

Supporting Information

Carbazole- and Triphenylamine-Substituted Pyrimidines: Synthesis and Photophysical Properties

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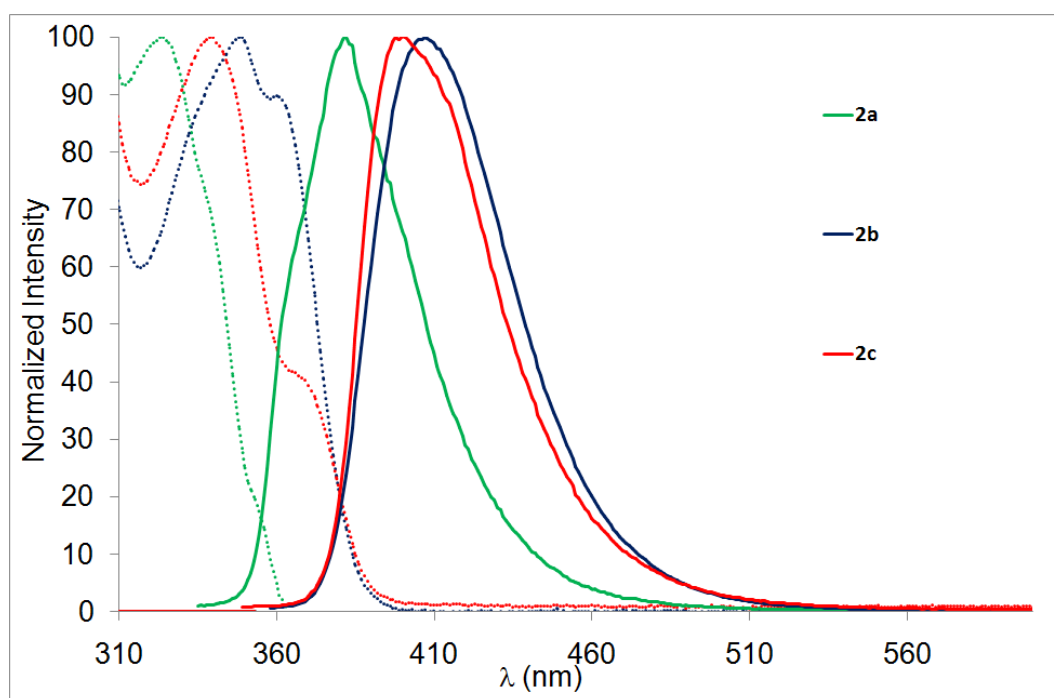


Figure S1. Normalized absorption (dashed lines) and emission (solid lines) spectra of compounds **2a** (green), **2b** (blue) and **2c** (red) in dichloromethane solution.

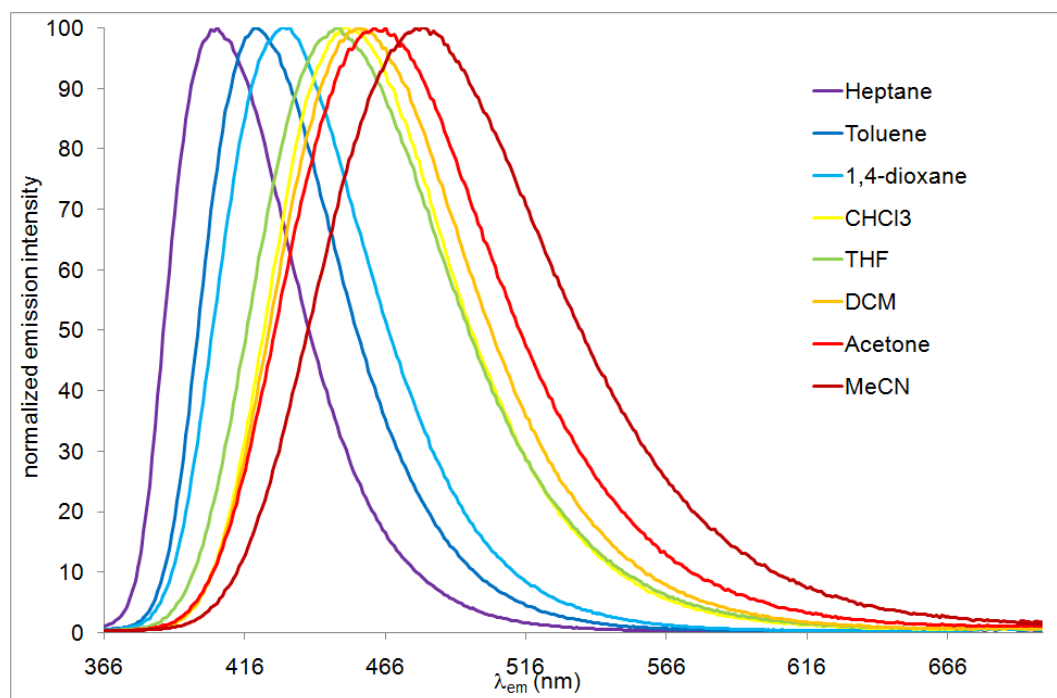


Figure S2. Normalized emission spectra of **1a** in different aprotic solvents.

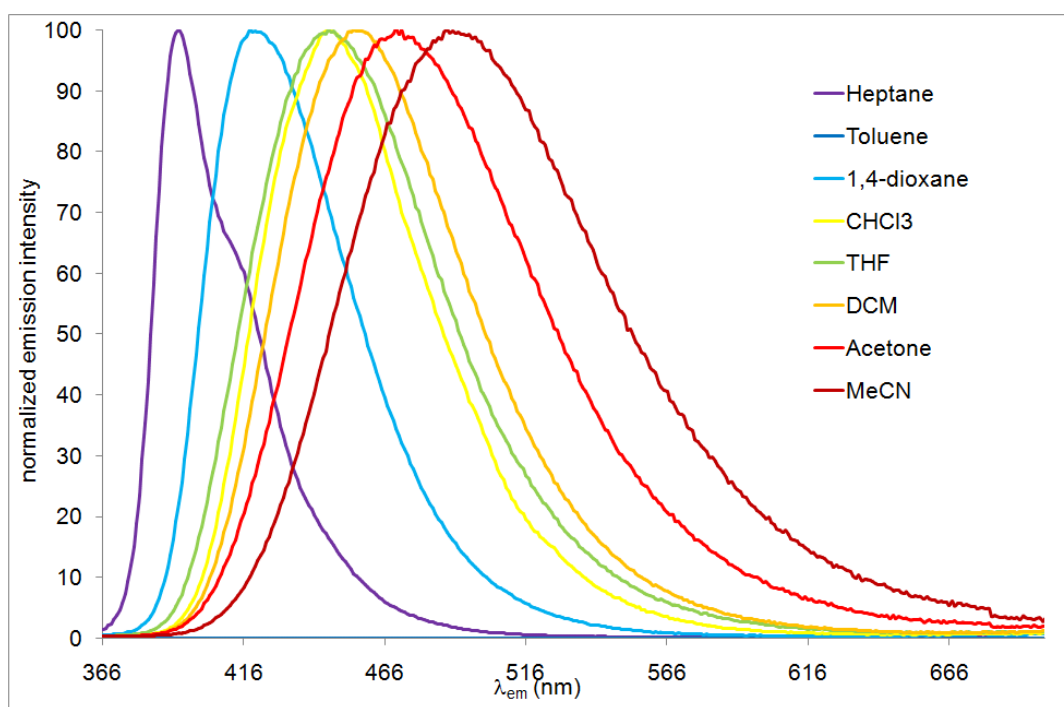


Figure S3. Normalized emission spectra of **1c** in different aprotic solvents.

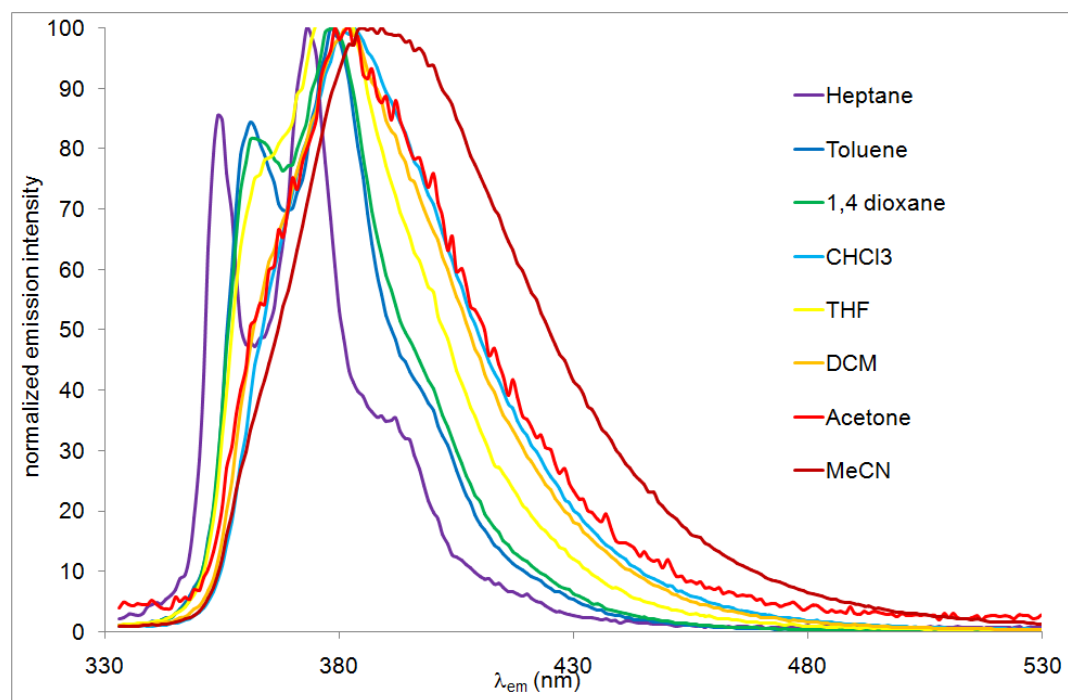


Figure S4. Normalized emission spectra of **2a** in different aprotic solvents.

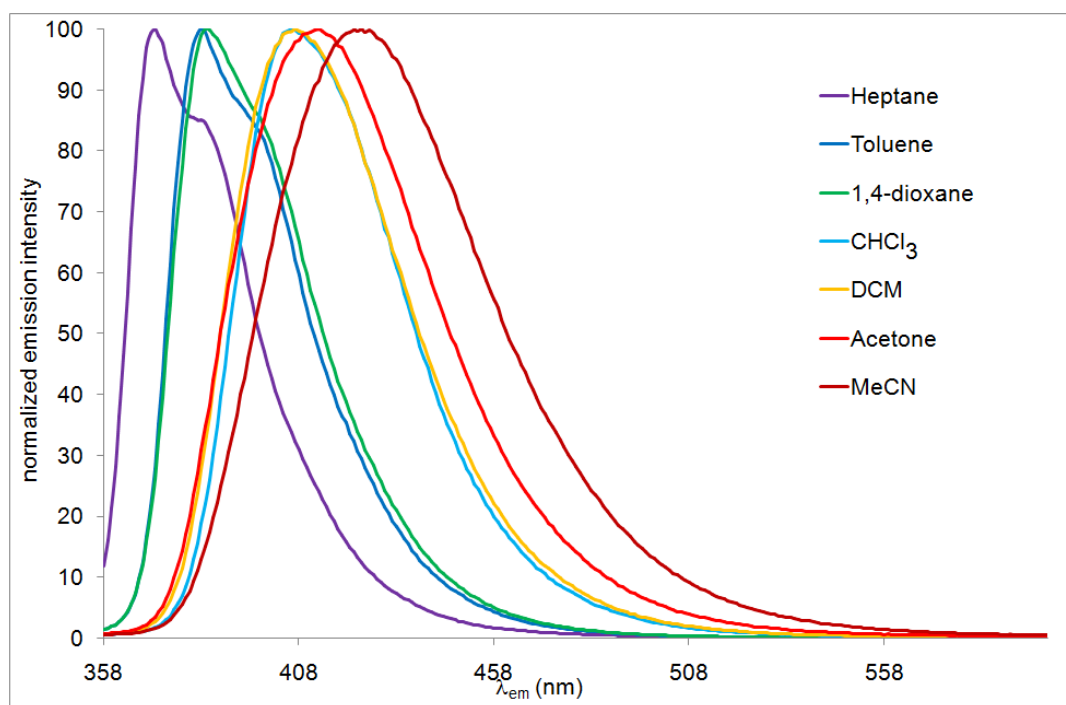


Figure S5. Normalized emission spectra of **2b** in different aprotic solvents.

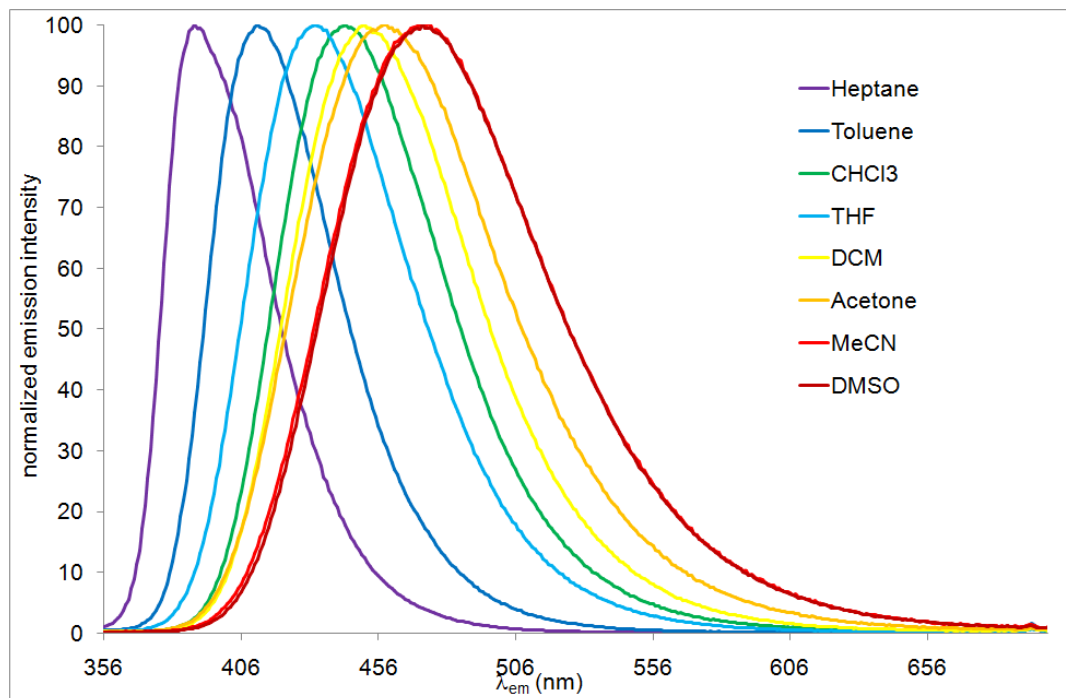


Figure S6. Normalized emission spectra of **2c** in different aprotic solvents.

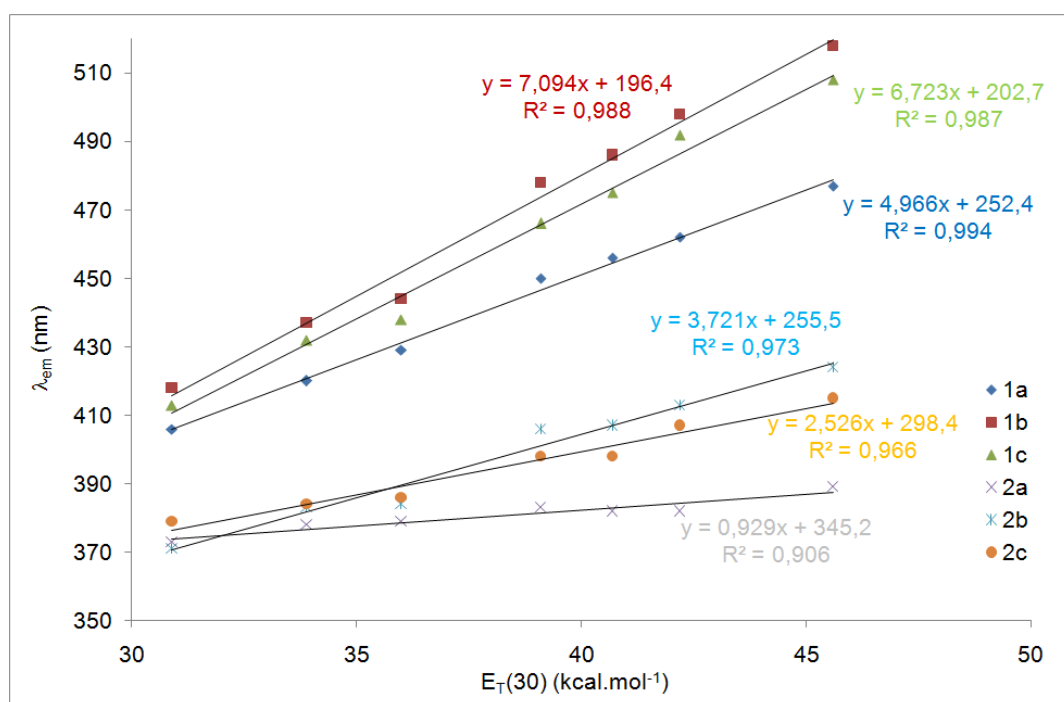


Figure S7. Emission maxima (λ_{em}) as a function of the Dimroth-Reichardt polarity parameter $E_T(30)$ for compounds **1** and **2**.

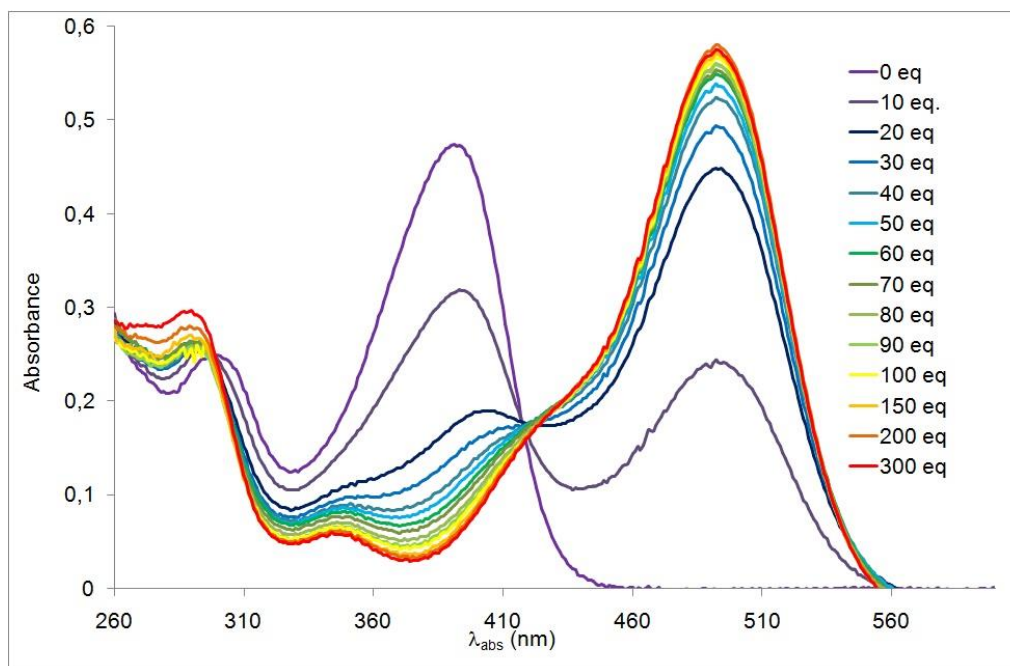


Figure S8. Changes in the absorption spectra of a chloroform solution of **1b** ($c = 9.76 \times 10^{-6}$ M) upon addition of CSA (0-300 equivalents).

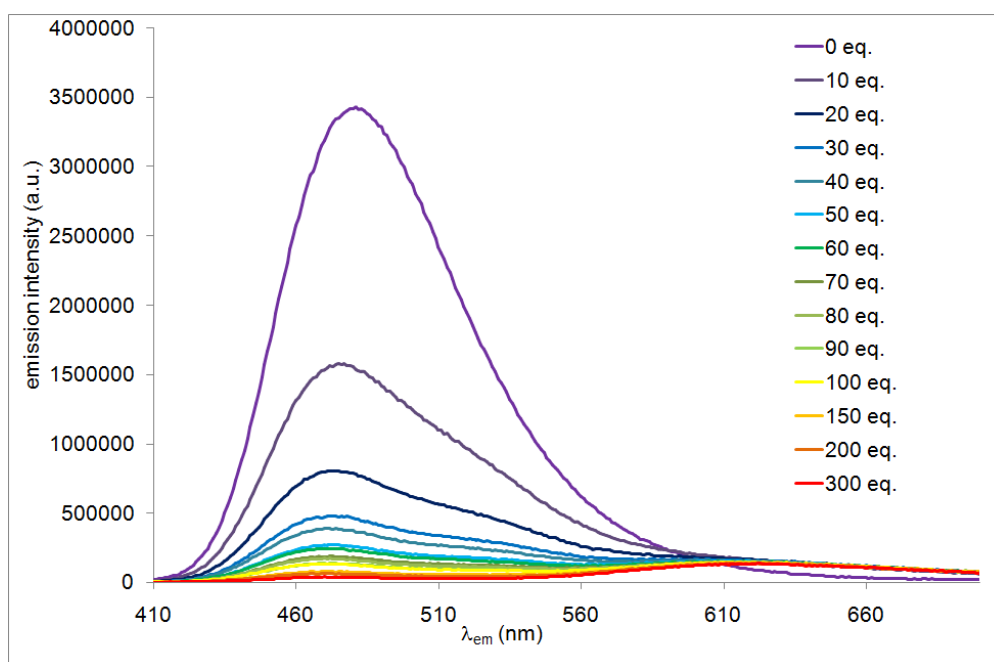


Figure S9. Changes in the emission spectra of a chloroform solution of **1b** ($c = 9.76 \times 10^{-6}$ M) upon addition of CSA (0-300 equivalents). $\lambda_{\text{exc}} = 400$ nm.

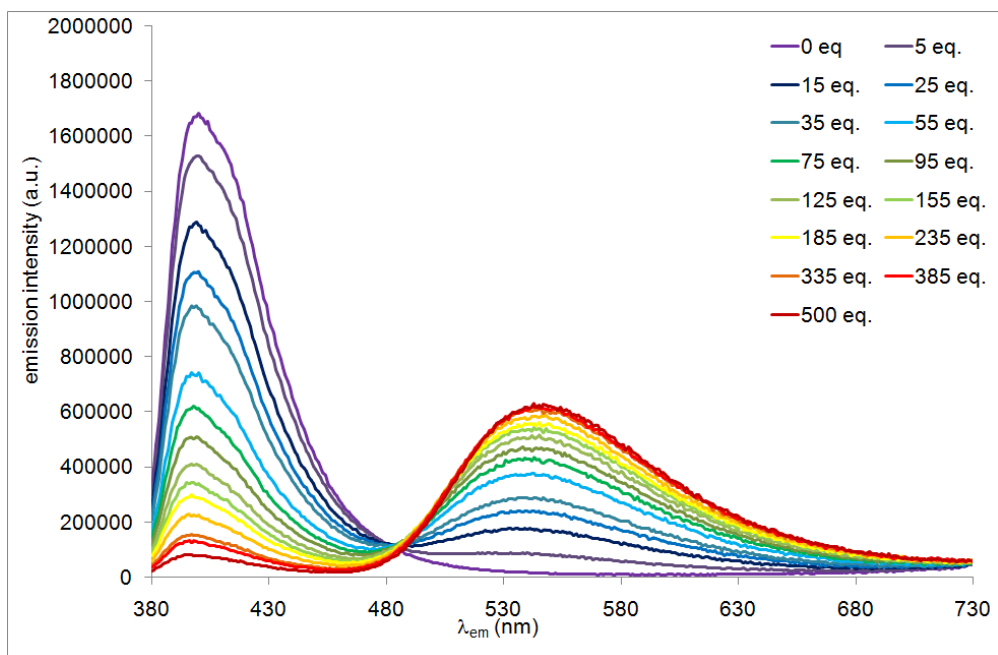


Figure S10. Changes in the emission spectra of a dichloromethane solution of **2c** ($c = 1.25 \times 10^{-5}$ M) upon addition of CSA (0-500 equivalents). $\lambda_{\text{exc}} = 370$ nm.



Figure S11. Changes in the colour of a chloroform solution of **1b** ($c = 9.76 \times 10^{-6}$ M) after the addition of 30 equivalents (middle) and in 10^{-2} M CSA (right). Photographs were taken in the dark upon irradiation with a hand-held UV lamp ($\lambda_{\text{em}} = 366$ nm).



Figure S12. Changes in the colour of a dichloromethane solution of **2b** ($c = 1.97 \times 10^{-5}$ M) after the addition of 2 equivalents (middle) and in 10^{-2} M CSA (right). Photographs were taken in the dark upon irradiation with a hand-held UV lamp ($\lambda_{\text{em}} = 366$ nm).

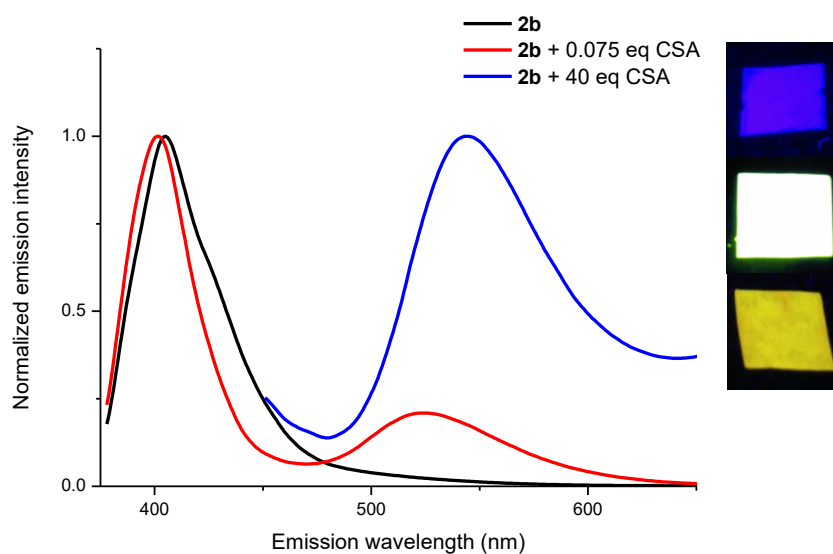


Figure S13. Fluorescence spectra ($\lambda_{\text{exc}} = 365$ nm) and changes in the colour of filter paper samples after immersion into a dichloromethane solution of polystyrene doped with **2b** (1 wt%) in the absence (top) and the presence of 0.075 equivalents (middle) and 40 equivalents (bottom) of CSA. Photographs were taken in the dark upon irradiation with a hand-held UV lamp ($\lambda_{\text{em}} = 366$ nm).

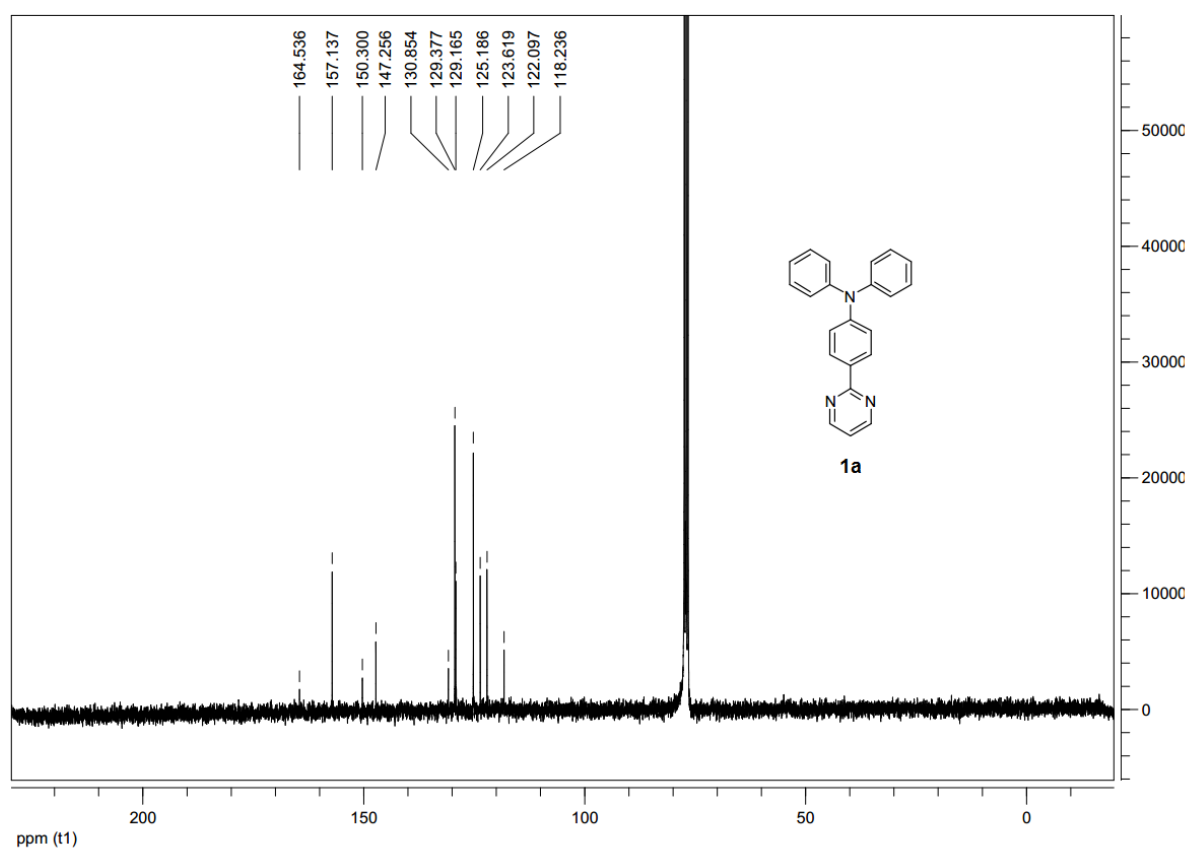
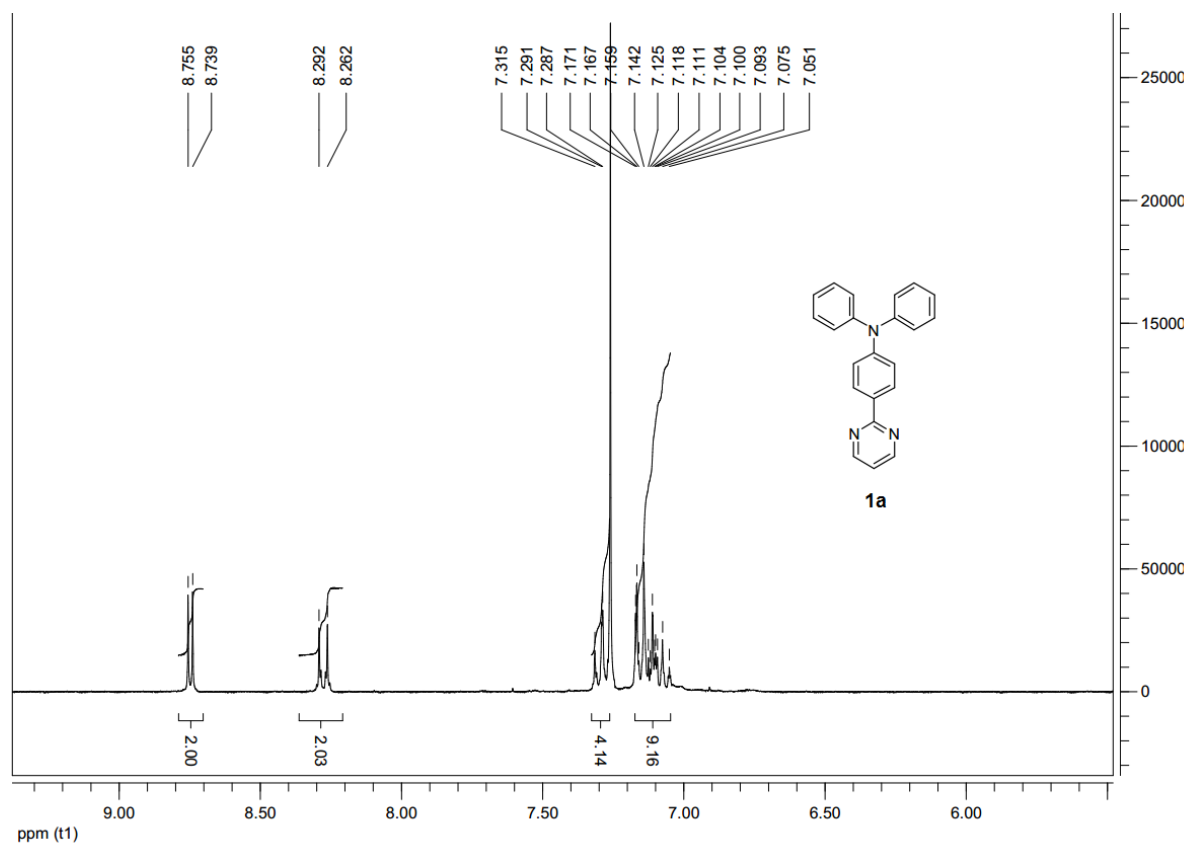


Figure S14. ¹H and ¹³C NMR spectra of compound **1a** in CDCl₃.

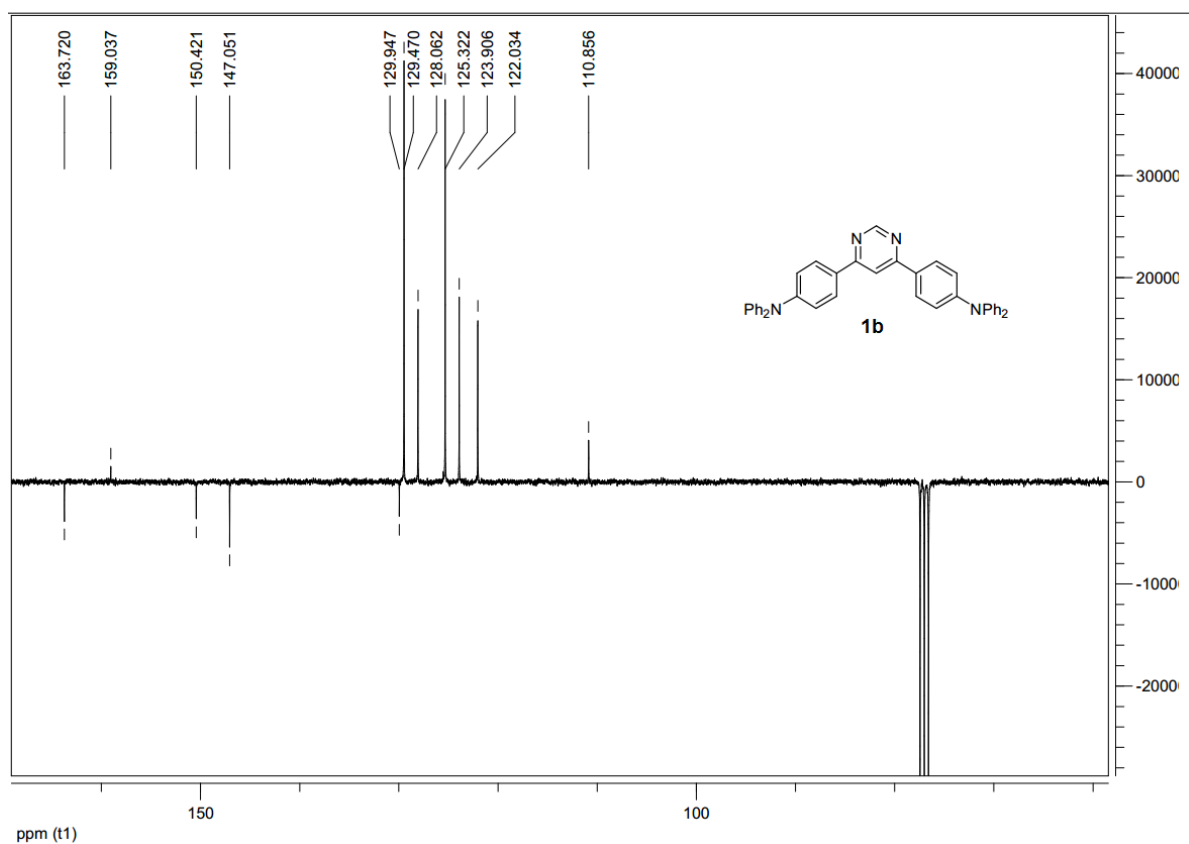
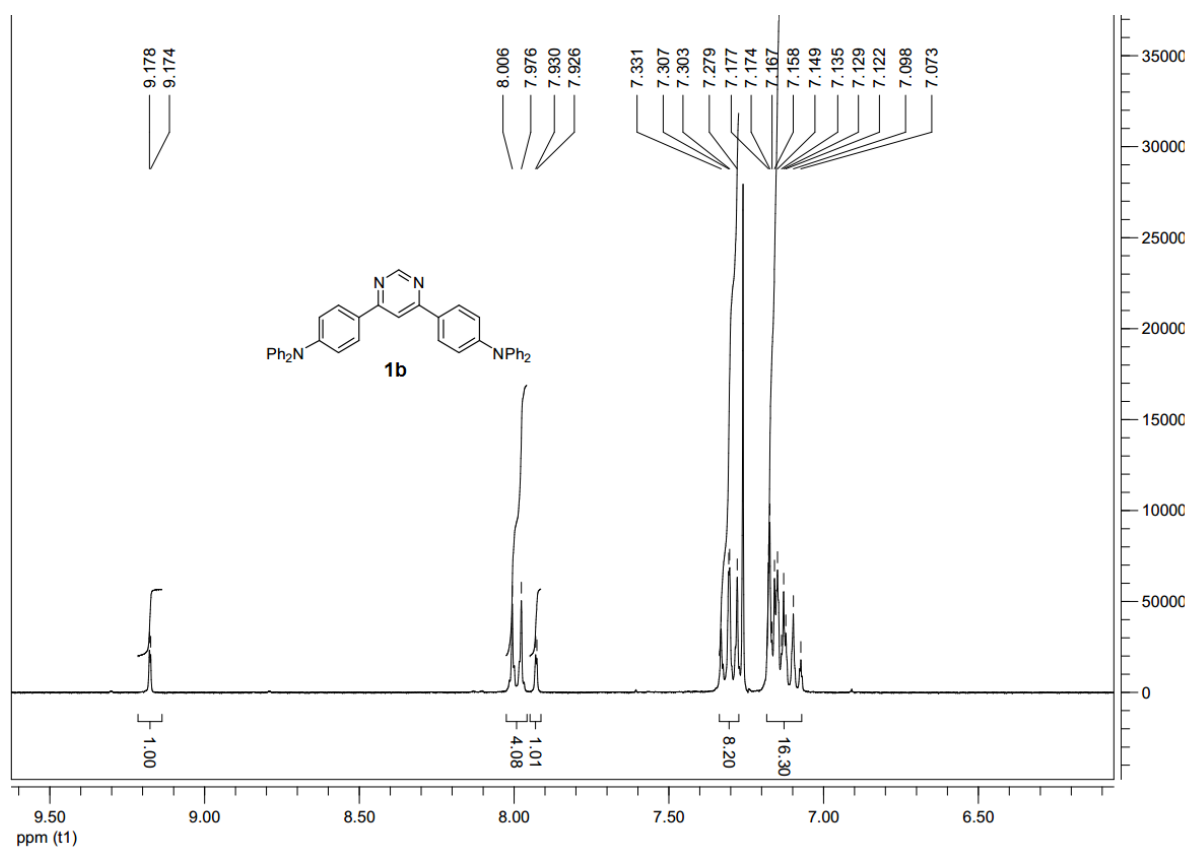


Figure S15. ¹H and ¹³C NMR spectra of compound **1b** in CDCl₃.

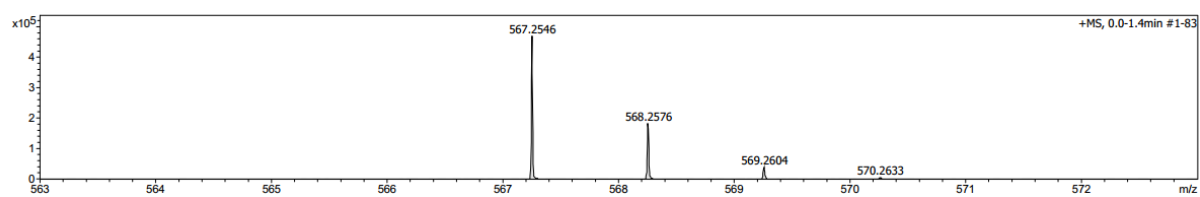


Figure S16. HRMS spectra of compound **1b**.

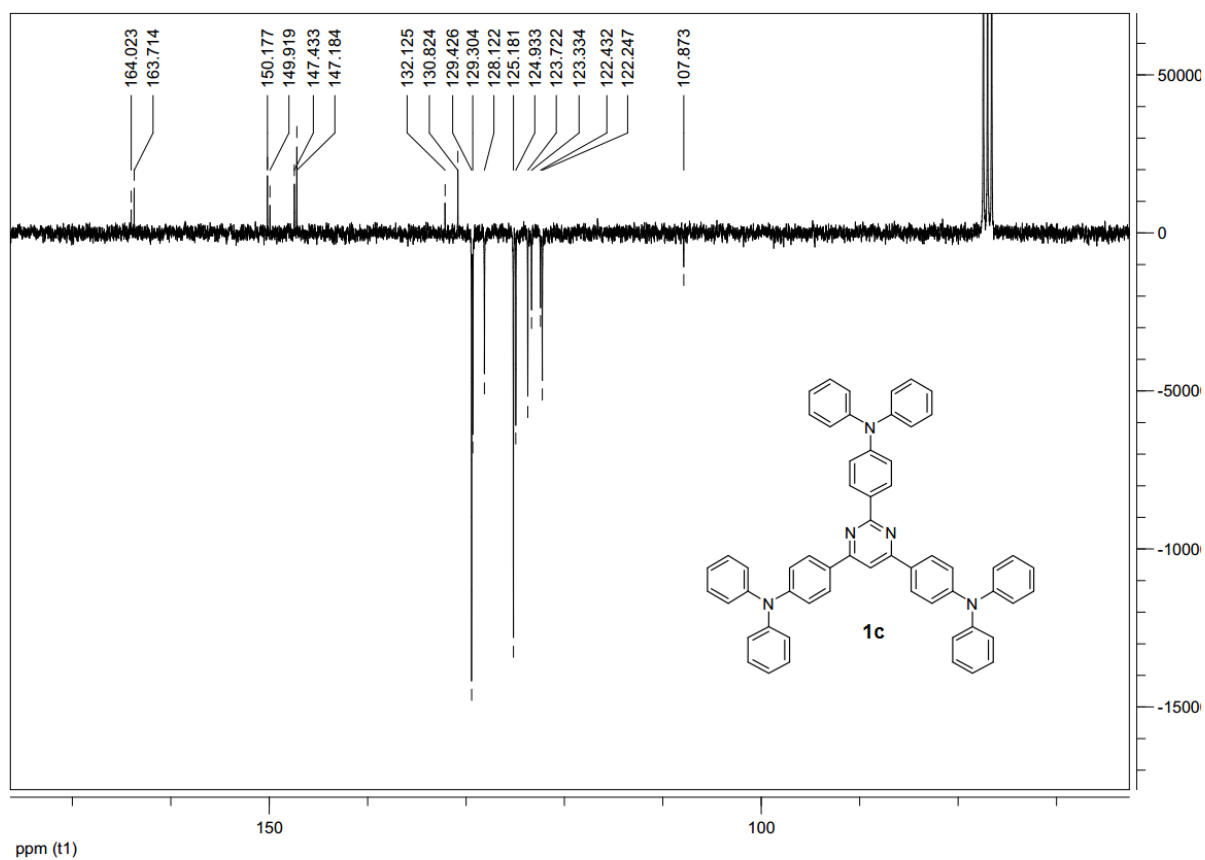
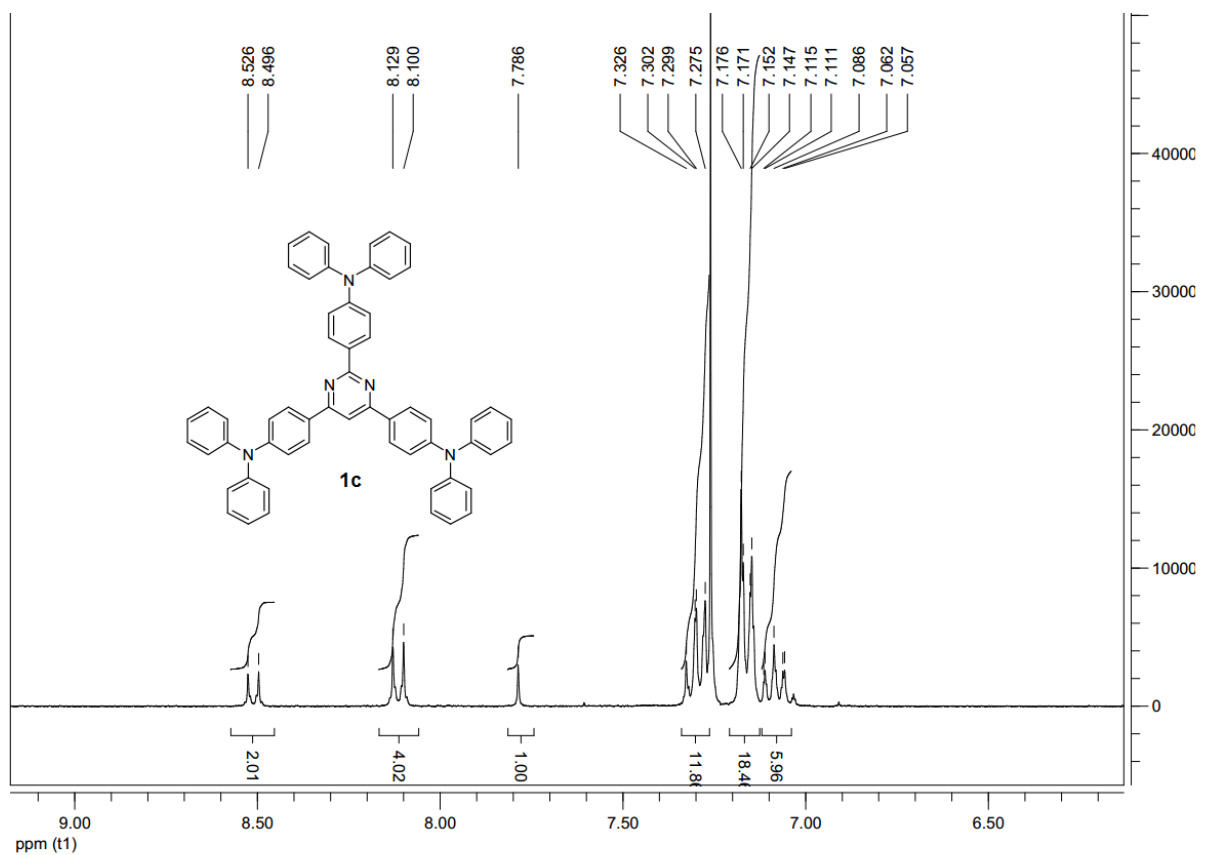


Figure S17. ¹H and ¹³C NMR spectra of compound **1c** in CDCl₃.

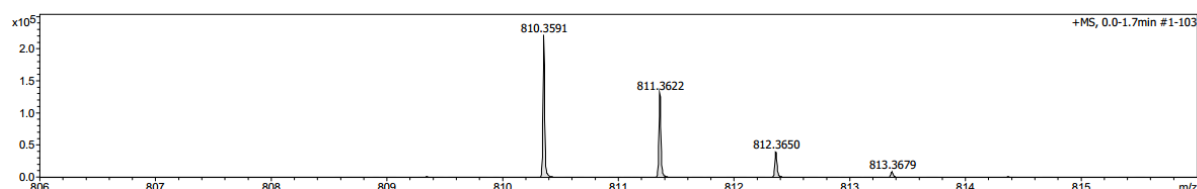


Figure S18. HRMS spectra of compound **1c**.

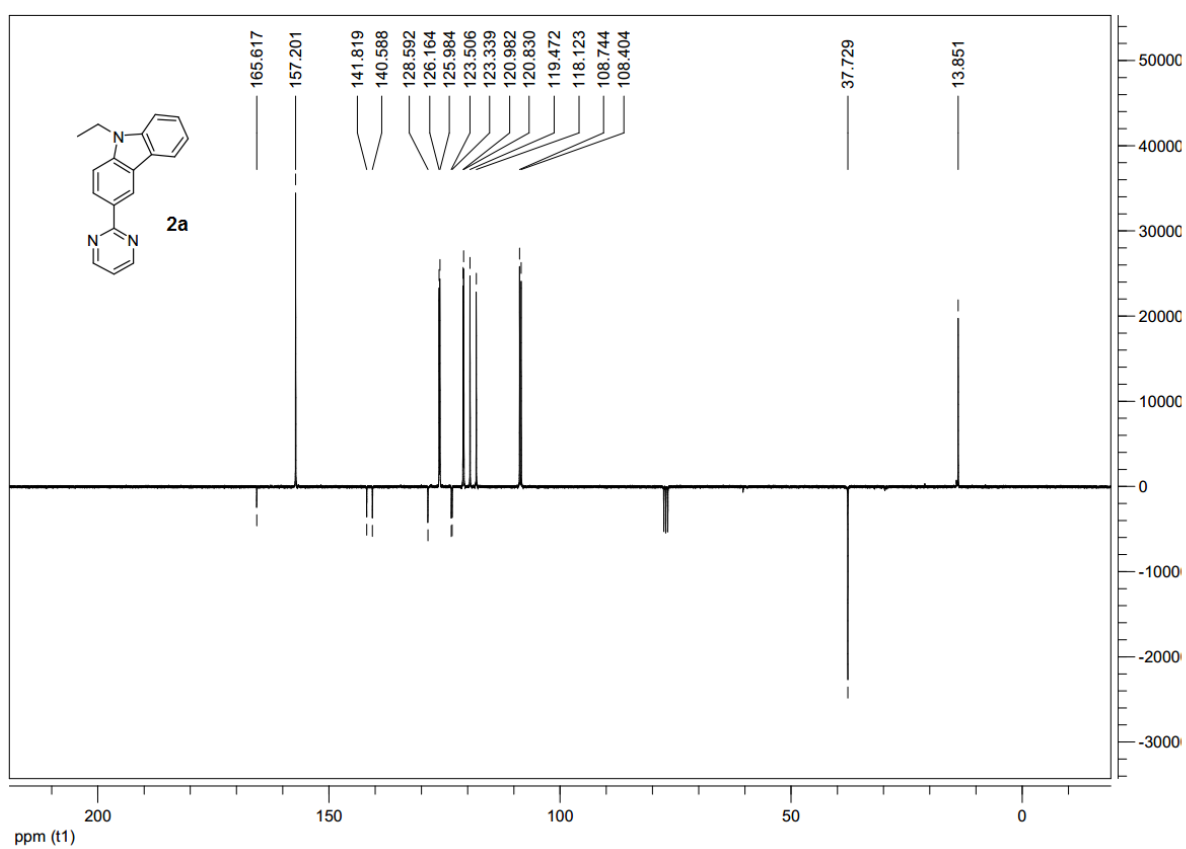
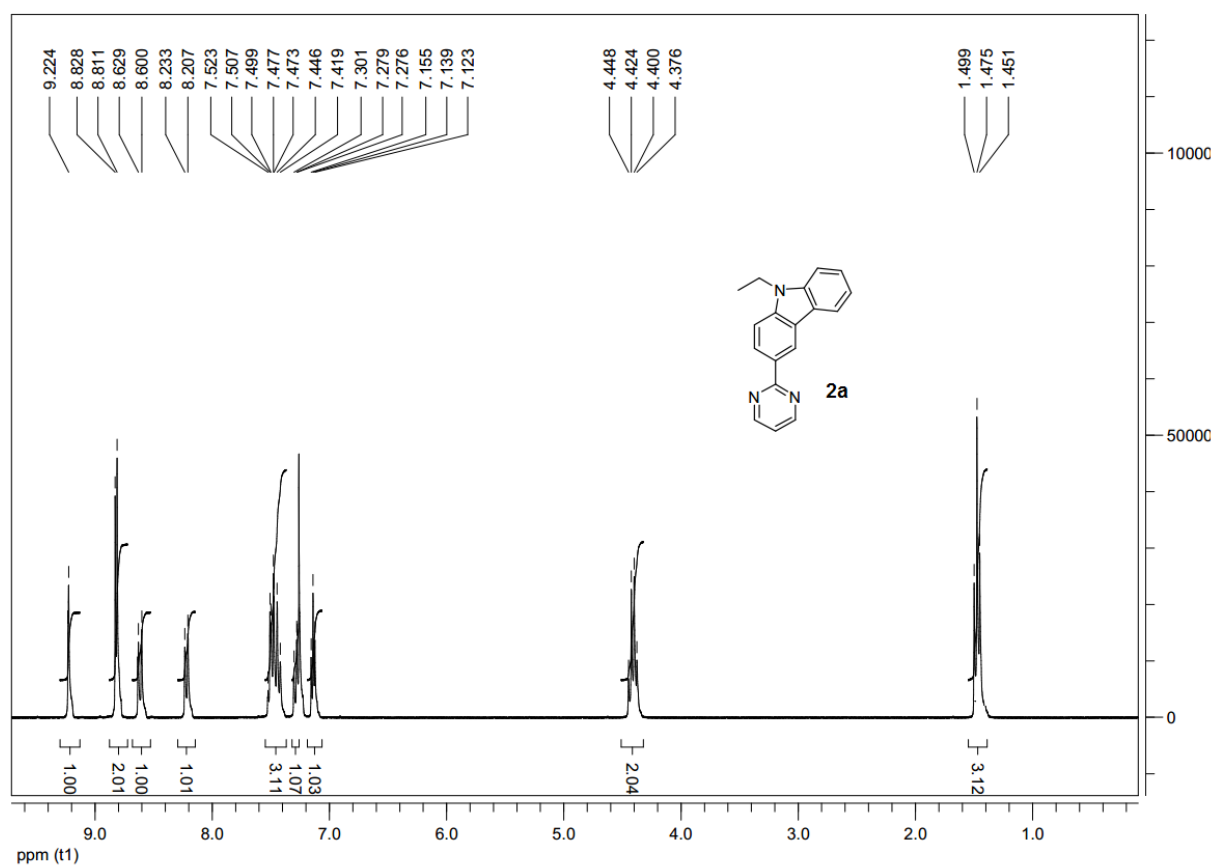


Figure S19. ¹H and ¹³C NMR spectra of compound **2a** in CDCl₃.

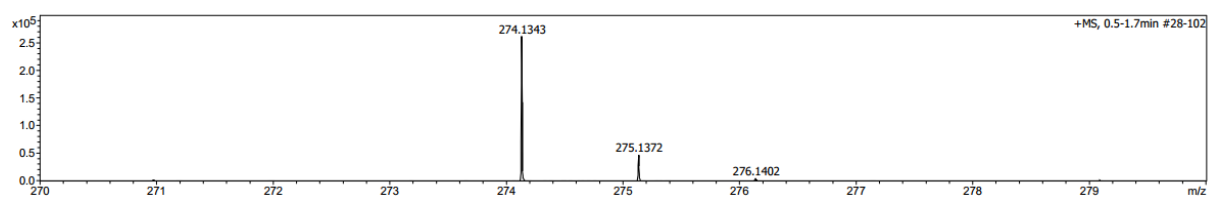


Figure S20. HRMS spectra of compound **2a**.

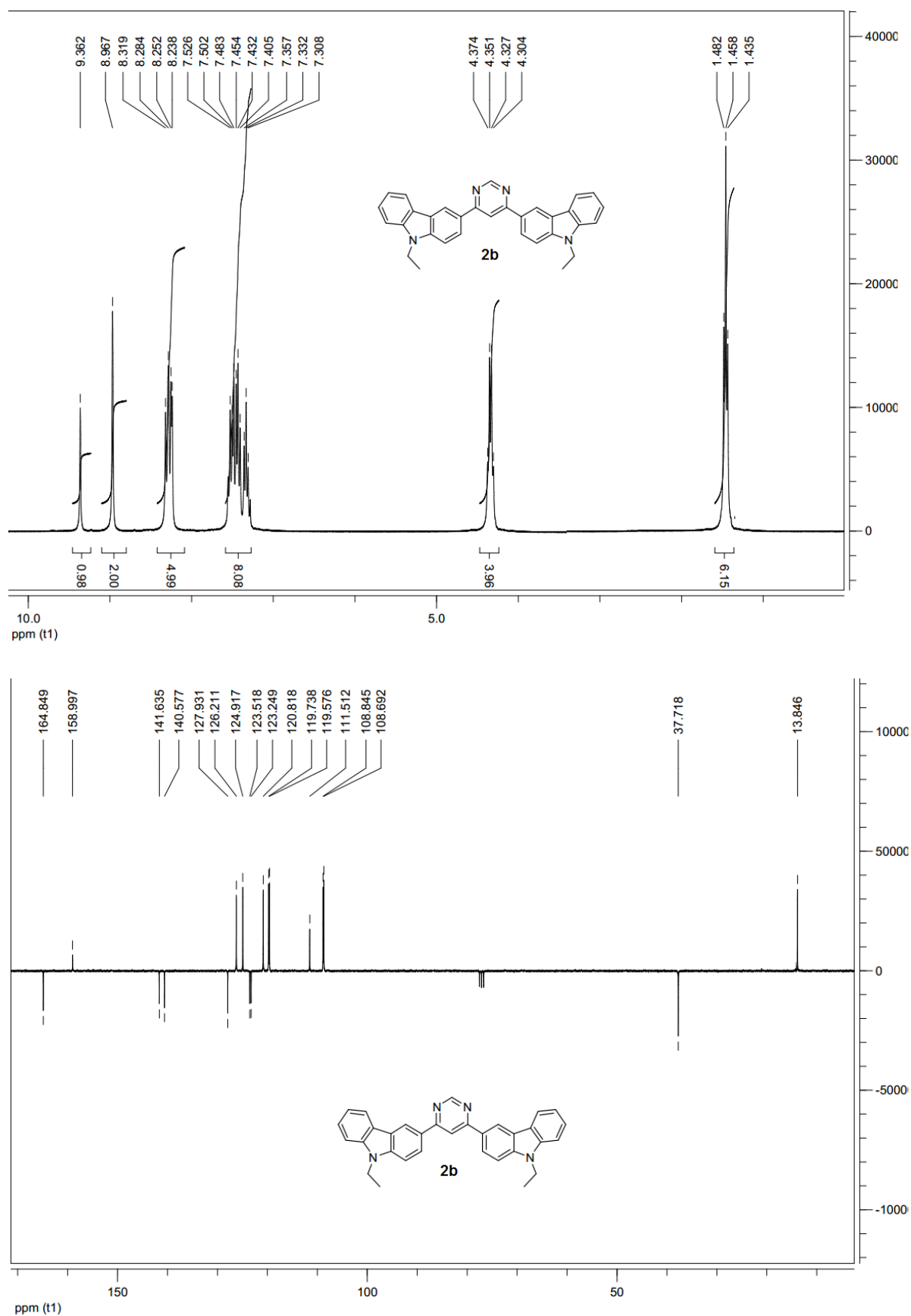


Figure S21. ¹H and ¹³C NMR spectra of compound **2b** in CDCl₃.

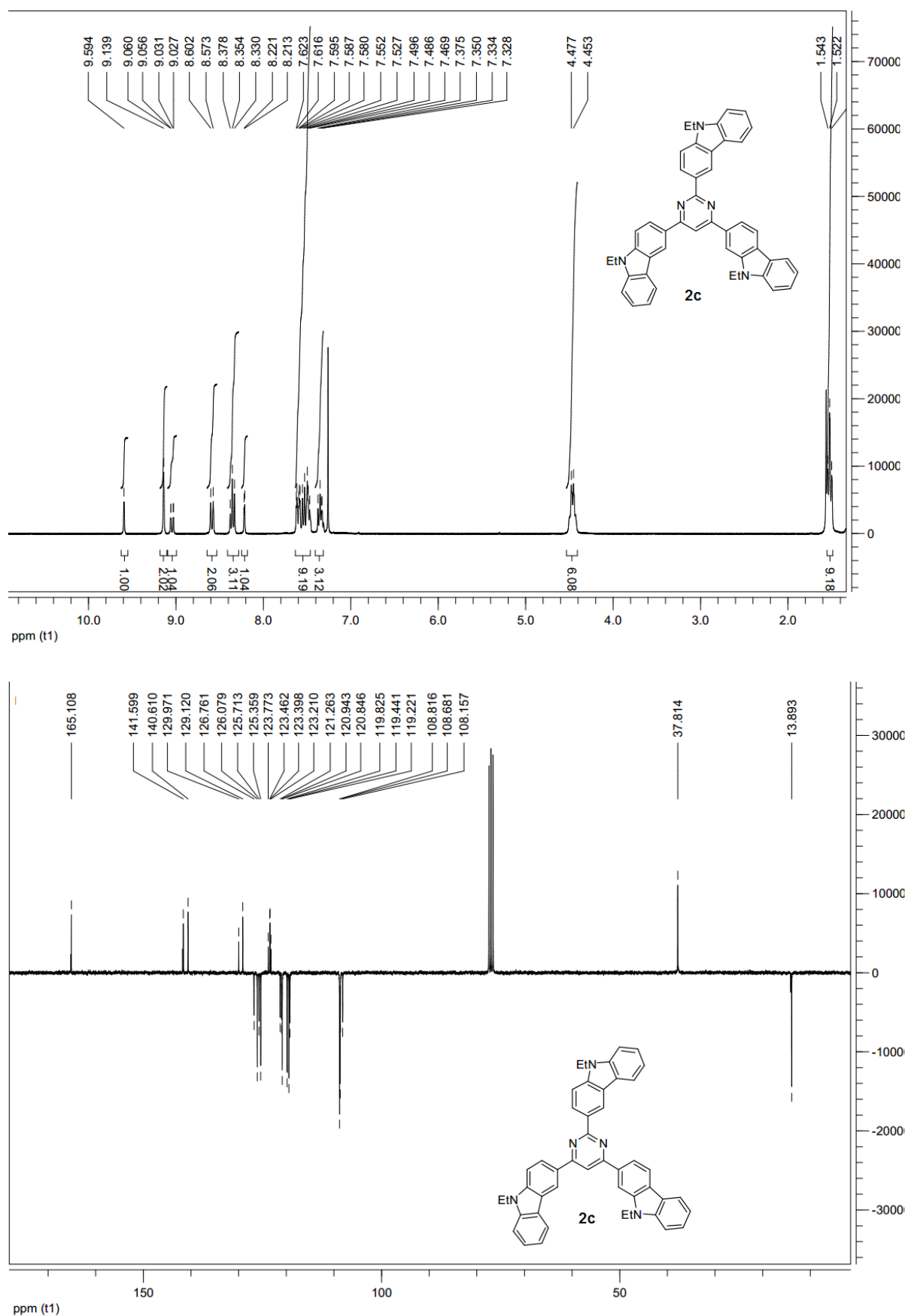


Figure S22. ¹H and ¹³C NMR spectra of compound **2c** in CDCl₃.

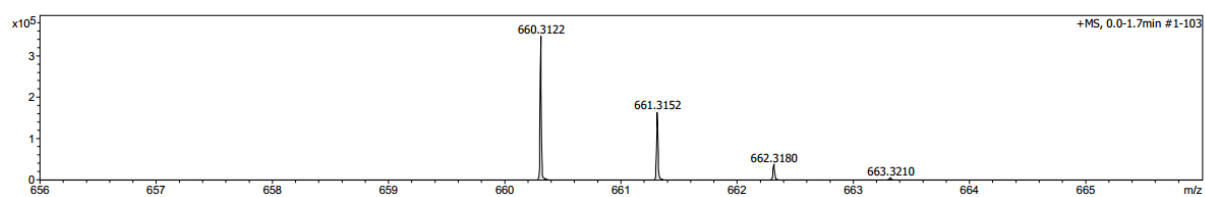


Figure S23. HRMS spectrum of compound 2c.