



# Article Conjugates of Iron-Transporting N-Hydroxylactams with Ciprofloxacin

Olga Bakulina<sup>1</sup>, Anton Bannykh<sup>1</sup>, Ekaterina Levashova<sup>1</sup> and Mikhail Krasavin<sup>1,2,\*</sup>

- <sup>1</sup> Saint Petersburg State University, 26 Universitetskii Prospect, 198504 Peterhof, Russia; o.bakulina@spbu.ru (O.B.); bannykh.anton.v@gmail.com (A.B.); levashovaekaterinayu@gmail.com (E.L.)
- <sup>2</sup> Immanuel Kant Baltic Federal University, 236041 Kaliningrad, Russia
  - \* Correspondence: m.krasavin@spbu.ru

**Abstract:** Screening of a library of novel *N*-hydroxylactams amenable by the Castagnoli-Cushman reaction identified four lead compounds that facilitated <sup>55</sup>Fe transport into *P. aeruginosa* cells (one of these synthetic siderophores was found to be as efficient at promoting iron uptake as the natural siderophores pyoverdine, pyochelin or enterobactin). Conjugates of the four lead siderophores with ciprofloxacin were tested for antibacterial activity against *P. aeruginosa* POA1 (wild type) and the  $\Delta pvdF\Delta pchA$  mutant strain. The antibacterial activity was found to be pronounced against the  $\Delta pvdF\Delta pchA$  mutant strain grown in CAA medium but not for the POA1 strain. This may be indicative of these compounds being 'Trojan horse' antibiotics. Further scrutiny of the mechanism of the antibacterial action of the newly developed conjugates is warranted.

**Keywords:** Castagnoli-Cushman reaction; cyclic hydroxamic acids; siderophores; bacterial iron transport; Trojan horse antibiotics



Citation: Bakulina, O.; Bannykh, A.; Levashova, E.; Krasavin, M. Conjugates of Iron-Transporting *N*-Hydroxylactams with Ciprofloxacin. *Molecules* **2022**, *27*, 3910. https://doi.org/10.3390/ molecules27123910

Academic Editor: Peter J. Rutledge

Received: 27 May 2022 Accepted: 16 June 2022 Published: 18 June 2022

**Publisher's Note:** MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



**Copyright:** © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/).

# 1. Introduction

Cyclic hydroxamic acid (*N*-hydroxylactam) moieties are frequently encountered in natural products and synthetic inhibitors of therapeutically relevant enzyme biotargets. The biological activity of *N*-hydroxylactams often has to do with their ability to chelate metal ions. Chelation of iron is the principal mechanism of action of microbial siderophores (metabolites excreted by iron-starved procaryotic microorganisms into extracellular space to scavenge iron needed for such processes as growth and differentiation [1,2]), examples of which include oxachelin (I) [3], scabichelin (II) [4] as well as structurally simple *N*-hydroxy-3,4-dihydroisoquinolin-1-one (III) which was isolated from *Streptomyces griseus* [5]. Anchoring to the prosthetic metal ion is central to the activity of small-molecule *N*-hydroxylactam inhibitors of such important zinc enzyme targets as matrix metalloprotease (e.g., IV [6] and V [7]), HIV integrase (e.g., VI [8]), histone deacetylases (e.g., VII [9]) as well as of (non-heme) iron-containing enzyme lipoxygenase (e.g., VIII or BMD 188 [10]) (Figure 1).

Some biological effects of *N*-hydroxylactams manifest themselves independently of metal chelation. Indeed, bacterial siderophores have been shown to possess a signaling function, antibacterial activity and to regulate oxidative stress [11]. At the same time, iron-loaded bacterial siderophores and their synthetic analogs were dubbed 'Trojan horse' carriers for antibiotics and fluorescent labels [12,13]. Conjugation of antibiotics or a fluorescent drugs otherwise eliminated by efflux-mediated resistance mechanisms to a siderophore molecule is a promising approach to circumventing drug resistance as well as to in vivo imaging and eradication of bacterial colonies [14]. This approach received clinical proof-of-concept in the 2019 approval of cefiderocol (Fetroja) for the treatment of complicated urinary tract infections [15].

Recently, we developed a facile access to a novel type of *N*-hydroxylactams (cyclic hydroxamic acids)—*trans*-2-hydroxy-1-oxo-3-aryl-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acids **1**—via a two-component condensation of oximes with homophthalic anhydride

(2) [16,17] or via a three-component condensation of aromatic aldehydes, hydroxylamine acetate and homophthalic acid (3) with azeotropic removal of water [18]. In addition to twenty-six hydroxamic acids **1a–z** prepared in the course of this investigation (for structures—see Supplementary Materials), we have recently prepared 2-hydroxy-3-(2-hydroxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (4) as a compound capable of providing an additional phenolic hydroxy group for metal chelation (Scheme 1).



Figure 1. Examples of N-hydroxylactams whose biological activity is exerted via metal chelation.



**Scheme 1.** Preparation of racemic *trans*-2-hydroxy-1-oxo-3-aryl-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acids **1** and **4** (relative stereochemistry shown).

3 of 14

Having amassed the library of novel *N*-hydroxylactams **1a–z** and **4**, we became interested in identifying, via screening, non-cytotoxic lead compounds capable of transporting iron into Gram-negative *Pseudomonas aeruginosa* bacterial cells. We reasoned that, considering that these compounds contain a conveniently positioned carboxylic acid functionality, the newly identified lead iron-complexing—and iron-transporting-compounds could be conjugated to a known antibiotic molecule (e.g., ciprofloxacin as was recently described [19,20]). Herein, we report our results obtained in the course of realization of this strategy.

# 2. Results and Discussion

In order to establish if *P. aeruginosa* cells are able to use any of the synthesized compounds **1a–z** or **4** to access extracellular iron, we used a growth assay in iron-restricted conditions, i. e. using iron-deficient casamino acid (CAA) medium and iron concentration around 20 nM, according to the previous work [21]. The *P. aeruginosa*  $\Delta pvdF\Delta pchA$  mutant unable to produce the siderophores pyoverdine [22] and pyochelin [23] was used [24] and was grown on a CAA medium in the absence or presence of 10 mM and 100 mM concentration of the compounds screened. This bacterial strain is able to grow in an iron-deficient environment likely due to citrate acting as siderophore [25] or owing to the functioning of the iron reduction systems which import ferrous ions through the *feoABC* system [26]. The assay is based on the premise that in the presence of an excess of a metal chelator, all iron traces present in the growth medium become scavenged, making this nutrient no longer available for low-affinity iron import systems. However, if the growth of the *P. aeruginosa*  $\Delta pvdF\Delta pchA$  mutant is still observed in the presence of the tested compounds, it would be indicative of the bacteria being able to use these iron chelators as siderophores.

A large portion of compounds 1a-z and 4 tested displayed nearly total growth inhibition at 10  $\mu$ M, indicating that such compounds cannot be used by *P. aeruginosa* as siderophores to access iron. For 12 compounds tested (1a, 1c-e, 1i-k, 1t-v, 1x, 1z, 4), the results were more promising, with a total inhibition at 100  $\mu$ M and no or a partial inhibition at 10  $\mu$ M (see Figure S1, Supplementary Materials). These compounds were then screened for the ability to transport <sup>55</sup>Fe into *P. aeruginosa* cells.

The compounds were loaded with <sup>55</sup>Fe and incubated with the *P. aeruginosa*  $\Delta pvdF\Delta pchA$ mutant. The latter was chosen so as to avoid any uptake of <sup>55</sup>Fe via the endogenous siderophores (pyoverdine and pyochelin). Reassuringly, for compounds 1a (DVD-000265), 1c (DVD-00627) and 4 (DVD-000468), an uptake of around 40–70 pmol of  $^{55}$ Fe/mL/OD<sub>600 nm</sub> after 30 min was observed and only in the absence of the protonophore carbonyl cyanide *m*-chlorophenyl hydrazone (CCCP) [27]. Because CCCP had been shown to inhibit any iron uptake by TonB-dependent transporters (bacterial outer membrane proteins that bind and transport siderophores [28]) in bacteria [29,30], this finding indicates that the uptake of iron by these compounds was not due to passive diffusion via porins [31], but rather due to a proton motive force-dependent transport relying on TonB. A special case was identified for compound **1i** (DVD-000304), as an uptake of about 150 pmol  $^{55}$ Fe/mL/OD<sub>600 nm</sub> was observed after 30 min incubation and was also abolished in the presence of CCCP indicating it being a TonB-dependent uptake. Such a high <sup>55</sup>Fe uptake by compound **1i** was judged to be as efficient as the one observed for natural siderophores pyoverdine, pyochelin or enterobactin in *P. aeruginosa* [30]. Altogether, these data show an iron uptake into *P. aeruginosa* cells in the presence of compounds **1a** (DVD-000265), **1c** (DVD-00627), **1i** (DVD-000304 and 4 (DVD-000468), which suggests that these compounds can be used as siderophores by this pathogen (Figure 2).

Having identified the four lead synthetic siderophores (**1a**, **1c**, **1i** and **4**), all of which contain a carboxylic acid functionality suitable for conjugation, we proceeded to synthesize the 'Trojan horse'-type conjugates of these siderophores with antibiotic ciprofloxacin. To this end, we synthesized three *N*-benzyloxylactams **5a–c** as depicted in Scheme 2.



**Figure 2.** Transport of <sup>55</sup>Fe via compounds **1a** (DVD-000265), **1c** (DVD-00627), **1i** (DVD-000304 and **4** (DVD-000468) in *P. aeruginosa* cells; structures of the four lead siderophores; *P. aeruginosa* pyoverdine and pyochelin deficient strain ( $\Delta pvdF\Delta pchA$ ) was grown in CAA medium overnight as described in Materials and Methods. Afterwards, cells were washed with 50 mM Tris-HCl pH 8.0 and re-suspended in the same buffer to and OD<sub>600 nm</sub> (optical density measured at 600 nm) of 1. The iron transport assay was started by the addition of 500 nM <sup>55</sup>Fe-loaded compounds **1a**, **1c**, **1i** and **4** and, at different time points, aliquots were taken, bacteria were pelleted and the radioactivity retained in the cells was monitored. The  $\Delta pvdF\Delta pchA$  strain was also treated with 200 mM of CCCP protonophore and incubated in the presence of <sup>55</sup>Fe-loaded compounds (CCCP inhibits any proton motive dependent uptake).



Scheme 2. Synthesis of *N*-benzyloxylactams 5a-c.

Compounds **5a–c** were amidated using HATU and methyl 4-aminobutanoate [32]. Following basic hydrolysis, carboxylic acid building blocks **6a–c** were obtained in excellent yields over two steps. HATU-mediated coupling with ciprofloxacin methyl or benzyl ester gave compounds **7a–c** (in case of **7c**, the benzyl ester group underwent hydrolysis during the HPLC purification). Removal of the benzyl group from the *N*-benzyloxy lactam moiety and ester hydrolysis of intermediate **8a'** gave compound **8a**. Treatment of compound **7b** with BBr<sub>3</sub> led to the simultaneous deprotection of the *N*-hydroxylactam, the ester and the phenolic hydroxy group to give compound **8b**. Hydrogenation of **7b** and **7c** gave siderophore-ciprofloxacin conjugates **8c** and **8d**, respectively (Scheme 3).



Scheme 3. Synthesis of ciprofloxacin-siderophore conjugates 8a-d.

Siderophore-ciprofloxacin conjugates **8a–d** were tested for antibacterial activity against *P. aeruginosa* PAO1 (wild type) strain and the  $\Delta pvdF\Delta pchA$  mutant. Since the latter are not able to produce pyoverdine and pyochelin, they may be able to more efficiently use the synthetic siderophore-ciprofloxacin conjugates to get access to iron compared to the PAO1 strain. The minimum inhibitory concentrations (MIC) were first determined for both strains grown in Müller-Hinton (MH) medium [33]. For both strains, no antibacterial activity was detected, which is not surprising for bacteria grown in MH medium since the outer membrane transporters involved in iron acquisition are only expressed under iron-restricted growth conditions [30]. Consequently, we also determined the MIC values in iron-restricted conditions (CAA medium). Again, no antibiotic activity was observed against the PAO1 strain. However, for the  $\Delta pvdF\Delta pchA$  mutant, MICs of 1 mM were determined for all four siderophore-ciprofloxacin conjugates (Table 1).

**Table 1.** MIC values determined for ciprofloxacin-siderophore conjugates **8a–d** against *P. aeruginosa* PAO1 and  $\Delta pvdF\Delta pchA$  strains grown in MH or CAA (iron-restricted) medium (both strains were grown in the presence of increasing concentrations of **8a–d**; the MIC values were measured in  $\mu$ M after 48 h in triplicates; ciprofloxacin was used as the positive control displaying an MIC value of 48 ± 1.2  $\mu$ M).

Compound	MH Medium		CAA Medium	
	PAO1	$\Delta pvdF\Delta pchA$	PAO1	$\Delta pvdF\Delta pchA$
8a	>64	64	>64	1
8b	>64	64	>64	1
8c	>64	64	>64	1
8d	>64	64	>64	1

The observed antibacterial effect from siderophore-ciprofloxacin conjugates **8a–d** solely on the *P. aeruginosa*  $\Delta pvdF\Delta pchA$  mutant strain grown in (iron-restricted) medium may be indicative of these conjugates indeed acting as 'Trojan horse' antibiotics. The iron-starved bacteria unable to produce and excrete their own siderophores (pyoverdine and pyochelin) are likely to utilize the siderophoric constructs **8a–d** to shuttle the iron inside the cells. Once inside the bacteria, these constructs may already exert their antibiotic activity due to the presence of a ciprofloxacin payload. Unfortunately, we were unable to monitor the <sup>55</sup>Fe uptake with these compounds because of the solubility problems encountered.

#### 3. Materials and Methods

## 3.1. Bacterial Strains and Growth Assays in Iron-Restricted Conditions

The *P. aeruginosa* PAO1 and  $\Delta pvdF\Delta pchA$  strains were used in this study. PAO1 is a wild type strain and  $\Delta pvdF\Delta pchA$  is a strain unable to produce the siderophores pyoverdine and pyochelin [22]. Bacteria were first grown in LB medium overnight at 30 °C, and then washed and re-suspended in 20 mL iron-deficient CAA (casamino acid) medium containing 5 g·L<sup>-1</sup> low-iron CAA (Difco), 1.46 g·L<sup>-1</sup> K<sub>2</sub>HPO<sub>4</sub>·3H<sub>2</sub>O and 0.25 g·L<sup>-1</sup> MgSO<sub>4</sub>·7H<sub>2</sub>O and grown overnight at 30 °C. Bacteria were then washed, re-suspended in CAA medium at an optical density of 0.02 at 600 nm and distributed into the wells of a 96-well plate (Greiner, U-bottomed microplate) in the absence or in the presence of 10 µM or 100 µM of the tested compounds (5 mM stock solutions in DMSO were used). The plate was incubated at 30 °C, with shaking, in a TECAN microplate reader (Infinite M200, Tecan) (Tecan, Männedorf, Switzerland) and bacterial growth was monitored at OD<sub>600 nm</sub>. The presented data are the mean of three replicates for each measurement.

#### 3.2. Iron Uptake

Compound-<sup>55</sup>Fe complexes were prepared at <sup>55</sup>Fe concentrations of 50  $\mu$ M, with a compound:iron (mol:mol) ratio of 200:1 as described previously [22]. *P. aeruginosa* strains were successively grown an overnight in LB broth, followed by an overnight in CAA medium and finally overnight in CAA medium. All of these successive cultures were

carried out at 30 °C. The bacteria were subsequently used for <sup>55</sup>Fe uptake kinetics as described previously [22], with a concentration of <sup>55</sup>Fe-compound complexes of 500 nM and with cells pretreated or not with 200  $\mu$ M CCCP (a proton motive force inhibitor) [27].

## 3.3. MIC Determination

Overnight cultures were grown at 37 °C in Lysogeny broth (LB) and diluted to obtain an opacity equivalent to 0.5 on the McFarland scale. Screening vials were filled with solutions of the test compounds in 0.5% DMSO as prepared above with three replications for each treatment. Ciprofloxacin (0.5–256  $\mu$ g/mL) and 0.5% DMSO served as positive and negative controls, respectively. The entire vial was incubated at 35 ± 2 °C for 18 h. After incubation, the antibacterial activity of the test compounds was determined by measuring the absorption of the solution with a spectrophotometer on 500 nm.

The MICs were measured using the twofold serial broth dilution method. The test organisms were grown in suitable broth for 18 h at 37 °C. Twofold serial dilutions of solutions of the test compounds were prepared at 256, 128, 64, 32, 16, 8, 4, 2, 1 and  $0.5 \,\mu g \, ml^{-1}$ . The tubes were then inoculated with the test microbe; each 5 mL received 0.1 mL of the above inoculums and were incubated at 37 °C. The vials were subsequently observed for the presence or absence of microbial growth.

#### 3.4. Compounds Synthesis

NMR spectra were acquired with a 400 MHz Bruker Avance III spectrometer (400.13 MHz for <sup>1</sup>H, 376.49 MHz for <sup>19</sup>F and 100.61 MHz for <sup>13</sup>C) or 500 MHz Bruker Avance III (500.03 MHz for <sup>1</sup>H and 125.73 MHz for <sup>13</sup>C) in CDCl<sub>3</sub> or DMSO- $d_6$  and were referenced to residual solvent proton signals ( $\delta_{\rm H}$  = 7.26 and 2.50, respectively) and solvent carbon signals  $(\delta_{\rm C} = 77.16 \text{ and } 39.52, \text{ respectively})$ . Multiplicities are abbreviated as follows: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad, dd = doublet of doublets, dt = doublet of triplets, ddd = doublet/doublets of doublets; coupling constants, J, are reported in Hz. Mass spectra were acquired with an HRMS-ESI-qTOF spectrometer Nexera LCMS9030 (Shimadzu Europa GmbH, Duisburg, Germany) or MaXis II Bruker Daltonic GmbH (electrospray ionization mode, positive ions detection) (Bremen, Germany). Flash column chromatography on silica (Merck, 230–400 mesh) was performed with a Biotage Isolera Prime instrument. TLC was performed on aluminum-backed pre-coated plates (0.25 mm) with silica gel 60 F<sub>254</sub> with a suitable solvent system and was visualized using UV fluorescence. Preparative HPLC was carried out in a compact preparative system ECOM ECS28P00 equipped with a spectrophotometric detector or a Shimadzu LC-20AP. Column: YMC-Pack SIL-06, 5  $\mu$ m, 250  $\times$  20 mm or Agilent Zorbax prepHT XDB-C18, 5  $\mu$ m,  $21.2 \times 150$  mm. Chlorobenzene was distilled from P<sub>2</sub>O<sub>5</sub> and stored over molecular sieves 4 Å (>24 h). Compound **5a** was prepared as described previously [18].

3.4.1. (3*RS*,4*RS*)-2-Hydroxy-3-(2-hydroxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (4)

A mixture of homophthalic anhydride (162 mg, 1 mmol) and 2-hydroxybenzaldehyde oxime (137 mg, 1 mmol) was refluxed in toluene (2 mL) for 16 h in a screw-cap vial under stirring. After cooling to room temperature, filtration of the solid and washing it with ether afforded 170 mg, 57% of pure title compound as white solid. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  12.95 (s, 1H), 10.15 (s, 1H), 9.90 (s, 1H), 8.00–7.91 (m, 1H), 7.43–7.34 (m, 2H), 7.30–7.21 (m, 1H), 7.08–6.97 (m, 1H), 6.84 (d, *J* = 7.9 Hz, 1H), 6.65–6.51 (m, 2H), 5.71 (s, 1H), 4.15 (s, 1H). <sup>13</sup>C NMR (101 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  172.1, 160.7, 154.3, 133.1, 131.6, 130.0, 128.5, 128.5, 127.8, 126.5, 125.8, 124.0, 118.6, 115.3, 60.7, 49.1. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>16</sub>H<sub>13</sub>NO<sub>5</sub>Na<sup>+</sup> 322.0686, found 322.0682.

3.4.2. (*3RS,4RS*)-2-(Benzyloxy)-3-(2-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (**5b**)

A mixture of homophthalic anhydride (320 mg, 2 mmol) and 2-methoxybenzaldehyde *O*-benzyl oxime (482 mg, 1 mmol) was refluxed in toluene (2 mL) for 72 h in a screw-cap vial under stirring. Cooling to room temperature, filtration of the solid and washing it with ether afforded 250 mg, 31% of pure title compound as white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub> + DMSO- $d_6$ )  $\delta$  13.17 (s, 1H), 7.99 (dd, *J* = 7.4, 1.8 Hz, 1H), 7.51–7.40 (m, 4H), 7.40–7.28 (m, 4H), 7.25–7.17 (m, 1H), 7.04 (dd, *J* = 8.3, 1.0 Hz, 1H), 6.78–6.71 (m, 1H), 6.66 (dd, *J* = 7.8, 1.7 Hz, 1H), 5.95 (d, *J* = 1.7 Hz, 1H), 5.02 (s, 2H), 4.24 (d, *J* = 1.7 Hz, 1H), 3.89 (s, 3H). <sup>13</sup>C NMR (101 MHz, DMSO- $d_6$ )  $\delta$  171.8, 161.4, 156.2, 135.0, 133.3, 132.3, 130.2, 129.2, 129.1, 128.5, 128.3, 128.0, 128.0, 126.8, 125.7, 125.2, 120.1, 111.4, 75.9, 59.1, 55.7, 49.5. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>24</sub>H<sub>21</sub>NO<sub>5</sub><sup>+</sup> 426.1312, found 426.1305.

3.4.3. (3*RS*,4*RS*)-2-(Benzyloxy)-3-(2-fluorophenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (**5c**)

A mixture of homophthalic acid (3.6 g, 20 mmol), *O*-benzylhydroxylamine (2.5 g, 20 mmol) and 2-fluorobenzaldehyde (2.5 g, 20 mmol) was suspended in toluene (20 mL) and refluxed under stirring for 48 h with the azeotropic removal of water (Dean-Stark head). Cooling to room temperature, filtration of the solid and its crystallization from ACN afforded 1.65 g, 21% of pure title compound as white solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  13.28 (s, 1H), 8.06–7.99 (m, 1H), 7.56–7.19 (m, 10H), 7.03 (t, *J* = 7.4 Hz, 1H), 6.84 (t, *J* = 7.4 Hz, 1H), 6.00 (s, 1H), 5.05 (d, *J* = 10.0 Hz, 1H), 5.00 (d, *J* = 10.0 Hz, 1H), 4.35 (d, *J* = 1.6 Hz, 1H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  171.8, 161.9, 160.1 (d, *J* = 245.3 Hz), 135.4, 133.6, 133.0, 130.6, 129.8, 129.7, 129.1, 128.8, 128.7, 128.3, 127.5 (d, *J* = 3.6 Hz), 127.4, 125.5 (d, *J* = 13.1 Hz), 125.0 (d, *J* = 3.1 Hz), 116.4 (d, *J* = 21.0 Hz), 76.5, 58.8 (d, *J* = 3.0 Hz), 50.6. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>23</sub>H<sub>18</sub>NFO<sub>4</sub>Na<sup>+</sup> 414.1112, found 414.1109.

3.4.4. 4-((3*RS*,4*RS*)-2-(Benzyloxy)-3-(4-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoic acid (**6a**)

To a solution of rac-(3R,4R)-2-(benzyloxy)-3-(4-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (5a, 500 mg, 1.2 mmol) in DMF (2 mL) DIPEA (351 mg, 2.7 mmol) and HATU (518 mg, 1.4 mmol) were added at room temperature. After stirring for 5 min, methyl 4-aminobutanoate hydrochloride (201 mg, 1.4 mmol) was added and the mixture was left overnight. The reaction mixture was diluted with DCM and washed with water. The organic phase was dried over sodium sulfate, filtered and concentrated to give a crude product which was purified by flash column chromatography on silica (mobile phase CHCl<sub>3</sub>-MeOH 25:1) to give 486 mg of methyl 4-((3RS,4RS)-2-(benzyloxy)-3-(4-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoate, which was taken into the next step without full characterization. The obtained material was dissolved in aqueous MeOH (5 mL, 50%), cooled to 0 °C and treated with KOH (108 mg, 1.94 mmol). After stirring at room temperature for 16 h, the reaction mixture was acidified to pH 1 with HCl conc. and partitioned between EtOAc (20 mL) and water (10 mL). The aqueous layer was extracted with EtOAc (10 mL) and combined organics were dried over sodium sulfate, filtered and concentrated to give pure title compound in 72% yield (422 mg) as white foam. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) & 8.28–8.21 (m, 1H), 7.53–7.47 (m, 2H), 7.46–7.38 (m, 2H), 7.36–7.28 (m, 3H), 7.18–7.11 (m, 1H), 7.06 (d, J = 8.8 Hz, 2H), 6.73 (d, J = 8.8 Hz, 2H), 5.83 (d, J = 2.1 Hz, 1H), 5.47 (t, J = 5.9 Hz, 1H), 5.05 (d, J = 9.9 Hz, 1H), 5.01 (d, J = 10.0 Hz, 1H), 3.95 (d, J = 2.2 Hz, 1H), 3.70 (s, 3H), 3.32–3.20 (m, 2H), 2.25 (t, J = 6.9 Hz, 2H), 1.71 (p, J = 6.8 Hz, 2H). <sup>13</sup>C NMR (101 MHz, DMSO- $d_6$ )  $\delta$  174.1, 169.6, 161.0, 158.8, 135.2, 134.4, 132.5, 130.8, 129.0, 128.8, 128.4, 128.2, 128.1, 127.8, 127.6, 127.1, 113.8, 79.2, 75.6, 63.7, 55.0, 52.5, 38.3, 38.2, 30.9, 24.4. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>28</sub>H<sub>28</sub>N<sub>2</sub>O<sub>6</sub>Na<sup>+</sup> 511.1840, found 511.1840.

3.4.5. 4-((3*RS*,4*RS*)-2-(Benzyloxy)-3-(2-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoic acid (**6b**)

This compound was obtained following the same two-step procedure as for compound **6a** starting from (3*RS*,4*RS*)-2-(benzyloxy)-3-(2-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroiso-quinoline-4-carboxylic acid (**5b**, 845 mg, 2.1 mmol) in 92% yield (941 mg). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.28–8.22 (m, 1H), 7.51–7.42 (m, 4H), 7.34–7.27 (m, 3H), 7.20–7.12 (m, 1H), 7.12–7.06 (m, 1H), 6.89–6.80 (m, 2H), 6.71 (td, *J* = 7.6, 1.0 Hz, 1H), 6.21 (d, *J* = 1.9 Hz, 1H), 5.45 (t, *J* = 5.9 Hz, 1H), 5.10 (d, *J* = 9.9 Hz, 1H), 5.04 (d, *J* = 10.0 Hz, 1H), 4.14 (d, *J* = 2.0 Hz, 1H), 3.91 (s, 2H), 3.35–3.18 (m, 2H), 2.26 (t, *J* = 7.0 Hz, 2H), 1.78–1.66 (m, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  176.8, 170.5, 162.9, 156.5, 135.1, 133.2, 133.0, 129.6, 129.6, 129.2, 129.1, 129.0, 129.0, 128.6, 128.4, 126.8, 125.2, 120.7, 110.7, 60.4, 55.6, 51.8, 39.6, 38.8, 31.3, 24.4. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>28</sub>H<sub>28</sub>N<sub>2</sub>O<sub>6</sub>Na<sup>+</sup> 511.1840, found 511.1841.

3.4.6. 4-((3*RS*,4*RS*)-2-(Benzyloxy)-3-(2-fluorophenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoic acid (**6**c)

This compound was obtained following the same two-step procedure as for compound **6a** starting from (3*RS*,4*RS*)-2-(benzyloxy)-3-(2-fluorophenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxylic acid (**5c**, 782 mg, 2.0 mmol) in 90% yield (856 mg). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.26 (dd, *J* = 5.8, 3.5 Hz, 1H), 7.54–7.43 (m, 4H), 7.36–7.28 (m, 3H), 7.21–7.12 (m, 2H), 7.07–6.98 (m, 1H), 6.94–6.86 (m, 2H), 6.24 (d, *J* = 2.0 Hz, 1H), 5.44 (t, *J* = 5.9 Hz, 1H), 5.11 (d, *J* = 10.0 Hz, 1H), 5.05 (d, *J* = 10.0 Hz, 1H), 4.07 (d, *J* = 2.0 Hz, 1H), 3.28 (q, *J* = 6.8 Hz, 2H), 2.27 (t, *J* = 6.9 Hz, 2H), 1.79–1.67 (m, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  176.4, 169.7, 162.7, 160.2 (d, *J* = 246.3 Hz), 134.9, 133.4, 132.4, 129.8 (d, *J* = 8.3 Hz), 129.6, 129.4, 129.2, 129.1, 129.0, 128.8, 128.5, 127.7 (d, *J* = 3.4 Hz), 124.9, 124.8, 124.5 (d, *J* = 3.5 Hz), 115.9 (d, *J* = 21.2 Hz), 59.6 (d, *J* = 2.5 Hz), 52.7 (d, *J* = 2.2 Hz), 39.8, 31.2, 24.3. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>27</sub>H<sub>25</sub>FN<sub>2</sub>O<sub>5</sub>Na<sup>+</sup> 499.1640, found 499.1644.

3.4.7. Methyl 7-(4-((3*RS*,4*RS*)-2-(benzyloxy)-3-(4-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-1-cyclopropyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylate (**7a**)

To a solution of compound 6a (100 mg, 0.2 mmol), DMF (1 mL) DIPEA (60 mg, 0.47 mmol) and HATU (84 mg, 0.22 mmol) were added at room temperature. After stirring for 5 min, methyl methyl 1-cyclopropyl-6-fluoro-4-oxo-7-(piperazin-1-yl)-1,4-dihydroquinoline-3carboxylate hydrochloride (76 mg, 0.2 mmol) was added and the mixture was left overnight. The reaction mixture was diluted with EtOAc and washed with water. The organic phase was dried over sodium sulfate, filtered and concentrated to give the crude product which was purified by flash column chromatography on silica (mobile phase hex-acetone 5–100%, then acetone-MeOH) to give 113 mg, 67% of pure title compound. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.45 (s, 1H), 8.31 (t, *J* = 5.7 Hz, 1H), 7.98 (dd, *J* = 7.7, 1.5 Hz, 1H), 7.78 (d, J = 13.3 Hz, 1H), 7.54–7.48 (m, 1H), 7.47–7.38 (m, 2H), 7.34–7.28 (m, 6H), 7.17–7.10 (m, 2H), 6.87–6.79 (m, 2H), 5.44 (d, J = 3.1 Hz, 1H), 5.00 (d, J = 9.8 Hz, 1H), 4.90 (d, J = 9.9 Hz, 1H), 4.20 (d, J = 3.2 Hz, 1H), 3.74 (s, 3H), 3.71–3.68 (m, 1H), 3.67 (s, 3H), 3.65–3.59 (m, 2H), 3.57–3.47 (m, 2H), 3.26–3.07 (m, 6H), 2.28 (t, J = 7.3 Hz, 2H), 1.70–1.57 (m, 2H), 1.29–1.19 (m, 2H), 1.12–1.02 (m, 2H).  $^{13}$ C NMR (101 MHz, DMSO- $d_6$ )  $\delta$  170.9 (d, J = 133.2 Hz), 169.7, 164.9, 160.7, 158.8, 152.6 (d, J = 246.9 Hz), 148.3, 143.6 (d, J = 10.3 Hz), 138.0, 135.1, 134.5, 132.4, 130.9, 129.0, 128.8, 128.3 (d, J = 20.3 Hz), 128.2, 127.8, 127.6, 127.0, 126.9, 122.1, 122.0, 113.8, 111.6 (d, J = 22.3 Hz), 109.0, 106.5, 75.6, 63.5, 55.0, 52.3, 51.2, 49.8, 49.3, 44.6, 40.7, 38.4, 34.7, 29.6, 29.3, 24.6, 7.5. HRMS m/z [M+Na]<sup>+</sup> calculated for  $C_{46}H_{46}FN_5O_8Na^+$  838.3223, found 838.3255.

3.4.8. Benzyl 7-(4-(4-((3*RS*,4*RS*)-2-(benzyloxy)-3-(2-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-1-cyclopropyl-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylate (**7b**)

To a solution of compound **6b** (630 mg, 1.15 mmol) in DCM (20 mL) NEt<sub>3</sub> (0.5 mL) and HATU (480 mg, 1.26 mmol) were added at room temperature. After stirring for 5 min,

benzyl 1-cyclopropyl-6-fluoro-4-oxo-7-(piperazin-1-yl)-1,4-dihydroquinoline-3-carboxylate (550 mg, 1.31 mmol) was added and the mixture was left stirring overnight. The reaction mixture was diluted with water (30 mL) + conc. HCl (1 mL) and the resulting suspension was filtered. The solid material was washed with water (30 × 3 mL) and ether (20 × 2 mL) and dried in air to give pure title compound. Yield 1.00 g, 98%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.66 (s, 1H), 8.26–8.16 (m, 1H), 8.05 (d, *J* = 13.0 Hz, 1H), 7.47 (d, *J* = 4.1 Hz, 4H), 7.44–7.30 (m, 6H), 7.29–7.21 (m, 3H), 7.20–7.10 (m, 2H), 6.90–6.78 (m, 2H), 6.71 (t, *J* = 7.5 Hz, 1H), 6.20 (d, *J* = 1.8 Hz, 1H), 6.05 (t, *J* = 5.5 Hz, 1H), 5.38 (s, 2H), 5.07 (d, *J* = 10.1 Hz, 1H), 5.02 (d, *J* = 10.1 Hz, 1H), 4.13 (d, *J* = 1.8 Hz, 1H), 3.91 (s, 3H), 3.77 (s, 1H), 3.72 (s, 1H), 3.65–3.53 (m, 3H), 3.40–3.21 (m, 6H), 2.40–2.19 (m, 2H), 1.89–1.69 (m, 2H), 1.38 (d, *J* = 6.9 Hz, 2H), 1.18–1.08 (m, 2H). 177.2, 171.3, 170.7, 167.4, 162.6, 156.4, 153.7 (d, *J* = 251.5 Hz), 147.7, 145.5 (d, *J* = 10.4 Hz), 140.9, 139.2, 135.0, 133.3, 133.1, 129.4, 129.3, 129.0, 128.7, 128.7, 128.5, 128.4, 127.8, 127.1, 126.7, 125.1, 120.7, 120.3 (d, *J* = 7.9 Hz), 112.7, 112.5, 110.7, 108.1, 105.5 (d, *J* = 2.5 Hz), 65.4, 60.3, 55.6, 51.8, 51.8, 50.0 (d, *J* = 4.7 Hz), 49.6, 45.4, 41.6, 40.2, 35.6, 30.5, 24.1, 8.4, 8.4. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>52</sub>H<sub>50</sub>FN<sub>5</sub>O<sub>8</sub>Na<sup>+</sup> 914.3536, found 914.3533.

3.4.9. 7-(4-(4-((3*SR*,4*SR*)-2-(Benzyloxy)-3-(2-fluorophenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-1-cyclopropyl-6-fluoro-4-oxo-1,4dihydroquinoline-3-carboxylic acid (7c)

To a solution of compound 6c (240 mg, 0.5 mmol) in DCM (20 mL) DIPEA (195 mg, 1.5 mmol) and HATU (210 mg, 0.55 mmol) were added at room temperature. After stirring for 5 min, benzyl 1-cyclopropyl-6-fluoro-4-oxo-7-(piperazin-1-yl)-1,4-dihydroquinoline-3carboxylate hydrochloride (230 mg, 0.5 mmol) was added and the mixture was left stirring overnight. The reaction mixture was washed with water and HCl 1N. The organic phase was separated, dried and concentrated to give crude product, which was purified by flash column chromatography (DCM-MeOH) go give 333 mg of product with ~80% purity. This sample was further purified by preparative RP-HPLC (ACN-water + 0.1%TFA; gradient 20–90% of ACN in 35 min; 40 °C, 12 mL/min) to give 65 mg, 16% of pure title compound. The hydrolysis of the benzyl ester moiety probably occurred during HPLC separation. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.62 (s, 1H), 8.09 (t, *J* = 5.7 Hz, 1H), 8.03 (d, *J* = 7.6 Hz, 1H), 7.87 (d, J = 13.0 Hz, 1H), 7.60–7.44 (m, 3H), 7.42–7.19 (m, 8H), 7.08 (d, J = 7.4 Hz, 1H), 7.03 (d, J = 7.6 Hz, 1H), 5.77 (d, J = 3.4 Hz, 1H), 5.06 (d, J = 9.8 Hz, 1H), 4.90 (d, J = 9.8 Hz, 1H), 4.21 (d, J = 3.4 Hz, 1H), 3.85–3.59 (m, 4H), 3.53 (s, 2H), 3.28 (s, 3H), 3.22–3.07 (m, 2H), 2.25 (t, J = 7.2 Hz, 2H), 1.65 (q, J = 7.3 Hz, 2H), 1.31 (d, J = 6.9 Hz, 2H), 1.23–1.09 (m, 2H). <sup>13</sup>C NMR (101 MHz, DMSO- $d_6$ ) δ 176.8 (d, J = 2.3 Hz), 170.7, 169.8, 166.3, 161.4, 160.2 (d, *J* = 244.8 Hz), 153.4 (d, *J* = 249.1 Hz), 148.4, 145.3 (d, *J* = 10.2 Hz), 139.5, 135.4, 134.5, 133.3, 130.6 (d, J = 7.9 Hz), 129.5, 128.9, 128.7, 128.6, 128.1, 127.8, 126.0 (d, J = 12.6 Hz), 125.0, 119.2 (d, J = 7.7 Hz), 116.3 (d, J = 21.1 Hz), 111.4 (d, J = 23.0 Hz), 107.2, 106.9, 76.2, 58.8, 51.4, 50.0, 49.6, 44.9, 41.1, 38.9, 36.3, 29.7, 25.0, 8.0.. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>44</sub>H<sub>41</sub>F<sub>2</sub>N<sub>5</sub>O<sub>7</sub>Na<sup>+</sup> 812.2866, found 812.2872.

3.4.10. 1-Cyclopropyl-6-fluoro-7-(4-(4-((3*RS*,4*RS*)-2-hydroxy-3-(4-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**8a**)

In a round bottom flask equipped with a rubber septum and magnetic stir bar, compound **7a** (180 mg, 0.22 mmol) was dissolved in THF/MeOH mixture (10/30 mL) followed by the addition of Pd/C (10% wt) as catalyst (180 mg, 0.75 eq). Hydrogen gas (1 atm) was slowly bubbled through the reaction mixture via needle for 4 h and then the hydrogen inlet was removed and the reaction mixture was left stirring under atmosphere of hydrogen overnight. The reaction progress was monitored via HPLC. After completion of the reaction the reaction mixture was filtered through a pad of Celite, which was washed with MeOH 20 mL and hot DMF 50 mL. The filtrate was concentrated and the residue was treated with DCM-Et<sub>2</sub>O mixture (1:3) under ultrasound. After cooling to -20 °C the resulting solid was filtered to give 131 mg, 81% of methyl 1-cyclopropyl-6-fluoro-7-(4-(4-((3RS,4RS)-2-hydroxy-3-(4-methoxyphenyl)-1-oxo-

1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylate: <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  9.98 (s, 1H), 8.45 (s, 1H), 8.08 (t, *J* = 5.7 Hz, 1H), 7.93 (d, *J* = 1.6 Hz, 1H), 7.78 (d, *J* = 13.3 Hz, 1H), 7.48–7.36 (m, 3H), 7.21 (dd, *J* = 7.5, 1.4 Hz, 1H), 7.11–7.05 (m, 2H), 6.86–6.79 (m, 2H), 5.18 (d, *J* = 2.9 Hz, 1H), 4.04 (d, *J* = 3.0 Hz, 1H), 3.74 (s, 3H), 3.71–3.68 (m, 1H), 3.68 (s, 3H), 3.66–3.61 (m, 2H), 3.60–3.53 (m, 2H), 3.20 (d, *J* = 6.6 Hz, 4H), 3.12 (q, *J* = 7.6, 7.2 Hz, 2H), 2.29 (t, *J* = 7.3 Hz, 2H), 1.74–1.58 (m, 2H), 1.28–1.21 (m, 2H), 1.13–1.04 (m, 2H). <sup>13</sup>C NMR (126 MHz, DMSO- $d_6$ )  $\delta$  171.6, 170.3, 169.7, 164.9, 160.3, 158.6, 152.6 (d, *J* = 246.8 Hz), 148.3, 143.6 (d, *J* = 10.3 Hz), 138.0, 133.9, 131.8, 131.2, 129.1, 128.1, 127.7, 127.4, 126.7, 122.1, 122.0, 113.7, 111.6 (d, *J* = 22.7 Hz), 109.0, 106.6, 65.3, 64.9, 55.0, 52.5, 51.3, 44.6, 40.7, 38.4, 34.7, 29.3, 24.6, 15.1, 7.5. HRMS m/z [M+H]<sup>+</sup> calculated for C<sub>39</sub>H<sub>41</sub>FN<sub>5</sub>O<sub>8</sub><sup>+</sup> 726.2934, found 726.2925.

The latter compound (90 mg, 0.12 mmol) was added to a stirred mixture of MeOH (20 mL), water (10 mL) and NaOH (28 mg, 0.7 mmol) at room temperature. After stirring for 8 h, MeOH was evaporated from the mixture, the residue was cooled to 0 °C and acidified to pH 1 with conc. HCl followed by extraction with EtOAc ( $3 \times 30$  mL) to give 25 mg of crude product, which was purified by crystallization from DCM-Et<sub>2</sub>O to give 10 mg, 12% of pure title compound. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  15.18 (s, 1H), 9.97 (s, 1H), 8.68 (s, 1H), 8.07 (t, *J* = 5.6 Hz, 1H), 7.99–7.90 (m, 2H), 7.57 (d, *J* = 7.4 Hz, 1H), 7.48–7.35 (m, 2H), 7.25–7.19 (m, 1H), 7.12–7.05 (m, 2H), 6.86–6.80 (m, 2H), 5.18 (d, *J* = 2.9 Hz, 1H), 4.04 (d, *J* = 3.0 Hz, 1H), 3.87–3.78 (m, 1H), 3.76–3.71 (m, 1H), 3.70–3.63 (m, 4H), 3.58 (s, 2H), 3.33 (s, 4H), 3.12 (q, *J* = 6.7 Hz, 2H), 2.29 (t, *J* = 7.3 Hz, 2H), 1.74–1.60 (m, 2H), 1.36–1.27 (m, 2H), 1.22–1.14 (m, 2H). <sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  176.4, 170.4, 169.7, 165.9, 160.3, 158.6, 153.0 (d, *J* = 249.4 Hz), 148.1, 144.9 (d, *J* = 10.2 Hz), 139.2, 133.9, 131.8, 131.2, 129.1, 128.1, 127.7, 127.4, 126.7, 118.9, 118.8, 113.7, 111.0 (d, *J* = 23.0 Hz), 106.8, 106.6, 65.3, 55.0, 52.5, 49.5, 49.2, 44.5, 40.7, 38.4, 35.9, 29.3, 24.6, 7.6. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>38</sub>H<sub>38</sub>FN<sub>5</sub>O<sub>8</sub>Na<sup>+</sup> 734.2597, found 734.2596.

3.4.11. 1-Cyclopropyl-6-fluoro-7-(4-((4-((3*RS*,4*RS*)-2-hydroxy-3-(2-hydroxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**8b**)

Compound 7b (489 mg, 0.55 mmol) was placed in a round-bottom flask equipped with a magnetic stir bar and a dropping funnel capped with a calcium chloride filled drying tube followed by the addition of 100 mL of dry DCM. After cooling to 0  $^{\circ}$ C a solution of BBr<sub>3</sub> (635 mg, 2.54 mmol) in DCM (50 mL) was added dropwise to the flask within 30 min. After stirring for four days at room temperature the reaction mixture was poured onto 100 mL of crushed ice and stirred for 2h. The resulting solid was filtered, washed with water (10 mL) and THF (5 mL) to give pure title compound. Yield 23 mg, 15%. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ 15.19 (s, 1H), 9.99 (s, 1H), 9.88 (s, 1H), 8.67 (s, 1H), 8.00–7.90 (m, 2H), 7.83 (t, J = 5.6 Hz, 1H), 7.57 (d, J = 7.4 Hz, 1H), 7.48–7.36 (m, 2H), 7.36–7.31 (m, 1H), 7.10–7.00 (m, 1H), 6.85 (d, J = 8.0 Hz, 1H), 6.63–6.57 (m, 2H), 5.49 (d, J = 1.4 Hz, 1H), 3.95 (d, J = 1.5 Hz, 1H), 3.87–3.77 (m, 1H), 3.74–3.61 (m, 2H), 3.58–3.43 (m, 3H), 3.29 (s, 3H), 3.18–3.05 (m, 2H), 2.31–2.21 (m, 2H), 1.74–1.57 (m, 2H), 1.30 (d, J = 7.2 Hz, 2H), 1.22–1.12 (m, 2H). <sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>) δ 176.4, 176.4, 170.3, 170.3, 165.9, 159.6, 154.1, 152.9 (d, J = 249.4 Hz), 148.1, 144.9 (d, J = 10.2 Hz), 139.2, 133.8, 131.7, 128.7, 128.4 (d, J = 2.9 Hz), 127.7, 126.7, 125.7, 125.2, 118.8 (d, J = 7.7 Hz), 118.7, 115.2, 111.0 (d, J = 23.1 Hz), 106.8, 106.6 (d, J = 3.0 Hz), 60.0, 50.1, 49.4 (d, J = 26.6 Hz), 44.4, 40.7, 40.4, 38.2, 35.9, 29.1, 24.6, 7.6, 7.6. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>37</sub>H<sub>36</sub>FN<sub>5</sub>O<sub>8</sub>Na<sup>+</sup> 720.2440, found 720.2445.

3.4.12. 1-Cyclopropyl-6-fluoro-7-(4-(4-((3*SR*,4*SR*)-2-hydroxy-3-(2-methoxyphenyl)-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**8***c*)

In a round bottom flask equipped with a rubber septum and magnetic stir bar, compound **7b** (100 mg, 0.11 mmol) was dissolved in THF/MeOH mixture (10/30 mL) followed by the addition of Pd/C (10% wt) catalyst (60 mg, 0.5 eq). Hydrogen gas (1 atm) was slowly bubbled through reaction mixture via needle for 4h and then the hydrogen inlet

was removed and the reaction mixture was left stirring under atmosphere of hydrogen overnight. The reaction progress was monitored via HPLC. After completion of the reaction the reaction mixture was filtered through a pad of Celite, which was washed with THF (20 mL) and ACN (20 mL). This filtrate contained admixture only and was discarded. Celite was further washed with hot DMF (20 mL) to give 65 mg of crude product, which was purified by precipitation from DCM solution (10 mL) with Et<sub>2</sub>O to give 8c. Yield 30 mg, 38%. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 15.18 (s, 1H), 10.02 (s, 1H), 8.67 (s, 1H), 8.00–7.89 (m, 2H), 7.84 (t, J = 5.7 Hz, 1H), 7.57 (d, J = 7.4 Hz, 1H), 7.47–7.36 (m, 2H), 7.29 (d, J = 7.2 Hz, 1H), 7.22 (t, J = 7.7 Hz, 1H), 7.04 (d, J = 8.2 Hz, 1H), 6.77 (t, J = 7.5 Hz, 1H), 6.68 (d, J = 7.6 Hz, 1H), 5.49 (s, 1H), 3.91 (s, 1H), 3.88 (s, 3H), 3.82 (s, 1H), 3.67 (s, 3H), 3.53 (s, 2H), 3.30 (s, 3H), 3.12 (q, J = 6.5 Hz, 2H), 2.28 (t, J = 7.3 Hz, 2H), 1.75–1.61 (m, 2H), 1.31 (d, J = 6.9 Hz, 2H), 1.21–1.13 (m, 2H). <sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>) δ 176.3, 170.4, 170.2, 165.9, 159.7, 156.0, 152.9 (d, J = 249.4 Hz), 148.0, 144.9 (d, J = 9.9 Hz), 139.1, 133.7, 131.8, 128.9, 128.7, 128.4, 127.7, 126.8, 126.7, 125.6, 120.1, 118.8 (d, J = 8.0 Hz), 111.2, 111.0 (d, J = 23.1 Hz), 106.8, 106.6, 60.1, 55.7, 50.3, 49.5, 49.2, 44.5, 40.7, 38.3, 35.9, 29.2, 24.7, 7.6, 7.6. HRMS m/z [M+Na]+ calculated for C<sub>38</sub>H<sub>38</sub>FN<sub>5</sub>O<sub>8</sub>Na<sup>+</sup> 734.2597, found 734.2591.

3.4.13. 1-Cyclopropyl-6-fluoro-7-(4-(4-((3S,4S)-3-(2-fluorophenyl)-2-hydroxy-1-oxo-1,2,3,4-tetrahydroisoquinoline-4-carboxamido)butanoyl)piperazin-1-yl)-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**8d**)

In a round bottom flask equipped with a rubber septum and magnetic stir bar, compound 7c (53 mg, 0.07 mmol) was dissolved in MeOH (20 mL) followed by the addition of Pd/C (10% wt) catalyst (50 mg, 0.7 eq). Hydrogen gas (1 atm) was slowly bubbled through the reaction mixture via needle for 4h and then the hydrogen inlet was removed and the reaction mixture was left stirring under atmosphere of hydrogen overnight. The reaction progress was monitored via HPLC. After completion of the reaction the reaction mixture was filtered through a pad of Celite, which was washed with THF (20 mL). The filtrate was concentrated and the residue was purified by preparative RP-HPLC (ACN-water + 0.1% TFA; gradient 20–90% of ACN in 35 min; 40 °C, 12 mL/min) to give 18 mg, 41% of pure title compound as yellow oil. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.67 (s, 1H), 8.02–7.96 (m, 1H), 7.96–7.89 (m, 2H), 7.57 (d, J = 7.4 Hz, 1H), 7.53–7.40 (m, 2H), 7.37–7.27 (m, 2H), 7.27–7.18 (m, 1H), 7.13–7.04 (m, 1H), 6.95–6.86 (m, 1H), 5.51 (d, J = 2.9 Hz, 1H), 4.07 (d, J = 3.0 Hz, 1H), 3.87–3.78 (m, 1H), 3.67 (d, J = 5.4 Hz, 2H), 3.55 (s, 2H), 3.30 (s, 4H), 3.11 (q, J = 6.4 Hz, 2H), 2.25 (t, J = 7.3 Hz, 2H), 1.72–1.54 (m, J = 6.8 Hz, 2H), 1.36–1.27 (m, 2H), 1.18 (d, J = 3.8 Hz, 2H). <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ )  $\delta$ 176.4 (d, J = 2.4 Hz), 170.3, 169.5, 165.9, 160.3, 159.6 (d, J = 245.3 Hz), 153.0 (d, J = 249.3 Hz), 148.1, 144.9 (d, *J* = 10.2 Hz), 139.2, 133.5, 132.2, 129.9 (d, *J* = 8.5 Hz), 128.5, 128.1, 128.0, 127.4, 127.4, 127.0, 126.1 (d, J = 12.9 Hz), 124.4 (d, J = 3.0 Hz), 118.8 (d, J = 7.6 Hz), 115.7 (d, J = 21.3 Hz), 111.0 (d, J = 22.8 Hz), 106.8, 106.6 (d, J = 3.2 Hz), 59.7 (d, J = 2.9 Hz), 50.9, 49.5, 49.2, 44.5, 40.7, 38.4, 35.9, 29.2, 24.5, 7.6. HRMS m/z [M+Na]<sup>+</sup> calculated for C<sub>37</sub>H<sub>35</sub>F<sub>2</sub>N<sub>5</sub>O<sub>7</sub>Na<sup>+</sup> 722.2397, found 722.2393.

#### 4. Conclusions

The screening of a library of novel *N*-hydroxylactams amenable by the Castagnoli-Cushman reaction identified four lead compounds that exerted a minimal effect on the growth of *P. aeruginosa*  $\Delta pvdF\Delta pchA$  mutant strain grown in (iron-restricted) CAA medium and transported <sup>55</sup>Fe into *P. aeruginosa* cells. Notably, one of these synthetic siderophores was found to be as efficient at promoting iron uptake as the natural siderophores pyoverdine, pyochelin or enterobactin. Conjugates of the four lead siderophores with ciprofloxacin were synthesized via sequential amide coupling and protecting group manipulation. These were tested for antibacterial activity against *P. aeruginosa* POA1 (wild type) and the  $\Delta pvdF\Delta pchA$  mutant strain. For bacteria grown under non-restricted iron conditions (in Müller-Hinton medium), no antibacterial activity was observed. Likewise, none of the siderophore-ciprofloxacin constructs was active against the POA1 strain grown in iron-restricted conditions (CAA medium). However, the antibacterial activity was found to be pronounced against the  $\Delta pvdF\Delta pchA$  mutant strain grown in CAA medium. This may be indicative of the uptake of these compounds by the iron-starved bacteria and of their being 'Trojan horse' antibiotics. Further scrutiny of the mechanism of the antibacterial action of the newly developed conjugates is warranted.

**Supplementary Materials:** The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/molecules27123910/s1. Structures of the compounds screened. Copies of NMR spectra. Bacterial growth graphs. Figure S1: Growth of  $\Delta$ pvdF $\Delta$ pchA P. aeruginosa mutant in the absence or presence of 10  $\mu$ M and 100  $\mu$ M of the different compounds in iron-restricted medium. The growth medium used was CAA and growth was followed by monitoring optical density (OD) at 600 nm at 30 °C.

**Author Contributions:** Conceptualization, M.K.; methodology, O.B.; investigation, E.L., A.B. and O.B.; data curation, O.B., M.K. and E.L.; writing—original draft preparation, M.K.; writing—review and editing, O.B.; supervision, O.B. and M.K.; funding acquisition, M.K. All authors have read and agreed to the published version of the manuscript.

**Funding:** We gratefully acknowledge Megagrant of the Government of Russian Federation (# 075-15-2021-637).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Data available from the corresponding authors upon reasonable request.

Acknowledgments: We are grateful to the Research Centre for Magnetic Resonance and the Centre for Chemical Analysis and Materials Research of Saint Petersburg State University Research Park for the analytical data. The authors are indebted to Sarah Fritsch and Isabelle J. Schalk of University of Strasbourg for their help in evaluating compounds' iron transport and antibacterial properties in mutant bacterial strains.

**Conflicts of Interest:** The authors declare that they have no conflicts of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results.

Sample Availability: Samples of the compounds 8a-d are available from the authors.

#### References

- Schalk, I.J.; Hannauer, M.; Braud, A. New roles for bacterial siderophores in metal transport and tolerance. *Environ. Microbiol.* 2011, 13, 2844–2854. [CrossRef] [PubMed]
- Saha, R.; Saha, N.; Donofrio, R.S.; Bestervelt, L.L. Microbial siderophores: A mini review. J. Basic Microbiol. 2013, 53, 303–317. [CrossRef] [PubMed]
- Sontag, B.; Gerlitz, M.; Paululat, T.; Rasser, H.F.; Grun-Wollny, I.; Hansske, F.G. Oxachelin, a Novel Iron Chelator and Antifungal Agent from *Streptomyces* sp. GW9/1258. J. Antibiot. 2006, 59, 659–663. [CrossRef] [PubMed]
- Kodani, S.; Bicz, J.; Song, L.; Deeth, R.J.; Ohnishi-Kameyama, M.; Yoshida, M.; Ochi, K.; Challis, G.L. Structure and biosynthesis of scabichelin, a novel tris-hydroxamate siderophore produced by the plant pathogen *Streptomyces scabies* 87.22. *Org. Biomol. Chem.* 2013, 11, 4686–4694. [CrossRef] [PubMed]
- Gräfe, U.; Ritzau, M.; Ihn, W.; Möllmann, U.; Fleck, W.F.; Groth, J.; Reissbrodt, R. A new microbial isoquinoline iron chelator from Streptomyces spec. 2002-104. J. Basic Microbiol. 1994, 34, 351–355. [CrossRef]
- Zhang, Y.-M.; Xiang, B.; Yang, S.-M.; Rhodes, K.; Scannevin, R.; Jackson, P.; Chakravarty, D.; Fan, X.; Wilson, L.J.; Karnachi, P. Heterocyclic Derived Metalloprotease Inhibitors. PCT Int. Appl. WO 2008045668. *Chem. Abstr.* 2008, 148, 471882.
- Zhang, Y.-M.; Fan, X.; Yang, S.-M.; Scannevin, R.H.; Burke, S.L.; Rhodes, K.J.; Jackson, P.F. Syntheses and in vitro evaluation of arylsulfone-based MMP inhibitors with heterocycle-derived zinc-binding groups (ZBGs). *Bioorg. Med. Chem. Lett.* 2008, 18, 405–408. [CrossRef]
- 8. Pryde, D.C.; Webster, R.; Butler, S.L.; Murray, E.J.; Whitby, K.; Pickford, C.; Westby, M.; Palmer, M.J.; Bull, D.J.; Vuong, H.; et al. Discovery of an HIV integrase inhibitor with an excellent resistance profile. *MedChemComm* **2013**, *4*, 709–719. [CrossRef]
- 9. Mutule, I.; Borovika, D.; Rozenberga, E.; Romanchikova, N.; Zalubovskis, R.; Shestakova, I.; Trapencieris, P. 5-Membered cyclic hydroxamic acids as HDAC inhibitors. *J. Enzyme Inhib. Med. Chem.* **2015**, *30*, 216–223. [CrossRef]
- Tang, D.G.; Li, L.; Zhu, Z.; Joshi, B.; Johnson, C.R.; Marnett, L.J.; Honn, K.V.; Crissman, J.D.; Krajewski, S.; Reed, J.C.; et al. BMD188, A novel hydroxamic acid compound, demonstrates potent anti-prostate cancer effects vitro and vivo by inducing apoptosis: Requirements for mitochondria, reactive oxygen species, and proteases. *Pathol. Oncol. Res.* 1998, 4, 179–190. [CrossRef]

- 11. Johnstone, T.C.; Nolan, E.M. Beyond iron: Non-classical biological functions of bacterial siderophores. *Dalton Trans.* **2015**, *44*, 6320–6339. [CrossRef] [PubMed]
- 12. Dassonville-Klimpt, A.; Sonnet, P. Advances in 'Trojan horse' strategies in antibiotic delivery systems. *Future Med. Chem.* **2022**, *12*, 983–986. [CrossRef] [PubMed]
- Schalk, I.J. A Trojan-Horse Strategy Including a Bacterial Suicide Action for the Efficient Use of a Specific Gram-Positive Antibiotic on Gram-Negative Bacteria. J. Med. Chem. 2018, 61, 3842–3844. [CrossRef] [PubMed]
- 14. Ferreira, K.; Hu, H.-Y.; Fetz, V.; Prochnow, H.; Rais, B.; Müller, P.P.; Brönstrup, M. Multivalent Siderophore-DOTAM Conjugates as Theranostics for Imaging and Treatment of Bacterial Infections. *Angew. Chem. Int. Ed.* **2017**, *56*, 8272–8276. [CrossRef]
- Katsube, T.; Echols, R.; Ferreira, J.C.A.; Krenz, H.K.; Berg, J.K.; Galloway, C. Cefiderocol, a Siderophore Cephalosporin for Gram-Negative Bacterial Infections: Pharmacokinetics and Safety in Subjects With Renal Impairment. *J. Clin. Pharmacol.* 2017, 57, 584–591. [CrossRef]
- Bakulina, O.; Bannykh, A.; Dar'in, D.; Krasavin, M. Cyclic Hydroxamic Acid Analogues of Bacterial Siderophores as Iron-Complexing Agents prepared through the Castagnoli–Cushman Reaction of Unprotected Oximes. *Chem. Eur. J.* 2017, 23, 17667–17673. [CrossRef]
- 17. Bannykh, A.V.; Bakulina, O.Y.; Dar'in, D.V.; Krasavin, M. Hydroxylamine as an ammonia equivalent: Access to NH-tetrahydroisoquinolonic derivatives from aldoximes by the Castagnoli–Cushman reaction followed by reduction. *Mendeleev Commun.* **2019**, *29*, 337–338. [CrossRef]
- Chupakhin, E.; Bakulina, O.; Dar'in, D.; Krasavin, M. Facile Access to Fe(III)-Complexing Cyclic Hydroxamic Acids in a Three-Component Format. *Molecules* 2019, 24, 864. [CrossRef]
- Negash, K.H.; Norris, J.K.S.; Hodgkinson, J.T. Siderophore–Antibiotic Conjugate Design: New Drugs for Bad Bugs? *Molecules* 2019, 24, 3314. [CrossRef]
- 20. Sanderson, T.J.; Black, C.M.; Southwell, J.W.; Wilde, E.J.; Pandey, A.; Herman, R.; Thomas, G.H.; Boros, E.; Duhme-Klair, A.-K.; Routledge, A. A Salmochelin S4-Inspired Ciprofloxacin Trojan Horse Conjugate. *ACS Infect. Dis.* **2020**, *6*, 2532. [CrossRef]
- Cunrath, O.; Gasser, V.; Hoegy, F.; Reimmann, C.; Guillon, L.; Schalk, I. A cell biological view of the siderophore pyochelin iron uptake pathway in *Pseudomonas aeruginosa*. J. Environ. Microbiol. 2015, 17, 171–185. [CrossRef] [PubMed]
- 22. Menhart, N.; Thariath, A.; Viswanatha, T. Characterization of the pyoverdines of *Azotobacter vinelandii* ATCC 12 837 with regard to heterogeneity. *Biol. Metals* 1991, 4, 223–232. [CrossRef] [PubMed]
- Brandel, J.; Humbert, N.; Elhabiri, M.; Schalk, I.J.; Mislin, G.L.A.; Albrecht-Gary, A.-M. Pyochelin, a siderophore of Pseudomonas aeruginosa: Physicochemical characterization of the iron(iii), copper(ii) and zinc(ii) complexes. *Dalton Trans.* 2012, 41, 2820–2834. [CrossRef] [PubMed]
- Gasser, V.; Baco, E.; Cunrath, O.; Saint August, P.; Perraud, Q.; Zill, N.; Schleberger, C.; Schmidt, A.; Paulen, A.; Bumann, D.; et al. Catechol siderophores repress the pyochelin pathway and activate the enterobactin pathway in *Pseudomonas aeruginosa*: An opportunity for siderophore–antibiotic conjugates development. *J. Environ. Microbiol.* 2016, *18*, 819–832. [CrossRef] [PubMed]
- 25. Marshall, B.; Stintzi, A.; Gilmour, C.; Meyer, J.-M.; Poole, K. Citrate-mediated iron uptake in *Pseudomonas aeruginosa*: Involvement of the citrate-inducible FecA receptor and the FeoB ferrous iron transporter. *Microbiology* **2009**, *155*, 305–315. [CrossRef] [PubMed]
- Lau, C.K.Y.; Krewulak, K.D.; Vogel, H.J. Bacterial ferrous iron transport: The Feo system. FEMS Microbiol. Rev. 2016, 40, 273–298. [CrossRef] [PubMed]
- Woronowicz, K.; Olubanjo, O.B.; Sha, D.; Kay, J.M.; Niederman, R.A. Effects of the protonophore carbonyl-cyanide mchlorophenylhydrazone on intracytoplasmic membrane assembly in *Rhodobacter sphaeroides*. *Biochim. Biophys. Acta* 2015, 1847, 1119–1128. [CrossRef]
- Noinaj, N.; Guillier, M.; Barnard, T.J.; Buchanan, S.K. TonB-Dependent Transporters: Regulation, Structure, and Function. Annu. Rev. Microbiol. 2010, 64, 43–60. [CrossRef]
- Clément, E.; Mesini, P.J.; Pattus, F.; Abdallah, M.A.; Schalk, I. The binding mechanism of pyoverdin with the outer membrane receptor FpvA in Pseudomonas aeruginosa is dependent on its iron-loaded status. J. Biochemistry 2004, 43, 7954–7965. [CrossRef]
- Perraud, Q.; Cantero, P.; Munier, M.; Hoegy, F.; Zill, N.; Gasser, V.; Mislin, G.L.A.; Ehret-Sabatier, L.; Schalk, I. Phenotypic Adaptation of *Pseudomonas aeruginosa* in the Presence of Siderophore-Antibiotic Conjugates during Epithelial Cell Infection. *J. Microorganisms* 2020, *8*, 1820. [CrossRef]
- 31. Schirmer, T. General and Specific Porins from Bacterial Outer Membranes. J. Struct. Biol. 1998, 121, 101–109. [CrossRef] [PubMed]
- Wencewicz, T.A.; Mollmann, U.; Long, T.E.; Miller, M.J. Syntheses and biological studies of the naturally occurring salmycin "Trojan Horse" antibiotics and synthetic desferridanoxamine-antibiotic conjugates. *Biometals* 2009, 22, 633–648. [CrossRef] [PubMed]
- Mueller, J.H.; Hinton, A. Protein-Free Medium for Primary Isolation of the Gonococcus and Meningococcus. J. Exp. Biol. Med. 1941, 48, 330–333. [CrossRef]