Supplementary Materials

Spherical Al-MCM-41 doped with copper by modified TIE method as effective catalyst for low-temperature NH₃-SCR

Aleksandra Jankowska¹, Andrzej Kowalczyk¹, Małgorzata Rutkowska¹,

Marek Michalik², Lucjan Chmielarz^{1*}

¹Jagiellonian University in Kraków, Faculty of Chemistry, Gronostajowa 2, 30-387

Kraków, Poland

²Jagiellonian University in Kraków, Institute of Geological Sciences, Gronostajowa 3a, 30-387

Kraków, Poland

*Corresponding author. Tel.: +48 126862417, fax: +48 12 686 2750.

E-mail address: chmielar@chemia.uj.edu.pl (L. Chmielarz)

For the 100Cu-A sample, which was found to be the most NH₃-SCR catalyst in, additonal verification of its catalytic stability was done by three subsequent catalytuc cycles. The catalytic runs were carried out in a flow fixed-bed quartz microreactor under atmospheric pressure uder thes same conditions as in standard catalytic tests. The flow rate and composition of gas mixture was adjusted and controlled by mass flow controllers (Brooks Instruments, 19440-0903 Hatfield, PA, USA). The reactant concentrations were continuously monitored using a quadrupole mass spectrometer (QMS, PREVAC, Rogów, Poland) connected directly to the reactor outlet. Prior to the catalytic tests, the sample of 100 mg with the particle's sizes in the range of 0.250-0.315 mm was placed in the quartz microreactor and outgassed in a flow of pure helium at 550°C for 30 min. The gas mixture containing 0.25 vol.% NO, 0.25 vol.% NH₃ and 2.5 vol.% O₂ diluted in pure helium (total flow rate of 40 mL·min⁻¹) was used. The reaction was studied in the temperature range from room temperature to 550°C with intervals of 25°C (only te region from 100 to 400°C is presnted).



Figure S1. Temperature dependence of the NO conversion and N₂ selectivity in NH₃-SCR for the 100Cu-A sample performed in the 3 cycles.