

Supplementary Material

Figure S1: Representative data and the fitting parameters

Since we obtained 4 apparent diffusion constants from the four peaks of the NaDS the averaged value was used for further analysis. We had 24 diffusion experiments for Figure 2 of the manuscript therefore almost one hundred data were fitted with the MestreNova software. In order to illustrate the quality of the data we show below randomly selected spectra form each concentration

This are the comment of Mnova software about the fitting results:

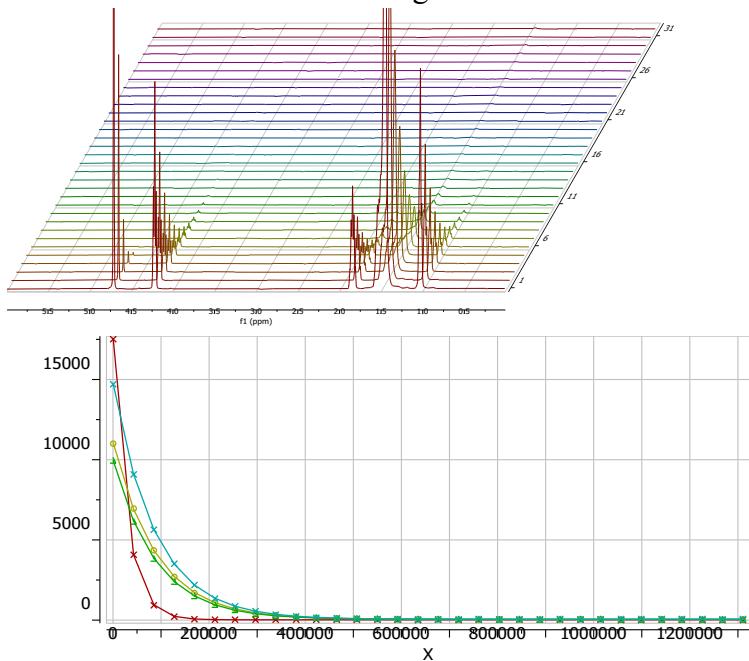
NOTE: The **Probnotmono** value is important to assess whether or not the model for the fit we are assuming, in this case, a three parameter-exponential model satisfies. For instance, if you assume a mono-exponential model but the experimental data follows a bi or multi-exponential behavior, then the analysis will not be fully correct. On the other hand, if we assume that the model is mono-exponential, the '**rError**' value gives the [absolute] probable error in the value of 'r', where 'r' is:
 $y = a \cdot \exp(-r \cdot x) + b$

We applied the above equation in the following form

$B + F \cdot \exp(-x \cdot G)$ therefore the rError is for G , the apparent diffusion coefficient.

We always checked the $B/F \cdot 100$ values (offset %). It was not more than 1% in either cases.

Diffusion of NaDS 0.015 mol kg⁻¹ solution in D₂O Δ=30 ms, δ =8 ms



Water peak (4.71 ppm)

B= 13.5723, F= 17857.2, G= 3.46422e-05

rError = 2.81893e-08, probnotmono = 0.983053

CH₂ 4.22 ppm

B= 34.7125, F= 11074.3, G= 1.11381e-05

rError = 1.89289e-08, probnotmono = 0.92135

CH₂ 1.84 ppm

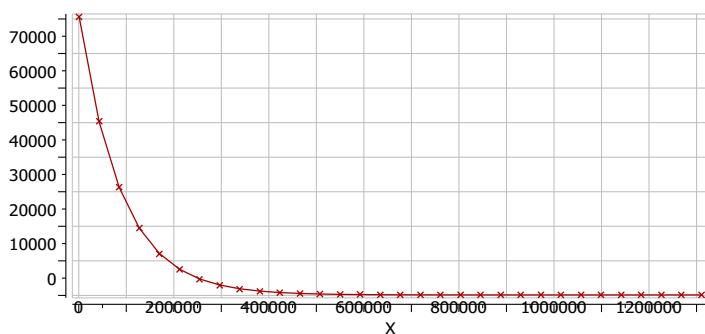
B= 39.9349, F= 9954.4, G= 1.13319e-05

rError = 1.68905e-08, probnotmono = 0.96145

CH₃ 1.03 ppm

B= -77.8792, F= 15200.9, G= 9.92067e-06

rError = 1.06481e-07, probnotmono = 0.993336

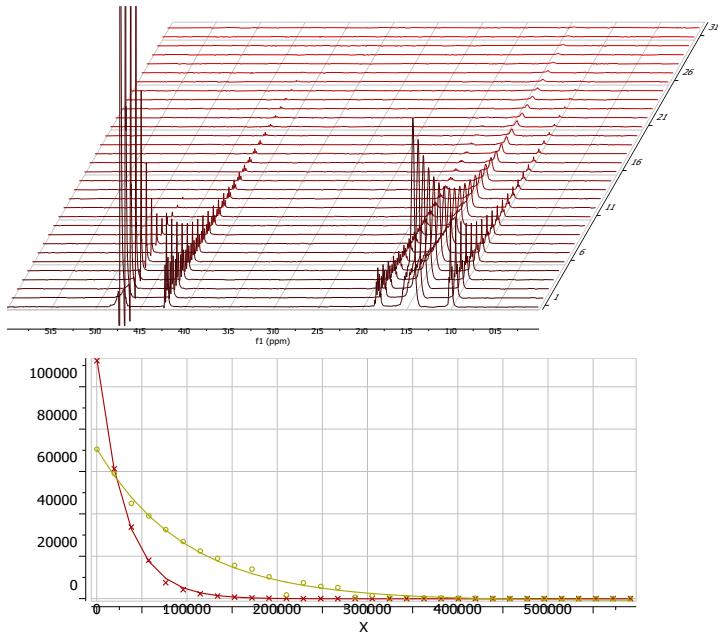


CH₂ peak (1.45-1.55 ppm)

B= -3475.41, F= 490054, G= 1.05753e-05

rError = 4.97833e-08, probnotmono = 1

Diffusion of NaDs 0.005 mol kg⁻¹ solution in D₂O Δ=15 ms, δ =8 ms



Water peak (4.71 ppm)

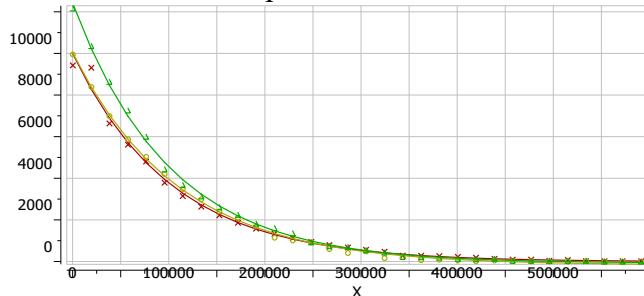
B= -104.773, F= 113720, G= 3.32785e-05

rError = 1.7288e-07, probnotmono = 0.9999998

CH₂ 1.42-1.53 ppm

B= -927.287, F= 71540.7, G= 1.00499e-05

rError = 2.32229e-07, probnotmono = 0.999796



CH₂ 4.22 ppm

B= -6.47894, F= 10001.7, G= 9.7169e-06

rError = 2.34496e-07, probnotmono = 0.92135

CH₂ 1.84 ppm

B= -172.298, F= 10207.8, G= 9.06488e-06

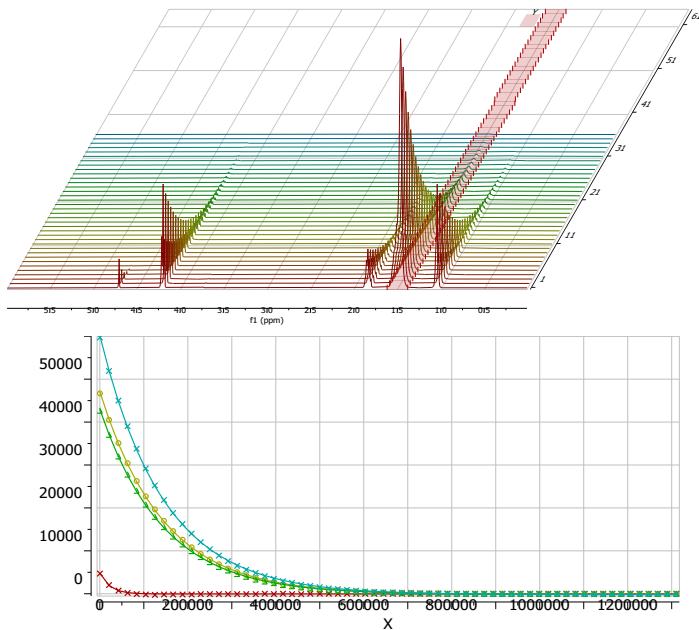
rError = 9.56075e-08, probnotmono = 0.999269

CH₃ 1.03 ppm

B= -77.8792, F= 15200.9, G= 9.92067e-06

rError = 1.06481e-07, probnotmono = 0.993336

Diffusion of NaDs 0.074 mol kg⁻¹ solution in D₂O Δ=30 ms, δ =8 ms



Water peak 4.71 ppm

B= -36.605, F= 4903.88, G= 3.36471e-05

rError = 8.87396e-07, probnotmono = 1

CH₂ 4.22 ppm

B= -124.228, F= 47283, G= 7.01776e-06

rError = 1.30441e-08, probnotmono = 1

CH₂ 1.84 ppm

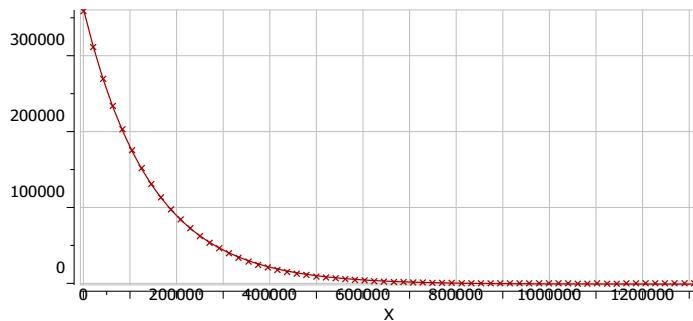
B= -121.582, F= 43192.5, G= 7.0181e-06

rError = 1.23635e-08, probnotmono = 1

CH₃ 1.03 ppm

B= -163.557, F= 60490.6, G= 6.96134e-06

rError = 1.22043e-08, probnotmono = 1

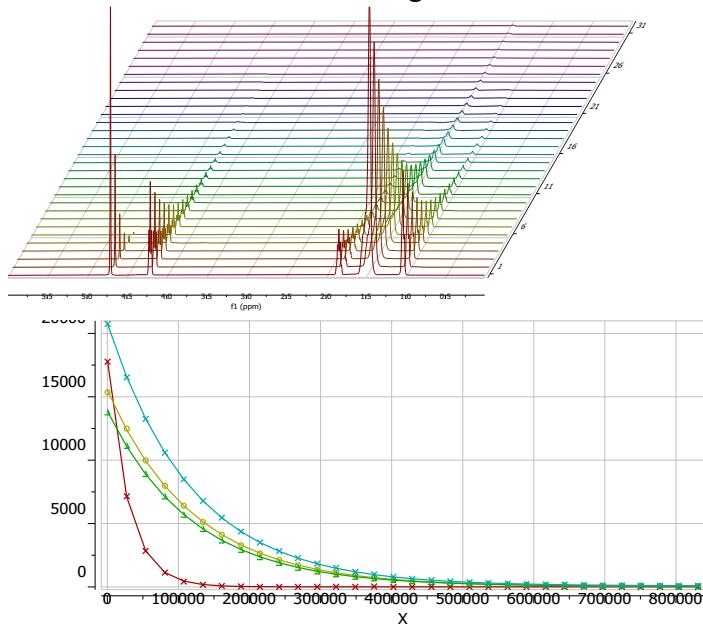


CH₂ peak 1.43-1.53 ppm

B= -999.561, F= 362961, G= 6.96679e-06

rError = 1.22093e-08, probnotmono = 1

Diffusion of NaDs 0.047 mol kg⁻¹ solution in D₂O Δ=20 ms, δ =8 ms



Water peak 4.71 ppm

B= 11.6092, F= 17995.2, G= 3.4249e-05

rError = 3.98459e-08, probnotmono = 0.855578

CH₂ 4.22 ppm

B= 45.0069, F= 15466.4, G= 8.2649e-06

rError = 1.68678e-08, probnotmono = 0.997661

CH₂ 1.84 ppm

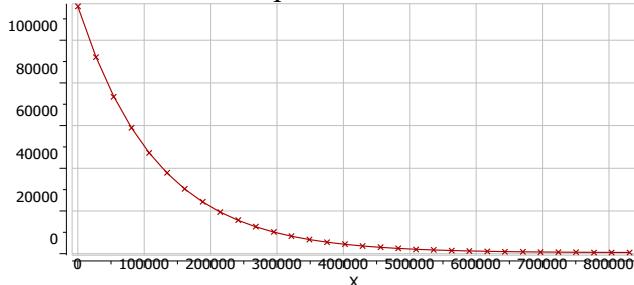
B= 53.8717, F= 13825.4, G= 8.29809e-06

rError = 1.00387e-08, probnotmono = 0.92135

CH₃ 1.03 ppm

B= 94.9812, F= 20698.5, G= 8.38078e-06

rError = 5.72934e-09, probnotmono = 0.855578

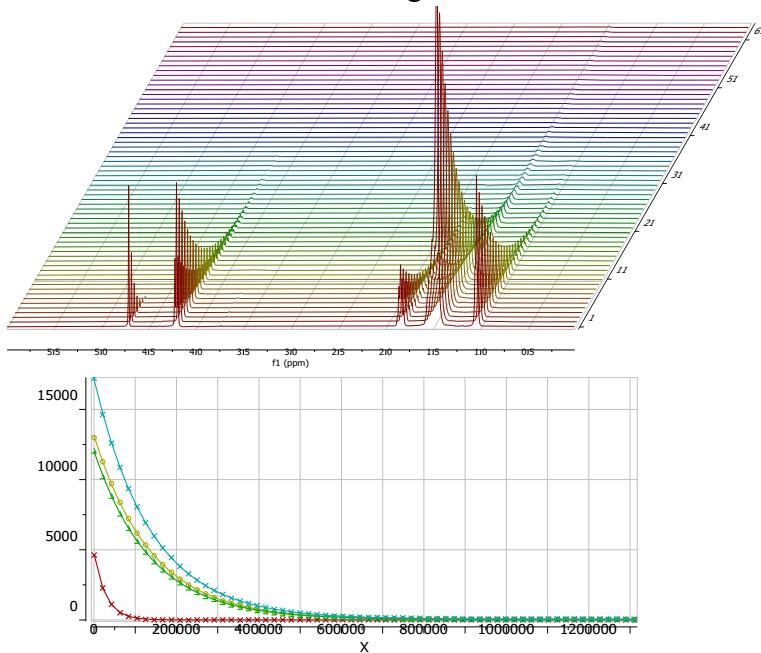


CH₂ peak 1.43-1.53 ppm

B= 455.465, F= 115408, G= 8.40559e-06

rError = 8.11708e-09, probnotmono = 0.99995

Diffusion of NaDs 0.059 mol kg⁻¹ solution in D₂O Δ=30 ms, δ =8 ms



Water peak 4.71 ppm

B= 0.388892, F= 4737.32, G= 3.47174e-05

rError = 5.7734e-08, probnotmono = 0.97725

CH₂ 4.22 ppm

B= 22.1, F= 13105, G= 7.24064e-06

rError = 7.68164e-09, probnotmono = 1

CH₂ 1.84 ppm

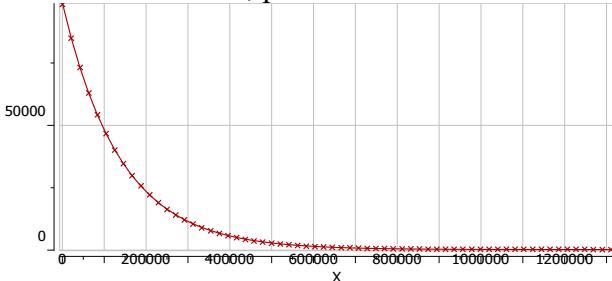
B= 28.7228, F= 11992.2, G= 7.32302e-06

rError = 6.8353e-09, probnotmono = 0.99702

CH₃ 1.03 ppm

B= 48.8479, F= 17119.7, G= 7.28536e-06

rError = 6.72099e-09, probnotmono = 0.987776

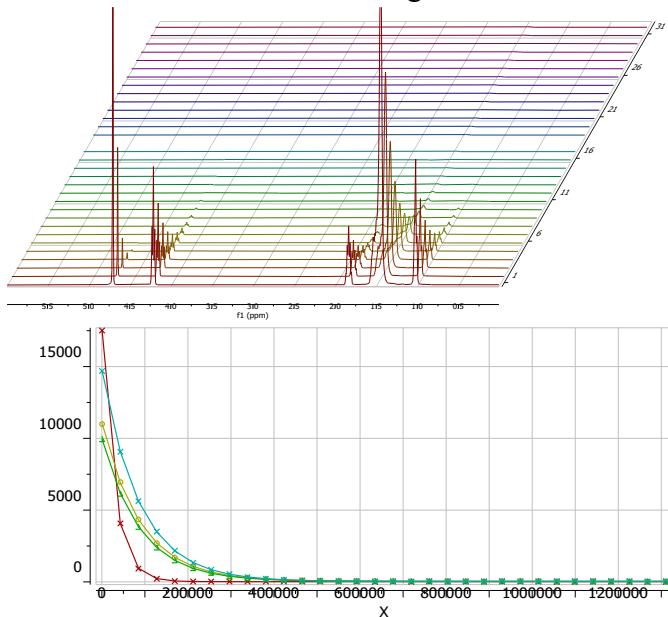


CH₂ peak 1.43-1.53 ppm

B= 213.327, F= 98876.2, G= 7.2341e-06

rError = 4.61981e-09, probnotmono = 1

Diffusion of NaDs 0.029 mol kg⁻¹ solution in D₂O Δ=30 ms, δ =8 ms



Water peak 4.71 ppm

B= 13.5723, F= 17857.2, G= 3.46422e-05

rError = 2.81893e-08, probnotmono = 0.983053

CH₂ 4.22 ppm

B= 34.7125, F= 11074.3, G= 1.11381e-05

rError = 1.89289e-08, probnotmono = 0.92135

CH₂ 1.84 ppm

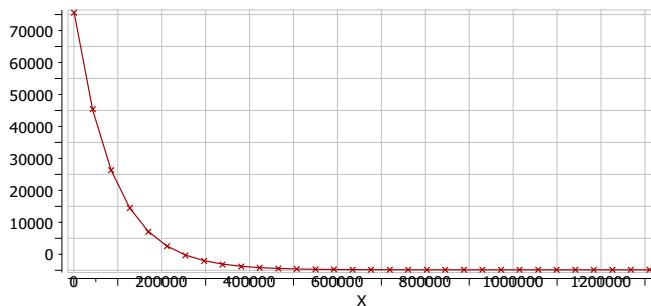
B= 39.9349, F= 9954.4, G= 1.13319e-05

rError = 1.68905e-08, probnotmono = 0.96145

CH₃ 1.03 ppm

B= 34.2499, F= 14751.2, G= 1.14127e-05

rError = 9.54809e-09, probnotmono = 0.92135



CH₂ peak 1.43-1.53 ppm

B= 136.632, F= 81115.2, G= 1.12706e-05

rError = 1.23238e-08, probnotmono = 0.999998

Illustration of the determination of real diffusion coefficients.

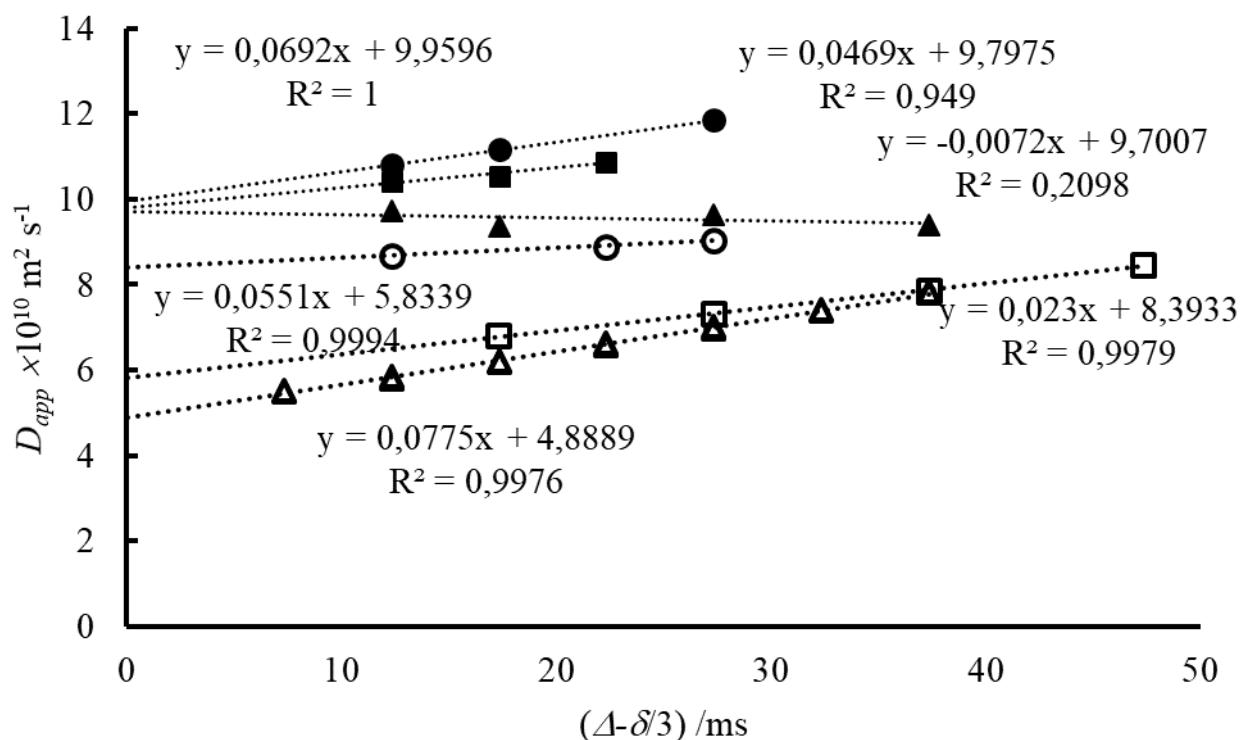


Figure S2. The apparent self-diffusion coefficients measured at different concentrations as a function of the diffusion time (Δ) at 319 K. Concentrations: \blacktriangle $0.005 \text{ mol kg}^{-1}$, \bullet $0.015 \text{ mol kg}^{-1}$, \blacksquare $0.029 \text{ mol kg}^{-1}$, \circ $0.047 \text{ mol kg}^{-1}$, \square $0.059 \text{ mol kg}^{-1}$, \triangle $0.074 \text{ mol kg}^{-1}$. The extrapolation of apparent diffusion coefficients to zero diffusion time. The values are written in the Figure in that order as the lines are drawn except the value of $0.047 \text{ mol kg}^{-1}$ which is at the most right position.