

Supplementary Materials

Detailed Procedures for the Determination of Association Constants

In the 1:2 ^1H -NMR titration experiment between **S** and **V1** there are 4 unknowns: K_{SV} , K_{VSV} , δ_{SV} , δ_{VSV} . In which δ_{SV} is the chemical shift of a proton resonance in the 1:1 complex between **S** and **V1** and δ_{VSV} is the chemical shift of a proton resonance in the 1:2 complex between **S** and **V1**. We found no computer program powerful enough to fit a cubic equation describing 1:2 binding to the data and extract the 4 unknowns with accuracy from the single binding isotherm obtained from the ^1H -NMR titration. For this reason we followed an indirect and rather elaborate procedure to obtain the most accurate values for K_{SV} and K_{VSV} .

The 1:2 binding equation was programmed in Mathematica[®]. Random values of K_{SV} and K_{VSV} were entered and the evolution of $[\text{S}]$ and complexes $[\text{SV}]$ and $[\text{VSV}]$ at the experimental concentrations of the ^1H -NMR experiment were calculated.

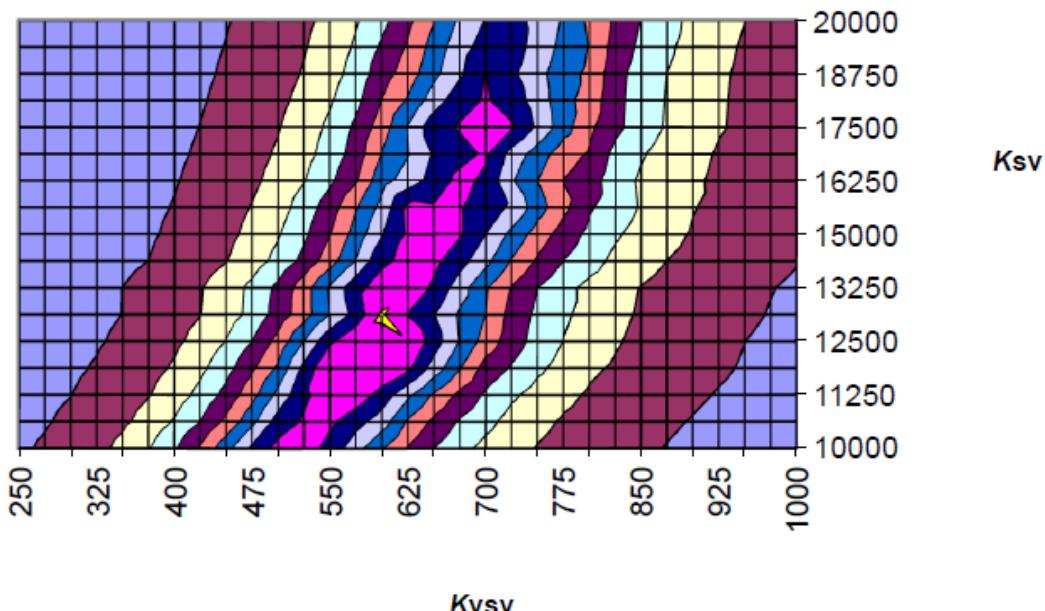
Using the known values of $[\text{S}]_0$, $[\text{V}]_0$ and δ_S and the by Mathematica[®] calculated values of $[\text{S}]$, $[\text{V}]$, and $[\text{VSV}]$ at the programmed association constant, the experimentally obtained binding curve from the ^1H -NMR titration (δ_{obs}) could be simply fitted to the following equation

$$\delta_{obs} = \frac{\partial_S \cdot [\text{S}] + \partial_{VS} \cdot [\text{SV}] + \partial_{VSV} \cdot [\text{VSV}]}{[\text{S}]_0} \quad (1)$$

This provided the complex chemical shifts δ_{VS} and δ_{VSV} as well as a reduced Chi squared value revealing the accuracy of the fit.

By repeating this procedure while changing the into Mathematica imported values of K_{SV} and K_{VSV} a plot of reduced Chi squared values versus K_{SV} and K_{VSV} was obtained. From this landscape (Figure S1), the absolute minimum in reduced Chi squared value, and hence the best fit to the binding model could be found: $K_{SV} = 12842 \text{ M}^{-1}$, $K_{VSV} = 604 \text{ M}^{-1}$.

Figure S1. Plot of reduced Chi squared values versus K_{SV} and K_{VSV} .



S + V1 (K_{SV} and K_{VSV}):

Figure S2. NMR titration at 298 K. $^1\text{H-NMR}$ Spectra of **S** in the presence of from bottom to top increasing quantities of **V1**.

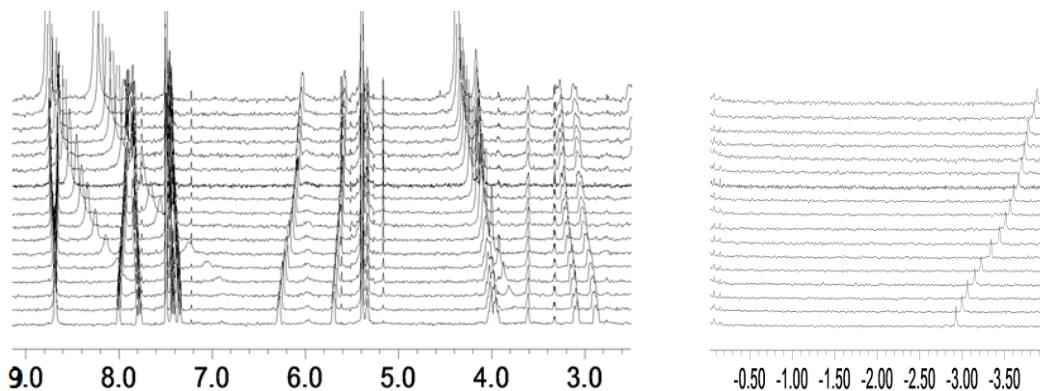


Figure S3. Binding curve of the titration between **S** and **V1** and the fits to 1:1 and 2:1 binding models.

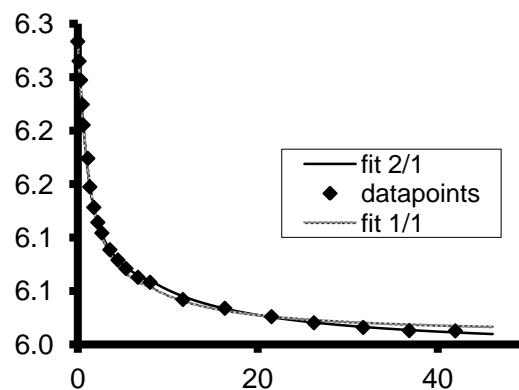


Figure S4. Fluorescence Titrations. Left: Drop in Fluorescence upon addition of **V1** to **S**. Right: normalized fluorescence emission of **S** upon addition of equivalents of **V1** with fits at different temperatures. (K_{SV} could be determined because at micromolar concentrations the 1:2 binding (K_{VSV}) is too low to significantly affect the experimental binding isotherm).

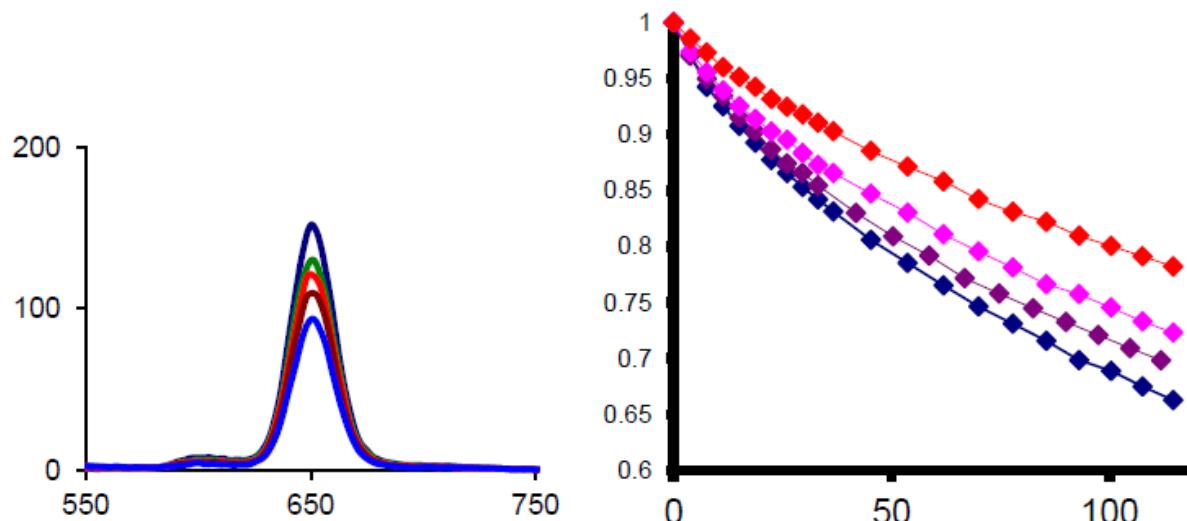


Table S1. Data Fluorescence Titrations.

<i>T</i> (K)	<i>K_{SV}</i>	<i>I/T</i>	<i>Ln(K_{SV})</i>
293	1.7×10^4	0.003413	9.740969
298	1.4×10^4	0.003356	9.546813
303	1.1×10^4	0.003300	9.341369
313	7.5×10^3	0.003195	8.922658

Figure S5. C + V1 (K_a): ^1H -NMR titration. ^1H -NMR Spectra of **C** in the presence of from bottom to top increasing quantities of **V1**.

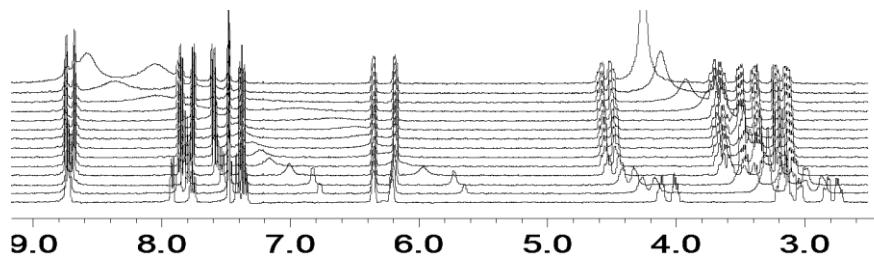


Figure S6. C + V1 (K_a): Fit of the NMR titration: Curve too steep to determine very accurately. (Association constant too high to determine accurately at milimolar concentrations. $K_{CV1} > 1 \times 10^5 \text{ M}^{-1}$).

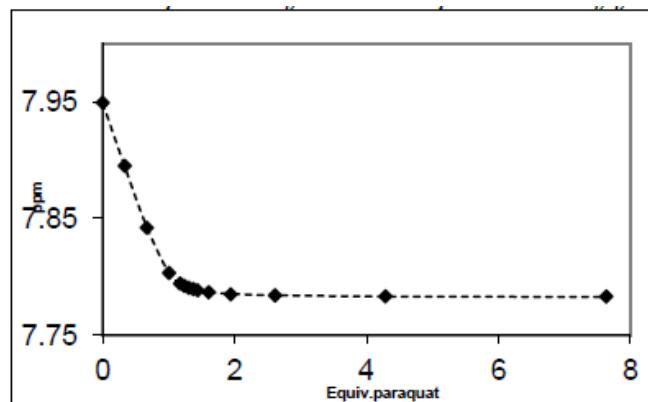


Figure S7. C + V1 (K_a): Fluorescence Titrations. **Left:** Drop in Fluorescence upon addition of V1 to C. **Right:** normalized fluorescence emission of C upon addition of equivalents of V1 with fits at different temperatures.

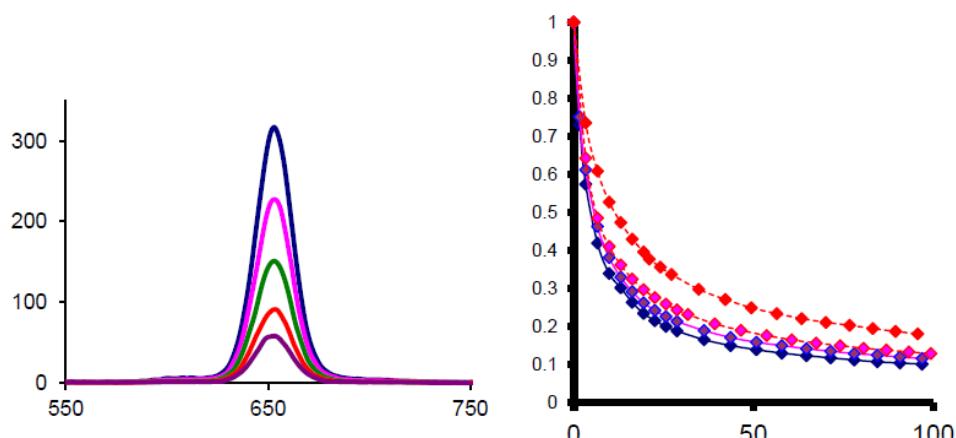
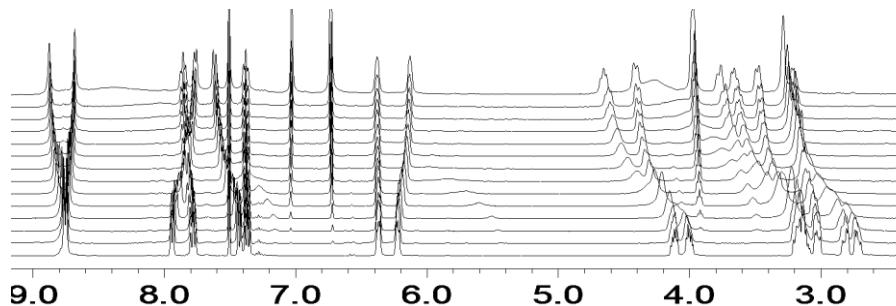
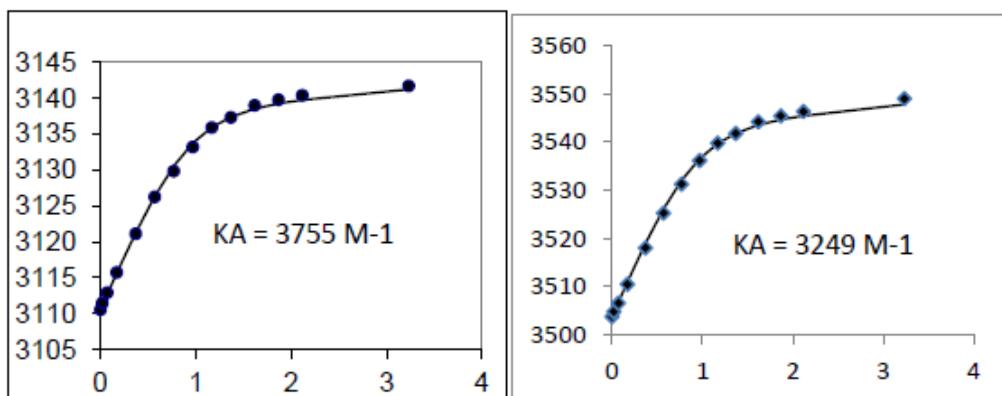


Table S2. Data Fluorescence titrations.

<i>T</i> (K)	<i>K_{CV}</i>	<i>I/T</i>	<i>Ln(K_{CV})</i>
293	3.4×10^5	0.003413	12.737
298	3.0×10^5	0.003356	12.612
303	2.7×10^5	0.003300	12.506
313	1.6×10^5	0.003195	11.951

Figure S8. C + V2 (K_{CV2}): ^1H -NMR titration.**Figure S9.** Curve fitting of ^1H -NMR titration data of binding between C and V2 (left H-8, right β pyrrole).

Mathematica: Example of 2:1 binding model as written into Mathematica which provides the values of the free host, free guest, the 1:1 and 1:2 complex at different values of K1, K2, total host and total guest.

```

ClearAll["Global`*"]
Clear[anul,bnul,cnul,k1,k2,k3,k4,a,b,c,anul,bnul,cnul,ac,ab,abc,abb]
k1 = 2*^4;
k2 = 1*^3;

anul = 1*^-4;
(*cnul = 2/10000;*) (*wordt nu niet gebruikt als constante, maar als variabele*)
cstart = 0.001*anul; (*kun je veranderen voor andere punten*)
cstop = 50*anul; (*kun je veranderen voor andere punten*)
caantalpunt = 200; (*kun je veranderen voor meer punten*)

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```

cinterval = (cstop-cstart)/caantalpunt; (*afblijven*)
cresultaat = Table[0,{caantalpunt},{4}];
cnultable = Table[0,{caantalpunt},{1}];

goedeoplossing = {0,0,0,0};
cpunt = 1;
While[cpunt<=caantalpunt,{goedeoplossing={0,0,0,0},Clear[cnul,a,b,ab,abb],cnul=cstart+cinterval*(cpunt-1),
Alex=NSolve[{k1 ab/(a*b), k2 abb/(ab*b), anul a+ab+abb, cnul b+ab+abb},{a,b,ab,abb}],
teller=1;While[teller<Length[Alex]+1,{
If[Positive[a]/.Alex[[teller]],
If[Positive[b]/.Alex[[teller]],
If[Positive[ab]/.Alex[[teller]],
If[Positive[abb]/.Alex[[teller]],
{goedeoplossing[[1]]=a/.Alex[[teller]],goedeoplossing[[2]]=b/.Alex[[teller]];goedeoplossing[[3]]=ab/.Alex[[teller]];goedeoplossing[[4]]=abb/.Alex[[teller]]}]
,teller=teller+1}],
cresultaat[[cpunt]]=goedeoplossing,
cnultable[[cpunt]]=cnul,
cpunt=cpunt+1}
],
cresultaat
cnultable
cplottable=Table[0,{caantalpunt},{2}]
plotteller=1;
resulttable=Table[0,{caantalpunt},{5}]
resulttable=Transpose[Prepend[Transpose[cresultaat],cnultable]]
NumberForm[resulttable,NumberFormat→(#1"E"#3&)];

(*resultaat wordt geëxporteerd als: cnul, a, b, ab, abb*)
SetDirectory["C:\Documents and Settings\Alexander\My Documents\Promotie/Theoretical models\2to1"];
(*hier wordt alles neergezet*)
filename="two to one.csv"; (*maak file met Mijn Computer en dan hier naam invullen*)
Export[filename,resulttable]; (* Shift Enter om te runnen*)

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