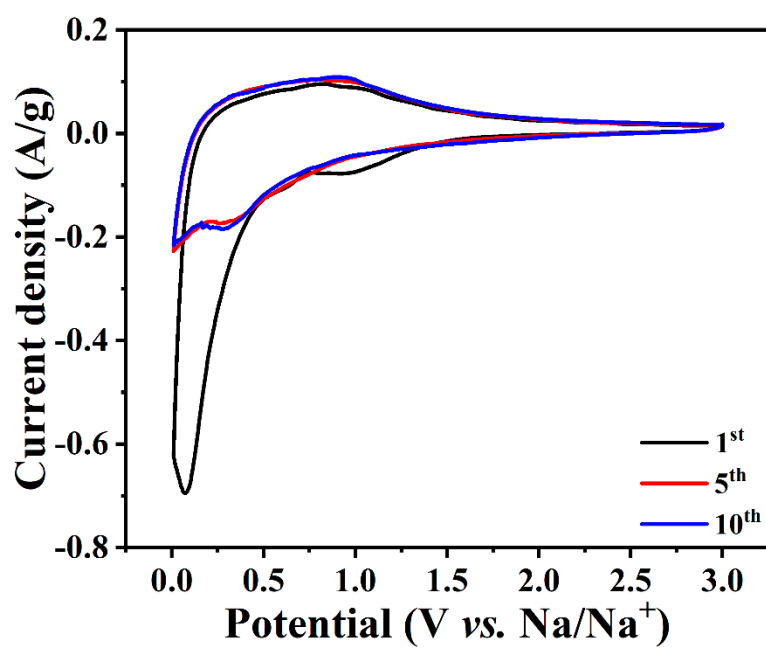


Figure S7. Cyclic voltammograms of *N,Z*-MPC collected in the potential range between 0.01 V and 3.0 V (vs. Na/Na⁺) at a scanning rate of 0.2 mV/sec.

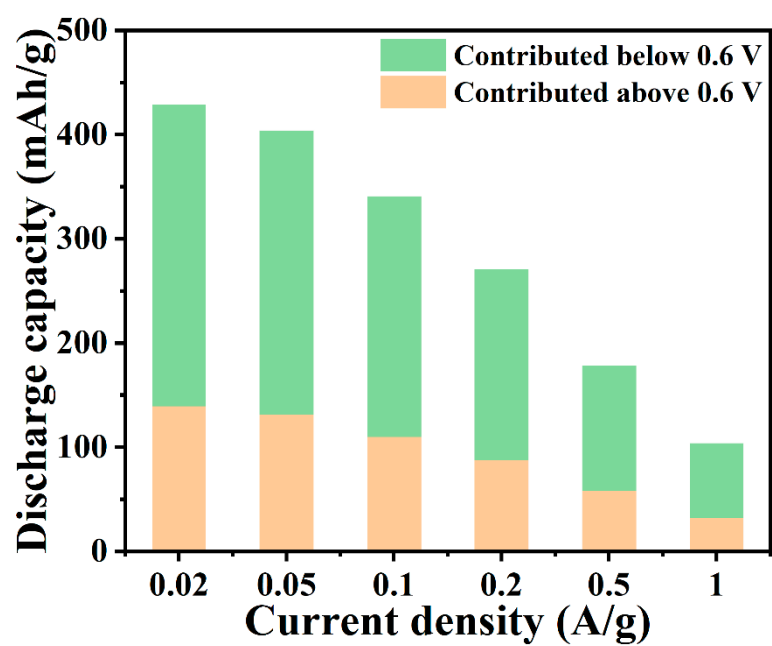


Figure S8. Relationship between the discharge capacity contributed from the potential below and above 0.6 V (vs. Na/Na⁺) and current density.

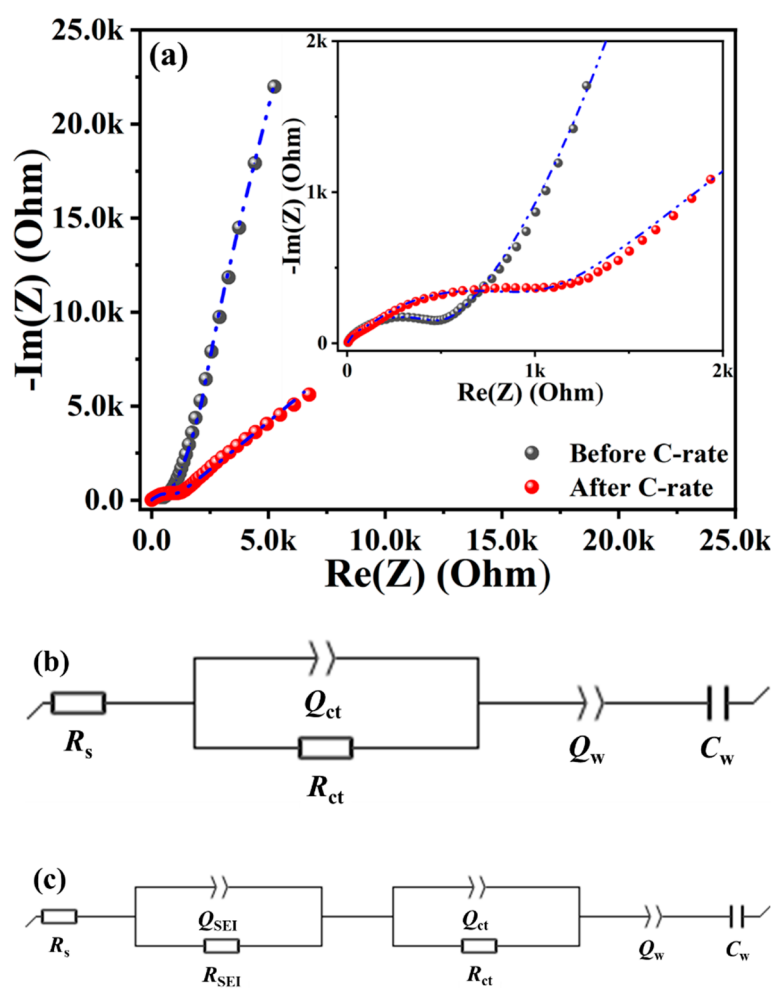


Figure S9. (a) Electrochemical impedance spectra of *N,Z*-MPC, and the equivalent circuit models used for parameter fitting: (b) before and (c) after C-rate measurements.

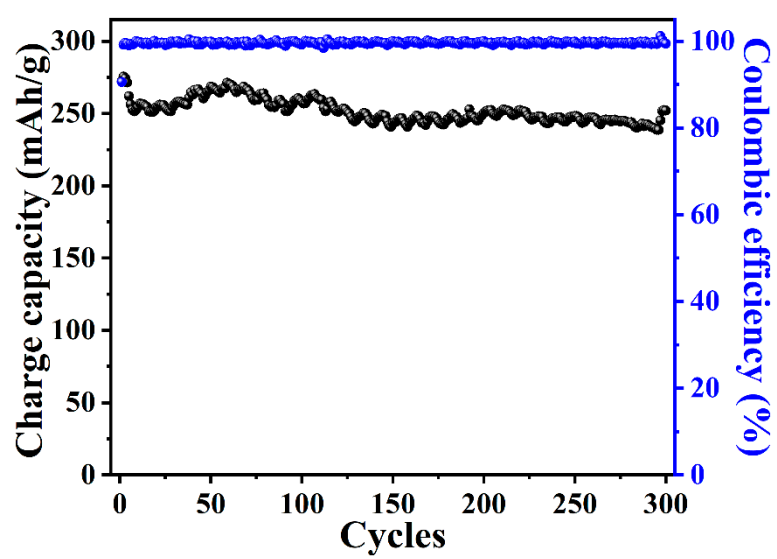


Figure S10. The cycling stability of *N,Z*-MPC measured at 0.2 A/g.

Table S1. Parameters of (002) peak from XRD pattern and related profile-fitting results.

(002) peak			Graphitic region			Disordered region			R^{\ddagger}
Sample	2θ (°)	d_{002}^{\dagger} (nm)	2θ (°)	d_{002}^{\dagger} (nm)	Area (%)	2θ (°)	d_{002}^{\dagger} (nm)	Area (%)	
<i>N,Z</i> -MPC	24.18	0.37	26.37	0.34	33	23.05	0.39	67	2.96

† Calculated by Bragg equation with the center of (002) diffraction peak.

‡ Defined as the ratio of the height of the (002) diffraction peak to the background.

Table S2. Comparisons of the electrochemical performances on previously reported nitrogen-doped carbon anodes for sodium-ion batteries using the carbonate-based electrolyte.

N content [†] (%)	S _{BET} (m ² /g)	Initial Coulombic efficiency (%)	Specific capacity (mAh/g at A/g)	Capacity retention (%)	Ref.
6.5	8	57	435 at 0.05, 147 at 2.0	74 at 1,200 cycles at 1 A/g	[20]
19.5	476	25	335 at 0.05, 94 at 5.0	95 at 265 cycles at 0.05 A/g	[55]
8.6	751	22	310 at 0.05, 54 at 5.0	90 at 2,000 cycles at 1 A/g [‡]	[56]
N/A	3	N/A	210 at 0.05, 135 at 5.0	71 at 600 cycles at 0.2 A/g	[57]
9	265	66	270 at 0.05, 185 at 1.0	88 at 300 cycles at 1 A/g	[58]
4.6	6	N/A	373 at 0.025, 120 at 1.0	95 at 1,000 cycles at 0.2 A/g	[62]
19.3	82	N/A	264 at 0.1, 57 at 40	99 at 2,000 cycles at 0.5 A/g	[43]
3.9	202	56	332 at 0.1, 135 at 1.6	87 at 300 cycles at 0.2 A/g	[63]
5.8	111	76	285 at 0.05, 153 at 5.0	94 at 600 cycles at 1 A/g	[59]
10.5	212	N/A	263 at 0.05, 92 at 10.0	84 at 2,000 cycles at 0.5 A/g [¶]	[60]
16.1	441	38	423 at 0.02, 392 at 0.05, 104 at 1.0	93 at 300 cycles at 0.2 A/g 97 at 3,000 cycles at 1.0 A/g	This study

[†]Determined by XPS, [‡]Calculated based on 100th cycle, [¶]Calculated based on 10th cycle.

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