

Essential Oils Encapsulated in Zeolite Structures as Delivery Systems (EODS): An Overview

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Table S1. Mechanisms of action of some EO against fungi

Plant source of EO		Target microorganism	Mechanism of action	References
scientific name	common name			
<i>Coriaria nepalensis</i>	Tannery tree	<i>Candida albicans</i>	Inhibition of ergosterol bio-synthesis and membrane in-tegrity disturbance	[1]
		<i>Candida glabrata</i>		
		<i>Candida parapsilosis</i>		
		<i>Candida krusei</i>		
<i>Curcuma longa</i>	Turmeric	<i>Aspergillus flavus</i>	Inhibition of ergosterol biosynthesis	
<i>Melaleuca alternifolia</i>	Tea tree	<i>Candida albicans</i>	Glucose-dependent external media acidification; Permea-bility and membrane fluidity change	
		<i>Candida glabrata</i>	Glucose-dependent external media acidification	
		<i>Saccharomyces cerevisiae</i>		
<i>Thymus vulgaris</i>	Thyme	<i>Aspergillus flavus</i> <i>Aspergillus parasiticus</i>	Reduction of mycelium growth and sporulation in-hibition	[2]
<i>Origanum vulgare</i>	Oregano			
<i>Eucalyptus globulus</i>	Eucalyptus	<i>Mucor hiemalis</i>	Effect on degradative and enzymatic ability	[3]
		<i>Penicillium glabrum</i>		
		<i>Fusarium roseum</i>		
<i>Mentha piperita</i>	Mint	<i>Candida albicans</i>	Reduction of ergosterol pro-duction, induction of intra-cellular acidification, mor-phological changes, and membrane disruption	[4]

Table S2. Synergic, additive or antagonistic interactions between EO components

Combined EO components	Microorganism	Method used to study interactions	Interaction	References
Thymol/ Carvacrol	<i>Staphylococcus aureus</i>	Half dilution	Additive	
	<i>Pseudomonas aeruginosa</i>			
	<i>Escherichia coli</i>	Checkerboard	Synergism/ Additive	
	<i>Staphylococcus aureus</i>		Antagonism	
	<i>Escherichia coli</i>			
	<i>Staphylococcus aureus</i>	Mixture	Additive	
	<i>Pseudomonas aeruginosa</i>			
Thymol/ Eugenol	<i>Escherichia coli</i>	Checkerboard	Synergism	
Carvacrol/ Eugenol	<i>Escherichia coli</i>	Checkerboard	Synergism	
	<i>Staphylococcus aureus</i>		Antagonism	
	<i>Bacillus cereus</i>			
	<i>Escherichia coli</i>			
	Carvacrol/ Linalool	<i>Listeria monocytogenes</i>	Checkerboard	Synergism
Cinnamaldehyde/ Carvacrol	<i>Escherichia coli</i>	Checkerboard	Additive	[5]
Cinnamaldehyde/ Thymol	<i>Escherichia coli</i>	Checkerboard	Synergism	
	<i>Salmonella typhimurium</i>	Mixture		
Cinnamaldehyde/ Eugenol	<i>Staphylococcus</i> sp.	Mixture	Additive	
	<i>Micrococcus</i> sp.			
	<i>Enterobacter</i> sp.			
Limonene/ 1,8-cineole	<i>Staphylococcus aureus</i>	Mixture	Synergism	
	<i>Pseudomonas aeruginosa</i>			
1,8-cineole/ Aromadendrene	Methicillin-resistant <i>Staphylococcus aureus</i>	Checkerboard	Additive	
	Vancomycin-resistant <i>Enterococci</i>			
	<i>Enterococcus faecalis</i>			
Carvacrol/ Thymol	<i>Aspergillus flavus</i>	Mixture	Synergism	[6]
	<i>Aspergillus alternata</i>			
	<i>Penicillium</i> sp.			
	<i>Fusarium</i> spp.			
Ascaridole/ Carvacrol	<i>Leishmania amazonensis</i>	Mixture	Synergism	[7]
Carvacrol/ Cinnamaldehyde	<i>Escherichia coli</i>	Mixture	Synergism	[8]
	<i>Listeria innocua</i>			

Note: In the checkerboard method, compounds are combined in various proportions and added to a matrix divided in sections, such as microtiter plates, to determine the fractional inhibitory concentration or the effect of the combination index of each compound. The mixture method requires the comparison between experimental data and reference values in which synergism, antagonism or additive interactions are absent.

Table S3. EO and antimicrobial drugs interactions

Microorganism	Plant source of EO (scientific name)	Antibiotic/ Antifungal	Effect	References
<i>Staphylococcus aureus</i>	<i>Foeniculum vulgare</i>	Cefoxitin	Enhancement of inhibition zone	[9]
		Mupirocin		
		Cotrimoxazole		
		Ciprofloxacin		
<i>Staphylococcus aureus</i>	<i>Eugenia uniflora</i>	Amikacin	Sinergy	[10]
<i>Escherichia coli</i>				
<i>Escherichia coli</i>	<i>Origanum vulgare</i>	Penicillin	Synergy	[11]
<i>Trichophyton rubrum</i>	<i>Melaleuca alternifolia</i>	Iitraconazole	Synergy and MIC reduction	[12]
		Ketoconazole		
<i>Trichophyton rubrum</i>	<i>Cinnamomum verum</i>	Fluconazole	Synergy	[13,14]
	<i>Syzygium aromaticum</i>			
	<i>Cymbopogon martini</i>			
	<i>Thymus vulgaris</i>			
<i>Trichophyton</i> spp.	<i>Allium sativum</i>	Ketoconazole	Synergy	[15]
<i>Pseudomonas aeruginosa</i>	<i>Prunus armeniaca</i>	Ciprofloxacin		
<i>Candida albicans</i>	<i>Laurus nobilis</i>	Fluconazole	MIC reduction	[16]
<i>Candida glabrata</i>				
<i>Escherichia coli</i>	<i>Mentha spicata</i> L.	Gentamicin	Gentamicin susceptibility amplified by 30x	[17]
<i>Candida</i> spp.		Fluconazole	Strong anti-candida effect	
		Ketoconazole		

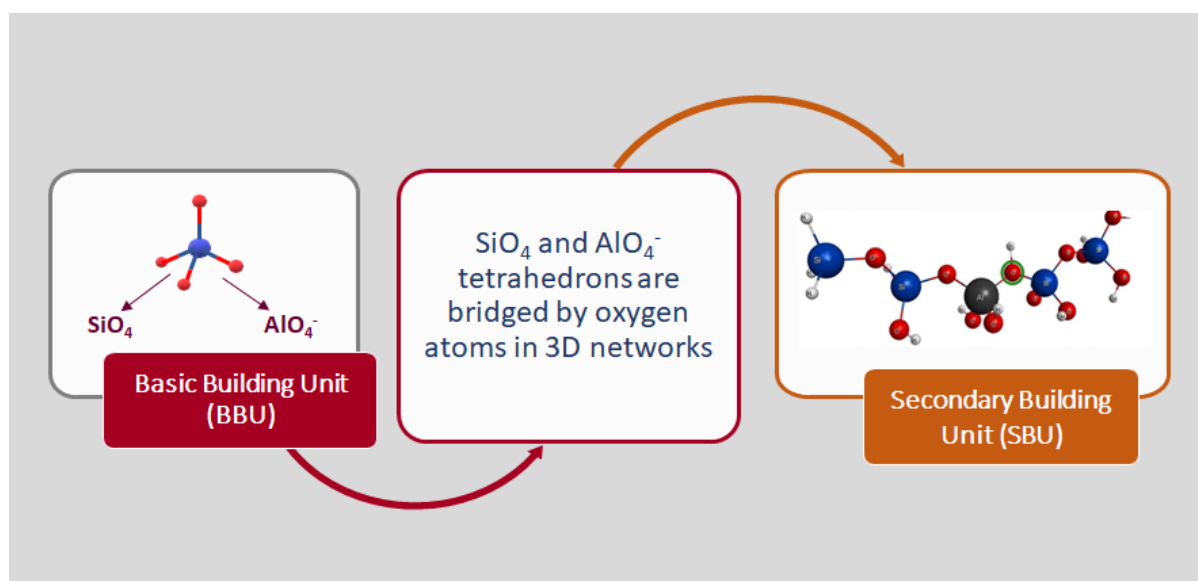


Figure S1. General scheme of the building pattern of zeolites' 3D framework, in which silicon atoms are color blue, oxygen atoms red, aluminum atoms black and hydrogen atoms white (adapted from [18,19]).

Table S4. Zeolites applicability and designated function (adapted from [19–21])

Application	Function
Construction	Cheap construction blocks with apparent low density, high porosity and homogenous texture
	Neutralization of excessive lime produced during concrete drying
	Low weight insulating material
Water and residual water treatment	Ammonium extraction from municipal residual water and agriculture
	Nitrifying bacteria growth medium
	Reduction of ammonium and ammonia contents in potable water
	Cheap and selective removal of Pb ²⁺ from potable water
Adsorption and Catalysis	Drying and purification of acidic gases
	Removal of water and carbon dioxide from sour natural gas
	Methane purification resulting from garbage decomposition
	Oxygen enriched air needed to hospitals, livestock and fish transport and restaurants with poor ventilation
Nuclear ashes and residues	Caesium and strontium removal from nuclear plants effluents
	Added to ⁹⁰ Sr and ¹³⁷ Cs contaminated soils, caused by tests or accidents, to reduce plants' uptake
	Added to pills and cookies to human consumption, in Bulgaria, to reduce intestinal absorption of Chernobyl's ashes
Agriculture and livestock	Poultry and swine food supplements
	Absorption of rations' contaminating aflatoxins
	Slow release of chemical fertilizers
	Soil conditioner in sandy soils and poor in clay
	Reduction of nitrate lixiviation in golf courses
	Extermination of fruit trees pests
	Zeoponic substrate
Aquaculture	Ammonium removal from incubator water, transports and aquariums
	Food supplements in fish feeds
Animals' residues treatment	Odor reduction
	Residues' humidity control
	Purification of methane gas produced
Consumer products	Deodorant agents for shoes, waste bins and fridges
	Water and ammonia removal from pets' urine
	Odor reduction in barns and stables
	Pillows filled with zeolites to remove odors from coffins
Medical applications	Hemodialysis filters to remove ammonia, allowing the reuse of the saline solution in portable dialysis systems
	Athlete's foot treatment
	Reduction of recuperation time of wounds and surgical incisions

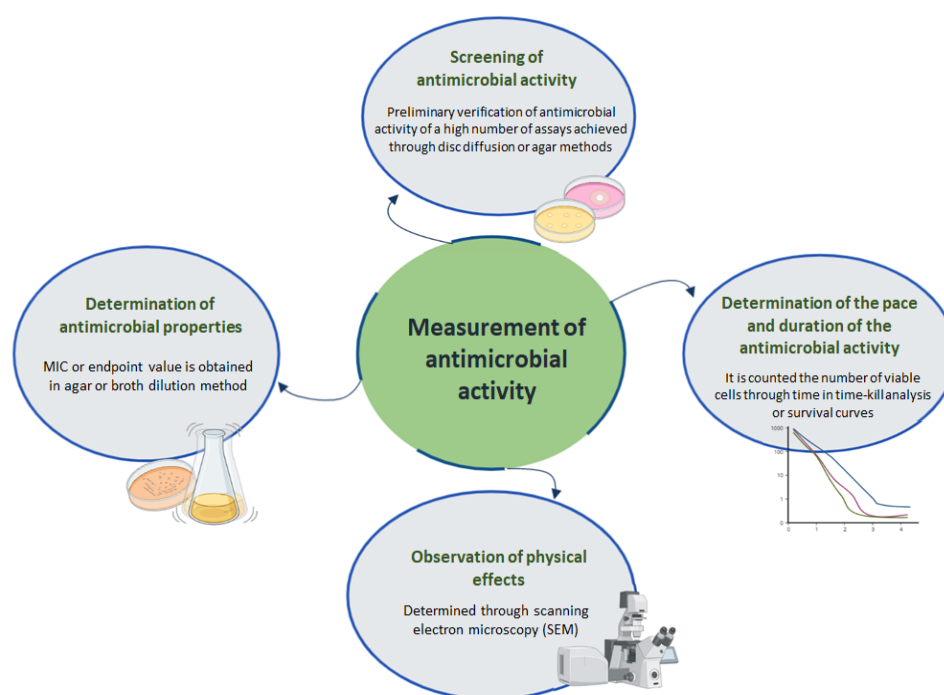


Figure S2. Methods used to determine antimicrobial activity of EO.

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