

## Supplementary Materials

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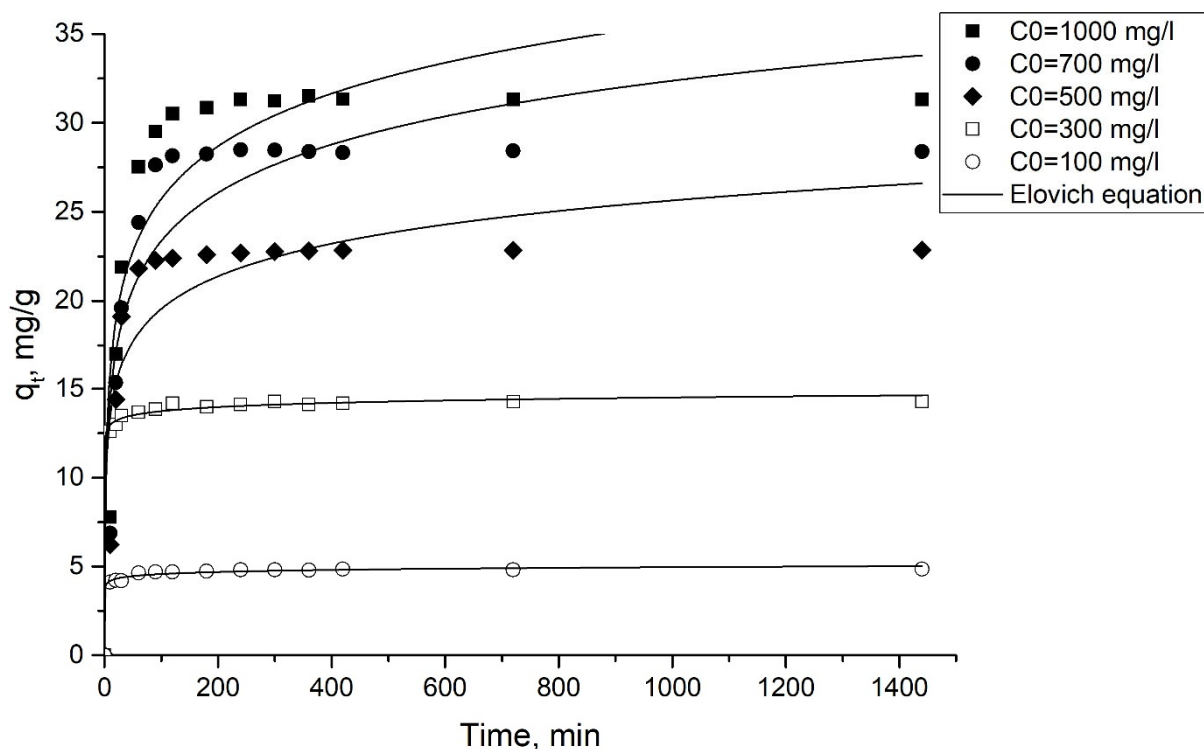
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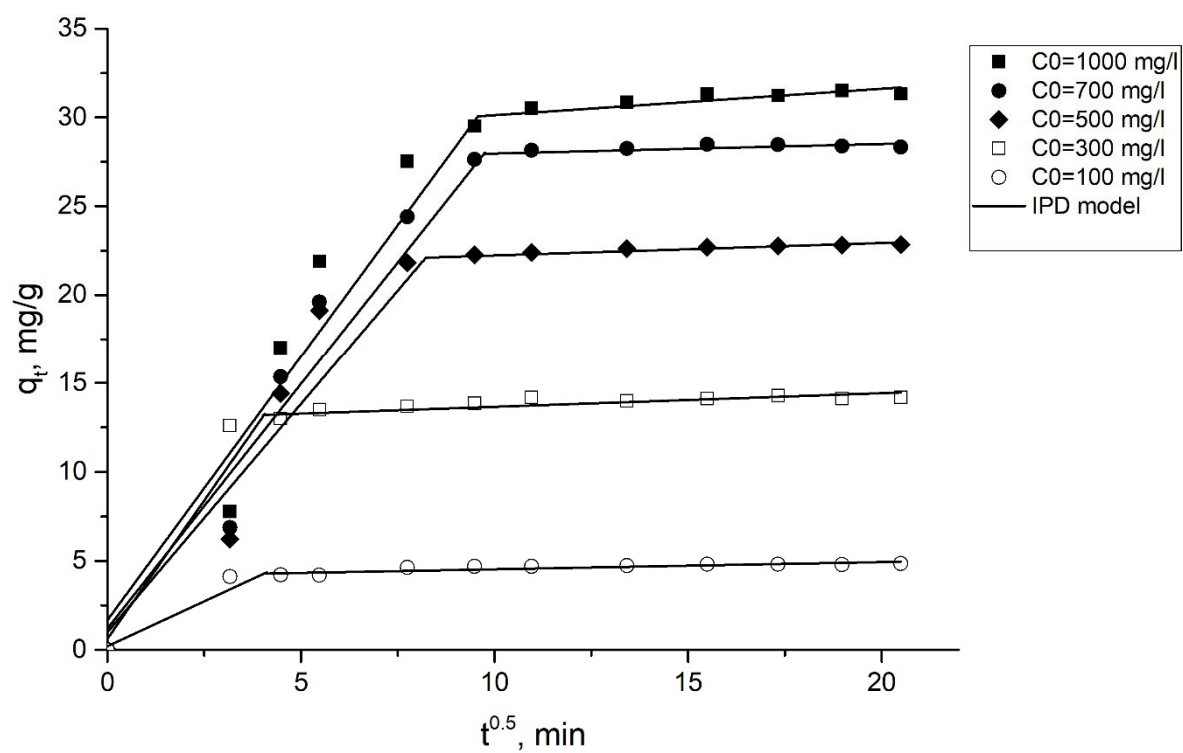
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**Table S1.** Kinetic models for the removal of molybdate using Al<sub>2</sub>O<sub>3</sub>.

C <sub>0</sub> Mo (VI), mg/l	IPD						Elovich equation		
	k <sub>p1</sub> , mg/g min <sup>1/2</sup>	C <sub>1</sub> , mg/g	R <sup>2</sup>	k <sub>p2</sub> , mg/g min <sup>1/2</sup>	C <sub>2</sub> , mg/g	R <sup>2</sup>	α, mg/(g·min)	β, g/mg	R <sup>2</sup>
100	1.009	0.208	0.931	0.04	4.125	0.784	1.143x10 <sup>9</sup>	5.96	0.993
300	3.106	0.629	0.933	0.077	12.89	0.722	2.186x10 <sup>15</sup>	2.985	0.997
500	2.566	0.993	0.903	0.07	21.524	0.857	42.437	0.378	0.838
700	2.75	1.204	0.943	0.054	27.429	0.552	15.24	0.255	0.86
1000	2.967	1.673	0.932	0.149	28.637	0.774	17.51	0.234	0.868



**Figure S1.** Elovich kinetic model for the adsorption of Mo(VI) on the Al<sub>2</sub>O<sub>3</sub>.



**Figure S2.** Intra-particle diffusion kinetic model for the adsorption of Mo(VI) on the Al<sub>2</sub>O<sub>3</sub>. Time up to 720 minutes.