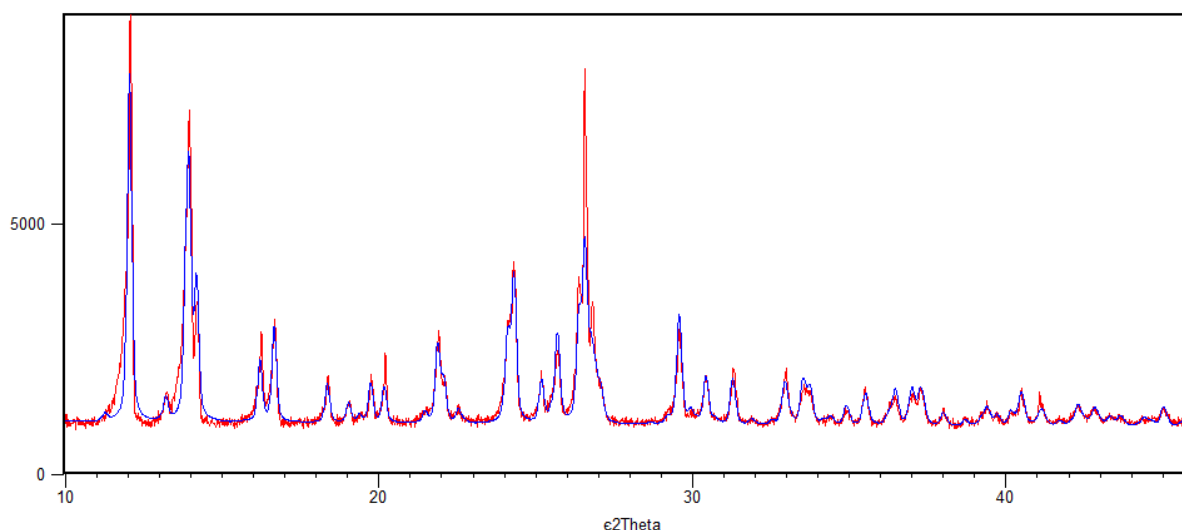


## Zn(II) heteroleptic halide complexes with 2-halopyridines: features of halogen bonding in solid state

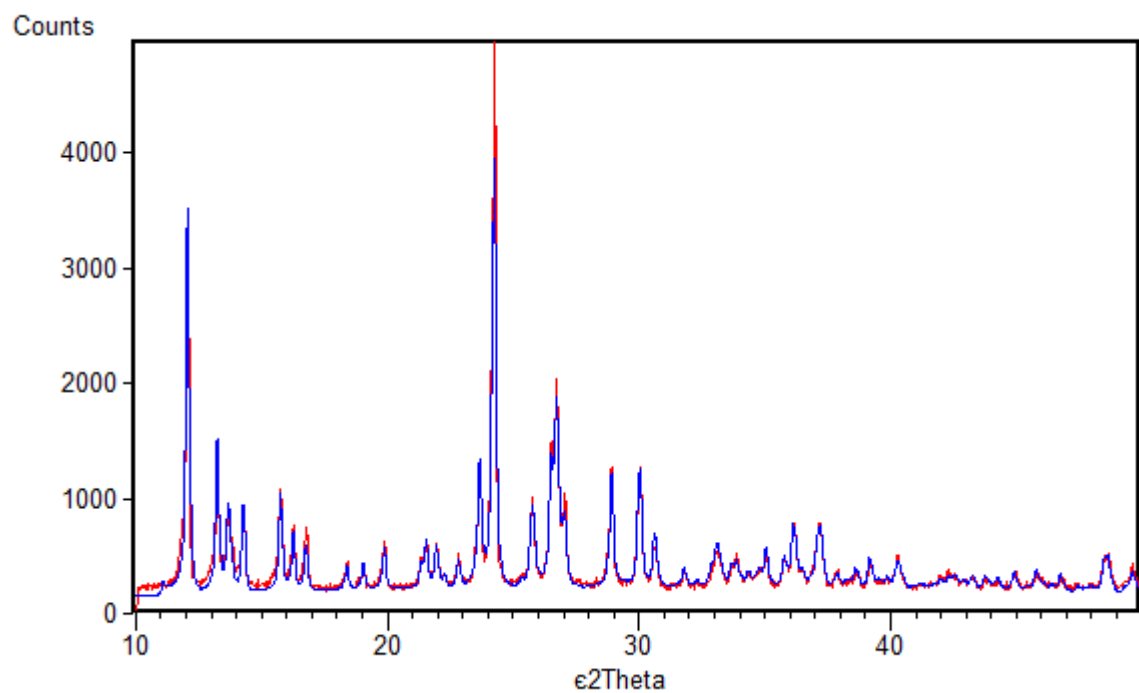
Mikhail A. Vershinin, Marianna I. Rakhmanova, Alexander S. Novikov, Maxim N. Sokolov and Sergey A. Adonin\*

**Powder X-ray diffractometry.** XRD analysis of polycrystals was performed on Shimadzu XRD-7000 diffractometer (CuK-alpha radiation, Ni – filter, linear One Sight detector, 5 – 50° 2 $\theta$  range, 0.0143° 2 $\theta$  step, 2s per step). A polycrystalline sample was slightly ground with hexane in an agate mortar, and the resulting suspension was deposited on the polished side of a standard quartz sample holder, and a smooth thin layer being formed after drying. Plotting of PXRD patterns and data treatment was performed using X'Pert Plus software.

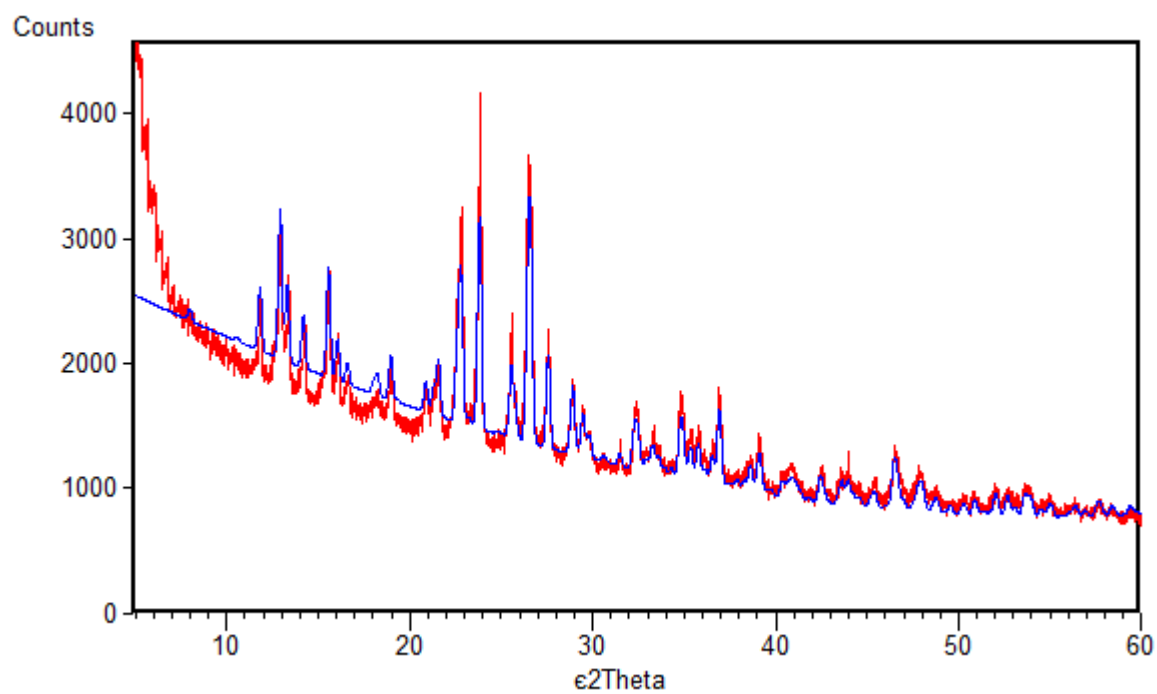


**Figure S1.** Experimental (blue) and calculated (red) PXRD patterns for **1**

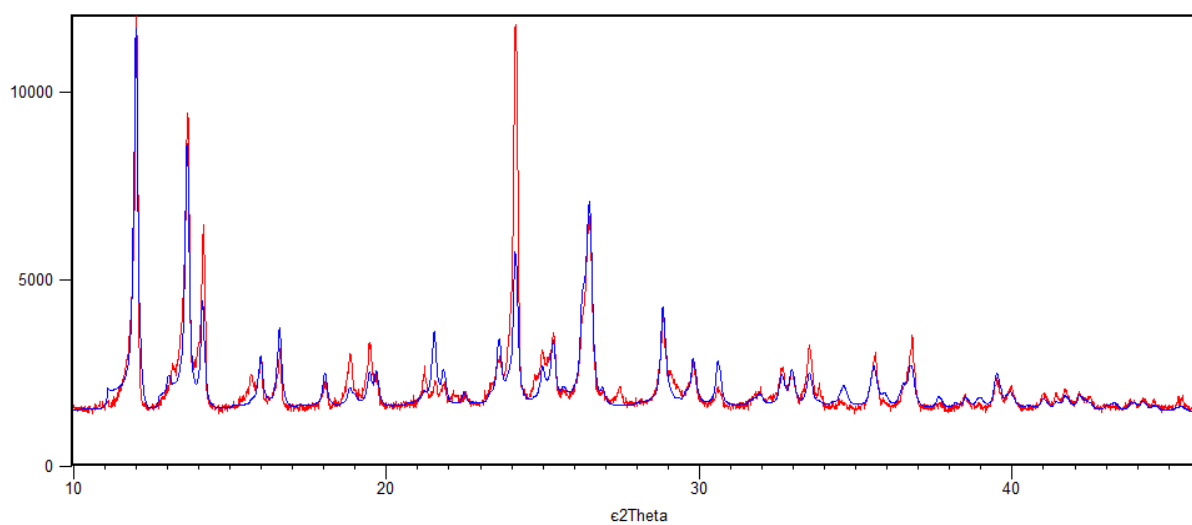
## S-2



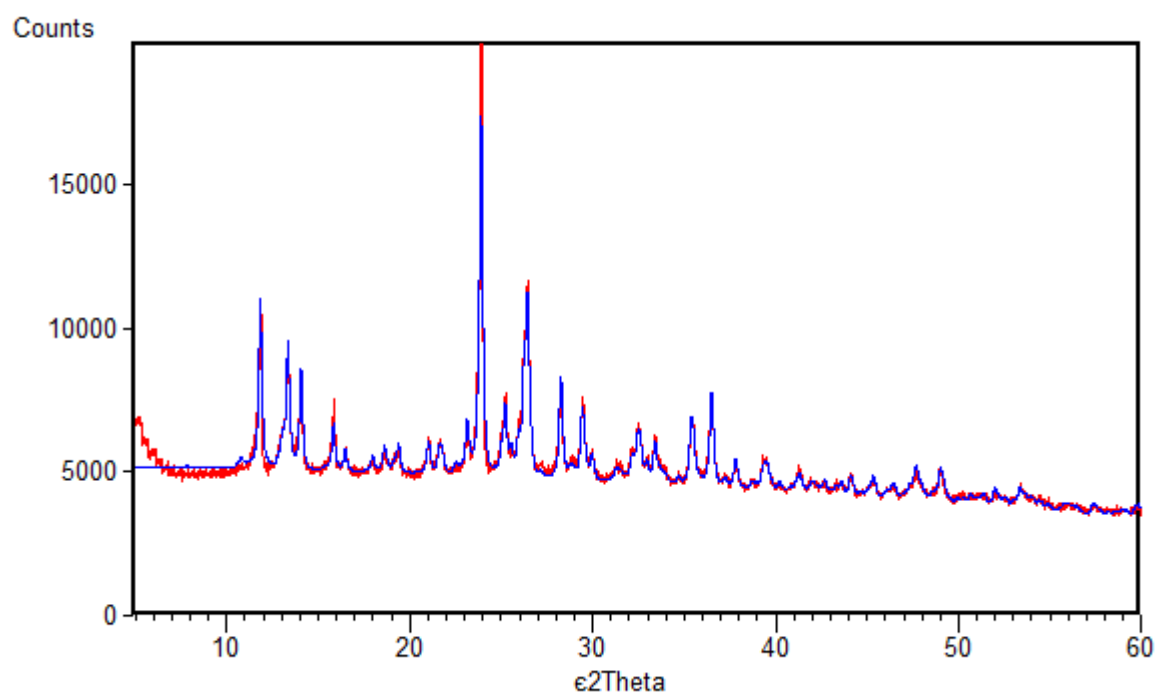
**Figure S2.** Experimental (blue) and calculated (red) PXRD patterns for **2**



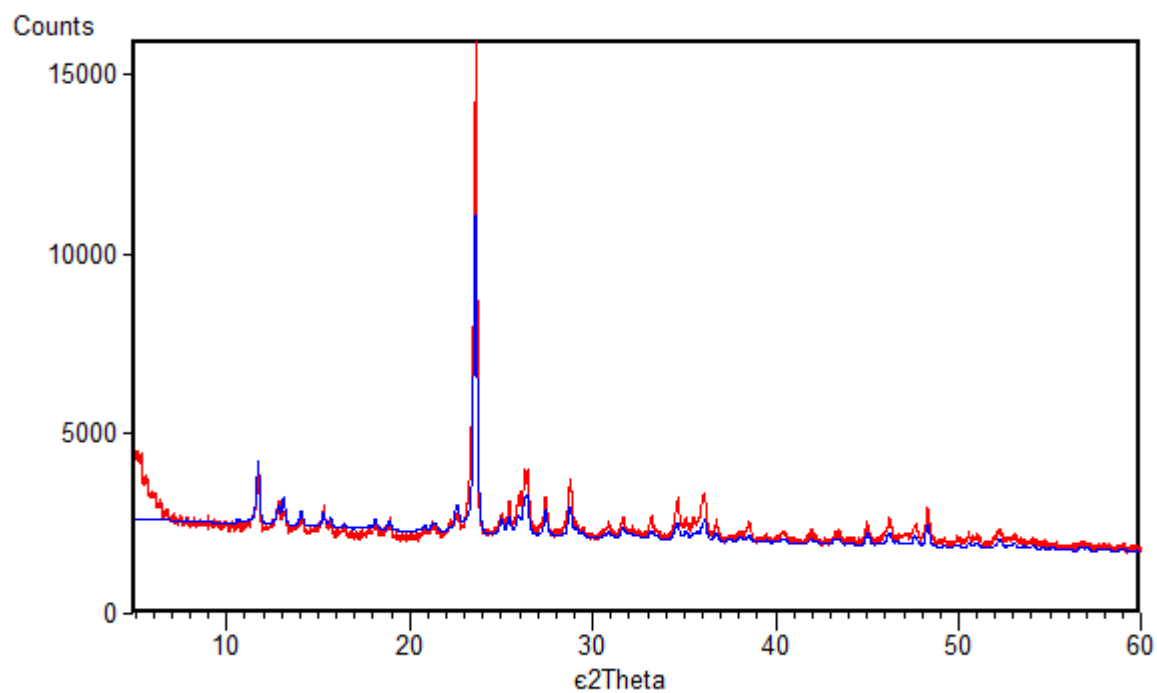
**Figure S3.** Experimental (blue) and calculated (red) PXRD patterns for **3**



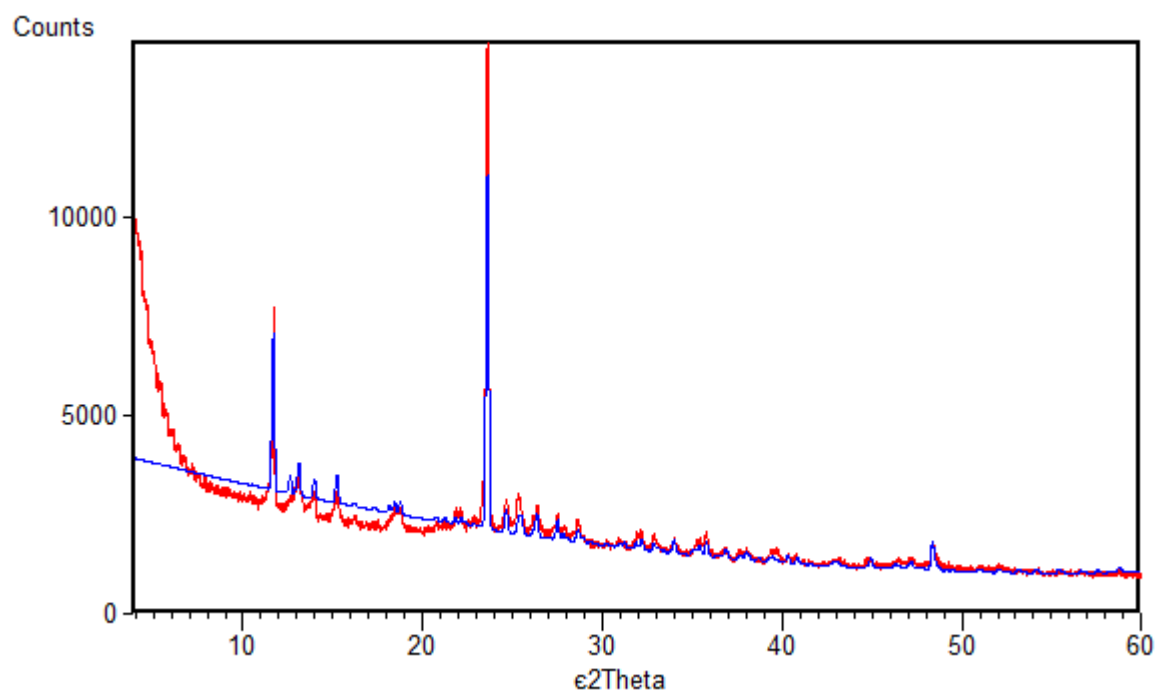
**Figure S4.** Experimental (blue) and calculated (red) PXRD patterns for **4**



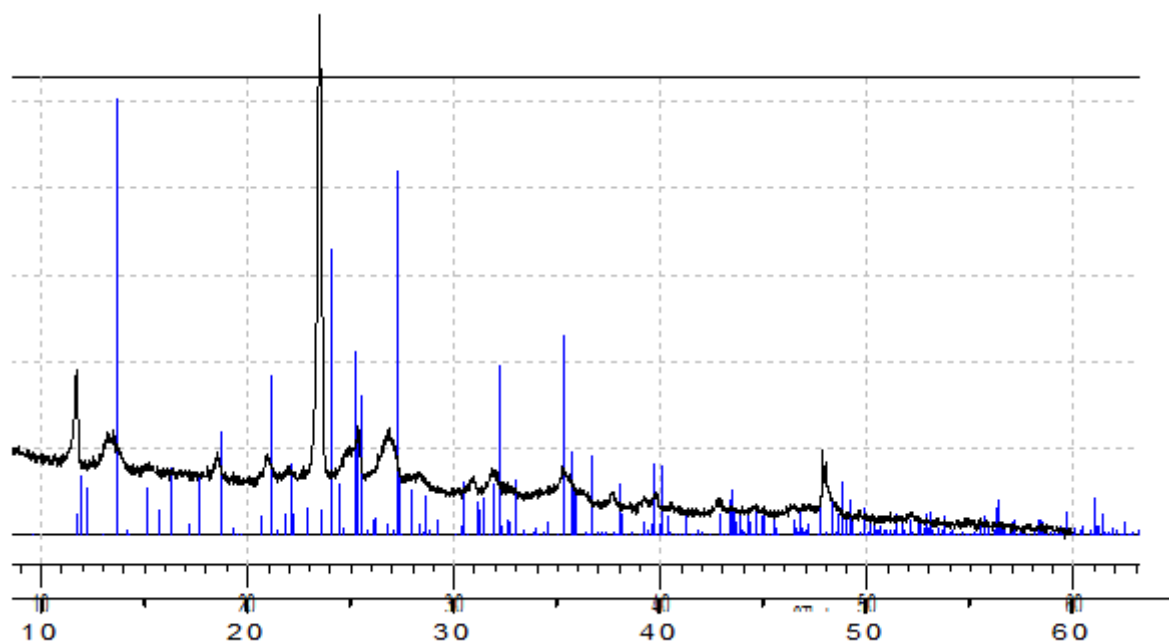
**Figure S5.** Experimental (blue) and calculated (red) PXRD patterns for **5**



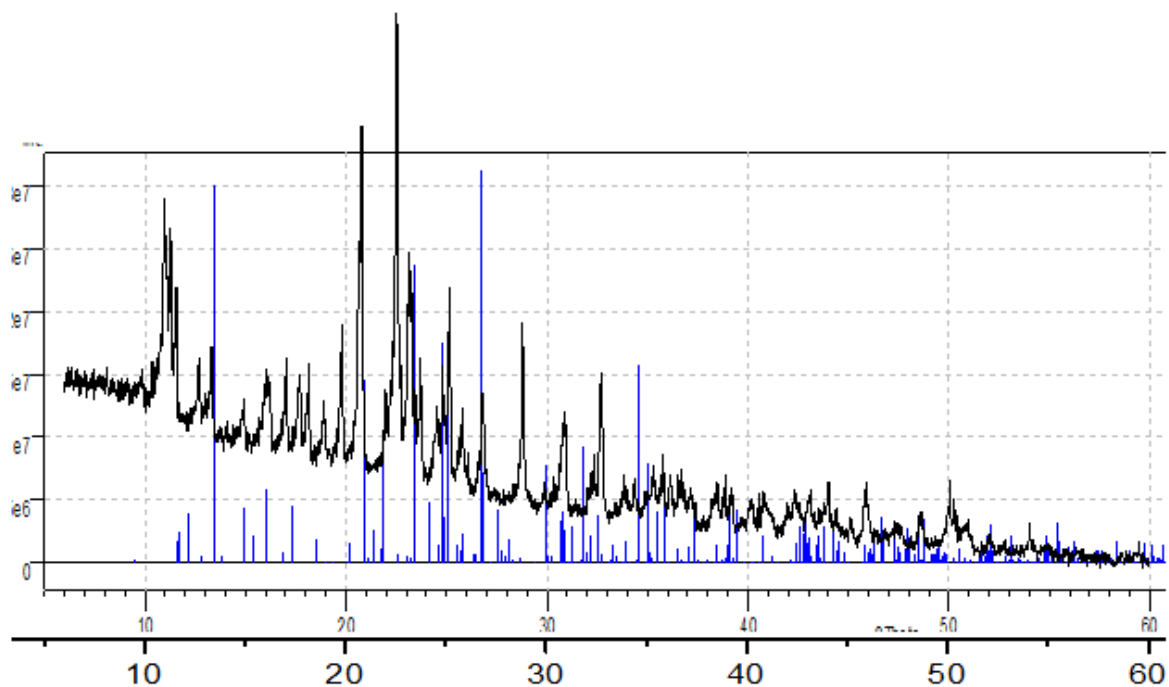
**Figure S6.** Experimental (blue) and calculated (red) PXRD patterns for **6**



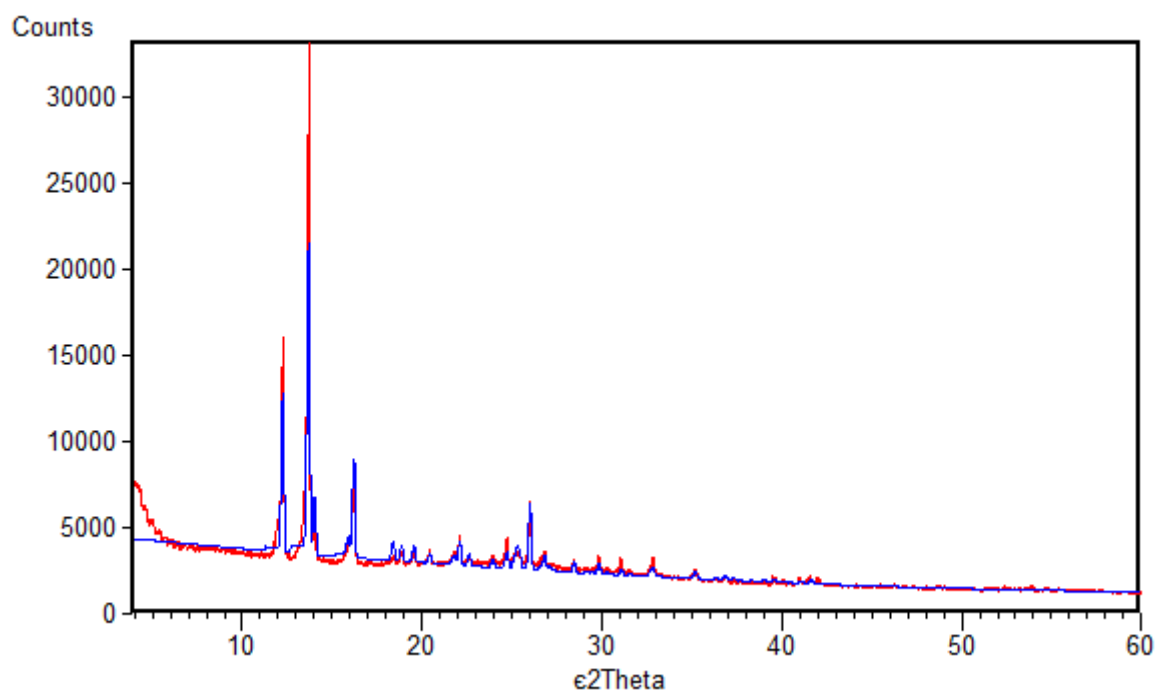
**Figure S7.** Experimental (blue) and calculated (red) PXRD patterns for **7**



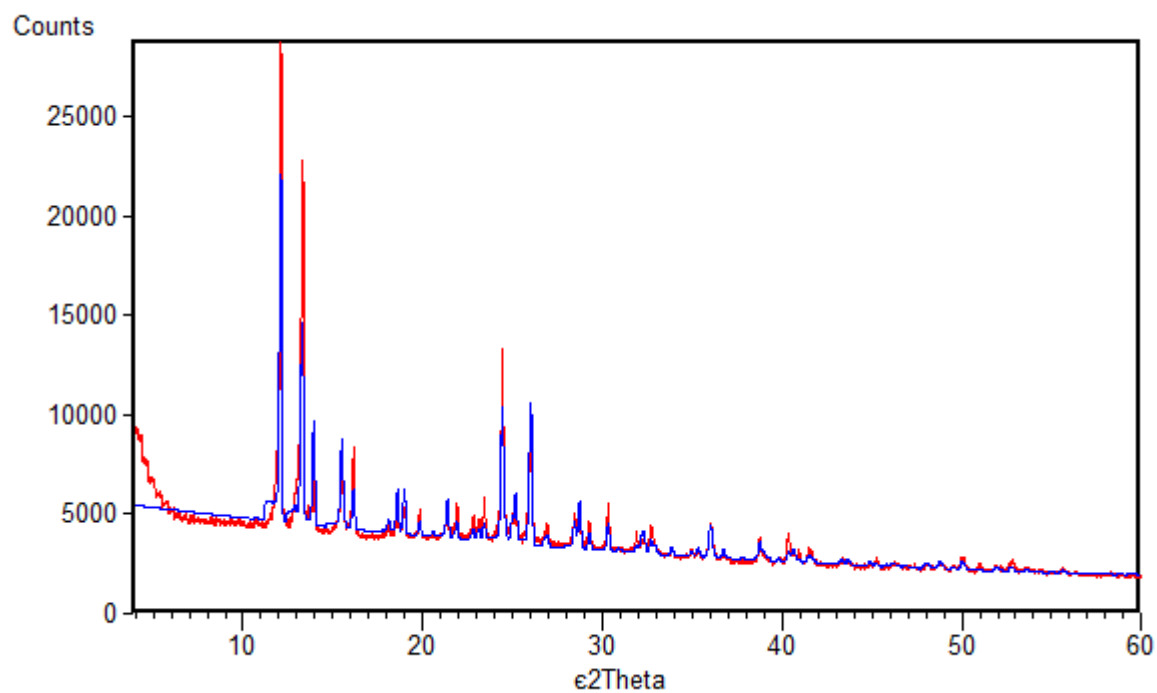
**Figure S8.** Experimental (black) and calculated (blue) PXRD patterns for **8** (calculation based on the structure of **9**)



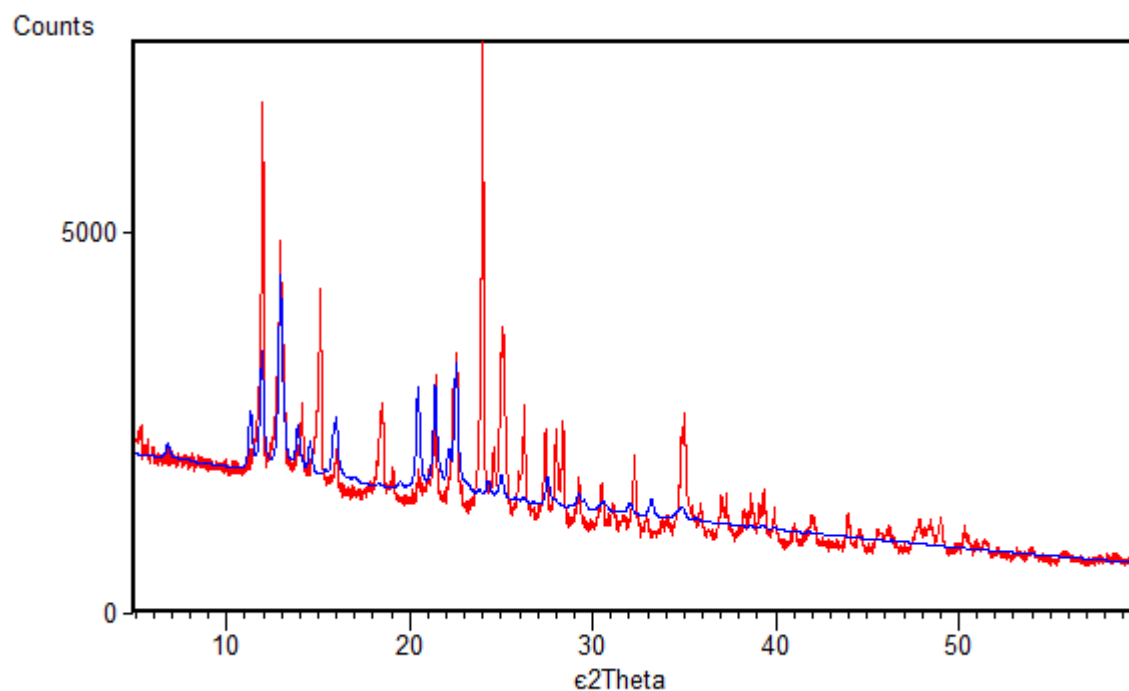
**Figure S9.** Experimental (black) and calculated (blue) PXRD patterns for **9**



**Figure S10.** Experimental (blue) and calculated (red) PXRD patterns for **10**



**Figure S11.** Experimental (blue) and calculated (red) PXRD patterns for **11**



**Figure S12.** Experimental (blue) and calculated (red) PXRD patterns for **12**

**Table 1S.** Element analysis data for **1-9**

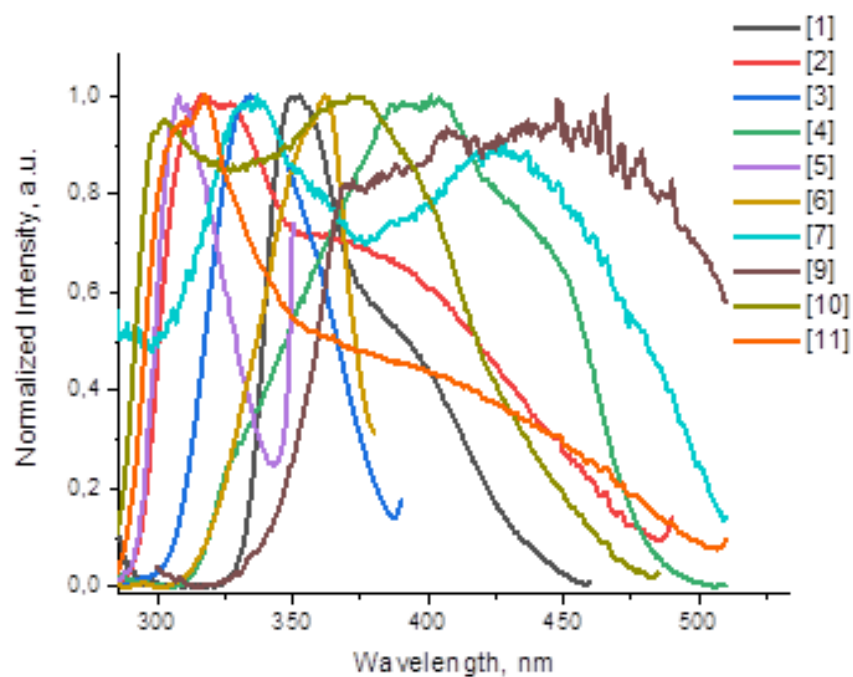
Compound	C, H, N, calcd	C, H, N, found
<b>1</b>	33.35, 2.24, 7.78	33.72; 2.29; 7.84
<b>2</b>	26.80, 1.80, 6.25	27.02; 1.84; 6.33
<b>3</b>	22.07; 1.48; 5.15	22.16; 1.57; 5.23
<b>4</b>	26.80, 1.80, 6.25	27.04; 1.87; 6.37
<b>5</b>	22.40; 1.51; 5.23	22.71; 1.58; 5.31
<b>6</b>	19.00; 1.28; 4.43	18.81; 1.32; 4.57
<b>7</b>	22.07; 1.48; 5.15	21.79; 1.46; 5.03
<b>8</b>	19.00; 1.28; 4.43	18.87; 1.29; 4.37
<b>9</b>	16.49; 1.11; 3.85	16.57; 1.12; 3.87

**Table 2S.** Crystal data and structure refinement for **1-7** and **9**

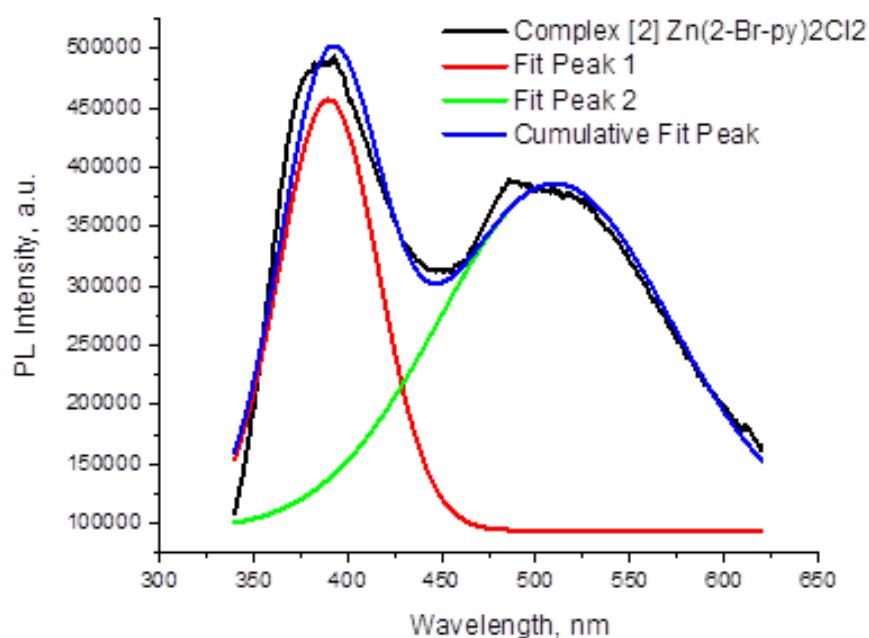
	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>9</b>
Empirical formula	C <sub>10</sub> H <sub>8</sub> Cl <sub>4</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Br <sub>2</sub> Cl <sub>2</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Cl <sub>2</sub> I <sub>2</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Br <sub>2</sub> Cl <sub>2</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Br <sub>4</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Br <sub>2</sub> I <sub>2</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> Cl <sub>2</sub> I <sub>2</sub> N <sub>2</sub> Zn	C <sub>10</sub> H <sub>8</sub> I <sub>4</sub> N <sub>2</sub> Zn
<i>M</i> , g/mol	363.35	452.27	546.25	452.27	541.19	635.18	546.25	729.15
Crystal system, space group	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Triclinic, <i>P</i> -1	Monoclinic, <i>P</i> 2 <sub>1</sub> / <i>n</i>
<i>a</i> , <i>b</i> , <i>c</i> , Å	7.9423 (5), 8.3152 (5), 11.0341 (5)	7.9146 (5), 8.4277 (6), 11.0498 (6)	7.9452 (4), 8.6864 (4), 11.1868 (5)	7.9347 (4), 8.4869 (8), 11.2381 (10)	7.9547 (4), 8.5643 (5), 11.2938 (4)	8.0186 (5), 8.7153 (4), 11.4729 (7)	7.9153 (5), 8.7199 (5), 11.6868 (7)	8.8619 (4), 15.1663 (7), 11.8971 (6)
$\alpha$ , $\beta$ , $\gamma$ , °	85.391 (4), 77.674 (4), 68.667 (5)	85.571 (5), 78.054 (5), 69.729 (6)	86.820 (4), 78.182 (4), 71.658 (5)	85.754 (7), 77.454 (6), 69.687 (6)	85.769 (4), 78.058 (4), 70.447 (5)	86.326 (4), 78.408 (5), 71.749 (5)	87.084 (4), 77.985 (5), 72.329 (5)	90, 92.969 (2), 90
<i>V</i> , Å <sup>3</sup>	663.13 (7)	676.41 (8)	717.27 (6)	692.77 (10)	709.34 (6)	745.91 (8)	751.67 (8)	1596.85 (13)
<i>Z</i>	2	2	2	2	2	2	2	4
<i>F</i> (000)	360	432	504	432	504	576	504	1296
<i>T</i> <sub>min</sub> , <i>T</i> <sub>max</sub>	0.873, 1.000	0.534, 1.000	0.623, 1.000	0.497, 1.000	0.602, 1.000	0.585, 1.000	0.871, 1.000	0.454, 0.746



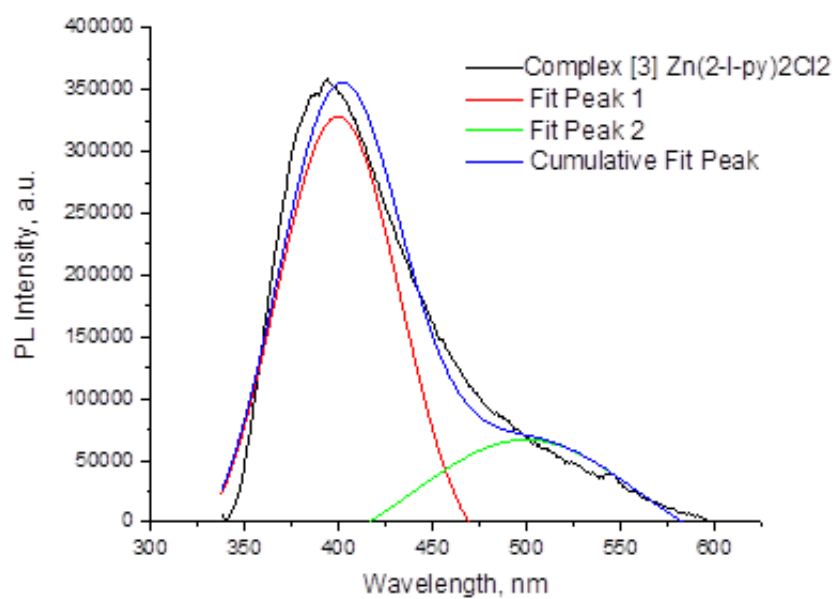
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	4886, 2908, 2468	4637, 2585, 2212	5309, 2717, 2380	4920, 2631, 2129	5098, 2693, 2352	5424, 2802, 2352	5427, 2854, 2376	27006, 2952, 2890
$R_{\text{int}}$	0.026	0.029	0.026	0.026	0.026	0.034	0.032	0.037
$\theta$ values ( $^\circ$ )	$\theta_{\text{max}} = 28.9, \theta_{\text{min}} = 1.9$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 1.9$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 2.5$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 1.9$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 1.8$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 2.5$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 2.5$	$\theta_{\text{max}} = 25.7, \theta_{\text{min}} = 2.2$
Range of $h, k, l$	$h = -10 \rightarrow 10, k = -10 \rightarrow 8, l = -14 \rightarrow 12$	$h = -7 \rightarrow 9, k = -10 \rightarrow 9, l = -13 \rightarrow 13$	$h = -9 \rightarrow 9, k = -10 \rightarrow 10, l = -13 \rightarrow 10$	$h = -9 \rightarrow 9, k = -10 \rightarrow 10, l = -11 \rightarrow 13$	$h = -8 \rightarrow 9, k = -10 \rightarrow 8, l = -13 \rightarrow 13$	$h = -9 \rightarrow 9, k = -8 \rightarrow 10, l = -13 \rightarrow 13$	$h = -9 \rightarrow 9, k = -10 \rightarrow 10, l = -14 \rightarrow 14$	$h = -10 \rightarrow 10, k = -18 \rightarrow 18, l = -14 \rightarrow 14$
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.035, 0.081, 1.05	0.034, 0.078, 1.01	0.030, 0.064, 1.06	0.035, 0.078, 1.07	0.026, 0.052, 1.01	0.054, 0.139, 1.09	0.033, 0.067, 1.01	0.040, 0.110, 1.10
No. of reflections	2908	2585	2717	2631	2693	2802	2854	2952
No. of parameters	154	154	154	154	154	154	154	154
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e $\text{\AA}^{-3}$ )	0.64, -0.57	0.87, -0.65	0.98, -0.99	0.77, -0.66	0.64, -0.67	1.93, -1.67	1.51, -0.91	3.49, -0.80



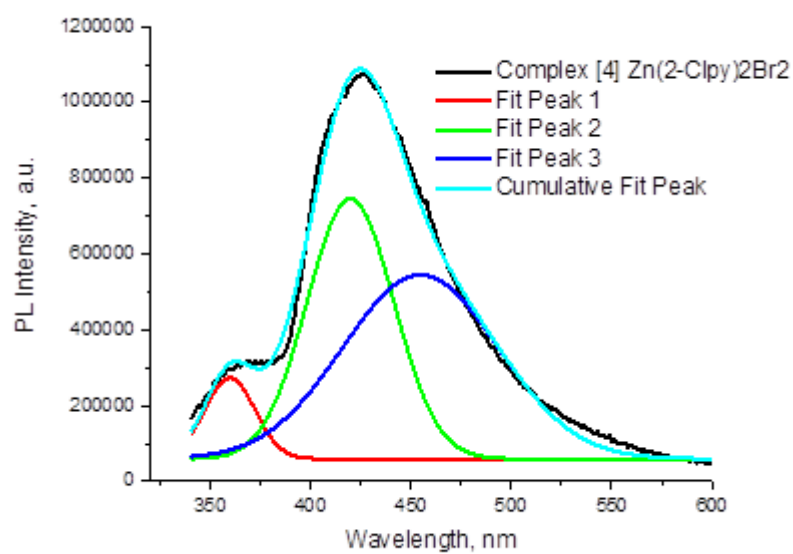
**Fig. S13.** Normalized excitation spectra of complexes in solid state at 298 K.



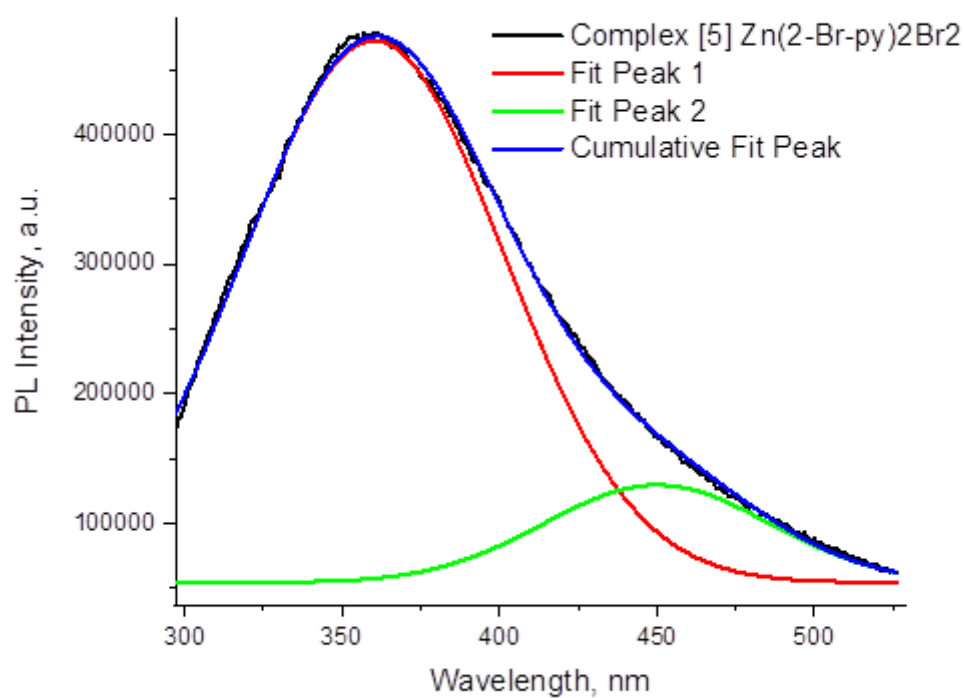
**Fig. S14.** Emission spectra of **2** and its decomposition on two and three Gauss components,  $T = 298$  K,  $\lambda_{\text{exc}} = 350$  nm. Gauss fit was performed in eV scale and then spectra were reported in nm scale. Conversion between eV and nm scales was performed with account of  $\lambda_2$  factor



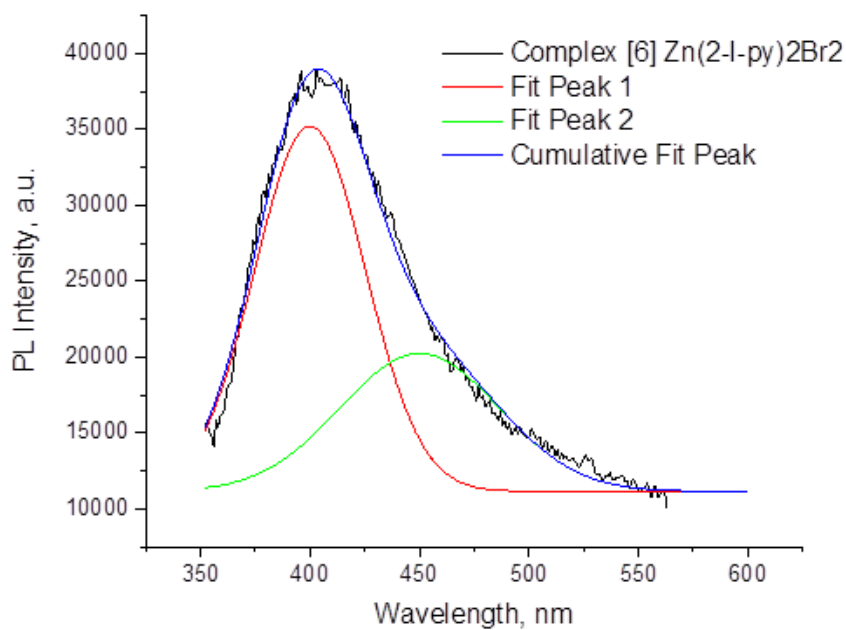
**Fig. S15.** Description see S14, for 3



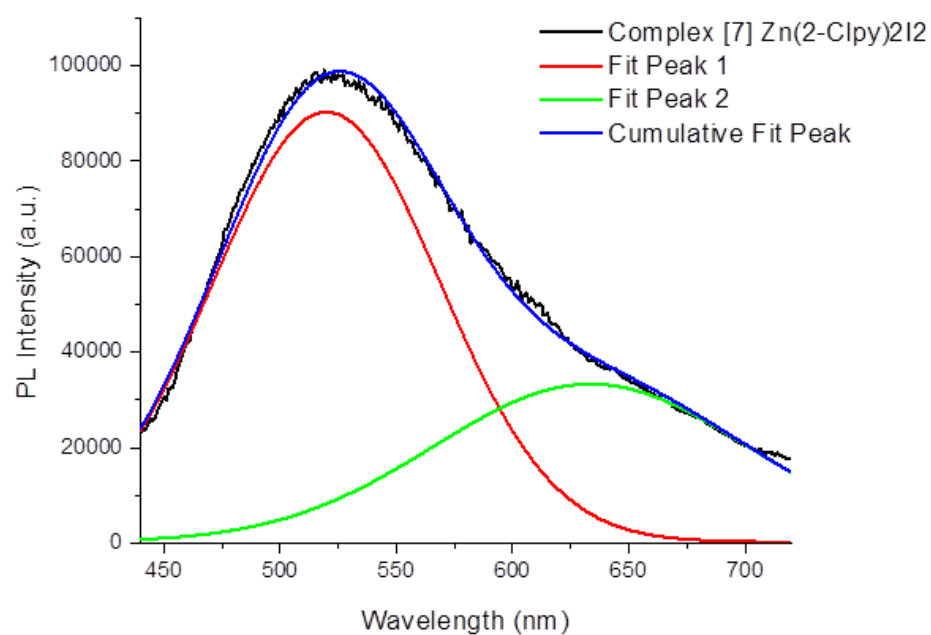
**Fig. S16.** Description see S14, for 4



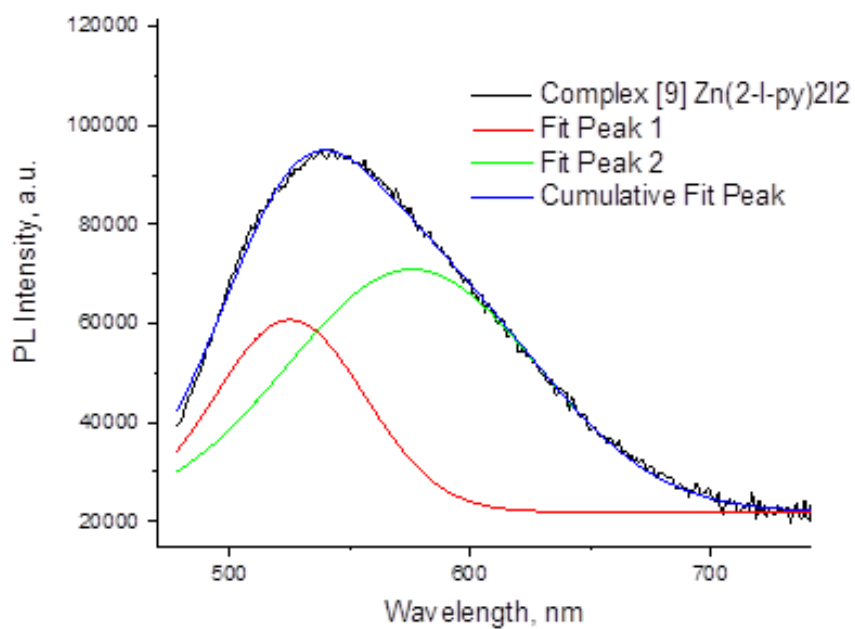
**Fig. S17.** Description see S14, for 5



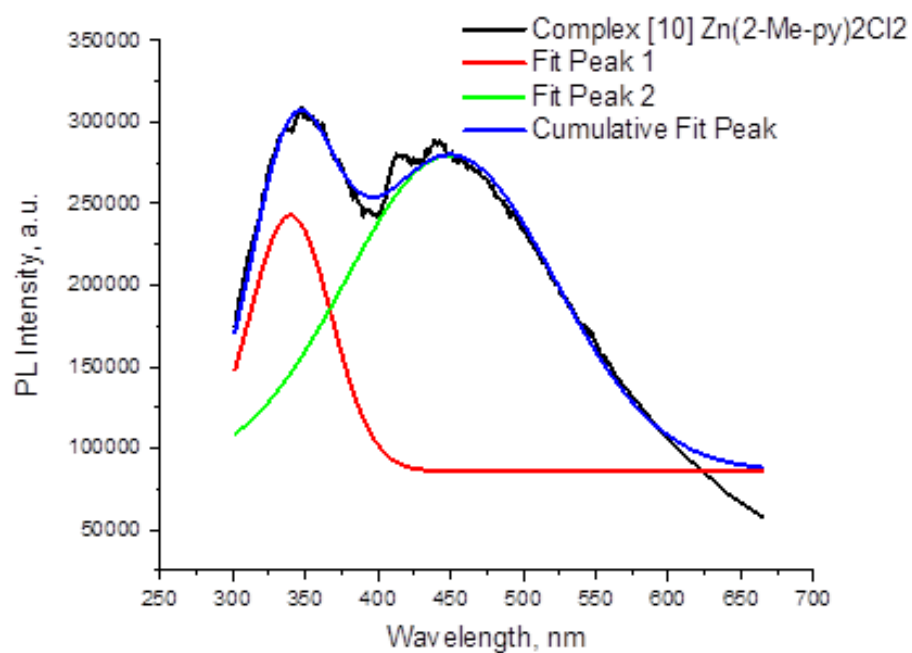
**Fig. S18.** Description see S14, for 6



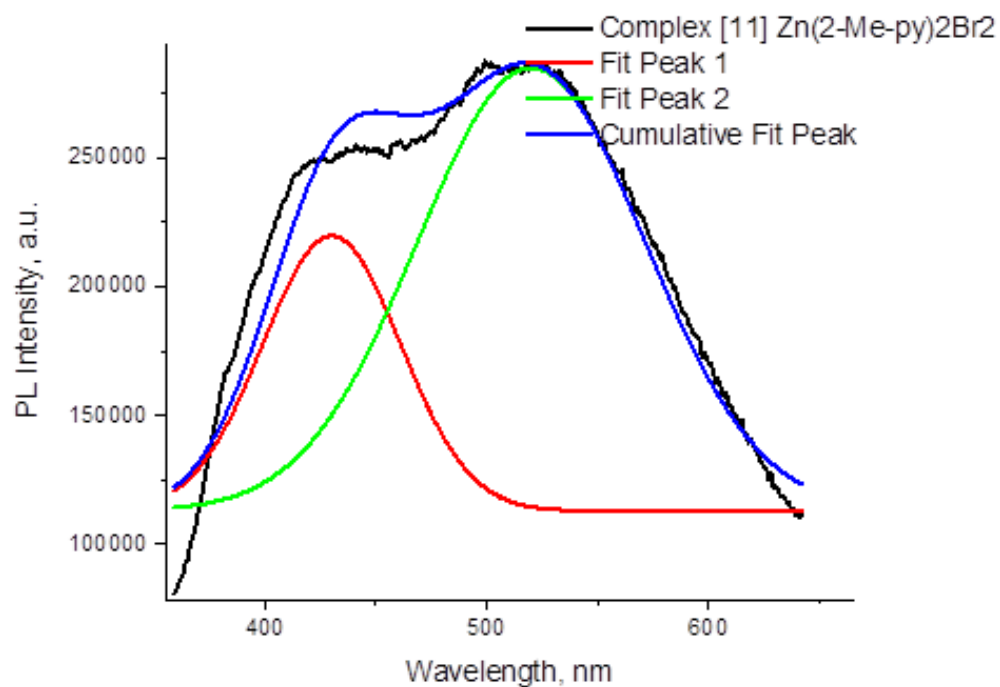
**Fig. S19.** Description see S14, for **7**



**Fig. S20.** Description see S14, for **9**



**Fig. S21.** Description see S14, for **10**



**Fig. S22.** Description see S14, for **11**

**Table 3S.** Cartesian atomic coordinates for model supramolecular trimeric associates.

Atom	X	Y	Z
<b>1</b>			
C	4.440975	4.226013	3.951839
C	4.121055	5.106585	4.968365
H	3.701437	5.940238	4.788344
C	4.437899	4.720047	6.261930
H	4.258331	5.300021	6.991716
C	5.016528	3.488396	6.481836
H	5.225201	3.205979	7.364693
C	5.287634	2.670387	5.406021
H	5.672115	1.816538	5.566639
C	3.884008	1.773042	0.297520
H	4.619745	2.239740	-0.083004
C	2.763193	1.557765	-0.471073
H	2.728628	1.861821	-1.370100
C	1.689221	0.893307	0.086238
H	0.908952	0.728348	-0.430110
C	1.755268	0.471191	1.397050
H	1.021838	0.032475	1.810990
C	2.917872	0.708678	2.080483
N	5.028145	3.036863	4.138328
N	3.971700	1.340003	1.581382
Cl	6.676803	0.025179	3.583389
Cl	7.048009	2.842922	1.212285
Cl	4.066612	4.653443	2.316882

Cl	3.057617	0.191137	3.731179
Zn	5.695180	1.787289	2.622271
C	6.526293	3.519445	-3.951839
C	6.846213	2.638873	-4.968365
H	7.265831	1.805220	-4.788344
C	6.529369	3.025411	-6.261930
H	6.708937	2.445437	-6.991716
C	5.950740	4.257062	-6.481836
H	5.742068	4.539478	-7.364693
C	5.679634	5.075071	-5.406021
H	5.295154	5.928919	-5.566639
C	7.083260	5.972416	-0.297520
H	6.347523	5.505717	0.083004
C	8.204075	6.187693	0.471073
H	8.238640	5.883637	1.370100
C	9.278047	6.852151	-0.086238
H	10.058316	7.017110	0.430110
C	9.212000	7.274267	-1.397050
H	9.945430	7.712983	-1.810990
C	8.049396	7.036780	-2.080483
N	5.939123	4.708595	-4.138328
N	6.995568	6.405455	-1.581382
Cl	4.290466	7.720279	-3.583389
Cl	3.919260	4.902536	-1.212285
Cl	6.900656	3.092015	-2.316882
Cl	7.909651	7.554320	-3.731179



Zn	5.272088	5.958169	-2.622271
C	5.856815	-4.194073	6.827864
C	6.176736	-5.074645	5.811338
H	6.596354	-5.908297	5.991359
C	5.859891	-4.688107	4.517774
H	6.039459	-5.268080	3.787988
C	5.281263	-3.456456	4.297868
H	5.072590	-3.174039	3.415010
C	5.010156	-2.638446	5.373682
H	4.625676	-1.784598	5.213065
C	6.413783	-1.741102	10.482184
H	5.678045	-2.207800	10.862707
C	7.534598	-1.525824	11.250777
H	7.569163	-1.829881	12.149804
C	8.608569	-0.861366	10.693466
H	9.388839	-0.696407	11.209814
C	8.542522	-0.439250	9.382654
H	9.275952	-0.000535	8.968713
C	7.379918	-0.676737	8.699221
N	5.269645	-3.004923	6.641375
N	6.326090	-1.308062	9.198321
Cl	3.620988	0.006761	7.196314
Cl	3.249782	-2.810982	9.567418
Cl	6.231178	-4.621502	8.462822
Cl	7.240173	-0.159197	7.048525
Zn	4.602611	-1.755348	8.157433

2			
C	9.227428	6.198175	10.548692
H	8.476952	5.754557	10.925971
C	10.327963	6.409213	11.331357
H	10.331078	6.130059	12.239423
C	11.437225	7.032075	10.786518
H	12.214512	7.183864	11.312980
C	11.394059	7.428911	9.464418
H	12.143282	7.840685	9.050384
C	10.233399	7.208108	8.767154
C	7.900934	5.224971	5.411637
H	7.497124	6.070746	5.257049
C	8.181496	4.428744	4.338174
H	7.960806	4.713601	3.458215
C	8.791566	3.206171	4.543569
H	8.986473	2.634923	3.809550
C	9.112993	2.826513	5.829995
H	9.553631	2.002991	6.004041
C	8.770810	3.688882	6.859136
Cl	6.463964	7.864815	7.222903
Cl	6.096383	5.053692	9.565927
Zn	7.472102	6.117100	8.184803
Br	10.142643	7.756378	6.951348
Br	9.199706	3.207261	8.636133
N	9.163091	6.596760	9.254699
N	8.170308	4.864183	6.680766

C	6.181431	1.837355	11.071910
H	6.931906	2.280973	10.694631
C	5.080895	1.626317	10.289244
H	5.077781	1.905471	9.381179
C	3.971634	1.003455	10.834084
H	3.194347	0.851666	10.307622
C	4.014800	0.606619	12.156183
H	3.265577	0.194845	12.570218
C	5.175460	0.827422	12.853448
C	7.507924	2.810559	16.208965
H	7.911734	1.964784	16.363553
C	7.227363	3.606786	17.282428
H	7.448053	3.321929	18.162387
C	6.617293	4.829359	17.077032
H	6.422386	5.400607	17.811052
C	6.295865	5.209017	15.790607
H	5.855227	6.032539	15.616561
C	6.638049	4.346648	14.761466
Cl	8.944894	0.170715	14.397699
Cl	9.312476	2.981838	12.054675
Zn	7.936757	1.918430	13.435799
Br	5.266215	0.279152	14.669254
Br	6.209153	4.828269	12.984469
N	6.245767	1.438770	12.365903
N	7.238551	3.171347	14.939836
C	6.814104	9.678179	0.261609

H	7.564580	10.121796	-0.115670
C	5.713569	9.467140	-0.521057
H	5.710454	9.746295	-1.429122
C	4.604307	8.844278	0.023783
H	3.827020	8.692489	-0.502679
C	4.647473	8.447442	1.345882
H	3.898250	8.035668	1.759917
C	5.808133	8.668245	2.043147
C	8.140598	10.651383	5.398664
H	8.544408	9.805608	5.553252
C	7.860036	11.447609	6.472127
H	8.080726	11.162752	7.352086
C	7.249966	12.670183	6.266731
H	7.055059	13.241430	7.000751
C	6.928539	13.049840	4.980306
H	6.487901	13.873362	4.806260
C	7.270722	12.187471	3.951165
Cl	9.577568	8.011538	3.587398
Cl	9.945149	10.822661	1.244374
Zn	8.569430	9.759253	2.625498
Br	5.898889	8.119975	3.858953
Br	6.841826	12.669092	2.174168
N	6.878441	9.279593	1.555602
N	7.871224	11.012170	4.129535
<b>3</b>			
C	10.104779	7.352647	8.908239

C	11.251424	7.545026	9.648403
H	12.016716	7.952791	9.259707
C	11.264938	7.136673	10.951353
H	12.042353	7.259331	11.483483
C	10.123648	6.536934	11.496622
H	10.099983	6.261837	12.405402
C	9.042374	6.360199	10.673244
H	8.271346	5.934613	11.029092
C	7.753991	5.378566	5.516189
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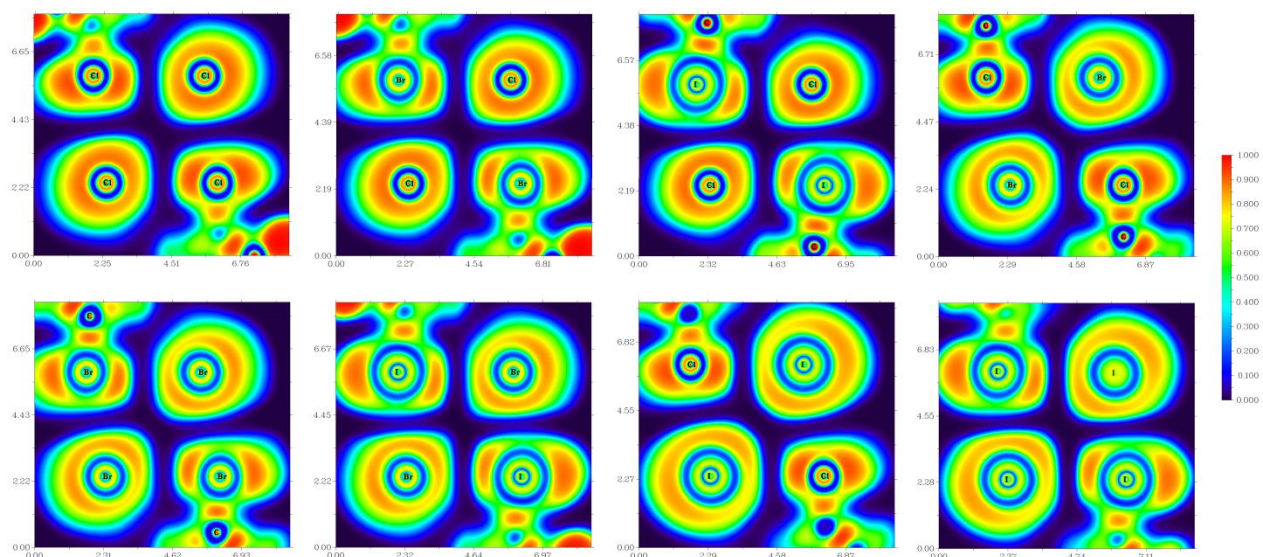
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H	1.206005	11.505155	0.736630
C	1.903360	12.057209	-4.832056
H	1.183199	11.480889	-5.060174
C	2.563912	12.727559	-5.846704
H	2.312283	12.592579	-6.753235
C	3.587846	13.590521	-5.529478

H	4.040427	14.077360	-6.207891
C	3.946797	13.734601	-4.197603
H	4.664077	14.304854	-3.948100
C	3.244956	13.035435	-3.241172
C	-0.114247	12.871639	1.485141
H	0.102450	12.732109	2.398800
N	0.261261	12.313519	-0.834055
N	2.242399	12.190672	-3.526320
Zn	1.217603	11.026507	-2.154168
I	-1.046994	13.573535	-3.190796
I	3.721359	13.319348	-1.205935
I	2.807927	9.399466	-0.926728
I	-0.495551	9.551432	-3.420815
C	7.619615	13.246246	10.778562
C	6.890207	13.995462	11.668458
H	6.230853	14.614247	11.378559
C	7.167653	13.796783	13.016967
H	6.703220	14.288171	13.684686
C	8.790361	12.139107	12.365881
H	9.451687	11.505155	12.617761
C	10.149042	12.057209	7.049075
H	9.428881	11.480889	6.820957
C	10.809594	12.727559	6.034426
H	10.557965	12.592579	5.127896
C	11.833528	13.590521	6.351652
H	12.286109	14.077360	5.673240

C	12.192479	13.734601	7.683527
H	12.909759	14.304854	7.933031
C	11.490638	13.035435	8.639958
C	8.131435	12.871639	13.366272
H	8.348132	12.732109	14.279931
N	8.506943	12.313519	11.047075
N	10.488081	12.190672	8.354811
Zn	9.463285	11.026507	9.726963
I	7.198688	13.573535	8.690334
I	11.967041	13.319348	10.675196
I	11.053609	9.399466	10.954402
I	7.750131	9.551432	8.460315



**Figure S23.** Visualization of electron localization function (ELF) in the areas of intermolecular non-covalent interactions  $X \cdots Y$  ( $X, Y = \text{Cl, Br, I}$ ) in **1–7** and **9** (from left to right, from top to bottom). Length units – Å, the color scale for the ELF maps presented in a.u.