

Supporting materials

Controllable Synthesis Photocatalytic Property and Mechanism of a Novel Direct Z-scheme POMs-based Nano Heterojunction α -Fe₂O₃/P₂Mo₁₈

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Characterization

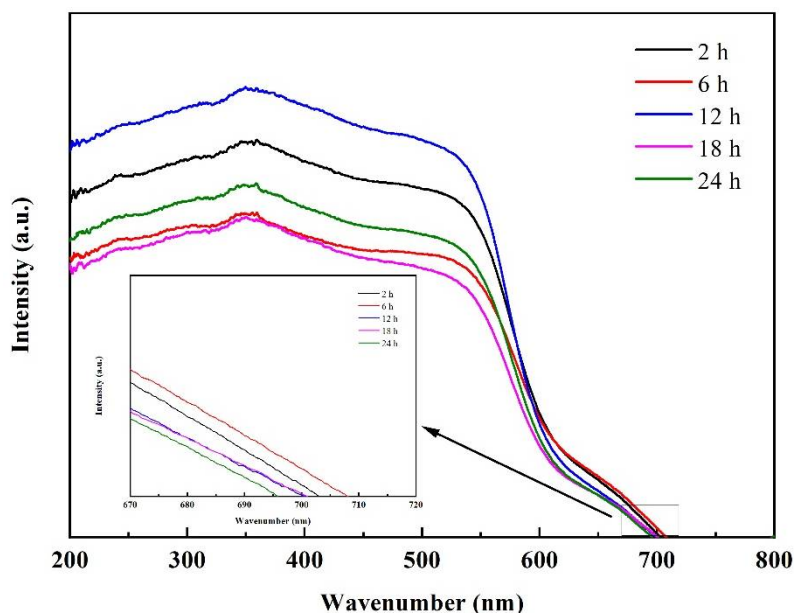


Fig. S1. DRS spectra of Fe₂O₃

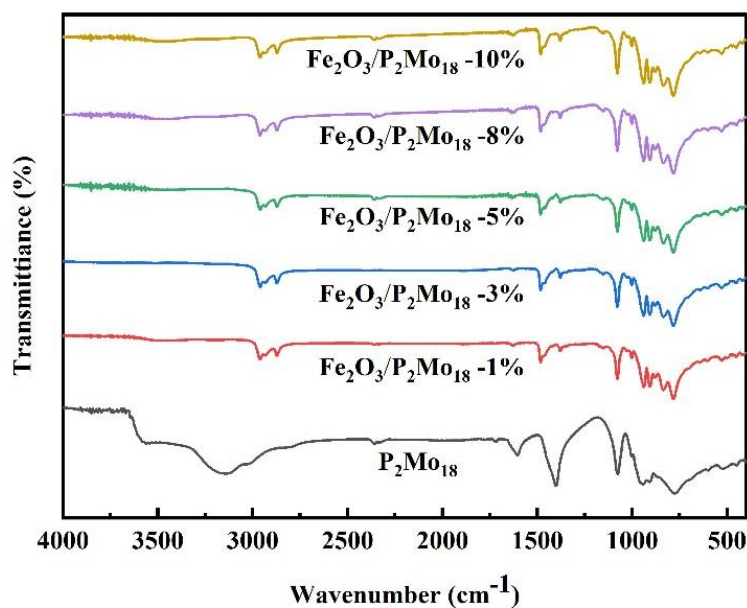


Fig S2. FT-IR spectra of the hybrid materials.

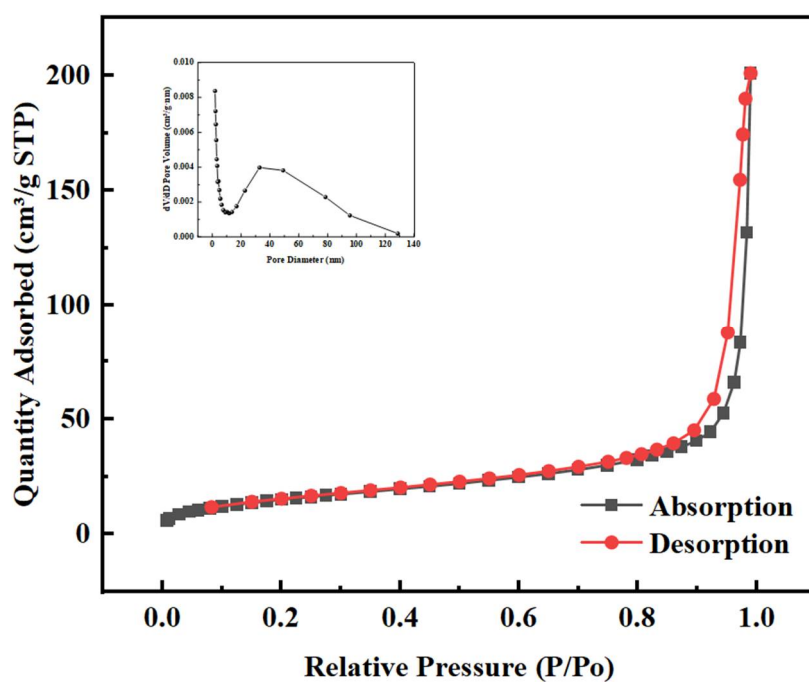


Fig. S3. N_2 adsorption desorption isotherms and pore size distribution curves (Insert)

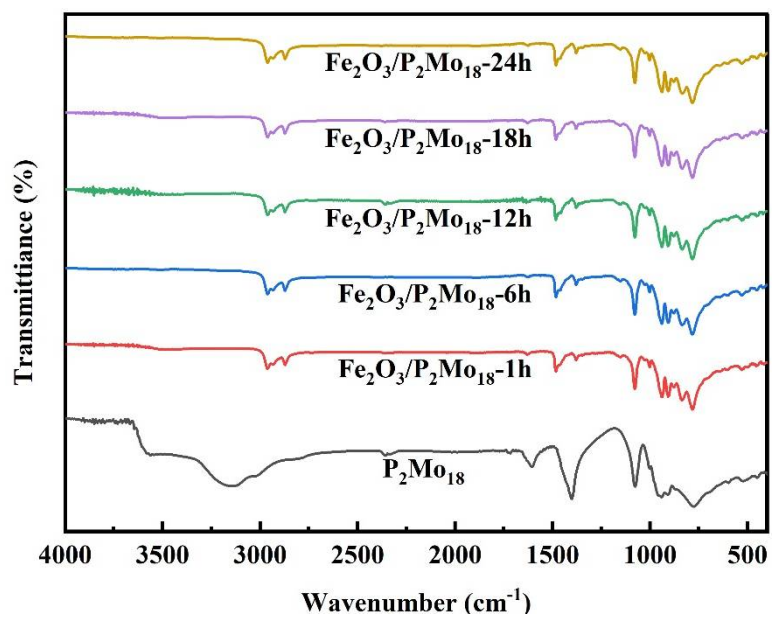


Fig. S4. FT-IR spectra of the hybrid materials with different reaction times of Fe_2O_3

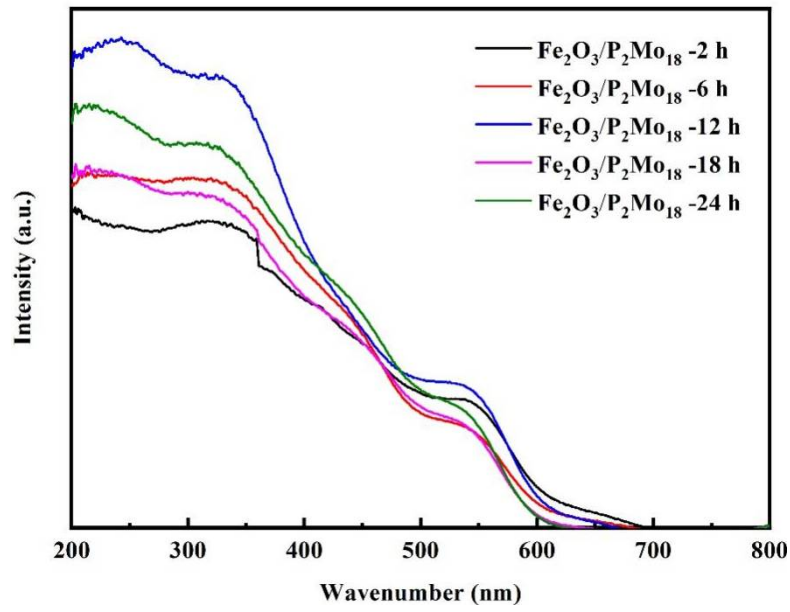


Fig. S5. UV-vis spectra of composites with different reaction times of Fe_2O_3

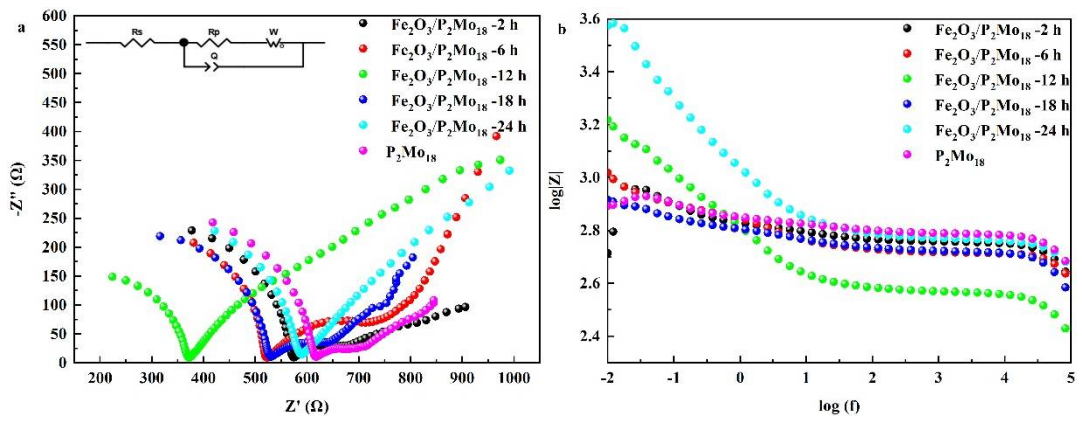


Fig. S6. Nyquist (a) and Bode (b) plots of composites made from Fe_2O_3 with different reaction times

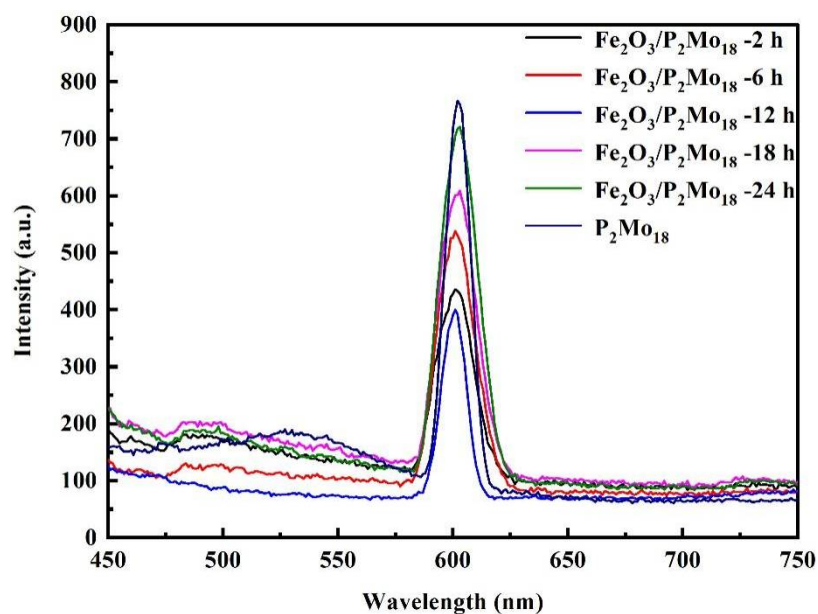


Fig. S7. Photoluminescence spectra of composites with different reaction times of Fe_2O_3

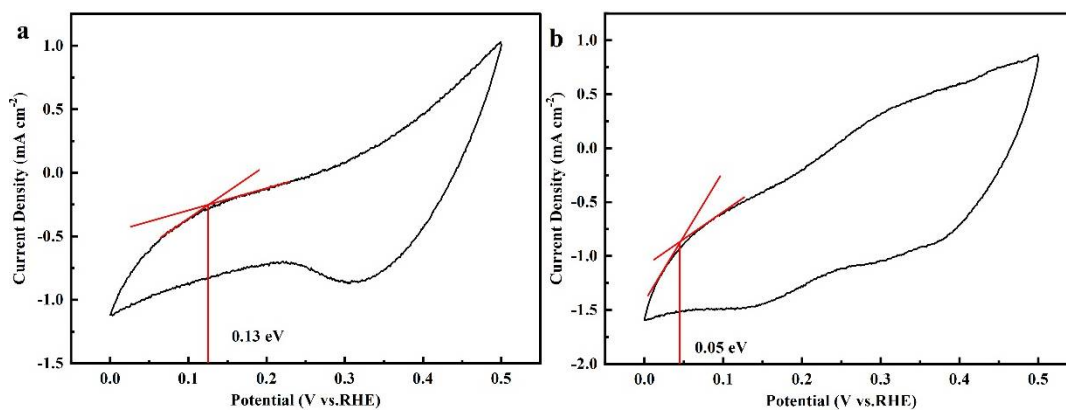


Fig. S8. the CV test diagram of polyoxoanion P_2Mo_{18} (a), Fe_2O_3 (b)

Table S1 The comparison of MB degradation activity of $\text{P}_2\text{Mo}_{18}/\text{Fe}_2\text{O}_3$ -5% with literature.

Photocatalyst	Concentration ($\text{mg}\cdot\text{L}^{-1}$)	Dosage ($\text{g}\cdot\text{L}^{-1}$)	Time (min)	Remove (%)	Reference
Fe_2O_3	10	0.5	120	73	1
TiO_2	13.5	0.2	60	78	2
$\alpha\text{-Fe}_2\text{O}_3/\text{BaTiO}_3$	20	0.5	180	94	3
$\alpha\text{-Fe}_2\text{O}_3/\text{Bi}_2\text{WO}_6$	5	0.2	25	98	4
$\text{K}_8/\text{EuW}_{10}$	2	0.25	2880	91	5
CdS@ZIF-8	10	1	120	72	6
ZnO/CeO_2	10	0.5	60	70	7
MoS_2/ZnO	10	0.05	180	93	8
$\text{Fe}_2\text{O}_3/\text{P}_2\text{Mo}_{18}$	20	0.1	90	98.9	This work

Table S2 The comparison of Cr reduction performance of P₂Mo₁₈/Fe₂O₃-5% with literature.

Photocatalyst	Concentration (mg·L ⁻¹)	Dosage (g·L ⁻¹)	Time (min)	Remove (%)	Reference
BiVO ₄ /FeVO ₄ @rGO	20	0.025	90	90.9	9
Co ₃ O ₄ /g-C ₃ N ₄	15	0.02	150	92.6	10
TiO ₂	10	0.01	180	62	11
TiO ₂ /SiO ₂	3	0.1	180	93	12
Fe ₃ O ₄ -ZnAl-LDH and TiO ₂	20	0.1	420	98	13
α-Fe ₂ O ₃ /g-C ₃ N ₄	10	0.1	150	98	14
AgI/BiVO ₄	15	0.02	100	71	15
Ag ₂ CO ₃ /BiVO ₄	15	0.02	150	70	16
Fe ₂ O ₃ /P ₂ Mo ₁₈	50	0.1	90	95.9	This work

Reference:

- [1] Zong M, Song D, Zhang X, et al. Environmental Science and Technology, 2021, 55(1):677-688
- [2] Maleki H, Bertola V. ACS Applied Nano Materials, 2019, 2(11):7237-7244
- [3] Yahia B, Faouzi S, Mohamed T. Journal of Photochemistry and Photobiology A: Chemistry, 2023, 439: 114634.
- [4] Wang W, Zhao W, Zhang H, et al. Chinese journal of catalysis, 2021, 42(1): 97-106.
- [5] Tong L, Wang Z, Xia C, et al. Journal of Physical Chemistry B, 2017, 121(46):10566-10573
- [6] Ankur M, Mala N, Shanid M, et al. Acs Omega, 2018, 3(7):8288-8308
- [7] Rajendran S, Khan M M, Gracia F, et al. Scientific reports, 2016, 6(1): 31641.
- [8] Chang Y C, Lin Y W, Lu M Y. Materials Chemistry and Physics, 2021, 266: 124560.
- [9] Yang R, Zhu Z, Hu C, et al. Chemical engineering journal, 2020, 390: 124522.
- [10] Zhao W, Li J, She T, et al. Journal of Hazardous Materials, 2021, 402: 123839.
- [11] Preethi R, Singh S. Journal of Environmental Management, 2023, 326: 116831.
- [12] Eddy D R, Ishmah S N, Permana M D, et al. Catalysts, 2020, 10(11): 1248.
- [13] Yang Y, Li J, Yan T, et al. Journal of colloid and interface science, 2020, 562: 493-501.
- [14] Xiao D, Dai K, Qu Y, et al. Applied Surface Science, 2015, 358: 181-187.
- [15] Huang Z, Dai X, Huang Z, et al. Chemosphere, 2019, 221: 824-833.
- [16] Yuan L, Weng B, Colmenares J C, et al. Small, 2017, 13(48): 1702253.