

Anion-induced Structural Diversity and Optical Chromism in a Series of Cyano-bridged Heterometallic 3d-4f Coordination Polymers

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Supporting Information

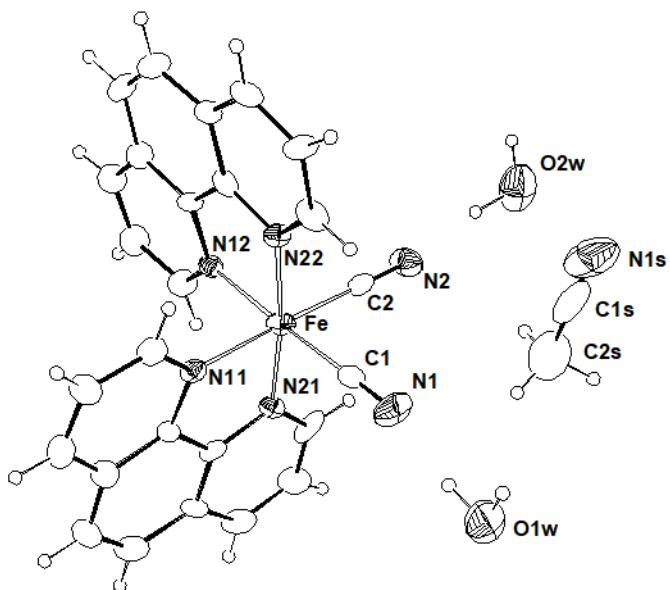


Figure S1. Ortep drawing of $[Fe(\text{Phen})_2(\text{CN})_2] \cdot 2\text{H}_2\text{O} \cdot \text{CH}_3\text{CN}$. Ellipsoids are drawn at 30% probability level.

Table S1. Summary of X-ray crystallographic data for [Fe(Phen)₂(CN)₂]·2H₂O·CH₃CN.

Empirical formula	C ₂₈ H ₂₃ FeN ₇ O ₂
Formula weight	545.38
Colour, habit	Prisms, purple
Crystal size, mm	0.10×0.08×0.05
Crystal system	Triclinic
Space group	P-1
<i>a</i> , Å	10.266(4)
<i>b</i> , Å	10.396(4)
<i>c</i> , Å	12.107(4)
α , deg.	80.92(1)
β , deg.	78.75(1)
γ , deg.	87.07(1)
<i>V</i> , Å ³	1251.1(8)
<i>Z</i>	2
<i>T</i> , K	293(2)
ρ (calc), Mg/m ³	1.448
μ , mm ⁻¹	0.644
θ range, deg.	1.73 to 24.04
No.of rflcn/obsv	39388 / 5692
GooF	0.892
<i>R</i> 1	0.0555
<i>wR</i> 2	0.0584

$R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$, $wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$, $w = 1 / [\sigma^2(F_o^2) + (aP)^2 + bP]$, where $P = [\max(F_o^2, 0) + 2F_c^2]/3$

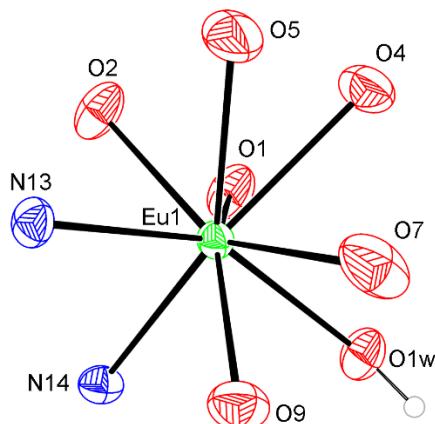


Figure S2. Capped square anti-prism geometry of the lanthanide ion in **1-Eu**. The square faces are represented by O(2)-O(4)-N(14)-O1w and by O(5)-O(7)-N(13)-O(9); O(1) is capping the former face.

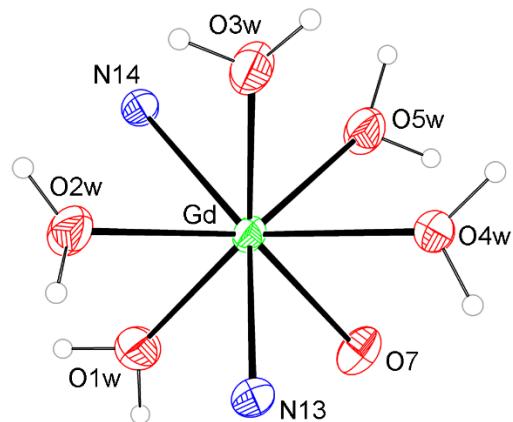


Figure S3. Square anti-prism geometry of the lanthanide ion in 2-Gd. The square faces are represented by O1w-N(14)-O5w-O(7) and by O2w-O3w-O4w-N(13).

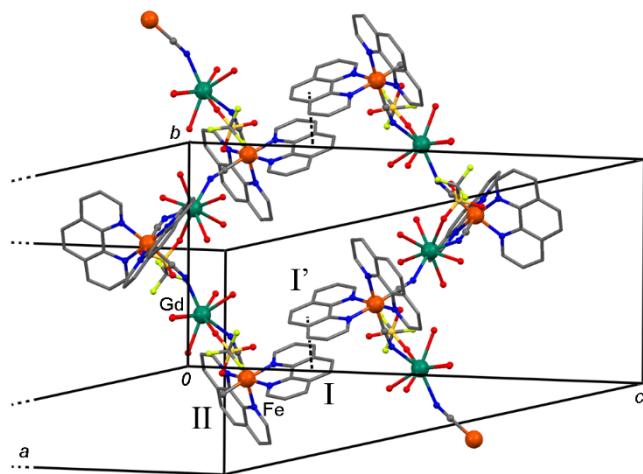


Figure S4. Portion of the crystal packing of 2-Eu. Only the coordination environment of the metal is reported for clarity. The π -stacking between I phenanthroline rings are reported as dashed bonds. Symmetry code ' = $\frac{1}{2}-x; \frac{1}{2}-y; 1-z$.

Table S2. Selected bond lengths (\AA) and angles ($^\circ$) for **1-Eu** and **1-Er**.

1-Eu			
Eu(1)-O(1)	2.538(2)	N(14)-Eu(1)-O(1)	75.03(8)
Eu(1)-O(2)	2.508(2)	N(14)-Eu(1)-O1w	78.68(9)
Eu(1)-O(4)	2.469(2)	O(1)-Eu(1)-O1w	70.09(8)
Eu(1)-O(5)	2.498(2)	N(13)-Eu(1)-O(5)	74.98(9)
Eu(1)-O(7)	2.480(3)	N(14)-Eu(1)-N(13)	83.05(8)
Eu(1)-O(9)	2.470(3)	O(1)-Eu(1)-O(5)	102.1(1)
Eu(1)-O1w	2.432(2)	O1w-Eu(1)-O(7)	75.8(1)
Eu(1)-N(13)	2.429(3)	N(13)-Eu(1)-O(2)	79.68(8)
Eu(1)-N(14)	2.482(2)	O(2)-Eu(1)-O(5)	74.7(1)
Fe(2)-N(11)	1.970(2)	O(5)-Eu(1)-O(4)	50.51(9)
Fe(2)-N(21)	1.996(2)	O(4)-Eu(1)-O(7)	70.92(9)
Fe(2)-N(12)	2.018(2)	O1w-Eu(1)-O(9)	79.80(9)
Fe(2)-N(22)	1.976(2)	O(9)-Eu(1)-N(14)	89.23(8)
Fe(2)-C(14)	1.911(3)	N(11)-Fe(2)-C(14)	94.4(1)
Fe(2)-C(13)'	1.901(3)	N(11)-Fe(2)-C(13)'	92.6(1)
Eu(1)-Fe(2)	5.5176(7)	N(22)-Fe(2)-C(14)	93.5(1)
Eu(1)-Fe(2)'	5.406(1)	N(22)-Fe(2)-C(13)'	91.4(1)
		N(12)-Fe(2)-C(13)'	91.9(1)
		N(21)-Fe(2)-C(14)	89.3(1)
		N(14)-C(14)-Fe(2)	176.7(3)
		N(13)'-C(13)'-Fe(2)	175.9(3)
		C(14)-N(14)-Eu(1)	169.4(2)
		C(13)-N(13)-Eu(1)	164.3(2)
1-Er			
Er(1)-O(1)	2.51(1)	N(14)-Er(1)-O(1)	74.7(4)
Er(1)-O(2)	2.45(1)	N(14)-Er(1)-O1w	79.1(4)
Er(1)-O(4)	2.41(1)	O(1)-Er(1)-O1w	69.8(4)
Er(1)-O(5)	2.44(1)	N(13)-Er(1)-O(5)	74.6(4)
Er(1)-O(7)	2.43(1)	N(14)-Er(1)-N(13)	82.2(4)
Er(1)-O(9)	2.41(1)	O(1)-Er(1)-O(5)	103.2(4)
Er(1)-O1w	2.36(1)	O1w-Er(1)-O(7)	76.3(4)
Er(1)-N(13)	2.37(1)	N(13)-Er(1)-O(2)	79.0(3)
Er(1)-N(14)	2.42(1)	O(2)-Er(1)-O(5)	73.9(5)
Fe(2)-N(11)	1.97(1)	O(5)-Er(1)-O(4)	51.7(3)
Fe(2)-N(21)	2.00(1)	O(4)-Er(1)-O(7)	71.0(3)
Fe(2)-N(12)	2.03(1)	O1w-Er(1)-O(9)	79.6(5)
Fe(2)-N(22)	1.98(1)	O(9)-Er(1)-N(14)	82.8(3)
Fe(2)-C(14)	1.93(1)	N(11)-Fe(2)-C(14)	94.9(4)
Fe(2)-C(13)'	1.91(1)	N(11)-Fe(2)-C(13)'	92.4(6)
Er(1)-Fe(2)	5.47(1)	N(22)-Fe(2)-C(14)	93.7(4)
Er(1)-Fe(2)'	5.36(1)	N(22)-Fe(2)-C(13)'	92.1(6)
		N(12)-Fe(2)-C(13)'	92.5(4)
		N(21)-Fe(2)-C(14)	88.8(4)

N(14)-C(14)-Fe(2)	177(1)
N(13)'-C(13)'-Fe(2)	175.3(9)
C(14)-N(14)-Er(1)	169.0(9)
C(13)-N(13)-Er(1)	166.0(8)

' = -x; -y; 1-z

Table S3. Selected bond lengths (\AA) and angles ($^\circ$) for **2-Ln**.

2-Gd

Gd-O1w	2.391(3)	O1w-Gd-O2w	76.4(1)
Gd-O2w	2.404(3)	O2w-Gd-N(14)	76.51(9)
Gd-O3w	2.398(3)	N(14)-Gd-O3w	78.6(1)
Gd-O4w	2.456(3)	O3w-Gd-O5w	84.1(1)
Gd-O5w	2.389(3)	O5w-Gd-O4w	76.7(1)
Gd-O(7)	2.417(3)	O4w-Gd-O(7)	76.7(1)
Gd-N(13)	2.426(3)	O(7)-Gd-N(13)	81.9(1)
Gd-N(14)	2.479(3)	N(13)-Gd-O1w	83.0(1)
Fe-N(11)	1.985(2)	N(13)-Gd-N(14)	147.07(9)
Fe-N(21)	2.012(3)	N(11)-Fe-C(14)'	94.0(1)
Fe-N(12)	2.006(3)	N(11)-Fe-C(13)	91.4(1)
Fe-N(22)	1.973(3)	N(22)-Fe-C(13)	93.4(1)
Fe-C(13)	1.884(3)	N(22)-Fe-C(14)'	92.5(1)
Fe-C(14)'	1.903(3)	N(12)-Fe-C(14)'	91.0(1)
Gd-Fe	5.2840(5)	N(21)-Fe-C(13)	92.9(1)
Gd-Fe'	5.4086(6)	N(13)-C(13)-Fe	174.6(3)
		N(14)'-C(14)'-Fe	178.4(3)
		C(13)-N(13)-Gd	153.1(2)
		C(14)-N(14)-Gd	156.1(2)

2-Eu

Eu-O1w	2.411(6)	O1w-Eu-O2w	76.7(2)
Eu-O2w	2.412(5)	O2w-Eu-N(14)	77.8(2)
Eu-O3w	2.417(6)	N(14)-Eu-O3w	78.4(2)
Eu-O4w	2.475(5)	O3w-Eu-O5w	83.7(2)
Eu-O5w	2.384(5)	O5w-Eu-O4w	74.6(2)
Eu-O(7)	2.424(5)	O4w-Eu-O(7)	76.9(2)
Eu-N(13)	2.435(6)	O(7)-Eu-N(13)	81.6(2)
Eu-N(14)	2.494(6)	N(13)-Eu-O1w	82.7(2)
Fe-N(11)	1.974(5)	N(13)-Eu-N(14)	146.8(2)
Fe-N(21)	2.013(6)	N(11)-Fe-C(14)'	94.0(2)
Fe-N(12)	2.016(6)	N(11)-Fe-C(13)	91.5(2)
Fe-N(22)	1.965(6)	N(22)-Fe-C(13)	93.7(3)
Fe-C(13)	1.893(7)	N(22)-Fe-C(14)'	91.9(3)
Fe-C(14)'	1.888(7)	N(12)-Fe-C(14)'	91.0(2)
Eu-Fe	5.288(1)	N(21)-Fe-C(13)	93.0(2)
Eu-Fe'	5.414(1)	N(13)-C(13)-Fe	174.4(6)
		N(14)'-C(14)'-Fe	177.4(6)
		C(13)-N(13)-Eu	153.2(5)

	C(14)-N(14)-Eu	155.9(5)	
2-Er			
Er-O1w	2.347(7)	O1w-Er-O2w	76.5(3)
Er-O2w	2.363(7)	O2w-Er-N(14)	75.9(3)
Er-O3w	2.405(6)	N(14)-Er-O3w	81.3(2)
Er-O4w	2.425(7)	O3w-Er-O5w	84.8(3)
Er-O5w	2.347(7)	O5w-Er-O4w	74.1(3)
Er-O(7)	2.373(7)	O4w-Er-O(7)	76.5(3)
Er-N(13)	2.361(8)	O(7)-Er-N(13)	81.7(3)
Er-N(14)	2.416(8)	N(13)-Er-O1w	83.7(3)
Fe-N(11)	1.970(8)	N(13)-Er-N(14)	147.1(3)
Fe-N(21)	2.025(8)	N(11)-Fe-C(14)'	95.1(3)
Fe-N(12)	2.002(8)	N(11)-Fe-C(13)	93.6(4)
Fe-N(22)	1.961(8)	N(22)-Fe-C(13)	93.8(4)
Fe-C(13)	1.89(1)	N(22)-Fe-C(14)'	91.1(3)
Fe-C(14)'	1.93(1)	N(12)-Fe-C(14)'	90.2(3)
Er-Fe	5.236(2)	N(21)-Fe-C(13)	93.6(3)
Er-Fe'	5.364(2)	N(13)-C(13)-Fe	174.1(8)
		N(14)'-C(14)'-Fe	174.7(8)
		C(13)-N(13)-Er	155.4(7)
		C(14)-N(14)-Er	156.8(7)

' = $\frac{1}{2}-x; \frac{1}{2}+y; \frac{1}{2}-z$

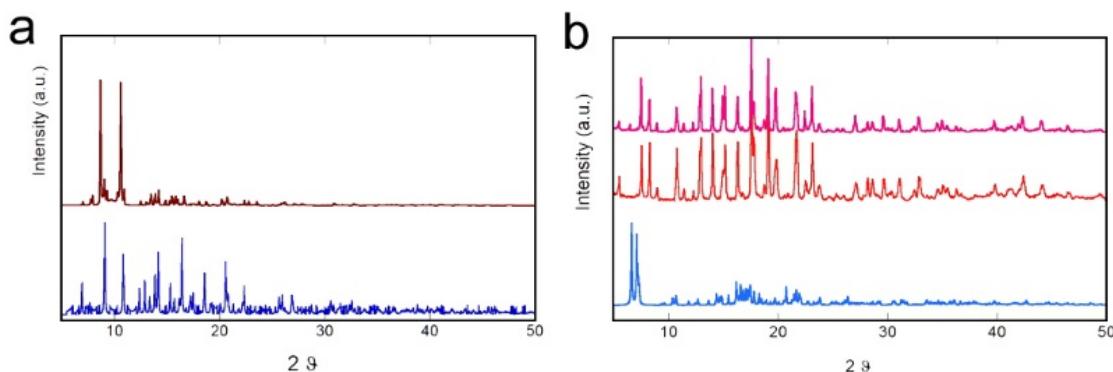


Figure S5. Powder-XRD patterns in the 2θ range from 5° to 50° (Cu K α). a) Experimental (above) and simulated (below) patterns of **1-Er**; b) Experimental patterns for **2-Gd** (pink), **2-Eu** (red) and simulated pattern of **2-Eu** (blue). Simulated patterns have been retrieved from single-crystal data. Some deviations of experimental data from the simulated ones in **2-Ln** are attributable to the loss of crystallization solvent over time and upon sample grinding.

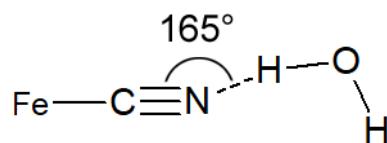


Figure S6. Schematic representation of the CN...H angle in $[\text{Fe}(\text{Phen})_2(\text{CN})_2] \cdot 2\text{H}_2\text{O} \cdot \text{CH}_3\text{CN}$.

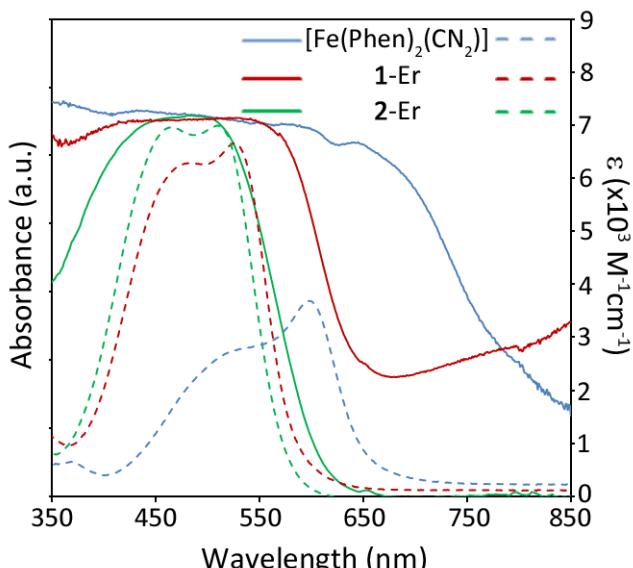


Figure S7. Electronic absorption of CH_3CN solutions (dashed lines) and solid-state diffuse reflectance (solid lines) spectra of **1-Er** and **2-Er** compared with the precursor $[\text{Fe}(\text{Phen})_2(\text{CN})_2]$ complex.

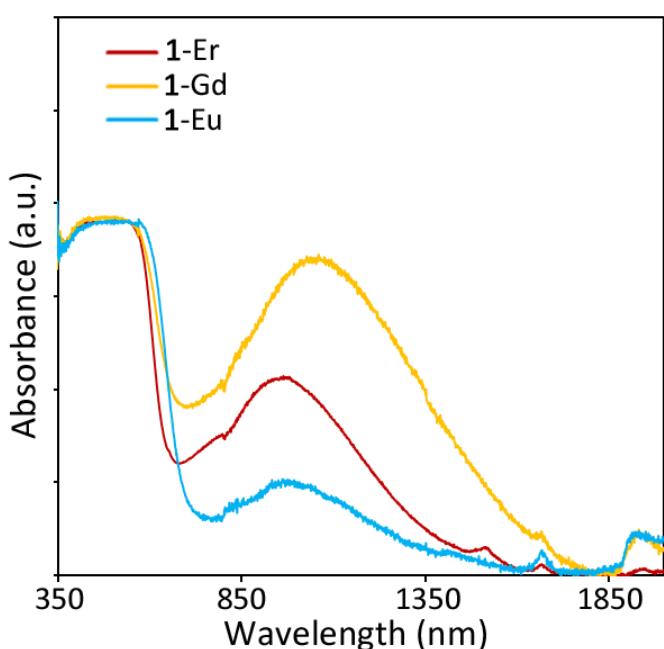


Figure S8. Solid-state diffuse reflectance spectra of crystalline samples of **1-Ln** ($\text{Ln} = \text{Er}, \text{Gd}, \text{Eu}$). The spectra are normalized for the absorption maximum in the visible range.

Table S4. DFT calculated molecular orbitals of $[\text{Fe}(\text{phen})_2\text{CN}_2]$ (isovalue plot 0.04).

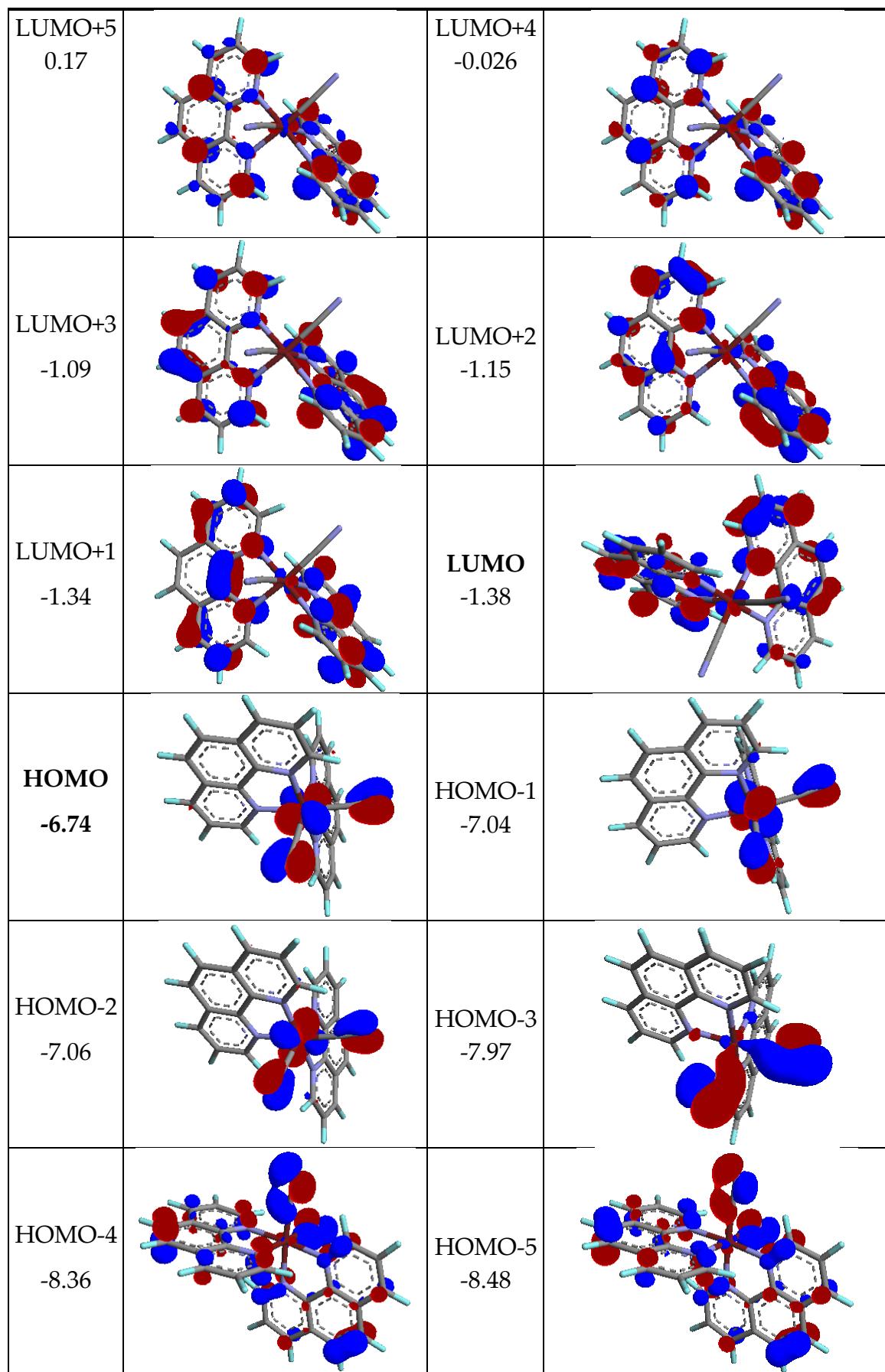


Table S5. DFT calculated molecular orbitals of **1-Er** (isovalue plot 0.04).

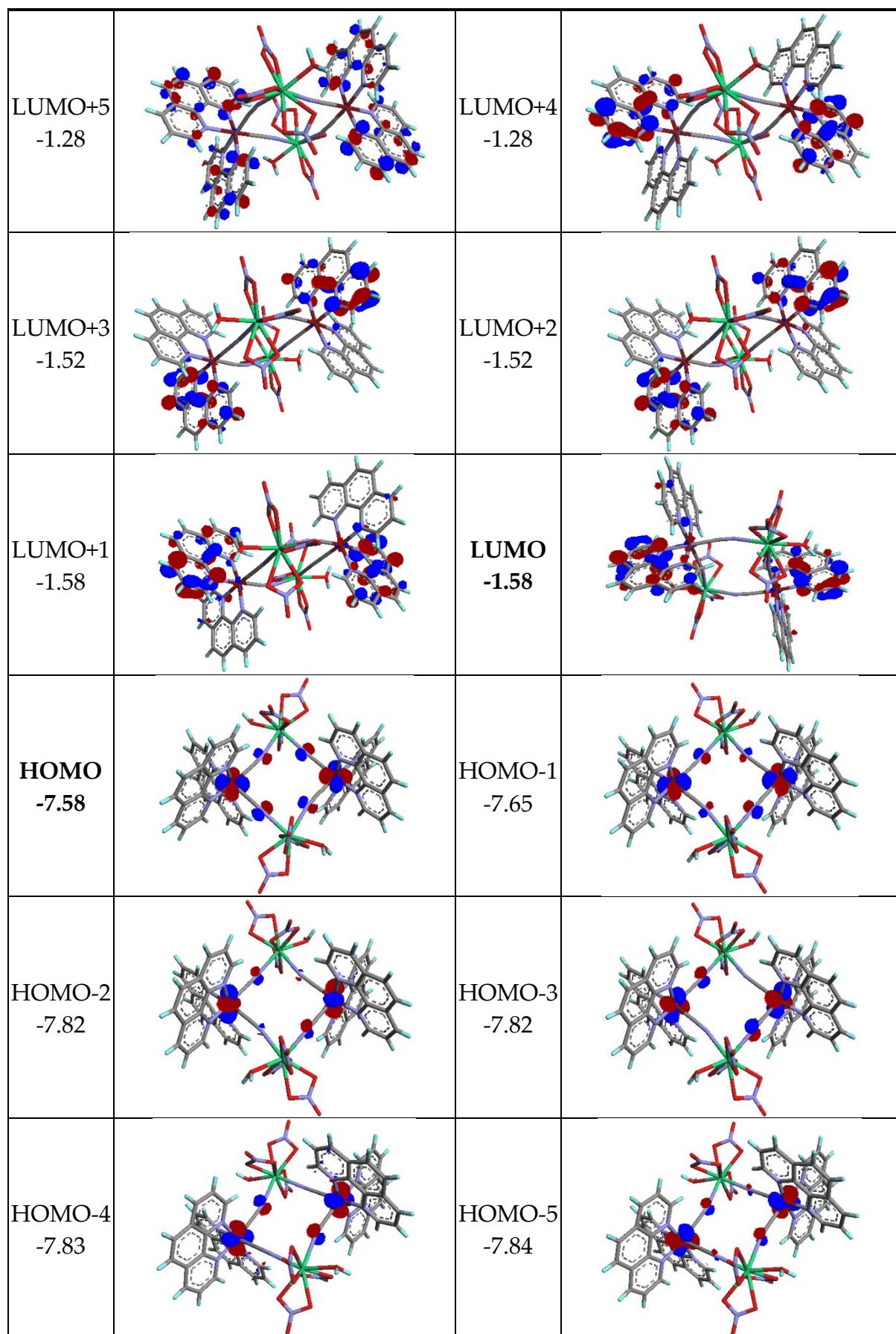


Table S6. DFT calculated molecular orbitals of **2-Er** (isovalue plot 0.04).

