SUPPLEMENTARY INFORMATION

Versatility of the Cyano Group in Intermolecular Interactions

Steve Scheiner* Department of Chemistry and Biochemistry Utah State University Logan, UT 84322-0300



Fig S1. AIM bond paths in complexes of NH₃ with (CN)₃C-C(CN)₃ in a) side-on and b) end-on complexes, c) (CN)₂HC-CH(CN)₂ and d) F₂HC-CHF₂



Fig S2. Overlap of NH₃ N lone pair orbital with a) $\pi^*(C=N)$, b) $\sigma^*(C-C)$, c) $\sigma^*(H-C)$, and d) $\sigma^*(C-C)$ orbital of cyclopropane derivative.



Fig S3. AIM bond paths in complexes of NH₃ with a) H₃C-C(CN)₃, b) H₃C-CH(CN)₂, c) (CN)₂C=C(CN)₂ and d) (CN)C=C(CN).



Fig S4. AIM bond paths in complexes of NH₃ with cyclic a) $CH_2\{C(CN)_2\}_2$, b) $CH_2CH_2C(CN)_2$, c) $CH_2\{CF_2\}_2$ and d) $CH_2CH_2\{C(CN)_2\}_2$.



Fig S5. Examples of crystal structures containing short contacts between C≡N groups and potential electron donors. Reproduced from Ref. 61 with permission from The Royal Society of Chemistry.