

# Supporting information

## Design and synthesis of fluorescent coumarin derivatives and their study for Cu<sup>2+</sup> sensing with an application on aqueous soil extracts

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## 1. Characterization of 2a

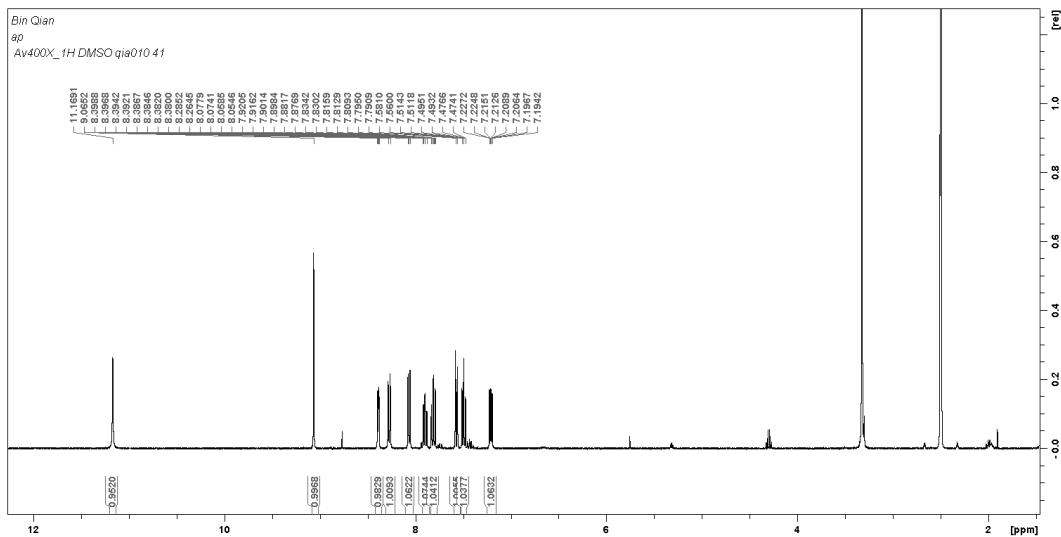
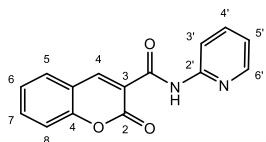


Figure S1  $^1\text{H}$  NMR spectrum of **2a**

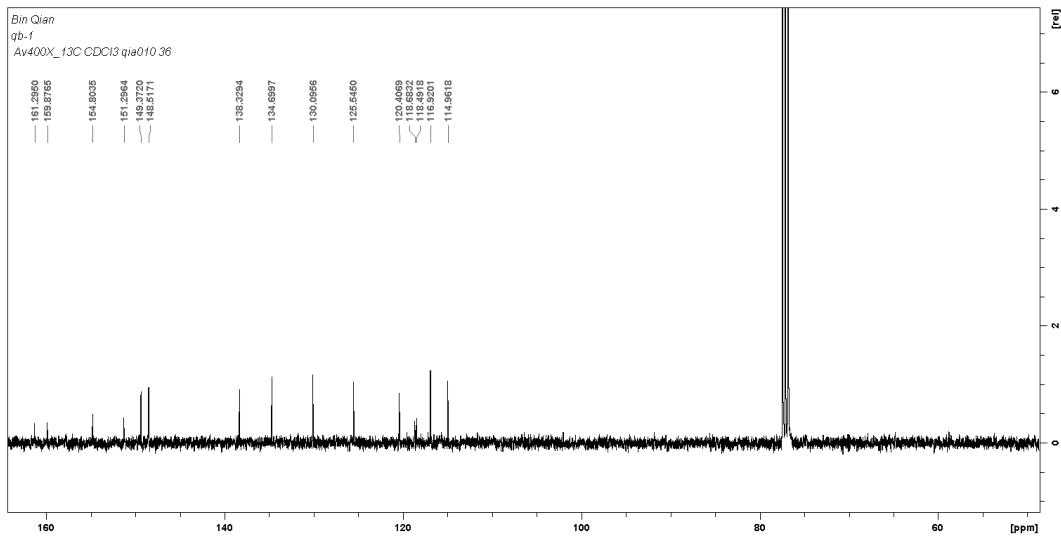


Figure S2  $^{13}\text{C}$  NMR spectrum of **2a**

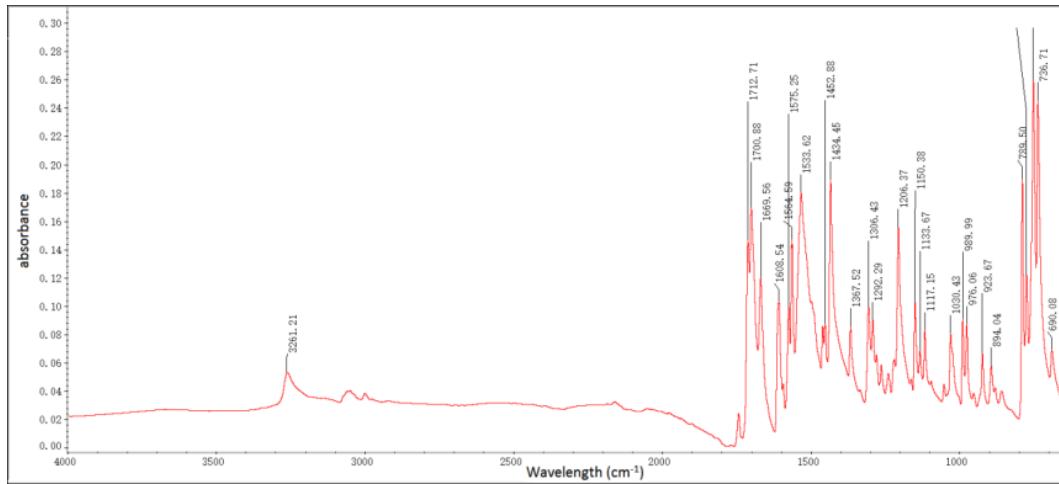


Figure S3 IR spectrum of **2a**

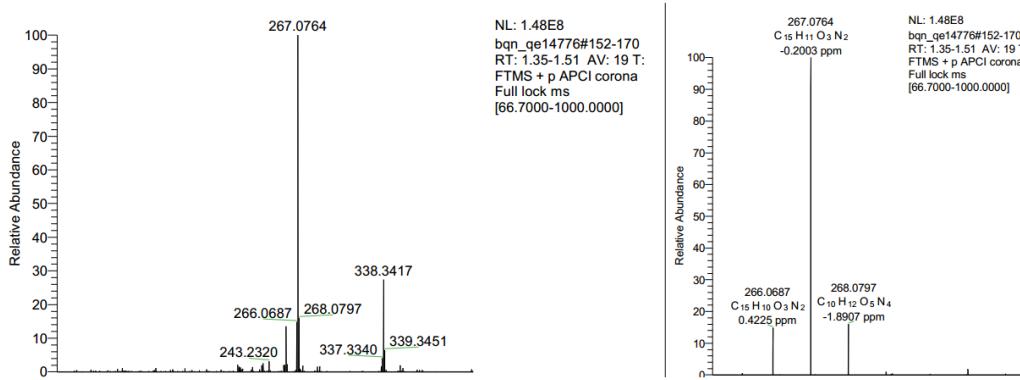
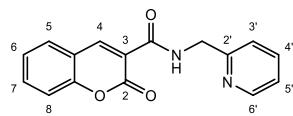


Figure S4 High resolution mass spectra of **2a**

## 2. Characterization of **2b**



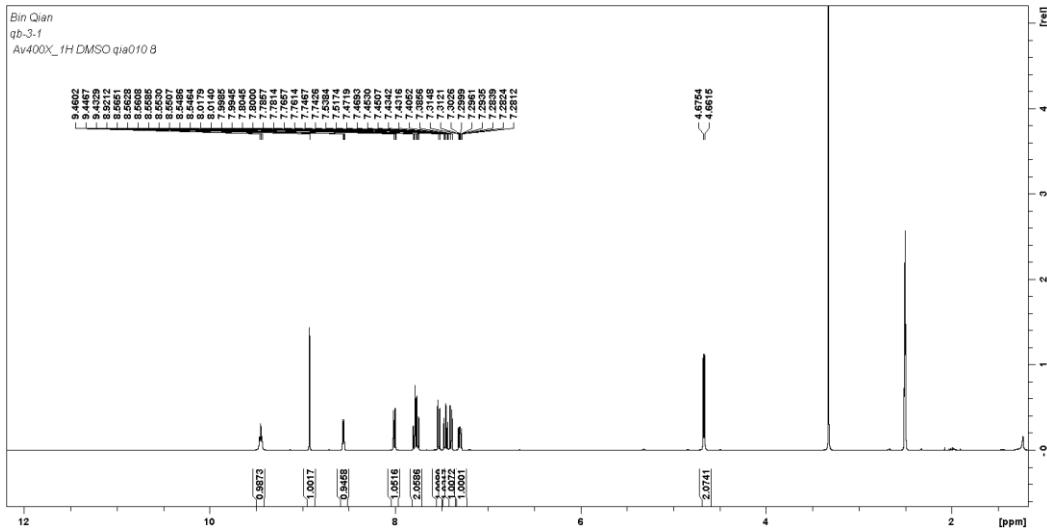


Figure S5  $^1\text{H}$  NMR spectrum of **2b**

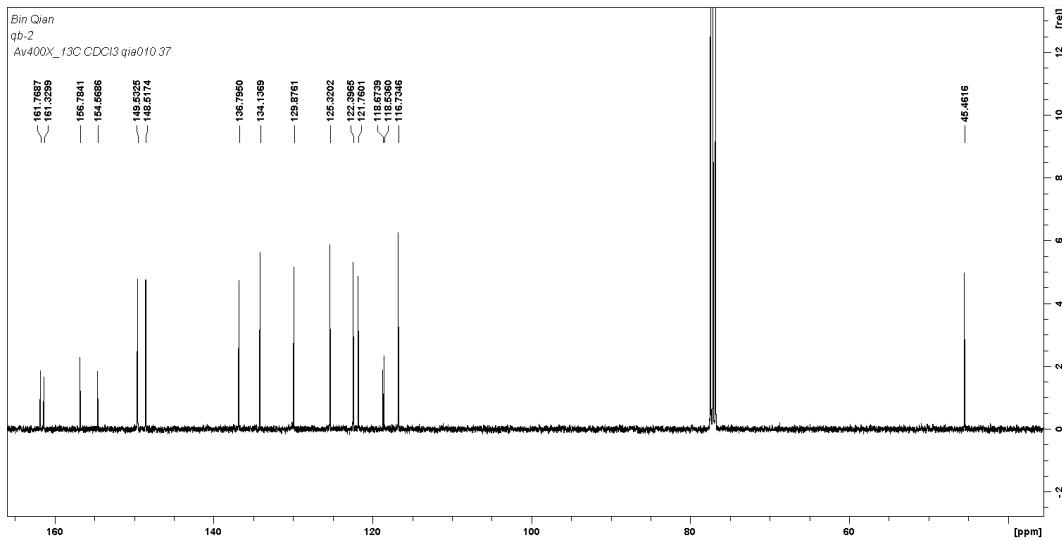


Figure S6  $^{13}\text{C}$  NMR spectrum of **2b**

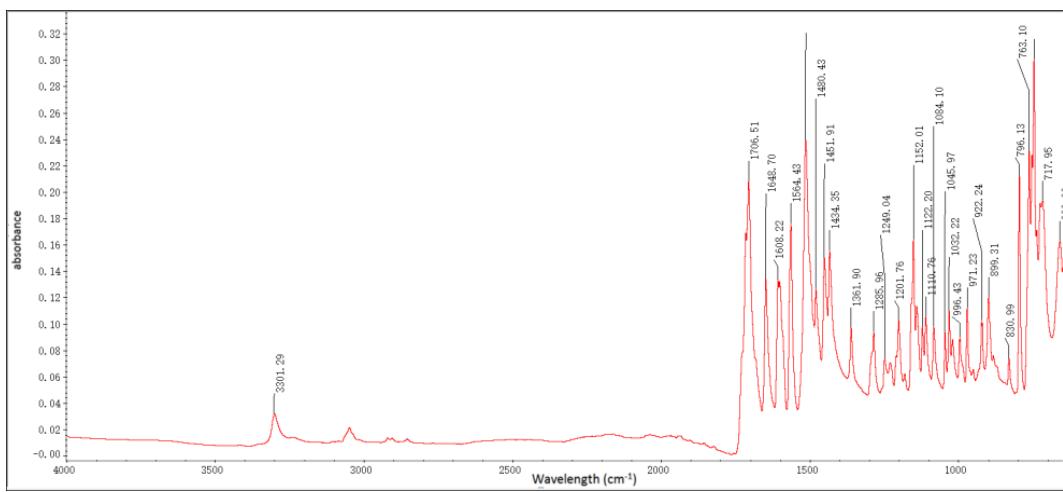


Figure S7 IR spectrum of **2b**

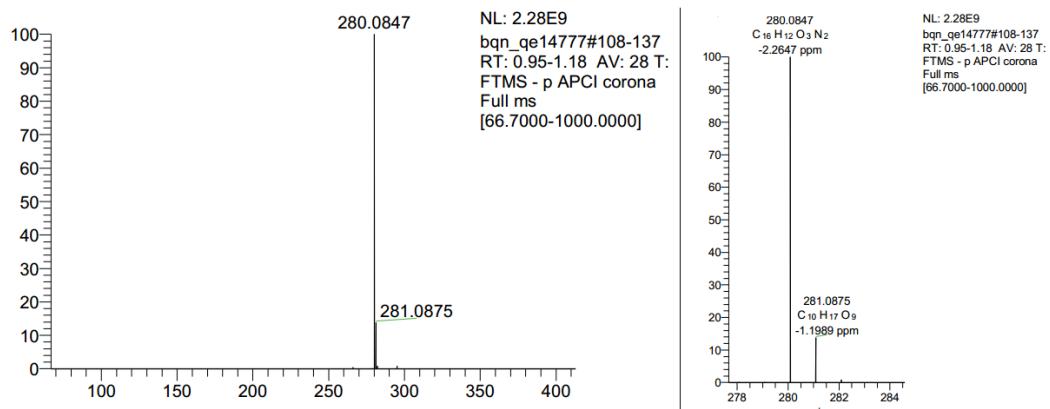
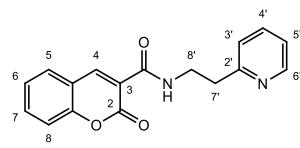


Figure S8 High resolution mass spectra of **2b**

### 3. Characterization of 2c



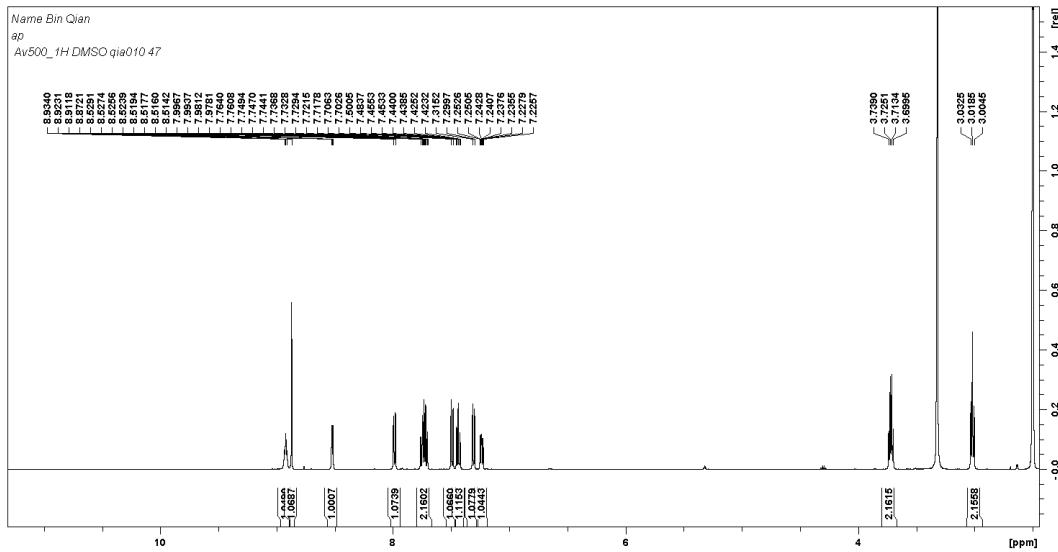


Figure S9  $^1\text{H}$  NMR spectrum of **2c**

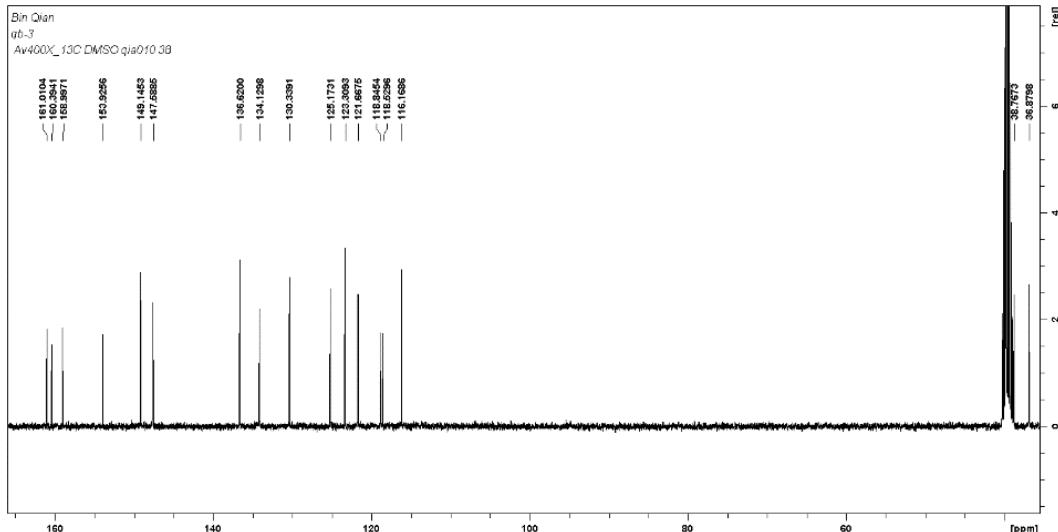


Figure S10  $^{13}\text{C}$  NMR spectrum of **2c**

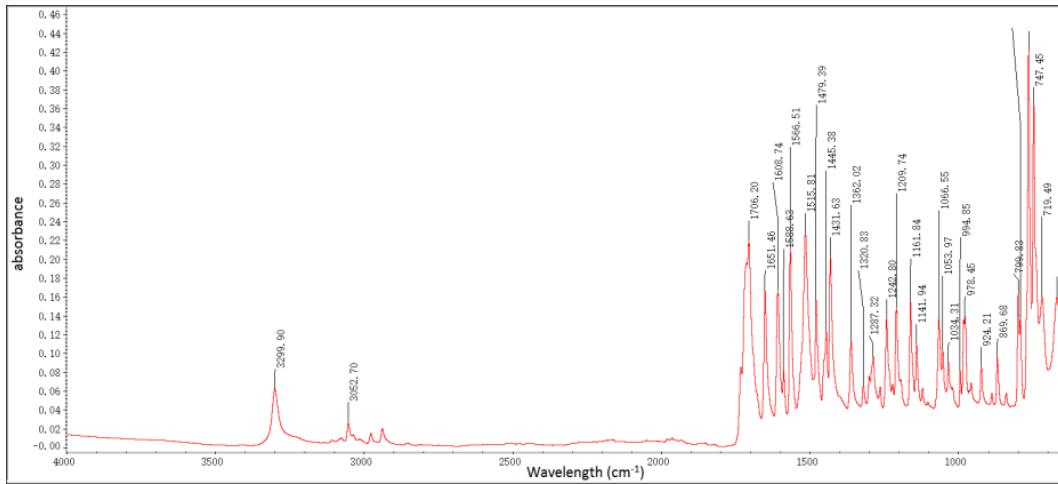


Figure S11 spectrum of **2c**

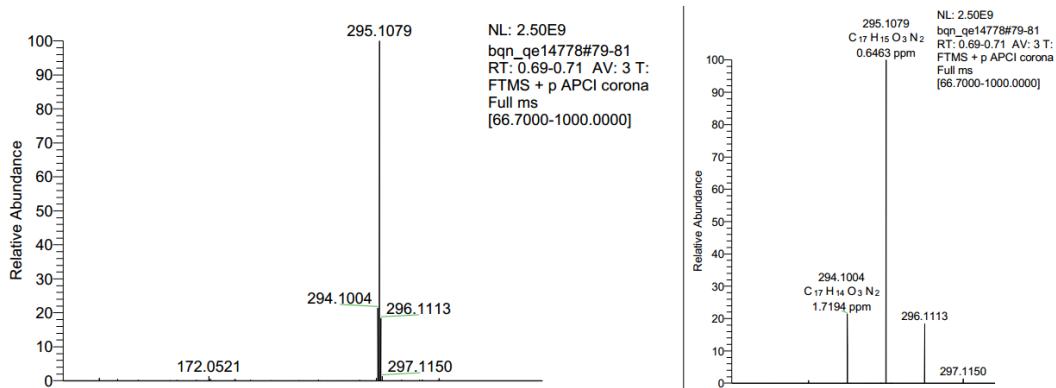
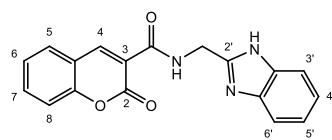


Figure S12 High resolution mass spectra of **2c**

#### 4. Characterization of **2d**



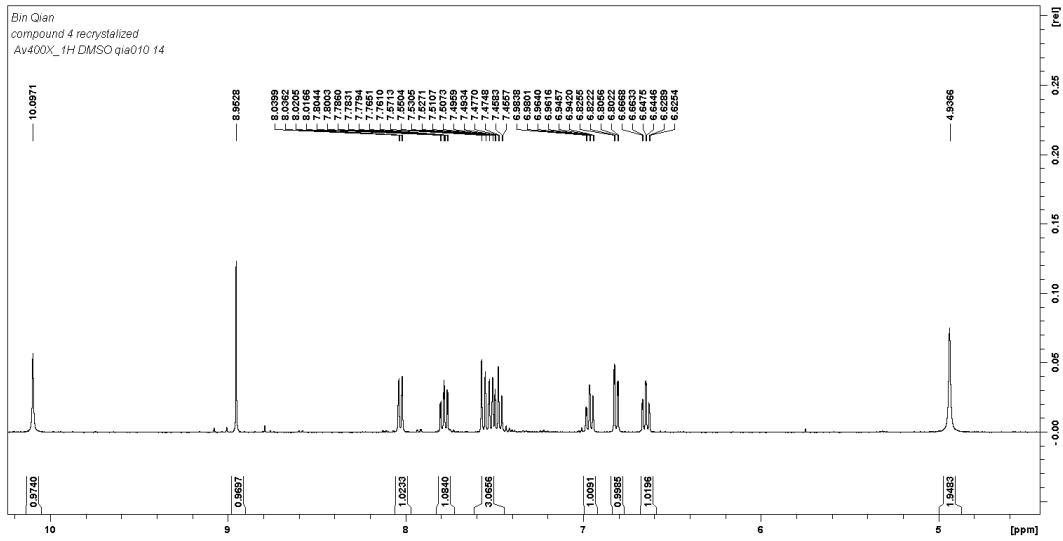


Figure S13  $^1\text{H}$  NMR spectrum of **2d**

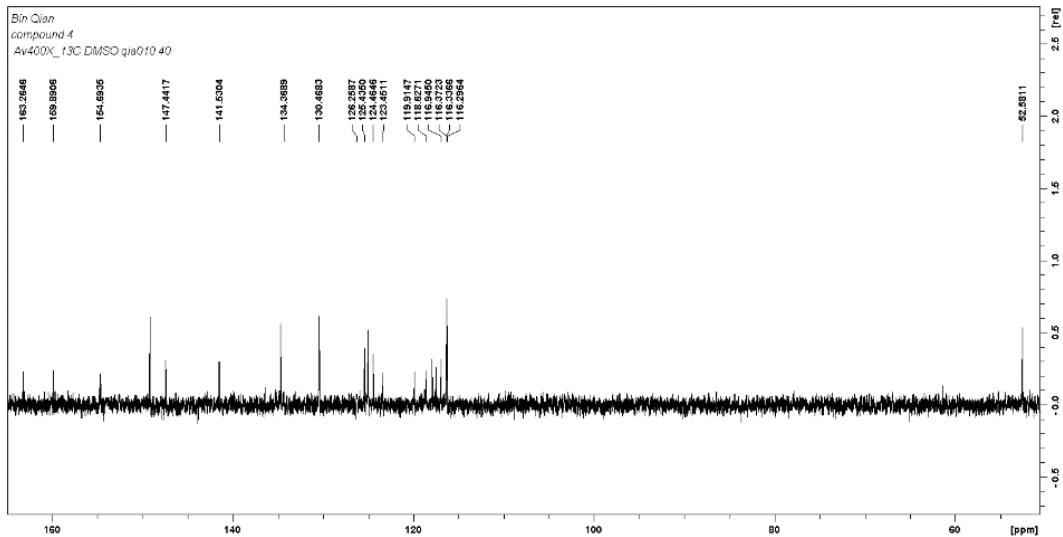


Figure S14  $^{13}\text{C}$  NMR spectrum of **2d**

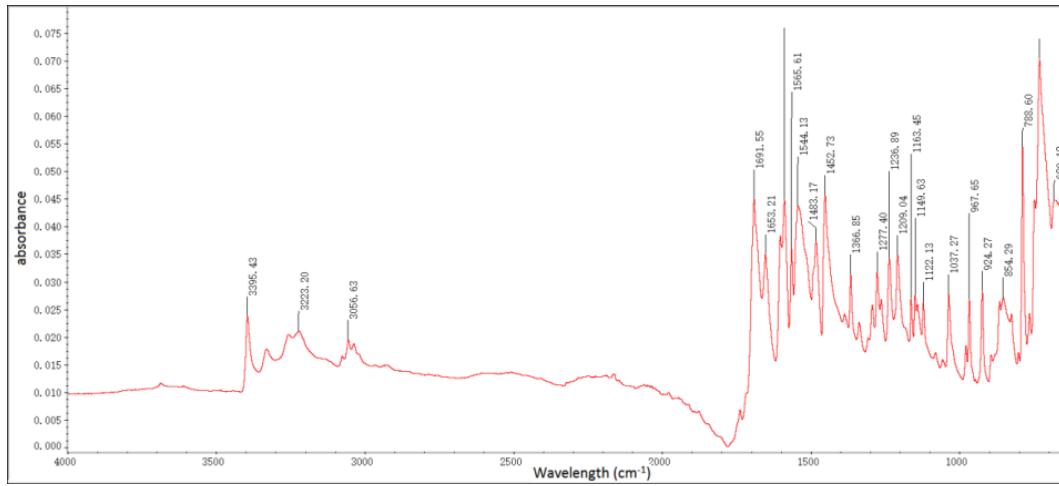


Figure S15 IR spectrum of **2d**

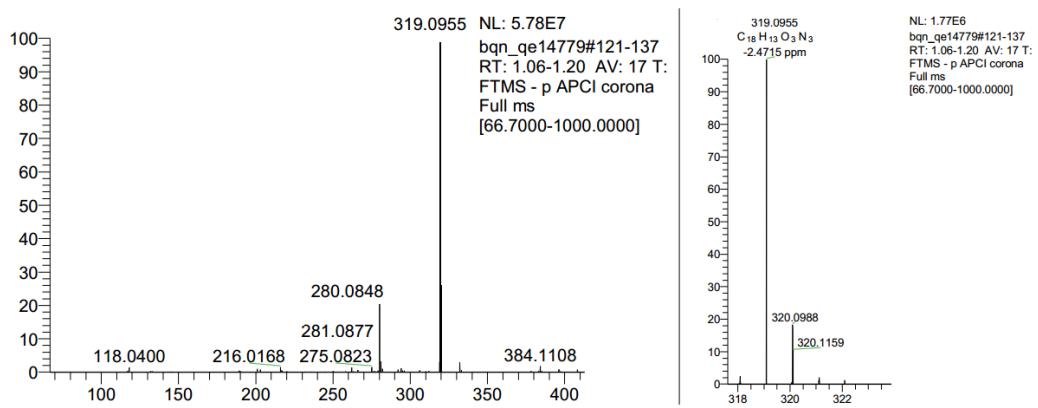
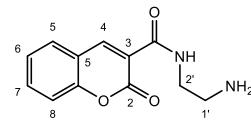


Figure S16 High resolution mass spectra of **2d**

## 5. Characterization of **2e**



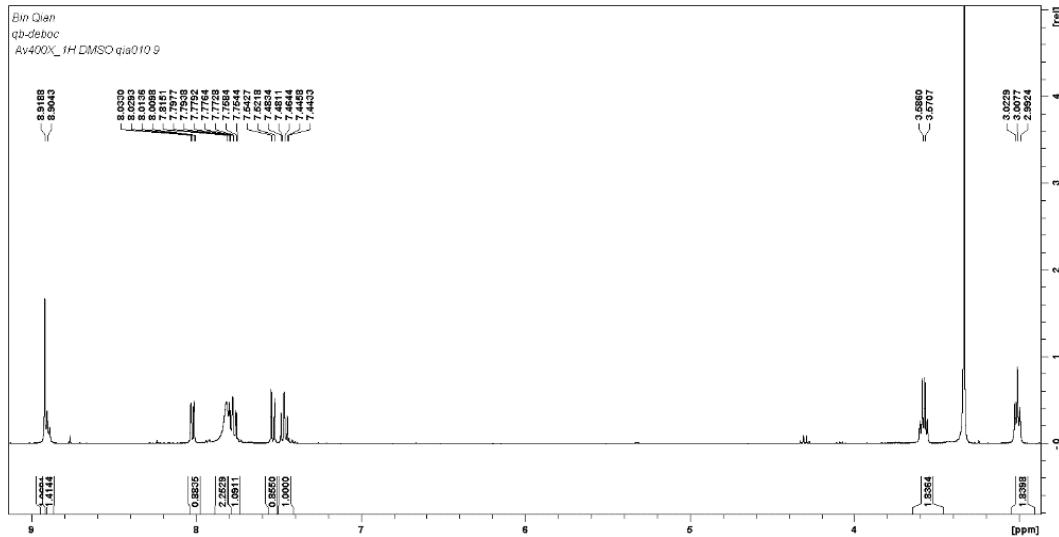


Figure S17  $^1\text{H}$  NMR spectrum of **2e**

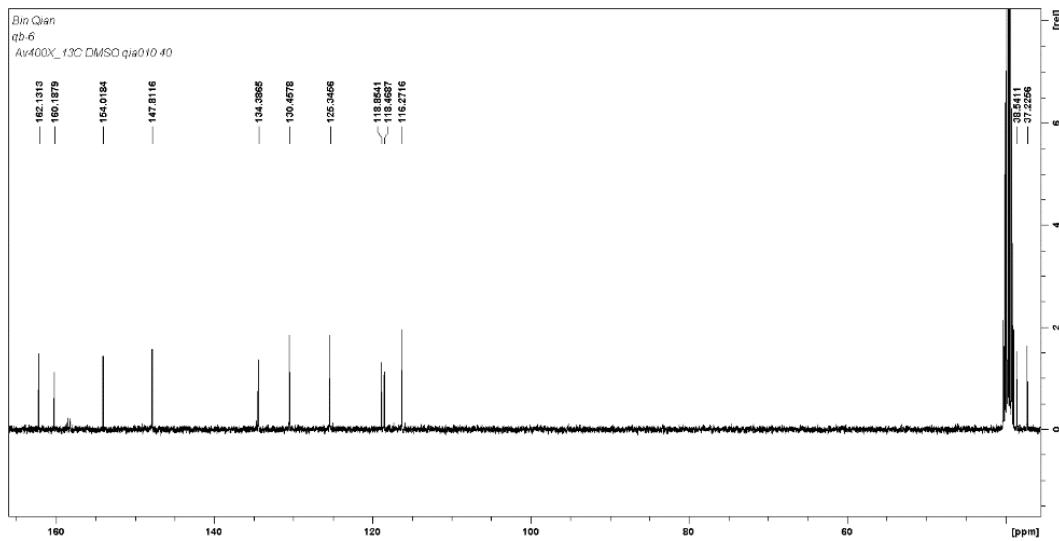


Figure S18  $^{13}\text{C}$  NMR spectrum of **2e**

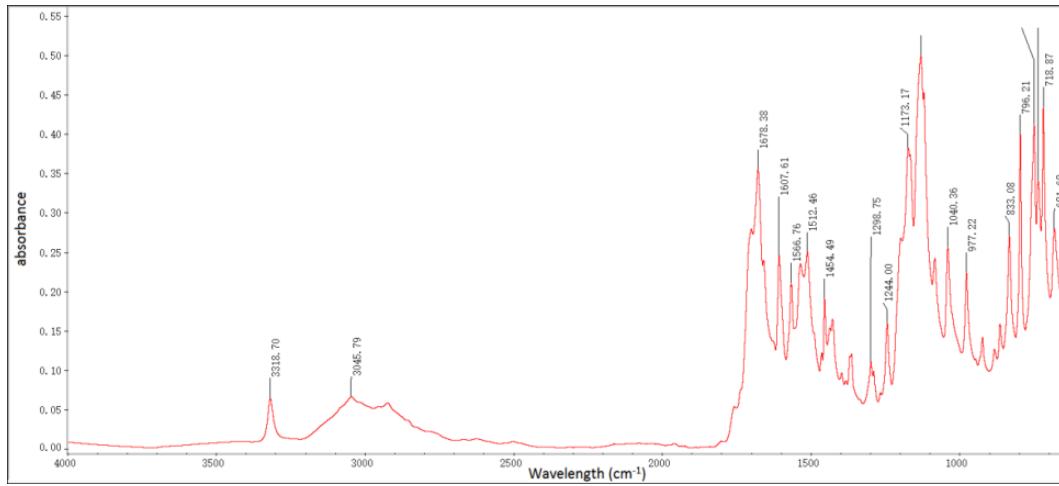


Figure S19 IR spectrum of **2e**

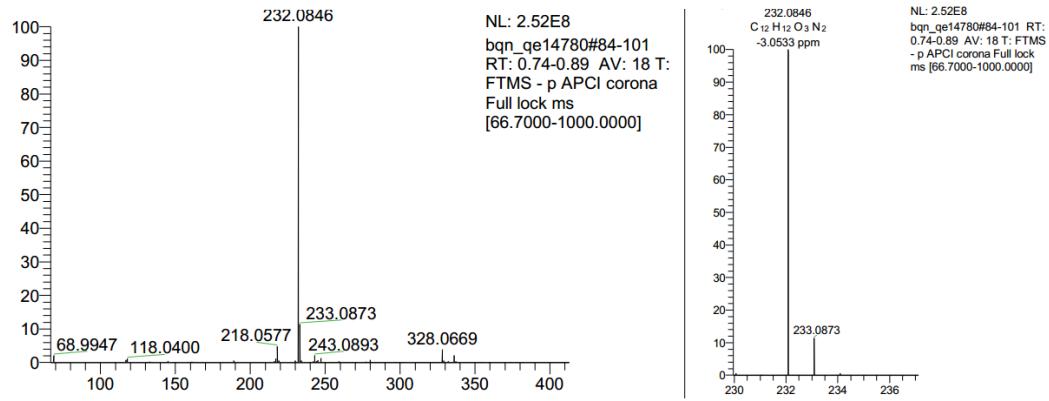


Figure S20 High resolution mass spectra of **2e**

## 6. UV-Vis absorbance for 2a-e with and without 1 eq. Cu<sup>2+</sup>

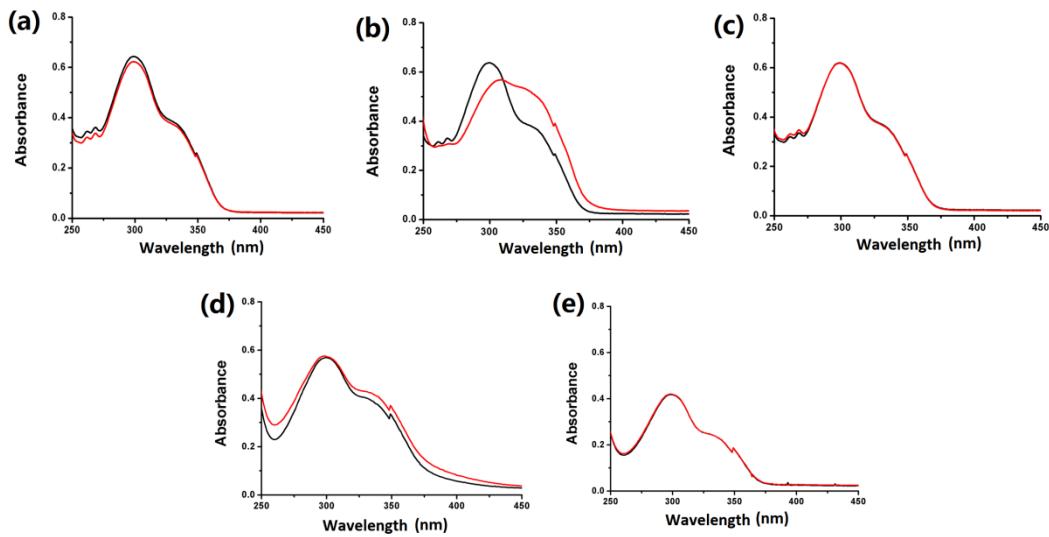


Figure S21 Uv-Vis absorbance **2a-e** (30  $\mu\text{M}$ ) in DMSO/HEPES buffer (v/v, 1/9) in the absence (black lines) and presence (red lines) of 1 equivalent of CuCl<sub>2</sub>: (a) **2a**; (b) **2b**; (c) **2c**; (d) **2d**; (e) **2e**

## 7. Kinetics study of 2b and 2d

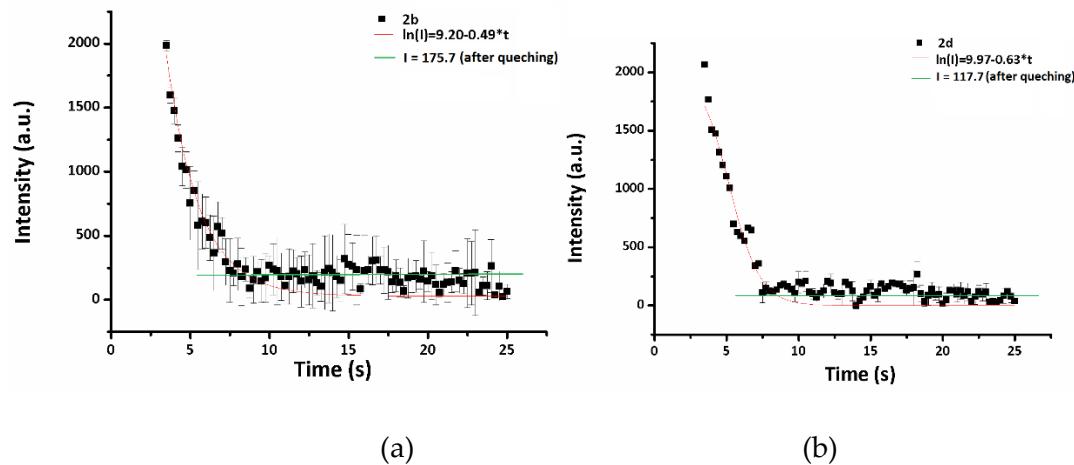


Figure S22 Kinetics study of (a) **2b** ( $\lambda_{\text{ex}}=303 \text{ nm}$ ) and (b) **2d** ( $\lambda_{\text{ex}}=325 \text{ nm}$ ) (30  $\mu\text{M}$ ) with the addition of 2 eq. of CuCl<sub>2</sub> in HEPES/DMSO buffer

## 8. Interference study of **2b** and **2d**

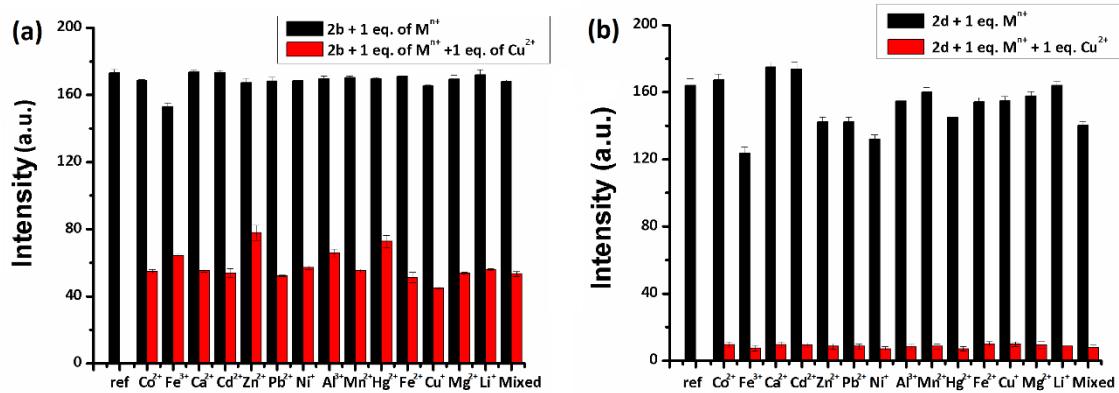


Figure S23 Fluorescence emission intensities of (a) **2b** ( $\lambda_{\text{ex}}=303 \text{ nm}$ ) and (b) **2d** ( $\lambda_{\text{ex}}=325 \text{ nm}$ ) both  $30 \mu\text{M}$  in the presence (red lines) and absence (black lines) of  $\text{Cu}^{2+}$  with 20 equivalents of various cations in DMSO/HEPES buffer.

Table S1 Fluorescence intensity variations of **2b** with adding 20 eq. metal ions

<b>2b</b>	Intensity (a.u.) after adding 20 eq. metal ions	$\Delta$ Intensity	$\Delta\%$
ref	168.6245		
$\text{Cu}^{2+}$	50.408		
$\text{Co}^{2+}$	160.9087	7.71575	4.58%
$\text{Fe}^{3+}$	144.838	23.78648	14.11%
$\text{Ca}^{2+}$	159.633	8.99144	5.33%
$\text{Cd}^{2+}$	151.3966	17.22786	10.22%
$\text{Zn}^{2+}$	160.2103	8.41412	4.99%
$\text{Pb}^{2+}$	164.7546	3.8699	2.29%
$\text{Ni}^{+}$	155.4868	13.13762	7.79%
$\text{Al}^{3+}$	154.2941	14.33038	8.50%
$\text{Mn}^{2+}$	157.4268	11.19762	6.64%
$\text{Hg}^{2+}$	159.9028	8.72169	5.17%
$\text{Fe}^{2+}$	165.4378	3.18669	1.89%
$\text{Cu}^{+}$	154.2277	14.39674	8.54%
$\text{Mg}^{2+}$	166.3161	2.30838	1.37%
$\text{Li}^{+}$	162.8126	5.81187	3.45%
Mixed	151.2277	17.39674	10.32%

Table S2 Fluorescence intensity variations of **2d** with adding 20 eq. metal ions

<b>2d</b>	Intensity (a.u.) after adding 20 eq. metal ions	$\Delta$ Intensity	$\Delta\%$
ref	164.274		
$\text{Cu}^{2+}$	7.43938		
$\text{Co}^{2+}$	124.0102	40.2638	24.51%
$\text{Fe}^{3+}$	80.86041	83.41356	50.78%

$\text{Ca}^{2+}$	150.4194	13.85461	8.43%
$\text{Cd}^{2+}$	159.4749	4.79904	2.92%
$\text{Zn}^{2+}$	159.7672	4.50674	2.74%
$\text{Pb}^{2+}$	151.5827	12.6913	7.73%
$\text{Ni}^{+}$	131.5143	32.75965	19.94%
$\text{Al}^{3+}$	154.5642	9.70979	5.91%
$\text{Mn}^{2+}$	160.1416	4.1324	2.52%
$\text{Hg}^{2+}$	128.713	35.561	21.65%
$\text{Fe}^{2+}$	156.2374	8.03659	4.89%
$\text{Cu}^{+}$	149.3737	14.90027	9.07%
$\text{Mg}^{2+}$	155.7565	8.51744	5.18%
$\text{Li}^{+}$	155.7248	8.54913	5.20%
Mixed	134.0102	30.2638	18.42%

## 9. Mass spectra of 2b- $\text{Cu}^{2+}$

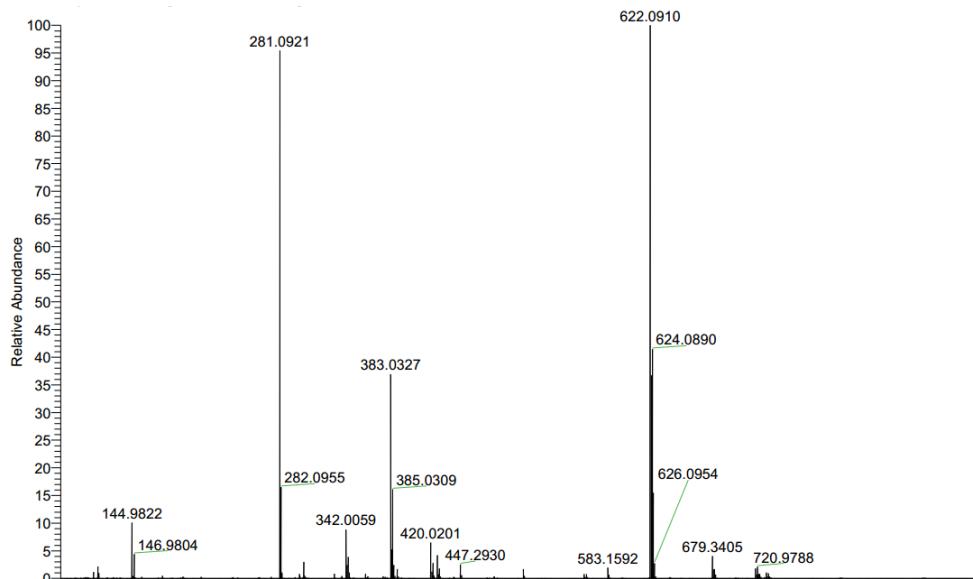


Figure S24 LRMS of 2b- $\text{Cu}^{2+}$

## 10. IR spectra of **2b** with Cu<sup>2+</sup>

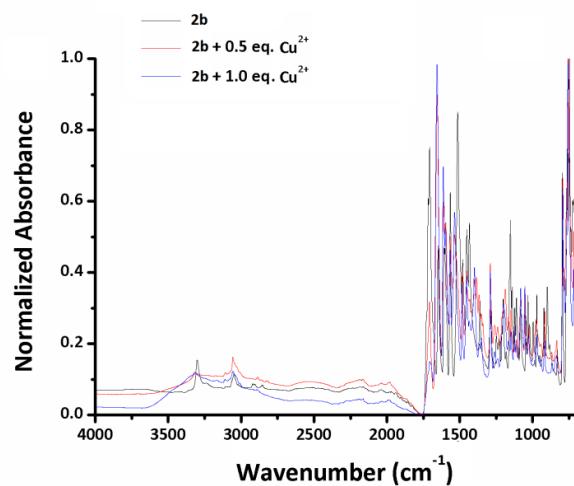


Figure S25 Normalized IR spectra of **2b** with different ratios of Cu<sup>2+</sup>

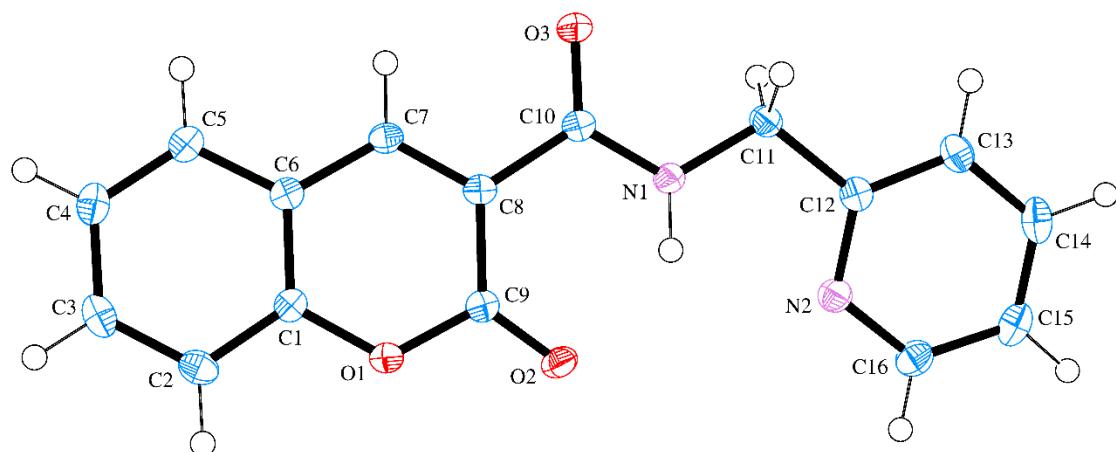


Figure S26 Single crystal structure of **2b**

## 11. Soil tests

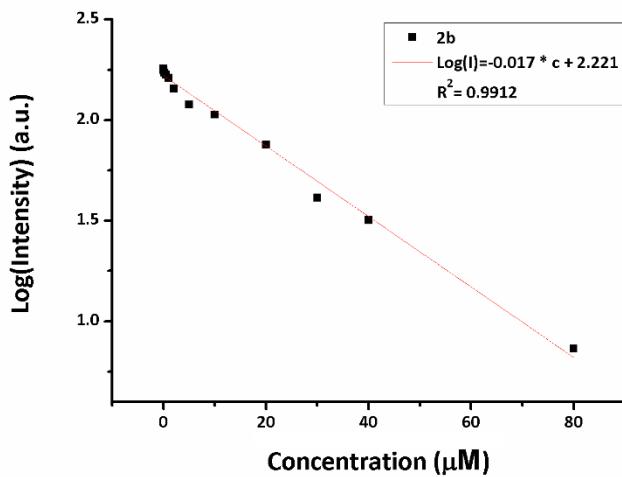


Figure S27 Standard curve for  $\text{Cu}^{2+}$  sensing by **2b** in DMSO/HEPES buffer (v/v, 1/9, 20 mM, pH=7)

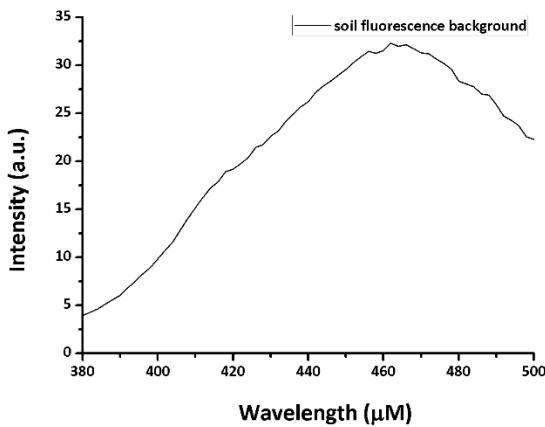


Figure S28 Fluorescence background for soil extracts in DMSO/HEPES buffer (v/v, 1/9, 20 mM, pH=7)

## 12. Single crystal data for **2b** and **2b-Cu<sup>2+</sup>**

Table S3 Crystal data and structure refinement for **2b**

Formula	$\text{C}_{16}\text{H}_{12}\text{N}_2\text{O}_3$			
Formula weight	280.28			
Crystal system	monoclinic			
Crystal size (mm³)	0.388 x 0.046 x 0.034			
Space group	P 21/c			
Unit cell dimensions	a (Å)	5.0235 (5)	$\alpha^\circ$	90

	<b>B (Å)</b>	21.729 (2)	<b>β°</b>	7.960 (2)
	<b>c (Å)</b>	12.0529 (12)	<b>γ°</b>	90
<b>Volume (Å<sup>3</sup>)</b>	1303.0(2)			
<b>Z</b>	4			
<b>Theta range for data collection (°)</b>	1.874 to 27.499			
<b>Index ranges</b>	-6<=h<=6, -23<=k<=28, -15<=l<=15			
<b>Reflections collected</b>	12755			
<b>Refinement method</b>	Full-matrix least-squares on F <sup>2</sup>			
<b>Data / restraints / parameters</b>	2994 / 0 / 193			
<b>Goodness-of-fit on F<sup>2</sup></b>	1.037			
<b>Calculated density (Mg/cm<sup>3</sup>)</b>	1.429			
<b>Absorption coefficient (mm<sup>-1</sup>)</b>	0.101			
<b>F(000)</b>	584			
<b>Max. and min. transmission</b>	0.746 and 0.714			
<b>Goodness-of-fit on F<sup>2</sup></b>	0.565			
<b>Final R indices</b>	<b>I&gt;2σ(I)</b>	R1 = 0.0365, wR2 = 0.0984		
	<b>all data</b>	R1 = 0.0433, wR2 = 0.1027		

Table S4 Crystal data and structure refinement for **2b**-Cu<sup>2+</sup>

<b>Formula</b>	C <sub>16</sub> H <sub>11</sub> ClCuN <sub>2</sub> O <sub>3</sub>			
<b>Formula weight</b>	378.26			
<b>Crystal system</b>	triclinic			
<b>Crystal size (mm)</b>	0.13 x 0.20 x 0.20			
<b>Space group</b>	P -1			
<b>Unit cell dimensions</b>	<b>a (Å)</b>	8.2691(14)	<b>α°</b>	92.736(4)
	<b>B (Å)</b>	8.9756(15)	<b>β°</b>	91.476(4)
	<b>c (Å)</b>	9.4832(16)	<b>γ°</b>	99.030(3)
<b>Volume (Å<sup>3</sup>)</b>	693.9(2)			
<b>Z</b>	2			
<b>Theta range for data collection (°)</b>	2.15 to 26.44			
<b>Index ranges</b>	-10<=h<=10, -11<=k<=11, -11<=l<=11			
<b>Reflections collected</b>	9247			
<b>Refinement method</b>	Full-matrix least-squares on F <sup>2</sup>			
<b>Data / restraints / parameters</b>	2839 / 0 / 208			
<b>Goodness-of-fit on F<sup>2</sup></b>	0.565			
<b>Calculated density (Mg/cm<sup>3</sup>)</b>	1.810			
<b>Absorption coefficient (mm<sup>-1</sup>)</b>	1.783			

<b>F(000)</b>	382	
<b>Reflections collected</b>	9247	
<b>Max. and min. transmission</b>	0.8053 and 0.7169	
<b>Data / restraints / parameters</b>	2839 / 0 / 208	
<b>Goodness-of-fit on F2</b>	0.565	
<b>Final R indices</b>	<b>I&gt;2σ(I)</b>	R1 = 0.0214, wR2 = 0.1102
	<b>all data</b>	R1 = 0.0218, wR2 = 0.1127