

Interaction study between ESIPT fluorescent lipophile-based benzazoles and BSA

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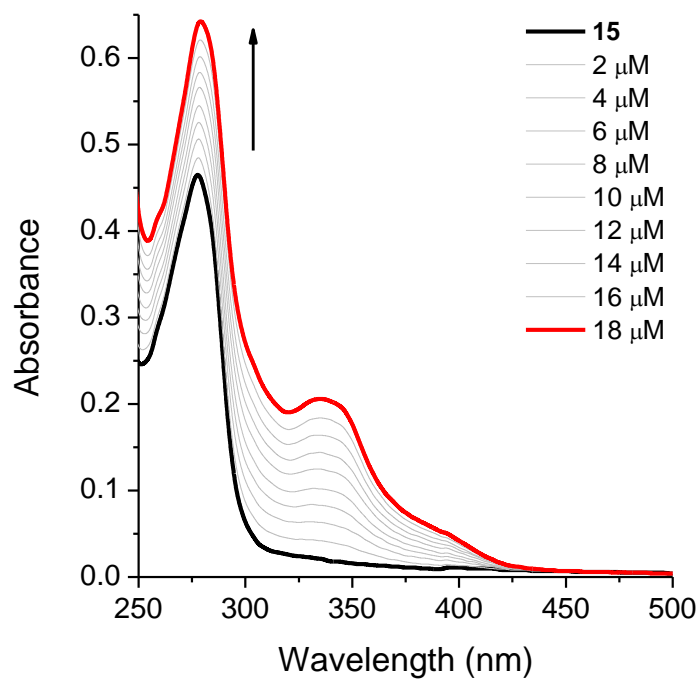


Figure S1. UV-Vis absorption spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **15** (0 to 20 μM).

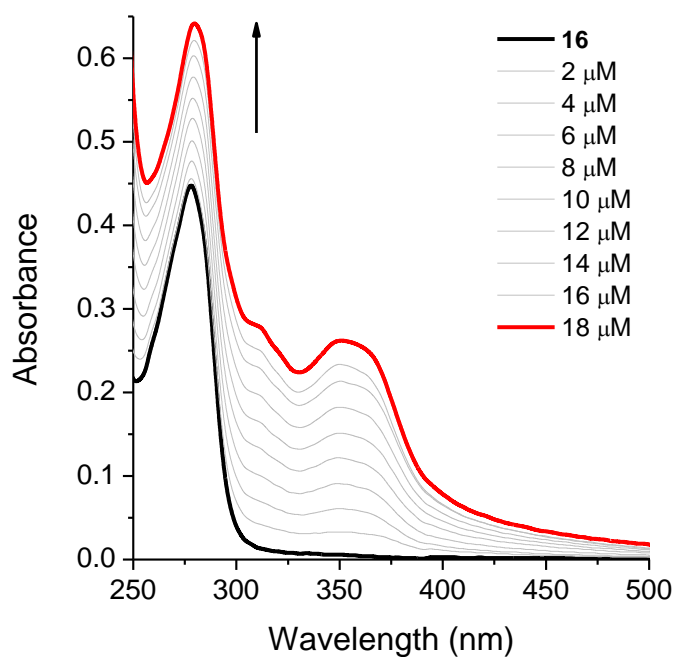


Figure S2. UV-Vis absorption spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **16** (0 to 20 μM).

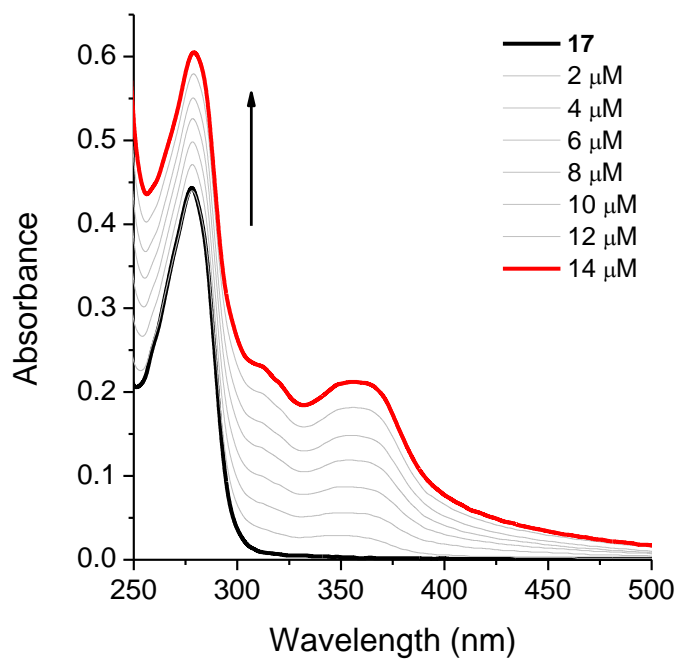


Figure S3. UV-Vis absorption spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **17** (0 to 20 μM).

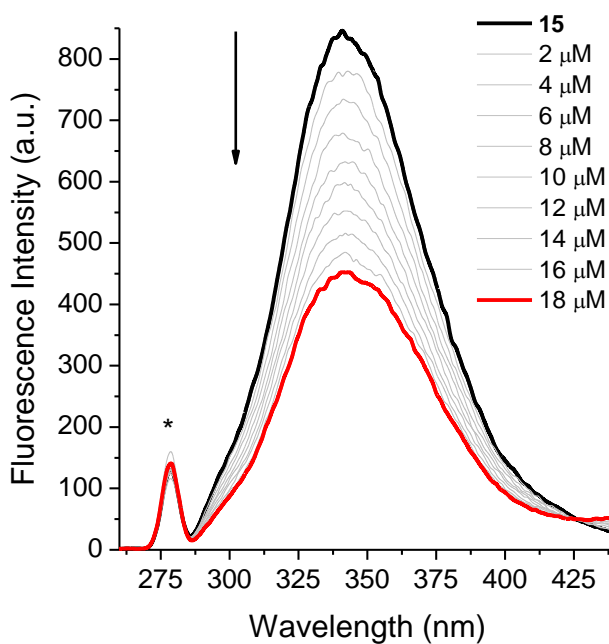


Figure S4. Steady-state fluorescence emission spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **15** (0 to 20 μM). The asterisk is related to the excitation wavelength. ($\lambda_{\text{ex}} = 277 \text{ nm}$, Slits (Ex/Em) = 5.0 nm/5.0 nm)

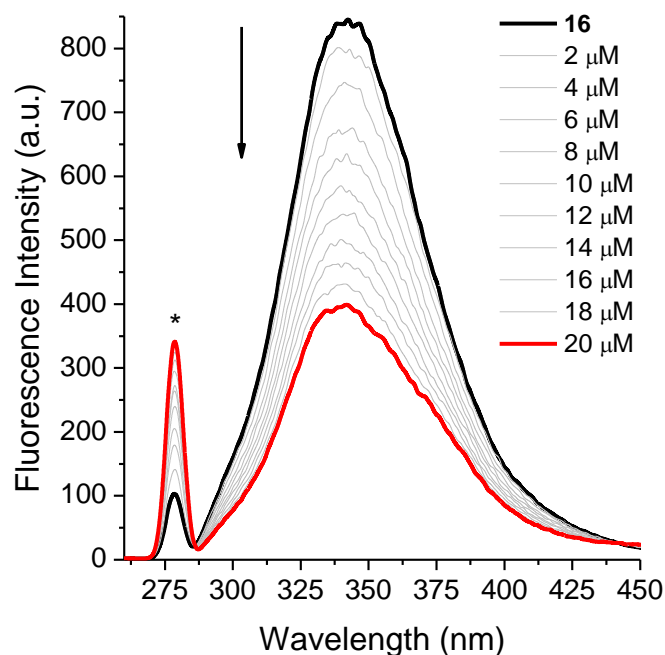


Figure S5. Steady-state fluorescence emission spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **16** (0 to 20 μM). The asterisk is related to the excitation wavelength. ($\lambda_{\text{ex}} = 277 \text{ nm}$, Slits (Ex/Em) = 5.0 nm/5.0 nm)

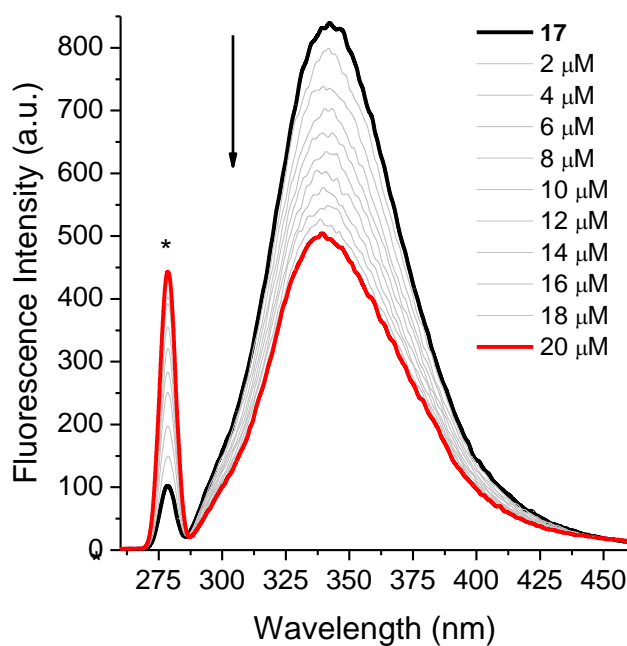


Figure S6. Steady-state fluorescence emission spectra for BSA (11 μM) in PBS (pH 7.2) in the absence and presence of different amounts of compound **17** (0 to 20 μM). The asterisk is related to the excitation wavelength. ($\lambda_{\text{ex}} = 277 \text{ nm}$, Slits (Ex/Em) = 5.0 nm/5.0 nm)

Table S1. Linear equations and respective coefficient of determination (R^2) from the plot F_0/F vs. $[Q]$ (Stern-Volmer relation) and $\log [(F_0-F)/F]$ vs. $\log [Q]$ (double log plot).

Benzazole	Stern-Volmer relation		Double log plot	
	Equation	R^2	Equation	R^2
12	$F_0/F=3.14 \times 10^4 [Q]+1.0022$	0.9969	$\log(F_0-F/F)=4.10+0.92 \log [Q]$	0.9928
13	$F_0/F=5.12 \times 10^4 [Q]+0.9746$	0.9986	$\log(F_0-F/F)=5.50+1.16 \log [Q]$	0.9953
14	$F_0/F=3.18 \times 10^4 [Q]+1.0180$	0.9927	$\log(F_0-F/F)=3.91+0.88 \log [Q]$	0.9878
15	$F_0/F=4.66 \times 10^4 [Q]+0.9663$	0.9916	$\log(F_0-F/F)=5.35+1.15 \log [Q]$	0.9983
16	$F_0/F=5.62 \times 10^4 [Q]+0.9237$	0.9859	$\log(F_0-F/F)=6.18+1.31 \log [Q]$	0.9989
17	$F_0/F=3.46 \times 10^4 [Q]+0.9890$	0.9975	$\log(F_0-F/F)=4.95+1.09 \log [Q]$	0.9933

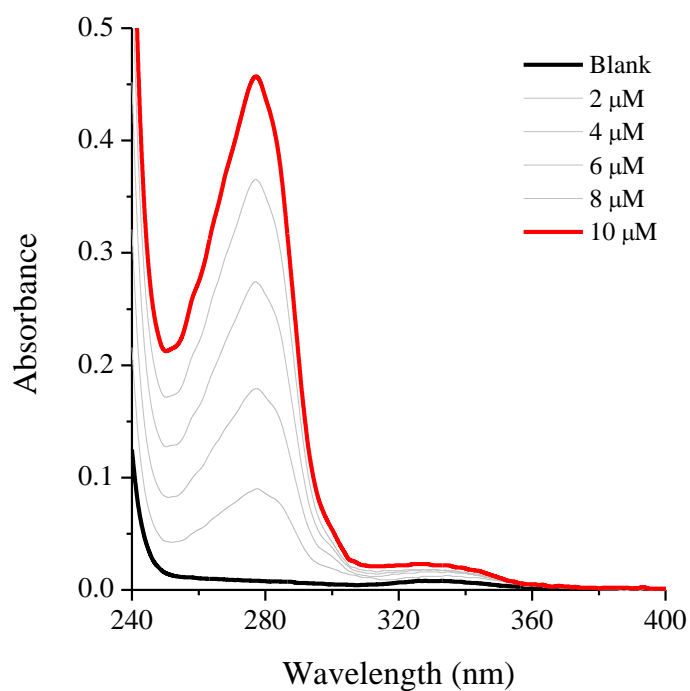


Figure S7. UV-Vis spectra of the benzazole 13 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

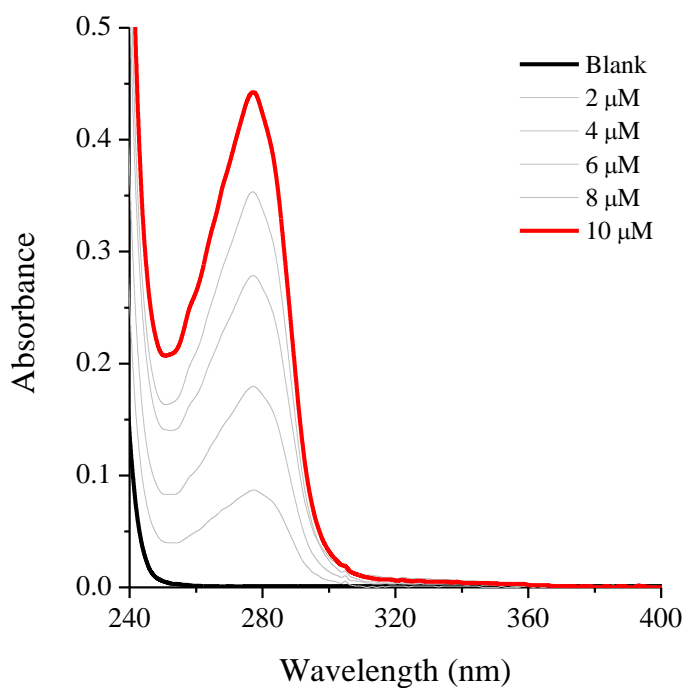


Figure S8. UV-Vis spectra of the benzazole 14 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

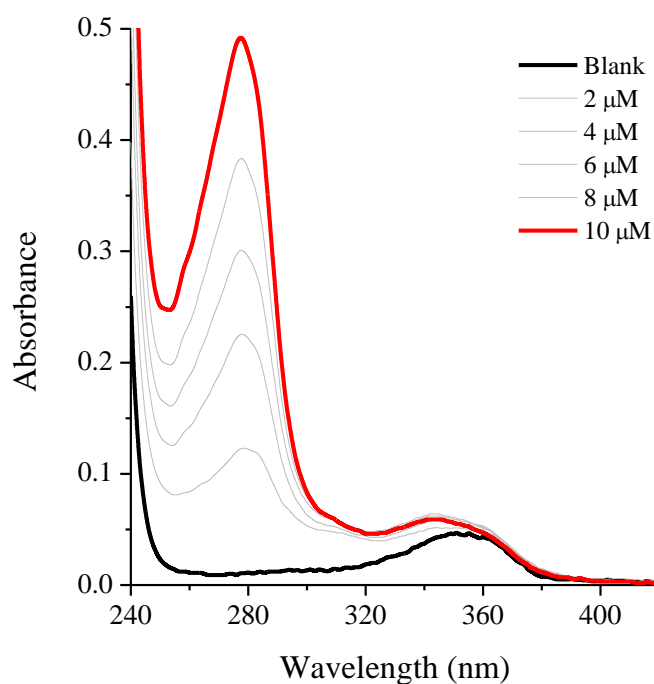


Figure S9. UV-Vis spectra of the benzazole 16 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

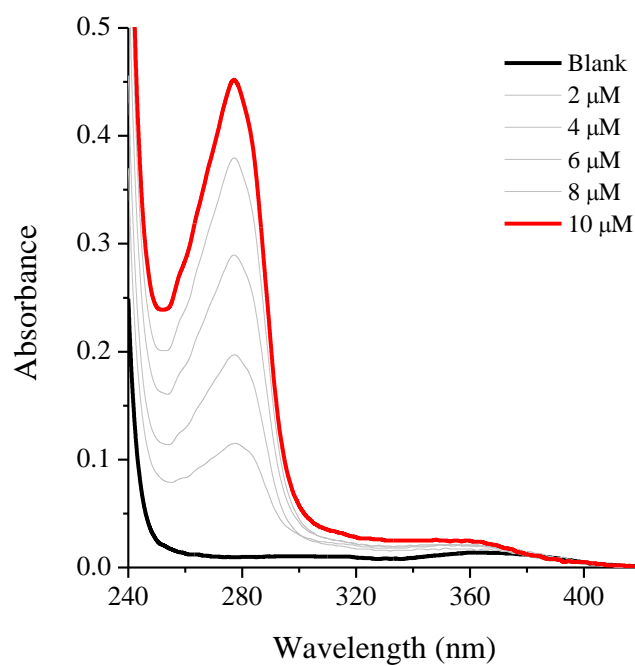


Figure S10. UV-Vis spectra of the benzazole 17 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

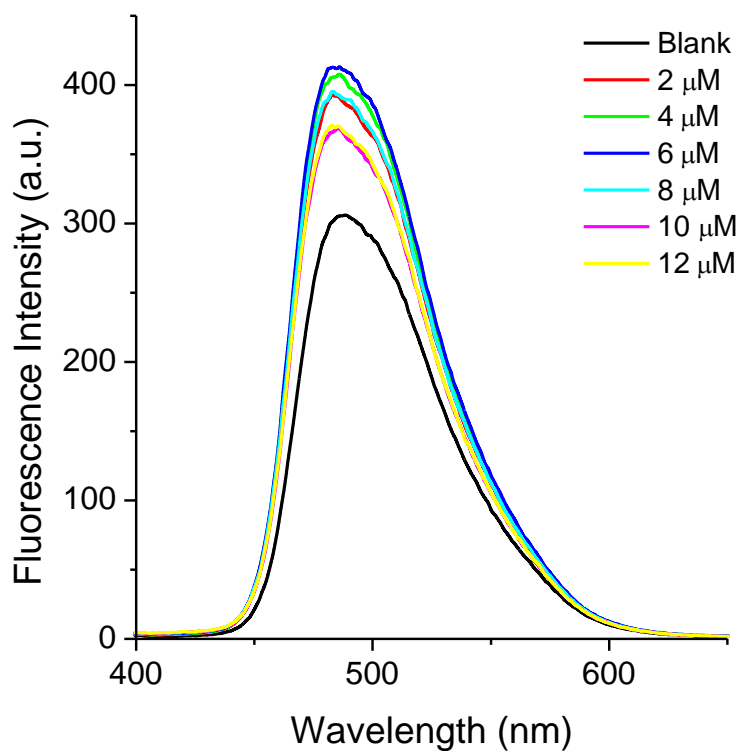


Figure S11. Fluorescence emission spectra of the benzazole 13 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

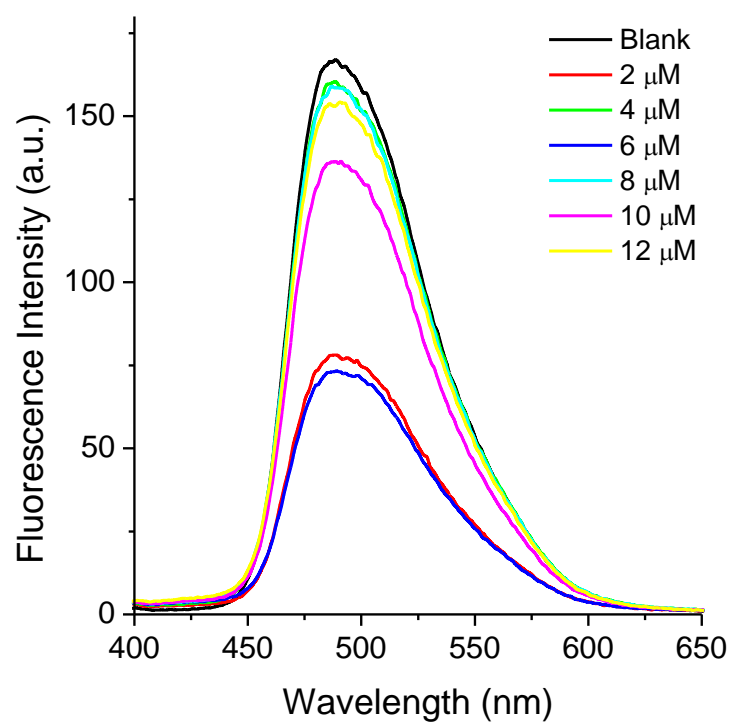


Figure S12. Fluorescence emission spectra of the benzazole 14 [2 μM] in different BSA concentrations [0-10 μM]. Blank sample is the UV-VIs in absence of BSA.

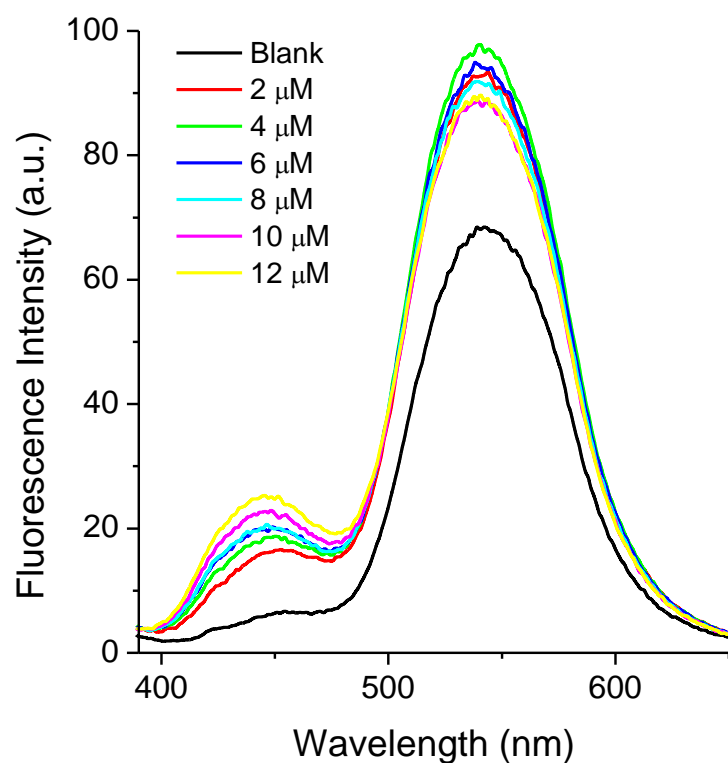


Figure S13. Fluorescence emission spectra of the benzazole 16 [2 μ M] in different BSA concentrations [0-10 μ M]. Blank sample is the UV-VIs in absence of BSA.

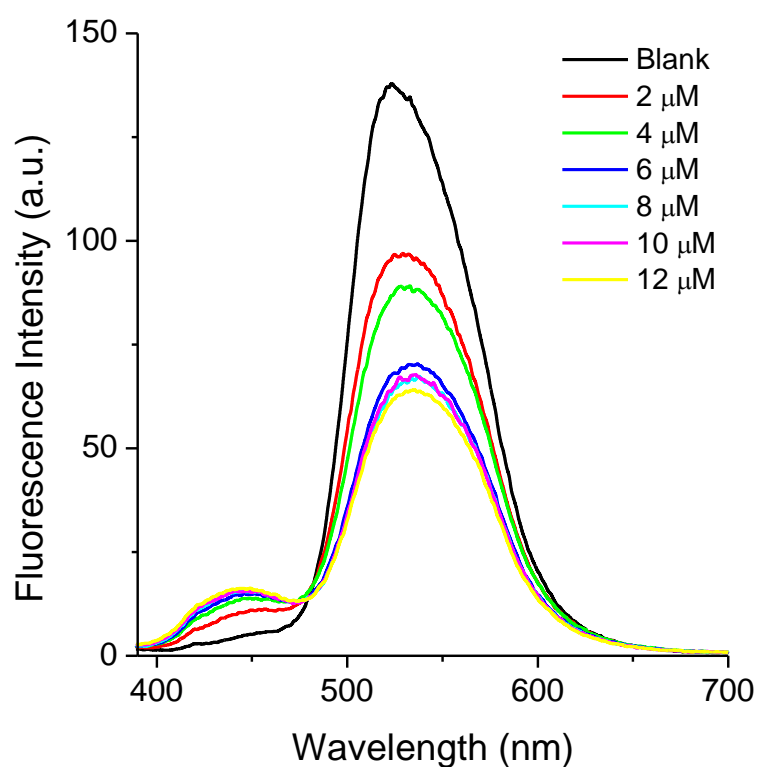


Figure S14. Fluorescence emission spectra of the benzazole 17 [2 μ M] in different BSA concentrations [0-10 μ M]. Blank sample is the UV-VIs in absence of BSA.

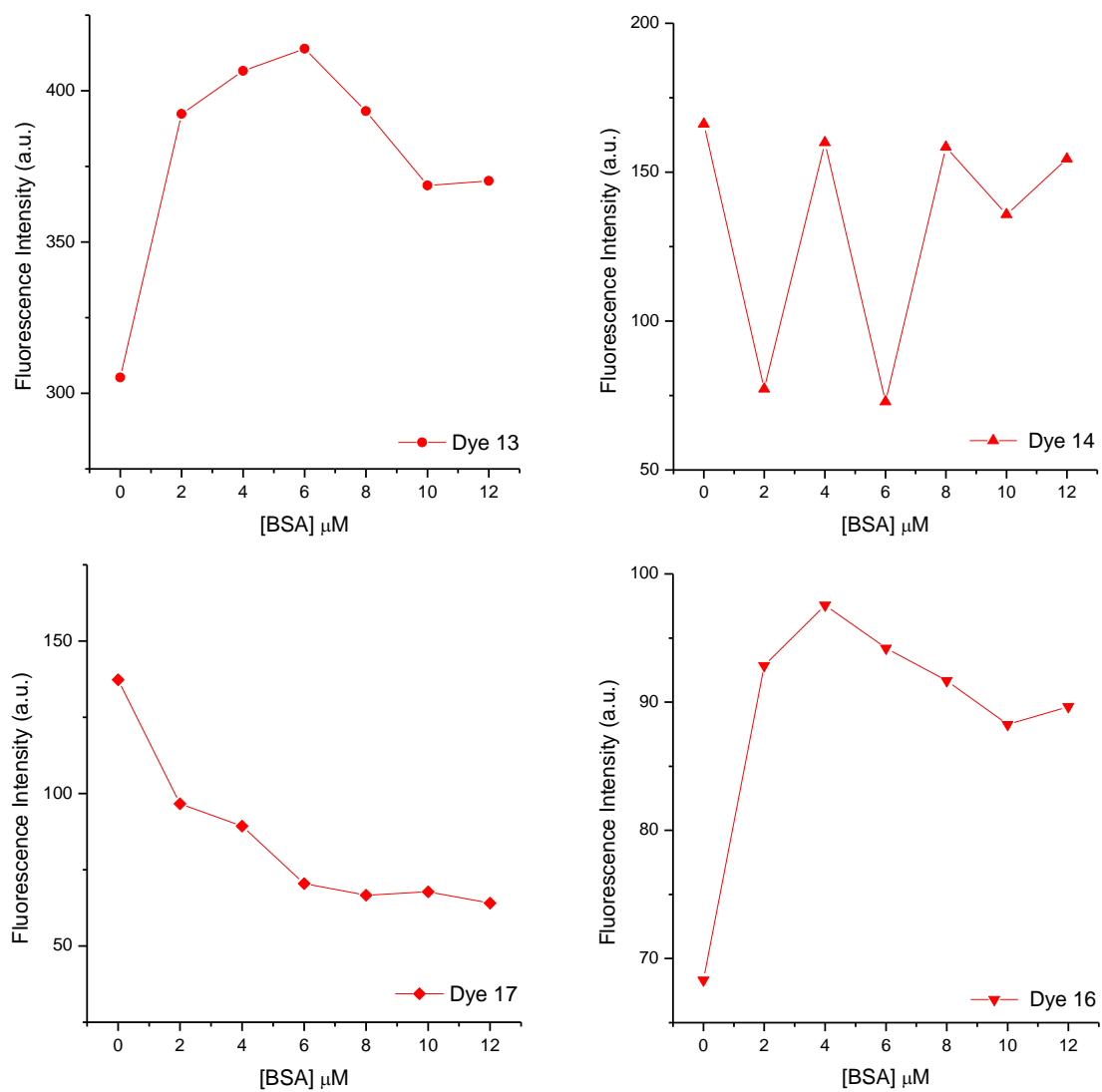


Figure S15. Plot of fluorescence intensity maxima as a function of BSA concentration.