## Supporting Information

## Mixed Two-Dimensional Organic-Inorganic Halide Perovskites for Highly Efficient and Stable Photovoltaic Application

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Table S1. Calculated lattice constants $(\AA)$ of $\mathrm{BA}_{2} \mathrm{MA}_{2} \mathrm{~B}_{3} \mathrm{X}_{10}\left(\mathrm{~B}=\mathrm{Pb}^{2+}\right.$ or $\mathrm{Sn}^{2+} ; \mathrm{X}=\mathrm{Br}^{-}$or $\left.\mathrm{I}^{-}\right)$with the mixed Cs atom.

| Rb atom | SnI-based |  |  | PbI-based |  |  | SnBr-based |  |  | PbBr-based |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | a | b | c | a | b | c | a | b | c | a | b | C |
| 0\% | 8.83 | 8.83 | 26.57 | 8.92 | 8.92 | 26.69 | 8.34 | 8.34 | 25.55 | 8.42 | 8.42 | 25.65 |
| 25\% | 8.79 | 8.79 | 26.5 | 8.91 | 8.91 | 26.63 | 8.33 | 8.33 | 25.28 | 8.4 | 8.4 | 25.61 |
| 50\% | 8.8 | 8.8 | 26.34 | 8.82 | 8.82 | 26.76 | 8.31 | 8.31 | 25.17 | 8.3 | 8.3 | 25.59 |
| 75\% | 8.77 | 8.77 | 26.3 | 8.76 | 8.76 | 26.64 | 8.24 | 8.24 | 25.18 | 8.26 | 8.26 | 25.58 |
| 100\% | 8.69 | 8.69 | 26.37 | 8.76 | 8.76 | 26.63 | 8.15 | 8.15 | 25.3 | 8.23 | 8.23 | 25.61 |

Table S2. Calculated lattice constants $(\AA)$ of $\mathrm{BA}_{2} \mathrm{MA}_{2} \mathrm{~B}_{3} \mathrm{X}_{10}\left(\mathrm{~B}=\mathrm{Pb}^{2+}\right.$ or $\mathrm{Sn}^{2+} ; \mathrm{X}=\mathrm{Br}^{-}$or $\left.\mathrm{I}^{-}\right)$with the mixed Rb atom.

| Cs atom | SnI-based |  |  | PbI-based |  |  | SnBr-based |  |  | PbBr-based |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | a | b | C | a | b | C | a | b | C | a | b | C |
| 0\% | 8.83 | 8.83 | 26.57 | 8.92 | 8.92 | 26.69 | 8.34 | 8.34 | 25.55 | 8.42 | 8.42 | 25.65 |
| 25\% | 8.84 | 8.84 | 26.49 | 8.85 | 8.85 | 26.75 | 8.33 | 8.33 | 25.44 | 8.4 | 8.4 | 25.65 |
| 50\% | 8.83 | 8.83 | 26.44 | 8.85 | 8.85 | 26.77 | 8.33 | 8.33 | 25.27 | 8.33 | 8.33 | 25.7 |
| 75\% | 8.82 | 8.82 | 26.4 | 8.83 | 8.83 | 26.97 | 8.32 | 8.32 | 25.24 | 8.33 | 8.33 | 25.7 |
| 100\% | 8.8 | 8.8 | 26.48 | 8.8 | 8.8 | 26.85 | 8.28 | 8.28 | 25.15 | 8.28 | 8.28 | 25.75 |

Table S3. Calculated lattice constants $(\AA)$ of $\mathrm{BA}_{2} \mathrm{MA}_{2} \mathrm{Sn}_{3} \mathrm{I}_{10}$ with the mixed Ge atom.

| Ge atom | SnI-based |  |  | SnBr-based |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{a}$ | $\mathbf{b}$ | $\mathbf{c}$ | $\mathbf{a}$ | $\mathbf{b}$ | $\mathbf{c}$ |
|  | 8.83 | 8.83 | 26.57 | 8.34 | 8.34 | 25.55 |
| $16.67 \%$ | 8.83 | 8.83 | 26.42 | 8.33 | 8.33 | 25.37 |
| $33.33 \%$ | 8.81 | 8.81 | 26.35 | 8.33 | 8.33 | 25.22 |
| $50 \%$ | 8.75 | 8.75 | 26.25 | 8.29 | 8.29 | 25.18 |
| $66.67 \%$ | 8.72 | 8.72 | 26.17 | 8.25 | 8.25 | 25.25 |
| $83.33 \%$ | 8.71 | 8.71 | 26.04 | 8.24 | 8.24 | 25.23 |

Table S4. Calculated lattice constants $(\AA)$ of $\mathrm{BA}_{2} \mathrm{MA}_{2} \mathrm{Sn}_{3} \mathrm{I}_{10}$ with the mixed Pb atom.

| $\mathbf{P b}$ atom | SnI-based |  |  | SnBr-based |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{a}$ | $\mathbf{b}$ | $\mathbf{c}$ | $\mathbf{a}$ | $\mathbf{b}$ | $\mathbf{c}$ |
| $0 \%$ | 8.83 | 8.83 | 26.57 | 8.34 | 8.34 | 25.55 |
| $16.67 \%$ | 8.86 | 8.86 | 26.54 | 8.31 | 8.31 | 25.51 |
| $33.33 \%$ | 8.88 | 8.88 | 26.57 | 8.38 | 8.38 | 25.53 |
| $50 \%$ | 8.89 | 8.89 | 26.59 | 8.38 | 8.38 | 25.53 |
| $66.67 \%$ | 8.89 | 8.89 | 26.65 | 8.41 | 8.41 | 25.53 |
| $83.33 \%$ | 8.91 | 8.91 | 26.77 | 8.4 | 8.4 | 25.63 |

Pbl-Cs


Figure S1. Calculated band structures of PbI-based 2D HOIP by mixing Cs atom with different concentrations: (a) pure ( $0 \%$ ), (b) Cs-1 ( $25 \%$ ), (c) Cs-2 ( $50 \%$ ), and (d) Cs-3 ( $75 \%$ ), (e) Cs-4 ( $100 \%$ ). Calculated partial densities of states (PDOSs) of PbI-based 2D HOIP by mixing Cs atom with different concentrations: (f) pure, (g) Cs-1, (h) Cs-2, (i) Cs-3, and (j) Cs-4.

## $\mathrm{PbBr}-\mathrm{Cs}$



Figure S2. Calculated band structures of PbBr -based 2D HOIP by mixing Cs atom with different concentrations: (a) pure, (b) Cs-1, (c) Cs-2, (d) Cs-3, and (e) Cs-4. Calculated partial densities of states (PDOSs) of PbBr-based 2D HOIP by mixing Cs atom with different concentrations: (f) pure, (g) Cs-1, (h) Cs-2, (i) Cs-3, and (j) Cs-4.

## Snl-Cs



Figure S3. Calculated band structures of SnI-based 2D HOIP by mixing Cs atom with different concentrations: (a) pure, (b) Cs-1, (c) Cs-2, (d) Cs-3, and (e) Cs-4. Calculated partial densities of states (PDOSs) of SnI-based 2D HOIP by mixing Cs atom with different concentrations: (f) pure, (g) Cs-1, (h) Cs-2, (i) Cs-3, and (j) Cs-4.

## SnBr-Cs



Figure S4. Calculated band structures of SnBr-based 2D HOIP by mixing Cs atom with different concentrations: (a) pure, (b) Cs-1, (c) Cs-2, (d) Cs-3, and (e) Cs-4. Calculated partial densities of states (PDOSs) of SnBr-based 2D HOIP by mixing Cs atom with different concentrations: (f) pure, (g) Cs-1, (h) Cs-2, (i) Cs-3, and (j) Cs-4.

Pbl-Rb


Figure S5. Calculated band structures of PbI-based 2D HOIP by mixing Rb atom with different concentrations: (a) Rb-1 (25\%), (b) Rb-2 (50\%), (c) Rb-3 (75\%), and (d) Rb-4 (100\%). Calculated partial densities of states (PDOSs) of PbI-based 2D HOIP by mixing Rb atom with different concentrations: (e) Rb-1, (f) Rb-2, (g) Rb-3, and (h) Rb-4.


Figure S6. Calculated band structures of PbBr -based 2D HOIP by mixing Rb atom with different concentrations: (a) Rb-1, (b) Rb-2, (c) Rb-3, and (d) Rb-4. Calculated partial densities of states (PDOSs) of PbBr-based 2D HOIP by mixing Rb atom with different concentrations: (e) Rb-1, (f) Rb-2, (g) Rb-3, and (h) Rb-4

## Snl-Rb



Figure S7. Calculated band structures of SnI-based 2D HOIP by mixing Rb atom with different concentrations: (a) pure, (b) Rb-1, (c) Rb-2, (d) Rb-3. Calculated partial densities of states (PDOSs) of SnI-based 2D HOIP by mixing Rb atom with different concentrations: (e) pure, (f) $\mathrm{Rb}-1$, (g) Rb-2, and (h) Rb-3.


Figure S8. Calculated band structures of SnBr-based 2D HOIP by mixing Rb atom with different concentrations: (a) Rb-1, (b) Rb-2, (c) Rb-3, (d) Rb-4. Calculated partial densities of states (PDOSs) of SnBr-based 2D HOIP by mixing Rb atom with different concentrations: (e) Rb-1, (f) Rb-2, (g) Rb-3, and (h) Rb-4.

## Snl-Ge



Figure S9. Calculated band structures of SnI-based 2D HOIP by mixing Ge atom with different concentrations: (a) Ge-1 ( $16.67 \%$ ), (b) Ge-2 (33.33\%), (c) Ge-3 (50\%), (d) Ge-4 (66.67\%), (e) Ge-5 ( $83.33 \%$ ). Calculated partial densities of states (PDOSs) of SnI-based 2D HOIP by mixing Ge atom with different concentrations: (f) Ge-1, (g) Ge-2, (h) Ge-3, (i) Ge-4, and (j) Ge-5.

## $\mathrm{SnBr}-\mathrm{Ge}$



Figure S10. Calculated band structures of SnBr-based 2D HOIP by mixing Ge atom with different concentrations: (a) Ge-1, (b) Ge-2, (c) Ge-3, (d) Ge-4, (e) Ge-5. Calculated partial densities of states (PDOSs) of SnBr -based 2D HOIP by mixing Ge atom with different concentrations: (f) Ge-1, (g) Ge-2, (h) Ge-3, (i) Ge-4, and (j) Ge-5.

## Snl-Pb



Figure S11. Calculated band structures of SnI-based 2D HOIP by mixing Pb atom with different concentrations: (a) $\mathrm{Pb}-1$ ( $16.67 \%$ ), (b) $\mathrm{Pb}-2$ ( $33.33 \%$ ), (c) $\mathrm{Pb}-3$ ( $50 \%$ ), (d) $\mathrm{Pb}-4$ ( $66.67 \%$ ), and (e) $\mathrm{Pb}-5$ (83.33\%). Calculated partial densities of states (PDOSs) of SnI-based 2D HOIP by mixing Pb atom with different concentrations: (f) $\mathrm{Pb}-1$, (g) $\mathrm{Pb}-2$, (h) $\mathrm{Pb}-3$, (i) $\mathrm{Pb}-4$, and (j) $\mathrm{Pb}-5$.

## $\mathrm{SnBr}-\mathrm{Pb}$




Figure S12. Calculated band structures of SnBr-based 2D HOIP by mixing Pb atom with different concentrations: (a) $\mathrm{Pb}-1$, (b) $\mathrm{Pb}-2$, (c) $\mathrm{Pb}-3$, (d) $\mathrm{Pb}-4$, (e) $\mathrm{Pb}-5$. Calculated partial densities of states (PDOSs) of SnBr -based 2D HOIP by mixing Pb atom with different concentrations: (f) $\mathrm{Pb}-1,(\mathrm{~g}) \mathrm{Pb}-2$, (h) $\mathrm{Pb}-3$, (i) $\mathrm{Pb}-4$, and (j) $\mathrm{Pb}-5$.


Figure S13. Calculated band structures of the various atoms ( $\mathrm{Cs}, \mathrm{Rb}, \mathrm{Ge}$, and Pb ) mixed 2D SnI-based system: (a) BA2Cs2Gel.5Sn1.5I10, (b) BA2Cs2Pb1.5Sn1.5I10, (c) BA2CsRbSn3I10, (d) BA2CsRbGe1.5Sn1.5I10, (e)
 $\mathrm{BA}_{2} \mathrm{Rb}_{2} \mathrm{~Pb}_{1.5} \mathrm{Sn}_{1.5 \mathrm{I}}^{10}$.



Figure S14. Calculated partial densities of states (PDOSs) of the various atoms ( $\mathrm{Cs}, \mathrm{Rb}, \mathrm{Ge}$, and Pb ) mixed 2D SnI-based system: (a) BA2Cs2Gel.5Sn1.510, (b) $\mathrm{BA}_{2} \mathrm{Cs}_{2} \mathrm{~Pb}_{1.5 \mathrm{Sn} 1.5 \mathrm{I} 10, ~(c) ~} \mathrm{BA}_{2} \mathrm{CsRbSn}_{3} \mathrm{I}_{10}$, (d)
 (h) $\mathrm{BA}_{2} \mathrm{Rb}_{2} \mathrm{~Pb}_{1.5} \mathrm{Sn}_{1.5} \mathrm{~S}_{10}$.


Figure S15. Calculated band structures with SOC effect for the various atoms ( $\mathrm{Cs}, \mathrm{Rb}, \mathrm{Ge}$, and Pb )

 $\mathrm{BA}_{2} \mathrm{Rb}_{2} \mathrm{Ge}_{1.5 \mathrm{Sn}}^{1.5 \mathrm{I} 10}$, and (i) $\mathrm{BA}_{2} \mathrm{Rb}_{2} \mathrm{~Pb}_{1.5 \mathrm{Sn}}^{1.5 \mathrm{I} 10}$.

